# KAUNAS UNIVERSITY OF TECHNOLOGY 

VAIDA MILIŠIŪNAITĖ

# SYNTHESIS OF NOVEL CONDENSED PYRAZOLE RING CONTAINING HETEROCYCLIC SYSTEMS AND INVESTIGATION OF THEIR PROPERTIES 

Doctoral dissertation
Physical Sciences, Chemistry (03P)

This doctoral dissertation was prepared at Kaunas University of Technology, Faculty of Chemical Technology, Department of Organic Chemistry during the period of 2014-2018. The studies have been supported by the Research Council of Lithuania.

## Scientific Supervisor:

Habil. Dr. Algirdas ŠAČKUS (Kaunas University of Technology, Physical Sciences, Chemistry, 03P),

Dr. Wolfgang Holzer (University of Vienna, Physical Sciences, Chemistry, 03P).

## Scientific Advisor:

Dr. Eglė ARBAČIAUSKIENĖ (Kaunas University of Technology, Physical Sciences, Chemistry, 03P).

Doctoral dissertation has been published in:
http://ktu.edu
Editor:

Brigita Brasienė (Publishing house "Technologija")
© V. Milišiūnaitè, 2018
ISBN 978-609-02-1539-5

The bibliographic information about the publication is available in the National Bibliographic Data Bank (NBDB) of the Martynas Mažvydas National Library of Lithuania.

# NAUJŲ KONDENSUOTŲJŲ HETEROCIKLINIŲ SISTEMŲ, TURINČIŲ PIRAZOLO ŽIEDĄ, SINTEZĖ IR JŲ SAVYBIŲ TYRIMAS 

Daktaro disertacija
Fiziniai mokslai, chemija (03P)

Disertacija rengta 2014-2018 metais Kauno technologijos universiteto Cheminės technologijos fakulteto Organinès chemijos katedroje. Mokslinius tyrimus rèmė Lietuvos mokslo taryba.

## Moksliniai vadovai:

Habil. dr. Algirdas ŠAČKUS (Kauno technologijos universitetas, fiziniai mokslai, chemija, 03P),

Dr. Wolfgang HOLZER (Vienos universitetas, fiziniai mokslai, chemija, 03P).

## Mokslinė konsultantė:

Dr. Eglė ARBAČIAUSKIENĖ (Kauno technologijos universitetas, fiziniai mokslai, chemija, 03P).

Interneto svetainės, kurioje skelbiama disertacija, adresas:
http://ktu.edu
Redagavo:

Brigita Brasienė (leidykla „Technologija")
© V. Milišiūnaitè, 2018

ISBN 978-609-02-1539-5
Leidinio bibliografinė informacija pateikiama Lietuvos nacionalinės Martyno Mažvydo bibliotekos Nacionalinės bibliografijos duomenų banke (NBDB)

## CONTENTS

1. INTRODUCTION ..... 9
2. LITERATURE REVIEW ..... 12
2.1. Synthesis of bicyclic pyrazole derivatives ..... 12
2.1.1. Pyrazolo[1,2-a]pyrazoles ..... 12
2.1.2. Pyrrolo[3,4-c]pyrazoles ..... 15
2.1.3. Construction of bicyclic pyrazole carboxamides. ..... 16
2.1.4. Construction of pyrazolopyrimidines ..... 16
2.1.4.1. Pyrazolo[4,3- $d$ ]pyrimidines ..... 16
2.1.4.2. Pyrazolo[4,5-d]pyrimidines ..... 17
2.1.5. Construction of pyrazolopyridines ..... 18
2.1.6. Dihydropyrano[2,3-c]pyrazoles ..... 20
2.1.7. Synthesis of pyrazolotriazines ..... 21
2.2. Synthesis of tricyclic condensed pyrazole derivatives ..... 22
2.2.1. Pyrazolo-quinoline derivatives ..... 22
2.2.2. Synthesis of chromenopyrazole compounds ..... 24
2.2.2.1. Chromeno[4,3-c]pyrazol-4-ones ..... 25
2.3. Synthesis of tetracyclic condensed pyrazole derivatives ..... 26
2.3.1. Construction of fused polycyclic pyrazolo[4,3-e]pyridines ..... 26
2.3.2. Tetracyclic tacrine analogs containing pyrano[2,3-c]pyrazole ..... 27
2.4. Conclusions ..... 27
3. RESULTS AND DISCUSSION ..... 28
3.1. Synthesis of $2 H$-furo[2,3-c]pyrazoles ..... 28
3.1.1. Synthesis of 4-alkynyl-1-phenyl-1 $H$-pyrazol-3-ols ..... 28
3.1.2. Cyclisation of the $2 H$-furo[2,3-c]pyrazoles from corresponding 4-alkynyl- pyrazol-3-ols ..... 30
3.2. Synthesis of substituted $2 H$-pyrazolo[4,3-c]pyridines ..... 33
3.2.1. Synthesis of starting materials ..... 33
3.2.2. Construction of $2 H$-pyrazolo[4,3-c]pyridines from corresponding 3-alkynyl- pyrazoles ..... 35
3.2.3. Evaluation of biological activity of synthesized pyrazolo[4,3-c]pyridines ..... 37
3.2.3.1. Anticancer activity in vitro ..... 38
3.2.3.2. Effect on the cell cycle and apoptosis ..... 39
3.3. Synthesis of substituted benzopyrano[2,3-c]pyrazol-4(2H)-ones ..... 41
3.4. Synthesis of pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazoles and related new ring systems by tandem cyclisation of vic-alkynylpyrazole-4-carbaldehydes with (het)aryl-1,2-diamines ..... 49
3.4.1. Construction of the $3 H$-pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles ..... 50
3.4.1.1. The synthesis of the starting 5-alkynylpyrazole-4-carbaldehydes ..... 50
3.4.1.2. Tandem cyclisations of the $3 H$-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazo- les from the corresponding 5-alkynylpyrazole-4-carbaldehydes ..... 51
3.4.1.3. Iodocyclisation of the 5-alkynylpyrazole-4-carbaldehyde ..... 53
3.4.2. Functionalization of $3 H$-pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles ..... 54
3.4.3. Construction of the $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4- $b^{\prime}$ ]dipyridines ..... 55
3.4.4. Construction of the 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyri-do[1,2-a]peri- midine ring system ..... 56
3.4.5. Construction of the $2 H$-pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles ..... 57
3.4.6. Single-crystal X-ray diffraction analysis ..... 59
3.4.7. Optical investigations ..... 60
4. EXPERIMENTAL PART ..... 63
4.1. Chemistry ..... 63
4.1.1. Instrumentation. ..... 63
4.1.2. Materials ..... 63
4.2. Biology ..... 111
4.2.1. Cancer cell lines and cytotoxicity assay ..... 111
4.2.2. Flow cytometry ..... 112
4.2.3. Caspase-3/7 assay ..... 112
5. THE MAIN RESULTS AND CONCLUSIONS ..... 114
REFERENCES ..... 115

## List of Abbreviations and Physical Units

[DABCO-H][AcO] - 1,4-diazobicyclo[2.2.2]octane (DABCO) based ionic liquids
A-549 - adenocarcinomic human alveolar basal epithelial cells
AChE - acetylcholinesterase
APT - attached proton test
BChE - butyrylcholinesterase
CB1R - cannabinoid type 1 receptor
CCR1 - C-C motif chemokine receptor 1
CCRF - human acute lymphoblastic leukemia cells
CV-B3 - coxsackie B4 virus
DCC - $N, N$ '-Dicyclohexylcarbodiimide
DIEA - $N, N$-Diisopropylethylamine
EV-A71 - the enterovirus A (sero)type enterovirus A71
G2 - gap 2 phase, which is the second subphase of interphase in the cell cycle directly preceding mitosis
GI50 - concentration of drug to cause $50 \%$ reduction in the proliferation of cancer cells
$h$ CAs IX and XII - carbonic anhydrase isoforms
$h$-e5`NT - human ecto-5'-nucleotidase
HeLA - immortal cell line
Hep G2 - human liver cancer cell line
HMBC - heteronuclear multiple bond coherence
HRMS - high resolution mass spectrometry
HSQC - heteronuclear single quantum coherence
Huh-7 - human liver cell line
IC50 - half maximal inhibitory concentration
K-562 - myelogenous leukemia cell line
LC - liquid chromatography
M-mitosis
MCF-7 - breast cancer cell line
MeOH - methanol
MS - mass spectrometry
MTBPS - mycobacterium tuberculosis pantothenate synthetase
MTT - colorimetric assay for assessing cell metabolic activity
NBS - $N$-bromosuccinimide
$\mathrm{NH}_{2} \mathrm{NHTs}$ - $p$-toluenesulfonyl hydrazide
NHC - $N$-heterocyclic carbene complex
NMR - nuclear magnetic resonance
NOESY - nuclear overhauser effect spectroscopy
OTf - trifluoromethanesulfonate
Panc-1 - human pancreatic cancer cell line isolated from a pancreatic carcinoma of ductal cell origin
ppm - parts per million
ROESY - rotating-frame overhauser effect spectroscopy
rt - room temperature
SAR - structure-activity relationship
SIRT1 - NAD-dependent deacetylase sirtuin-1
SIRT2 - NAD-dependent deacetylase sirtuin 2
SKNSH - human neuroblastoma cell line
$t \mathrm{Bu}$ - tert-butylate
TLC - thin layer chromatography
TMS - trimethylsilane
TMS - trimethylsilyl
TOCSY - total correlation spectroscopy
$\delta$ - chemical shift

## 1. INTRODUCTION

Pyrazoles and their derivatives represent a class of important nitrogen-containing heterocyclic compounds that are covering a broad range of synthetic as well as natural products that display innumerable chemical, biological, agrochemical, and pharmacological properties.

Fused compounds that are containing pyrazole moiety possess a broad spectrum of biological activities. It is known that derivatives containing pyrazolopyrimidine scaffold show pharmacological properties such as cyclin-dependent kinase (CDK) inhibitory, antiproliferative, anticancer, antituberculosis, sedative, antibacterial, antifungal, antileishmanial, and act like antiviral agents, antidepressants ${ }^{1}$. Furopyrazole based molecules demonstrate antimicrobial and antitumor activities ${ }^{2}$, while benzopyranopyrazole derivatives have anxiolytic, hypnotic, and anticonvulsant activity to the nervous system; moreover, they can be used as antihypertensive, antibacterial, diuretic, uricosuric, and vasodilative agents ${ }^{3}$. Pyrazole-benzimidazole hybrids show potential anti-proliferative activity against lung (A-549), breast (MCF-7), and cervical (HeLa) carcinoma cell lines. ${ }^{4}$ Several drugs have been developed from pyrazole derivatives for the past two decades. For instance, Allopurinol (pyrazolo[3,4-d]pyrim-idin-4-one) decreases high blood uric acid level, Sildenafil (pyrazolo[4,3- $d$ ]pyrimidine) is used to treat erectile dysfunction and pulmonary arterial hypertension, Zaleplon (pyrazolo[1,5-a]pyrimidine) is effective in reducing latency to sleep without evidence of undesired effects in elderly patients with insomnia, and Zolazepam (py-razolo[3,4-e]diazepine) is used as an anesthetic for a wide range of animals in veterinary medicine (Figure 1.1). There is a growing body of evidence that pyrazole and its derivatives provide a viable and valuable area for drug discovery. In view of the diverse pharmacological profile of condensed pyrazoles, design and synthesis of new biologically active condensed pyrazole derivatives have been a subject of many research studies for medicinal chemists.


Figure 1.1 Pyrazole nucleus containing pharmaceutical drugs
Because pyrazolo[3,4-b]pyridine skeleton is a donor- $\Pi$-acceptor system, some research ${ }^{1 \mathrm{~b}}$ concerning the fluorescent properties of its derivatives has been disclosed as well. In addition, the fused heterocyclic systems such as dipyrazolo[3,4-b:3', $4^{\prime}$ $d]$ pyridines exhibit strong fluorescence and have been applied in the preparation of blue light-emitting diodes ${ }^{16}$. Furthermore, pyrazoles are pluripotent ligands in coordination chemistry, optical brighteners and UV stabilizers, photoinduced electron transfer systems, and units in supramolecular entities ${ }^{1 \mathrm{~b}}$. Among those, pyrazole compounds
have attracted much attention in recent years owing to its wide application in chemistry and material science as well.

The majority of annulated pyrazoles can be achieved by multi-component reactions. Zhang et al. reported synthesis of trifluoromethylated trans-4,5-dihydro-furo[2,3-c]pyrazoles via the high stereoselective reaction of arsonium bromides and the electron-deficient alkenes in the presence of $\mathrm{K}_{2} \mathrm{CO}_{3}$ as a base ${ }^{5}$. A literature survey suggested that $2 H$-pyrazolo[4,3-c]pyridines are usually accomplished by the treatment of 4-chloro or 4-iodopyridines possessing a carbonyl moiety at 3-position with various hydrazines or heating mixtures of $(E)$-1-(4-azidopyridin-3-yl)- $N$-phenylmethanimines in toluene ${ }^{6}$. Turki et al. obtained benzopyrano[2,3-c]pyrazoles by treating 3-cyano iminocoumarins with hydrazines or hydrazides under the acidic catalyst conditions ${ }^{3}$.

As annulated pyrazole systems are of interest to medicinal chemists and materials scientists, and their synthesis is a worthwhile task; therefore, it has been decided to explore new synthetic approaches of various condensed pyrazoles systems and investigate the biological and optical activities of the obtained compounds.

## The main aim of this dissertation:

Design, synthesis, and investigation of novel polycyclic heterocyclic systems by employing ring-closure reactions of $v i c$-bifunctionalized pyrazoles.

The tasks proposed for the achievement of the above stated aim were as follows:

1. Synthesize 4-alkynyl-1-phenyl-1H-pyrazol-3-ols and investigate their transformation to 2 H -furo[2,3-c]pyrazoles.
2. Synthesize 3-alkynylpyrazole-4-carbaldehydes and the corresponding alkanones and examine their transformation to 2 H -pyrazolo[4,3-c]pyridines.
3. Synthesize novel 4-aroylpyrazol-3-ols and investigate their transformation to benzopyrano[2,3-c]pyrazol-4(2H)-ones.
4. Synthesize 3-and 5-alkynylpyrazole-4-carbaldehydes and examine their condensation with (het)aryl diamines.
5. Investigate the structure of novel heterocyclic compounds by advanced methods of NMR spectroscopy.
6. Investigate functional properties of novel heterocyclic compounds.

Scientific novelty: Bicyclic $2 H$-furo[2,3-c]pyrazoles were synthesized from commercially available 3-hydroxypyrazole via three strep synthetic approach by preparing 4-alkynyl- and 4-(arylethynyl)pyrazol-3-ols which participated in silver (I) or gold (I) mediated 5-endo-dig cyclization. Pyrazole-4-carbaldehydes that are carrying an alkynyl function group adjacent to the formyl moiety are valuable starting materials for the construction of condensed pyrazole systems. The prepared precursors were tested in various cyclisation reactions with dry ammonia or (het)aryl 1,2-diamines. A series of novel pyrazolo[4,3-c]pyridines were tested in structure-activity relationship (SAR) research for cytotoxicity against K-562 and MCF-7 cancer cell lines. The obtained 2 H - and 3 H -pyrazolo[4', $3^{\prime}: 3,4$ ]pyrido[1,2-a]benzimidazoles, 3 H -pyrazolo[4,3$c$ ]imidazo[1,2-a:5,4-b']dipyridines, and 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyr-ido[1,2-a]perimidines showed good optical properties. Moreover, the transformation
of commercially available 3-hydroxypyrazole to pyrazole esters underwent Fries rearrangement reaction to provide 4-aroylpyrazol-3-ols that were excellent building blocks for benzopyrano[2,3-c]pyrazol-4( 2 H )-ones.

Main statements for the defense:

1. 4-Alkynyl- and 4-(arylethynyl)pyrazol-3-ols undergo 5-endo-dig cyclization mediated by a silver (I) or gold (I) catalyst to form $2 H$-furo[2,3-c]pyrazole ring system.
2. Treating 3-alkynyl-1H-pyrazole-4-carbaldehydes, 4-ethanones and 4-propanones with dry ammonia affords a versatile library of $2 H$-pyrazolo[4,3-c]pyridines.
3. 4-Aroylpyrazol-3-ols, bearing halogen atom at 2 nd position of aryl substituent, are convenient synthons for the synthesis of benzopyrano[2,3-c]pyrazol-4(2H)ones.
4. Vic-alkynylpyrazole-4-carbaldehydes are suitable precursors for tandem cyclisations with (het)aryl diamines.

The results of the doctoral dissertation were presented at 7 scientific conferences, and 2 articles were published in reviewed scientific journals.

## 2. LITERATURE REVIEW

The aim of this literature review is to provide an overview concerning synthesis and application of various condensed pyrazole systems.

### 2.1. Synthesis of bicyclic pyrazole derivatives

Recently, much attention has been paid to the development of newer approaches to obtain bicyclic pyrazole systems ${ }^{7}$. For example, Lilly has presented a bicyclic pyrazolidinone LY 186826 exhibiting antibiotic activity better than that of several penicillins and cephalosporins ${ }^{8}$. Moreover, new promising herbicides ${ }^{9}$ and potent drug molecules for treatment of cognitive dysfunctions such as Alzheimer disease were introduced (Figure 2.1) ${ }^{10}$.


Pyrazolo[1,2-a]pyrazole-1,3(2H)-dione



Pyrazolo[1,2-a]pyrazole-1,7-dione

Figure 2.1 Biologically active bicyclic pyrazolidinone and pyrazolo[1,2-a]pyrazoles
It is known that pyrazolo[1,5-b]pyrazoles are used as hair dyes ${ }^{11,12}$, and 3-oxo3 H -pyrazolo[1,2-a]pyrazol-4-ium-1-olates can be used with fertilizers as nitrification inhibitors (Figure 2.2) ${ }^{13}$. It is essential to mention that pyrazolo[3,4-c]pyrazoles are useful for the treatment of esophageal, gastrointestinal mucosa injury ${ }^{14}$, and brain injury ${ }^{14}$; moreover, they are known as immunostimulatory, ${ }^{14}$ antianginal, ${ }^{15}$ and antitumor ${ }^{16}$ agents.


Pyrazolo[1,5-b]pyrazoles


Pyrazolo[1,2-a]pyrazol-4-ium-1-olates

Figure 2.2 Bicyclic fused pyrazole compounds used as hair dyes and nitrification inhibitors

### 2.1.1. Pyrazolo[1,2-a]pyrazoles

There are three general approaches to synthesize pyrazolo[1,2-a]pyrazoles:
-1,3-dipolar cycloaddition of various acetylenes to azomethinimines,
-cycloaddition of azines to dipolarphiles,
-reaction of pyrazoles with ketene, 1,3-dicarbonyl, or dinitrile compounds ${ }^{10}$.
For example, 1,3-dipolar cycloaddition of methyl propiolate with [(Z)-arylmethylene]dimethylpyrazolidinone azomethine imine 2, which was formed through
the condensation of dimethylpyrazolidinone 1 with aromatic aldehydes, gave a mixture of the regioisomeric pyrazolo[1,2-a]pyrazoles 3 and $\mathbf{4}^{10}$, whereas pyrazolo[1,2a]pyrazoles 5 were synthesized by using 1,3-dipolar cycloaddition of azomethine imines to dimethyl acetylenedicarboxylate (DMAD) (Scheme 2.1) ${ }^{10}$.


Scheme 2.1 1,3-Dipolar cycloaddition of [(Z)-arylmethylene]dimethylpyrazolidinone azomethine imine 2

Bicyclic pyrazolidinone 7 can be obtained as well by cycloaddition of the ylide 6 and diallyl acetylenedicarboxylate (Scheme 2.2) ${ }^{17}$.


Scheme 2.2 Cycloaddition of the ylide 6
Another method to synthesize pyrazolpyrazoles is to perform a cycloaddition of azines to dipolarophiles. For instance, pyrazolopyrazoles $\mathbf{1 0}-\mathbf{1 2}$ were obtained by a "crisscross" cycloaddition reaction of 1,2-bis(perfluoropropan-2-ylidene)hydrazine $\mathbf{8}$ with olefins 9 ; the desired products formed in approximately $65 \%$ yields (Scheme $2.3)^{18}$.




10
11
12
Scheme 2.3 Cycloaddition between the azines $\mathbf{8}$ and dipolarophiles 9

El-Alali and Al-Kamali reported a [2+3] cycloaddition reaction of aldazines or ketazines $\mathbf{1 3}$ with DMAD to afford pyrazolopyrazoles 14 (Scheme 2.4) ${ }^{19}$.


Scheme 2.4 [2+3] Cycloaddition of aldazines or ketazines 13
In 2005, Adib et al. ${ }^{20}$ presented a novel synthesis strategy of functionalized 7-oxo- $1 \mathrm{H}, 7 \mathrm{H}$-pyrazolo[ $1,2-a$ ] pyrazoles. 2,4-dihidro-3 H -pyrazol-3-ones $\mathbf{1 7}$ were treated with isocyanides $\mathbf{1 5}$ and dialkyl acetylenedicarboxylates $\mathbf{1 6}$ in a smooth 1:1:1 addition reaction to obtain highly functionalized desired pyrazole derivatives 18 in $69-81 \%$ yields (Scheme 2.5) ${ }^{20}$.


Scheme 2.5 Synthesis of 7-oxo-1H,7H-pyrazolo[1,2-a]pyrazoles 18
An efficient method for pyrazolpyrazoles synthesis is to perform a reaction between pyrazoles and ketene, 1,3-dicarbonyl, or dinitrile compounds. For example, pyrazoles 19 were treated with propa-1,2-diene-1,3-dione or 3-oxo-2-phenylacryloyl chlorides $\mathbf{2 0}$ in order to get cross-conjugated pyrazolium hydroxides 21 (Scheme $2.6)^{21}$.


Scheme 2.6 Reaction between pyrazoles 19 and compounds 20

### 2.1.2. Pyrrolo[3,4-c]pyrazoles

It is well known that pyrrolopyrazole has been exploited as a hinge binder for a number of protein kinase targets ${ }^{22}$. Steeghs et al. reported that an Aurora kinase inhibitor PHA-739358 (Figure 2.3) has advanced into the phase II clinical trials for the treatment of cancer because of very good pharmacokinetic properties and general safety profiles in phase I clinical study ${ }^{23}$. Brasca et al. presented a potent cyclin dependent kinases (CDK) inhibitor PHA-793887 which inhibits the tumor growth in preclinical xenograft tumor models. For this reason, it was further chosen as an anticancer agent for clinical evaluation ${ }^{24}$. In 2010, Shi et al. ${ }^{22}$ reported a synthesis of novel 1,4,5,6-tetrahydropyrrolo[3,4-c]pyrazoles as Aurora-A kinase inhibitors. It is worth pointing out that two of these compounds have ideal anti-proliferative activities in vitro ${ }^{22 \mathrm{c}}$.



PHA-793887
Figure 2.3 Aurora kinase inhibitors
The desired compounds were obtained according to the methods described in literature ${ }^{22 c}$ with minor revision by Shi et al. Firstly, 2-(2-cyanoethylamino)acetic acid 22 was treated with sulfuric acid and methanol to afford methyl ester 23 (Scheme 2.7). The amino group of compound $\mathbf{2 3}$ was further protected in order to get compound 24, which was treated with sodium methoxide.


Reagents and conditions: (a) $\mathrm{MeOH}, \mathrm{H}_{2} \mathrm{SO}_{4}, 7 \mathrm{~h}$, reflux; (b) di-tert-butyl dicarbonate, DCM/aq $\mathrm{NaHCO}_{3}$ (1:1), $24 \mathrm{~h}, 22^{\circ} \mathrm{C}$; (c) MeOH , toluene, $3 \mathrm{~h}, 80^{\circ} \mathrm{C}$, then 2 N HCl ; (d) hydrazine hydrochloride, EtOH, $3 \mathrm{~h}, 6{ }^{\circ} \mathrm{C}$, then $\mathrm{NaHCO}_{3}$, EtCOOCl, THF, $20 \mathrm{~h}, 0-5{ }^{\circ} \mathrm{C}$; (f) $\mathrm{R}^{1} \mathrm{COCl}$, DIEA, THF, $12 \mathrm{~h}, 22^{\circ} \mathrm{C}$; (g) TFA/DCM (1:1) 10 equiv, $1 \mathrm{~h}, 22^{\circ} \mathrm{C}$; (h) $\mathrm{R}^{2} \mathrm{COCl}$, DIEA, THF, $6 \mathrm{~h}, 22^{\circ} \mathrm{C}$.

Scheme 2.7 Preparation of pyrrolo[3,4-c]pyrazole 30

The obtained 4-oxo-pyrrolidine-3-carbonitrile $\mathbf{2 5}$ was cyclised to tetrahydropyrrolopyrazole 26 using hydrazine. Intermediate compound 26 was treated with ethyl chlorocarbonate in order to get compound 27. Later, after acylation, the compound 27 was transformed to derivative 28. After the amino group deprotection reaction of the latter compound, the obtained pyrazoles 29 were employed in the reaction with acyl chloride to afford pyrazole systems $\mathbf{3 0}$.

### 2.1.3. Construction of bicyclic pyrazole carboxamides

Therrien et al. ${ }^{25}$ presented synthesis and biological evaluation of an unprecedented series of bicyclic pyrazole derivatives as SIRT1 and SIRT2 inhibitors. Firstly, commercially available ethyl 2-oxocyclohexanecarboxylate $\mathbf{3 2}$ was acylated with either acyl chloride or methyl carboxylate to afford 1,3-diketone intermediates that were reacted with hydrazine hydrate in THF to obtain tetrahydroindazoles 33 . They were further converted to carboxamides 34 in the presence of a 7 N solution of ammonia in methanol upon heating in a sealed tube or treating ester $\mathbf{3 3}$ with methylamine in methanol (Scheme 2.8) ${ }^{25}$.


Scheme 2.8 Synthesis of bicyclic pyrazole carboxamides 34

### 2.1.4. Construction of pyrazolopyrimidines

### 2.1.4.1. Pyrazolo[4,3-d]pyrimidines

The synthesis of pyrazolo[4,3-d]pyrimidines was firstly reported in 1956. A study of biological activity of numerous compounds which were synthesized by using different synthetic pathways was represented ${ }^{1 \mathrm{a}}$. In 1956, Robins et al. presented the synthesis of pyrazolo[4,3-d]pyrimidines as potential purine antagonists ${ }^{26}$. Pyrazolo[4,3- $d$ ]pyrimidines received attention from medicinal chemists due to the broad spectrum of pharmacological activities such as adenosine receptor antagonists ${ }^{27}$, cytokinin antagonists ${ }^{28}$, corticotrophin releasing factor receptor antagonists ${ }^{29}$, anti-leishmanial ${ }^{30}$, phosphodiesterase 5 (PDE5) inhibitors ${ }^{31}$, anti-viral and anti-funga ${ }^{32}$, antiinflammatory ${ }^{33}$, agents in male and female sexual dysfunctions ${ }^{34}$ etc.

In 2005, Reddy et al. ${ }^{1 c}$ reported that pyrazolo[4,3-d]pyrimidin-7-ones 36 were synthesized by treating starting metarials $\mathbf{3 5}$ with appropriate aldehydes in acetic acid in the presence of catalytic amount of $p$-toluensulfonic acid (Scheme 2.9).


Scheme 2.9 Synthesis of pyrazolo[4,3-d]pyrimidines 36 and 37
In addition, tetrahydropyrazolo[4,3-d]pyrimidin-7-ones 37 were obtained in the reaction of starting material 35 with $4-N, N$-dimethylamino benzaldehydes, where acetic acid is used as a solvent. It is important to mention that arylideneaniline may as well be used for these reactions in order to get corresponding pyrazolo[4,3$d]$ pyrimidines 36. In this case, the excess of 2 equivalents of arylideneaniline was used, because performing an equimolar reaction provided product 37. Moreover, pyrazolo $[4,3-d]$ pyrimidines $\mathbf{3 6}$ were as well synthesized by heating starting material 35 with aromatic carboxylic acids in polyphosphoric acid (PPA) (Scheme 2.9) ${ }^{\text {1c }}$. Reddy with coworkers demonstrated that triethyl orthoesters were used to afford the directly desired pyrazolopyrimidines by refluxing them with pyrazolo carboxamides in xylene (Scheme 2.9) ${ }^{\text {1c }}$.

### 2.1.4.2. Pyrazolo[4,5- $d$ ]pyrimidines

Xing et al. ${ }^{35}$ have reported the synthesis and investigation of novel pyrazolo[4,5$d$ ]pyrimidines 41. The authors have achieved target compounds by using general synthetic route where 2-(ethoxyvinylidene)-malononitrile $\mathbf{3 8}$ was treated with isopropylhydrazine; later, the reaction was proceeded in the presence of basic hydrogen peroxide to obtain aminoamide 39. Pyrazole 39 was condensated with methyl thiophene-2-carboxylate 40 to get the desired $1 H$-pyrazolo[4,5- $d$ ] pyrimidinol 41 (Scheme 2.10) ${ }^{35}$. The activities against replication of poliovirus-1, EV-A71, and CV-B3 enteroviruses of the synthesized compounds $\mathbf{4 1}$ were tested.


Scheme 2.10 Synthesis of $1 H$-pyrazolo[4,5- $d$ ] pyrimidinol 41
Later, the synthesized compounds were functionalized to provide 1 H -pyrazolo[3,4- $d$ ]pyrimidine-4-carboxamides which were evaluated in the primary biological screen, and 22 of these derivatives were confirmed to be active against the
used enterovirus strain (CVB3-H3) ${ }^{35}$. Xing et al. reported that inhibitors with the highest selectivity indicators at $50 \%$ inhibition of viral replication $\left(\mathrm{SI}_{50}\right)$ were those with the isopropyl group at the N1 position and thiophenyl-2-yl unit at C6 position. Moreover, the promising improvement could be influenced by C 4 position because many different $N$-aryl groups showed better antiviral activities and compatibilities than the lead compound. Finally, the authors have noticed that some compounds showed good antienteroviral activity, which was important for the antienteroviral drug design ${ }^{35}$.

### 2.1.5. Construction of pyrazolopyridines

Evidently, the nitrogen-based heterocyclic compounds incorporating the pyridine moiety is an essential scaffold known for a wide range of biological activities. El-Gohary and Shaabanb have reported that pyridines were found to have DNA binding affinity, and notable, they highlighted the significance of compounds containing pyridine nucleus as antimicrobial and antitumor agents ${ }^{36}$. In 2017, Pelit noted that pyrazolopyridines exhibit numerous biological properties such as anti-viral agent, HIV inhibitors, CCR1 antagonists, protein kinase inhibitors, and they as well exhibit parasiticidal properties and antimalarial activities ${ }^{37}$. In 2013, Ganesh et al. reported that pyrazolo[4,3-c]pyridine derivatives such as 3-phenyl-4,5,6,7-tetrahydro$1 H$-pyrazolo[4,3-c]pyridines act as MTB PS inhibitors ${ }^{38}$. Furthemore, El-Gohary, and Shaabanb have marked that pyrazolopyridine nucleus are structurally similar to purine scaffold; for this reason, pyrazolopyridines might compete with purines and prohibit the synthesis of nucleic acids ${ }^{36}$. It is important to mention that several 4-aryl-5substitued pyrazolo[3,4-b]pyridines were reported to be efficient antitumor agents ${ }^{36}$. Due to the aforementioned facts, the molecular hybridization involving the combining of two heterocycles could reinforce biological acitivity.

Arlan et al. ${ }^{39}$ have published a one-pot three-component synthetic approach to obtain a group of new 4-aroyl-1,6-diaryl-3-methyl-1 H -pyrazlo[3,4-b]pyridine-5-carbonitriles (Scheme 2.11). The authors have chosen 1-(3-chlorophenyl)-3-methyl-1 H -pyrazol-5-amine 42, 3-(4-chlorophenyl)-3-oxopropanenitrile 43, and phenylglyoxal hydrate 44 as starting compounds to explore this reaction; unfortunately, the formation of the desired product 45 was not observed even after 24 h of stirring without a catalyst at room temperature. Later, scientists proceeded the reaction under the refluxing conditions in the presence of different catalysts and different solvent systems, and finally, product 45 was obtained with an isolated $89 \%$ yield $^{39}$.


Scheme 2.11 Synthesis of bicyclic 3-methyl-1H-pyrazlo[3,4-b]pyridine-5-carbonitriles

Arlan et al. as well have tested the effect of the amount of catalyst performing reaction with different amounts of nano- $\mathrm{Al}_{2} \mathrm{O}_{3}$; however, the increasing amount of catalyst did not improve the yield. Scientists have claimed that the above mentioned pyrazolopyridines could serve as intermediates for new planar polycyclic heterocycles ${ }^{39}$.

In 2018, Miliutina et al. ${ }^{40}$ presented a study concerning the novel domino reaction between 3-chlorochromones 46 and electron-rich aminoheterocycles, providing various new pyrazolo[3,4-b]pyridines 48 (Scheme 2.12).


Scheme 2.12 Synthesis of pyrazolo[3,4-b]pyridines 48
As chlorine could behave as an activating group for the regioselective formation of products and an applicable leaving group in intramolecular cyclization reaction, scientists have chosen 3 -chlorochromones $\mathbf{4 6}$ as starting materials. The best results were obtained when reactions were proceeded under acidic conditions in the presence of phosphoric acid, using DMF as a solvent. It has been noticed as well that the temperature made a huge influence (it should not be too high), and additional activation was indispensable, in spite of the fact that chromone reactivity was not sufficient (Scheme 2.12) ${ }^{40}$.

The obtained pyrazolo[3,4-b]pyridines 48 were further examined as promising inhibitors against ecto- $5^{\prime}$-nucleotidases (e5'NT). ${ }^{44}$ The scientists analyzed compounds 48 towards both human ( $h-$ ) and rat ( $r$-) isozymes and discovered that the majority of them inhibited $h$-e5'NT at very low concentrations ${ }^{40}$.

Miliutina et al. examined the synthesized pyrazolo[3,4-b]pyridines 48 for cytotoxic activity on cervical cancer cell lines (HeLa) in the presence of carboplatin as a standard drug. Interestingly, all compounds were significant inhibitors towards cancer cell lines because they showed more than $80 \%$ inhibition against cervical cancer $(\mathrm{HeLa})^{40}$.

Metwally and Deeb reported ${ }^{41}$ that compounds 54 could be synthesized by refluxing aminopyrazolo[4,3-c]pyridine 50 with aromatic aldehydes $\mathbf{5 2}$ in DMF (Scheme 2.13). The acetylation of compounds $\mathbf{5 4}$ with acetic anhydride provided products which were completely identical in all aspects (melting point, mixed melting point, and IR) with compounds 53 (Scheme 2.13) ${ }^{41}$.

Some of the newly synthesized pyrazolo[4,3-c]pyridine derivatives ${ }^{41}$ were investigated for anticancer activity. The results of the cytotoxic activity revealed that compound $54\left(\mathrm{Ar}=4-\mathrm{MeO}-\mathrm{C}_{6} \mathrm{H}_{4}\right)$ was the most active compound against the breast and liver carcinoma cell lines with $\mathrm{IC}_{50}$ values of 1.937 and $3.695 \mu \mathrm{~g} / \mathrm{mL}$ compared to a reference drug (doxorubicin) with $\mathrm{IC}_{50}$ values of 2.527 and $4.749 \mu \mathrm{~g} / \mathrm{ml}$, respectively. Moreover, compound $54\left(\mathrm{Ar}=4-\mathrm{Cl}-\mathrm{C}_{6} \mathrm{H}_{4}\right)$ was a potent compound against the
colon carcinoma cell line with $\mathrm{IC}_{50}=2.914 \mu \mathrm{~g} / \mathrm{ml}$ compared to the doxorubicin with $\mathrm{IC}_{50}$ value of $3.641 \mu \mathrm{~g} / \mathrm{ml} .^{41}$


Scheme 2.13 Synthesis of 4 H -pyrazolo[4,3-c]pyridine-4,6-(5H)-dione

### 2.1.6. Dihydropyrano[2,3-c]pyrazoles

Dihydropyrano[2,3-c] pyrazoles are an important part of numerous biologically active compounds and are attractive templates for medicinal chemistry. Pyrano[2,3c]pyrazoles have several biological activities such as molluscicidal ${ }^{42}$, insecticidal ${ }^{43}$, antiinflammatory ${ }^{44}$, analgesic ${ }^{48}$, and anticancer ${ }^{45}$. In 1973, Junek and Aigner were the first who obtained pyranopyrazoles; the reaction was done by treating 3-methyl-1-phenylpyrazolin-5-one with tetracyanoethylene ${ }^{46}$.

Kiyani et al. ${ }^{47}$ reported the synthesis of pyranopyrazoles $\mathbf{5 8}$ by treating aryl aldehydes 54 with hydrazines 55 , ethyl acetoacetate 56 and malonitrile 57 in the presence of sodium benzoate as the mild basic catalyst (Scheme 2.14).


Scheme 2.14 Synthesis of pyranopyrazoles 58
The authors demonstrated an efficient method to obtain pyranopyrazoles via four-component reaction, to get the desired products 58 without column purification by using environmentally friendly reaction conditions ${ }^{47}$.

Liu and coauthors ${ }^{48}$ demonstrated the synthesis of pyranopyrazoles 63 by using ketones 61 instead of aldehydes (Scheme 2.15). The reaction of ethyl acetoacetates 59, hydrazines 60, malononitrile 61, and 3-pentanone 61 in the presence of [Dabco$\mathrm{H}][\mathrm{AcO}]$ as a catalyst gave ketone-derived dihydropyrano[2,3-c]pyrazoles 63 in good yields ${ }^{48}$.


Scheme 2.15 Synthesis of dihydropyranopyrazoles 63
Abdelrazek et al. ${ }^{42}$ published another synthetic route to afford pyrano[2,3-c]pyrazole derivatives 68 (Scheme 2.16). In this case, pyrazol-5-ones 64 underwent a cyclization reaction with 3-furfurylidene or 3-thienylidene malononitriles 65 .


Scheme 2.16 Cyclisation of pyrazol-5-ones 64 with 3-furfurylidene/3-thienylidene malononitriles 65

### 2.1.7. Synthesis of pyrazolotriazines

It is well known that pyrazolotriazines are interesting heterocyclic structures due to their broad spectrum of biological activities such as anticancer ${ }^{49}$, cyclin-dependent protein kinase 7 (CDK7) inhibitory ${ }^{50}$, antifungal ${ }^{51}$, and antioxidant ${ }^{52}$.

Nasr et al. published synthesis of a novel group of pyrazolo[3,4-d][1,2,3]triazines and investigated their anticancer activity in vitro against three different cell lines, namely, Huh-7, Panc-1, and CCRF ${ }^{53}$. Shchegol'kov et al. ${ }^{52}$ reported the synthesis of novel 7-hydroxy-7-polyfluoroalkyl-4,7-dihydroazolo[5,1-c][1,2,4]triazines and tested them as AChE, BChE, and CaE inhibitors. Shchegol'kov have previously published a method to obtain 7-polyfluoroalkyl-4,7-dihydroazolo[5,1-c][1,2,4]triazines by one-pot reaction, treating azocoupled polyfluoroalkyl-containing 1,3-dicarbonyl reagents with hetaryldiazonium salts (Scheme 2.17) ${ }^{52}$.


Scheme 2.17 Synthesis of dihydropyrazolo[5,1-c][1,2,4]triazine-3-carboxylate 70
Scientists used this developed methodology to synthesize a novel series of dihydroazolotriazines by treating polyfluoroalkyl-1,3-dicarbonyl compounds 69 with pyrazolyldiazonium chlorides to afford ethyl-4-hydroxy-7-methyl(phenyl)-4-polyfluoroalkyl-1,4-dihydropyrazolo[5,1-c][1,2,4]triazine-3-carboxylates

In order to understand the influence of compounds 70 substituents on AChE , BChE, and CaE inhibition, Shchegol'kov et al. ${ }^{52}$ synthesized dialkyl-4-hydroxy-4-polyfluoroalkyl-1,4-dihydropyrazolo[5,1-c][1,2,4]triazine-3,8-dicarboxylates (Scheme 2.17) and investigated inhibitory activity against $\mathrm{AChE}, \mathrm{BChE}$, and CaE of all the synthesized compounds. The authors have noticed that derivatives $\mathbf{7 0}$ did not act as AChE and BChE inhibitors; however, they acted as selective CaE inhibitors ${ }^{52}$.

Hassan et al. ${ }^{54}$ published an investigation of $3 H$-pyrazolo[3,4- $\left.d\right][1,2,3]$ triazin$4(7 H)$-ones 73 anticancer activity in vitro. The authors synthesized the novel compounds from diazotized 5-aminopyrazoles 71 (Scheme 2.18) ${ }^{54}$.


Scheme 2.18 Construction of 3H-pyrazolo[3,4- $d$ ][1,2,3]triazin-4(7H)-ones 73

### 2.2. Synthesis of tricyclic condensed pyrazole derivatives

### 2.2.1. Pyrazolo-quinoline derivatives

Kasaboina et al. ${ }^{55}$ presented an efficient approach to synthesize pyrazolo-quinoline derivatives and the investigation of the anti-profilerative activity against different cancer lines. The intermediate aniline 78 was obtained in three-step synthesis. First, 2-nitrobenzaldehyde 74 was transformed to acetophenone 75 by using $40 \%$ aqueous NaOH solution and ethanol as a solvent to give compound 76 (Scheme 2.19). The latter compound was cyclised with phenyl hydrazine in the presence of iodine in acetic acid to give pyrazole 77. In the next reaction, nitro group was reducted to amine with iron powder, concentrated HCl performing reaction in a methanol.


Reagents and conditions: (a) $40 \% \mathrm{NaOH}$, EtOH , rt, 30 min ; (b) $\mathrm{PhNHNH}_{2}, \mathrm{I}_{2}, \mathrm{AcOH}$, reflux, 3 h ; (c) Fe powder, $\mathrm{HCl}_{\text {conc }}$, MeOH , reflux, 3 h ; (d) benzaldehyde, $\mathrm{I}_{2}$, DMSO, reflux, 3 h .

Scheme 2.19 Construction of 1 H -pyrazolo[4,3-c]quinolones 79
Finally, the desired products 79 were obtained by treating 2-(1,3-diphenyl-1 H -pyrazol-5-yl)aniline 78 with different aldehydes in the presence of molecular iodine and DMSO as a solvent (Scheme 2.19) ${ }^{55}$.

These synthesized novel compounds were tested in vitro for anti-proliferative activity against different cancer cell lines such as A-549, HeLa, SKNSH, HepG2, and MCF-7. The authors have noticed that 3-(4-bromophenyl)-4-(2,4-dichlorophenyl)-1-phenyl-1H-pyrazolo[4,3-c] quinolones 79 exhibited potent activity against MCF-7 compared with standard Daxorubicin. These results may lead to the discovery of new potent antitumor agents ${ }^{55}$.

In 2017, Ezzati et al. ${ }^{56}$ in Tetrahedron published a synthetic study of new series of pyrazolo[3,4-b]quinolin-5-ones $\mathbf{8 2}$ and pyrazolo[ 4 ',3':5,6]pyrido[2,3- $d$ ]pyrimidin5,7 -diones $\mathbf{8 3}$. The authors found out that pyrazolo[3,4-b]quinolones $\mathbf{8 2 - 8 3}$ could be obtained in a one-pot three-component reaction, treating arylglyoxals $\mathbf{8 2}$ with 1,3diketones 81 (cyclohexane-1,3-dione or dimedone) and 3-methyl-1-phenyl-1 H -pyra-zol-5-amine in $\mathrm{H}_{2} \mathrm{O} / \mathrm{EtOH}$ (Scheme 2.20).


Scheme 2.20 Construction of pyrazolo[3,4-b]quinolin-5-ones 82 and pyrazolo[4',3':5,6]pyr-ido[2,3- $d$ ]pyrimidin-5,7-diones 83

Depending on the reaction temperature, a selective formation of $\mathbf{8 2}$ or $\mathbf{8 3}$ could be achieved ${ }^{56}$ (Scheme 2.20).


Scheme 2.21 Construction of pyrazolopyrido[2,3-d]pyridimine-5,7(6H,8H)-diones 86
Ezzati and coworkers reported that pyrazolo[4',3':5,6]pyrido[2,3- $d$ ]pyridimine$5,7(6 H, 8 H)$-diones 86 could be obtained by the reaction of arylglyoxals 84 with 1,3dimethylbarbituric acid 85 and 3-methyl-1-phenyl-1 $H$-pyrazol-5-amine in $\mathrm{H}_{2} \mathrm{O} / \mathrm{EtOH}$ system under reflux in very good yields (73-96\%) (Scheme 2.21) ${ }^{56}$.

Pyrazolo[5,1-a]isoquinolines, which incorporate both isoquinoline and pyrazole scaffolds, represent one of the most attractive structures of the fused $N$-heterocyclic compounds, which may show promising biological activity ${ }^{57}$. In 2009, Chen et al. ${ }^{57}$ presented an investigation of AgOTf catalyzed tandem cyclization between $N^{\prime}$-(2-al-kynylbenzylidene)-hydrazides 87 and various alkynes 88 to obtain 1,2-dihydroisoquinolines $\mathbf{8 9}$ in very good yields (Scheme 2.22).


Scheme 2.22 Reaction between $N^{\prime}$-(2-alkynylbenzylidene)-hydrazides 87 and various alkynes 88

In 2007, Liu et al. ${ }^{58}$ reported an efficient synthetic strategy for 2-arylpyra-zolo[5,1-a]isoquinolines 92 catalyzed by [Cu(maloNHC)] catalyst. In this synthesis approach, $N^{\prime}-(2-(($ trimethylsilyl $)$ ethynyl)-benzylidene)hydrazides 90 were annulated with terminal aromatic alkynes 91 in the presence of $[\mathrm{Cu}($ maloNHC $)]$ catalyst, CuI , $\mathrm{KO} t \mathrm{Bu}$, and $\mathrm{NH}_{2} \mathrm{NHTs}$ in 1,4-dioxane (Scheme 2.23).


Scheme 2.23 Synthesis of 2-substituted pyrazolo[1,5-a]isoquinolines 92
This study extends the implementation of $\mathrm{Cu} / \mathrm{NHC}$ as a catalyst in N -heterocyclic compounds synthesis, where alkyne is reactant; however, it as well expands the library of 2-substituted pyrazolo[1,5-a]isoquinolines ${ }^{58}$.

### 2.2.2. Synthesis of chromenopyrazole compounds

Chromeno derivatives are interesting to scientists due to their wide range of biological activities. For instance, chromeno compounds are used as anticonvulsant, antimicrobial, antiinflammatory, TNF- $\alpha$ inhibition, antimalarial, and anticancer agents ${ }^{59}$. Thus, compounds that are containing chromeno and pyrazole scaffolds are much more attractive because pyrazole nucleus is significantly known due to the diversified applications in different areas. For example, compound $\mathbf{9 3}$ is known as $\mathrm{A}_{2}-$ subtype selective adenosine receptor antagonist (Figure 2.4) ${ }^{60}$.

It is known that chromenopyrazoles are non-psychoactive CB1R agonists with peripheral antinociceptive properties ${ }^{61}$. For these reasons, the synthesis of novel compounds that are containing both chromeno and pyrazole rings have always been attractive ${ }^{61}$.


93
Figure 2.4 A2-subtype selective adenosine receptor antagonist

### 2.2.2.1. Chromeno[4,3-c]pyrazol-4-ones

In 2018, Bonardi et al. ${ }^{62}$ reported the synthesis of new class inhibitors of tumorassociated carbonic anhydrases hCAs IX and XII: the chromeno[4,3-c]pyrazol-4-ones and pyrano[4,3-c]pyrazol-4-ones. Chromeno[4,3-c]pyrazol-4-ones 96 were synthesised from 5-substituted 2-hydroxybenzaldehydes by Knoevenagel condensation with ethyl acetoacetate in the presence of piperidine under microwave irradation following the reaction with suitable arylhydrazines in boiling ethanol and cyclization in refluxing xylene in the presence of para-toluensulfonic acid (Scheme $2.24)^{62}$.


Scheme 2.24 Synthesis of chromeno[4,3-c]pyrazol-4-ones 96
Morales et al. ${ }^{61}$ have synthesized various novel chromenopyrazoles 100 (Scheme 2.25) and screened them for biological activities. The desired products $\mathbf{1 0 0}$ were synthesized from the corresponding resorcinols 97 (Scheme 2.25). Firstly, the starting materials 97 were obtained by demethylation of 5-(1', 1 '-dimethyl-n-hepthyl)-1,3-dimethoxybenzene. Later, the corresponding resorcinols 97 were treated with 3,3dimethylacrylic acid in methansulfonic acid by using phosphorus pentoxide to get 7-alkyl-5-hydroxy-2,2-dimethylchroman-4-ones 98 . This reaction was performed under the microwave irradiation, applying the reaction conditions that were reported by Lim et al ${ }^{63}$. (Z)-7-alkyl-5-hydroxy-3-(hydroxymethylen)-2,2-dimethylchroman-4-ones 99 were obtained by applying $\alpha$-formation synthetic approach proposed by Press et al. ${ }^{64}$. Finally, $\beta$-ketoaldehydes of intermediate compounds $\mathbf{9 9}$ underwent condensation with appropriate hydrazines to yield 7-alkyl-1(2),4-dihydro-4,4-dimethylchromeno[4,3-
c]pyrazol-9-oles 100 (Scheme 2.25). It has been noticed that $\beta$-reactions of ketoaldehydes with methyl- and ethylhydrazines yielded the formation of two isomers ( $N^{1}$ and $N^{2}$-substituted pyrazoles); however, the analogous reaction with arylhydrazines resulted in the formation of only one isomer. The scientists explained that for $N^{\prime}$ arylhydrazines, $N^{\prime}$-hydrazine was much more nucleophilic than $N$-hydrazine, and for this reason, only one isomer was obtained in the reaction ${ }^{65}$.


Reaction conditions: (i) 3,3-dimethylacrylic acid, methanesulfonic acid, $\mathrm{P}_{2} \mathrm{O}_{5}, 8 \mathrm{~h}, 70^{\circ} \mathrm{C}, 81 \%$, (ii) NaH, THF, MW, $25 \mathrm{~min}, 45^{\circ} \mathrm{C}$, (iii) 1) ethyl formate, MW, $25 \mathrm{~min}, 45^{\circ} \mathrm{C}, 76 \%, 2$ ) corresponding hydrazine, $\mathrm{EtOH}, 1-4 \mathrm{~h}, 40^{\circ} \mathrm{C}, 36-50 \%$ )

Scheme 2.25 Novel chromenopyrazoles $\mathbf{1 0 0}$ synthesized by Morales et al.

### 2.3. Synthesis of tetracyclic condensed pyrazole derivatives

### 2.3.1. Construction of fused polycyclic pyrazolo[4,3-e]pyridines

In 2017, Pelit et al. ${ }^{37}$ published her study of one-pot three-component synthetic approach to provide novel fused polycyclic pyrazolopyridines 103 and 104 with excellent yields. Under this protocol, the reaction between 1,3-dimethyl-1 H -pyrazol-5-amine or 3-phenyl-1H-pyrazol-5-amine 102, indan-1,3-dione 101, and various aromatic aldehydes was proceeded in ethanol in the presence of organocatalyst, for instance, camphor-10-sulfonic acid (Scheme 2.26) ${ }^{37}$.


Scheme 2.26 Construction of fused polycyclic pyrazolo[4,3-e]pyridines
Pelit et al. as well investigated the antioxidant activity of synthesized novel fused polycyclic pyrazolo[4,3-e]pyridines $\mathbf{1 0 3}$. When product bearing substituent $\mathrm{Ar}=4$-cyanophenyl, it exhibited excellent antioxidant activity better than that of BHT, resorcinol, and ascorbic acid ${ }^{37}$.

### 2.3.2. Tetracyclic tacrine analogs containing pyrano[2,3-c]pyrazole

In 2014, Khoobi et al. ${ }^{66}$ reported a novel two-step synthetic approach for tacrinebased acetylcholinesterase ( AChE ) inhibitors, replacing the benzene ring of tacrine with aryl-dihydropyrano[2,3-c]pyrazoles. Firstly, the authors investigated one-pot four-component reaction of ethylacetoacetate $\mathbf{1 0 5}$, hydrazine hydrate, malonitrile, and aromatic aldehydes, varying different catalysts ((S)-proline, piperidine, pyridine, $\mathrm{NaOH}, \mathrm{K}_{2} \mathrm{CO}_{3}$ ) and environmentally benign solvents such as water, EtOH , polyethylene glycol. Khoobi and coworkers found out that the best results were achieved by using ( $S$ )-proline as a catalyst and water-ethanol medium (1:1) (Scheme 2.27). It was observed that ultrasonic irradiation accelerated the reaction (shortened it to 20 minutes) and improved the yield (the yield increased from $81 \%$ to $92 \%$ ). In the next step - Friedländer reaction - obtained pyrano[2,3-c]pyrazoles 106 were treated with cyclohexanone in the presence of $\mathrm{AlCl}_{3}$ to obtain the target compounds $\mathbf{1 0 7}$. Intermediate and target compounds were isolated as a mixture of enantiomers with no detectable optical activity.



106


105


Scheme 2.27 Novel tetracyclic tacrine analogs containing pyrano[2,3-c]pyrazole 107
Novel compounds $\mathbf{1 0 7}$ were tested for anti-AChE activity, and it has been noticed that the majority of them showed potent and selective anti-AChE activity in submicromolar range. In particular, the most potent compound (where $\mathrm{Ar}=3$, 4 -dimethoxyphenyl group) was more active than the reference drug tacrine: this derivative could protect neurons against oxidative stress ${ }^{66}$.

### 2.4. Conclusions

The synthesis of fused heterocycles that are containing pyrazole unit have been systematically reviewed and discussed. It could be concluded that the synthesis of annulated pyrazole derivatives is significantly important among the medicinal and materials chemistry scientists, and it is a worthwhile and challenging task from the point of view of organic chemistry.

## 3. RESULTS AND DISCUSSION

### 3.1. Synthesis of $\mathbf{2 H}$-furo[ $\mathbf{2 , 3 - c}$ ]pyrazoles

Pyrazole nuclei is widely found in a lot of pharmaceutically active compounds. It is reported that pyrazole compounds possess anticancer, antibacterial analgesic, antiinflammatory, antimicrobial, and antipriliferative activities ${ }^{67}$. Moreover, the compounds containing furan ring are significantly attractive because of their biological and pharmalogical properties, and they are widely found in numerous natural products ${ }^{5}$. For instance, Bandock et al. ${ }^{67}$ reported that Nitrofurantoin and Furazolidone are potent nitrofuran drugs, which possess various antimicrobial properties, including activity against trypanosomes. Nevertheless, the access to $2 \mathrm{H}-$ furo[2,3-c]pyrazoles, where both fused pyrazole and furan moieties are aromatic, is still quite limited, and their chemistry and functional properties remain largely unexplored. Due to the mentioned facts, it has been decided to synthesize a group of various novel 2 H -furo[2,3-c]pyrazoles.

### 3.1.1. Synthesis of 4-alkynyl-1-phenyl-1H-pyrazol-3-ols

As a starting material for the synthesis of 4-alkinyl-3-hydroxy-1-phenyl-1 H pyrazole derivatives 4-14, 1-phenyl-1H-pyrazol-3-ol (1), which was readily accessible from the oxidation of 1-phenyl-3-pyrazolidinone ${ }^{68}$, was used. 1-Phenylpyrazolidin-3-one (1) was treated with excess of aqueous acidic $\mathrm{FeCl}_{3}$ solution in refluxing EtOH to give pyrazole 2 in $70 \%$ yield. The iodination of compound $\mathbf{2}$ with iodine in the presence of KOH in DMF afforded 4-iodo-1H-pyrazol-3-ol 3 (Scheme 3.1) ${ }^{69}$. It has been shown previously that the Sonogashira-type coupling of the aforementioned iodinated compound with phenylacetylene under the standard reaction conditions $\left(\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}, \mathrm{CuI}\right.$, and TEA) gives 4-(phenylethynyl)-substituted pyrazolol 4 (Scheme 3.1) ${ }^{69}$.


1
2
3


4-14

Reagents and conditions: (i) $\mathrm{FeCl}_{3}, \mathrm{EtOH}, 100{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}$; (ii) $\mathrm{I}_{2}, \mathrm{KOH}, \mathrm{DMF}, \mathrm{rt}, 2 \mathrm{~h}, 80 \%$; (iii) $\mathrm{RC} \equiv \mathrm{CH}, \mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}$, CuI, TEA, DMF, $60^{\circ} \mathrm{C}, 1-12 \mathrm{~h}, 64-85 \%$.

Scheme 3.1 Synthesis of 4-alkynyl-3-hydroxy-1-phenyl-1 H -pyrazoles 4-14
When various het(aryl)- and alkyl acetylenes were used in the coupling with $\mathbf{3}$ under the same reaction conditions, the compounds $\mathbf{4} \mathbf{- 1 3}$ were obtained in 64-85\% yield (Scheme 3.1, Table 3.1). Compound $\mathbf{1 4}$ was detected by LC-MS analysis; however, it was not obtained because of a very low reaction yield (Table 3.1). It was found that the temperature for 'Sonogashira' cross coupling reaction could not be higher than $60{ }^{\circ} \mathrm{C}$.

Table 3.1 Synthesized various 4-alkynyl-3-hydroxy-1-phenyl-1 H -pyrazoles 4-14

| Compound | $\mathbf{R}$ | Yield, $\%$ | Compound | R | Yield, $\%$ |
| :---: | :--- | :---: | :---: | :--- | :---: |
| $\mathbf{4}$ | Ph | 76 | $\mathbf{1 0}$ | $n$-Bu | 85 |
| $\mathbf{5}$ | 4-MePh | 75 | $\mathbf{1 1}$ | $n$-Pent | 78 |
| $\mathbf{6}$ | 4-EtPh | 78 | $\mathbf{1 2}$ | cyclopropyl | 64 |
| $\mathbf{7}$ | 4-FPh | 73 | $\mathbf{1 3}$ | TMS | 50 |
| $\mathbf{8}$ | 4-EtOPh | 79 | $\mathbf{1 4}$ | $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OH}$ | traces |
| $\mathbf{9}$ | 3-Thienyl | 75 |  |  |  |

It was noticed that at higher than $60{ }^{\circ} \mathrm{C}$ temperature, compound $\mathbf{3}$ was deiodinated, and cross-coupling reaction did not work, or products were formed in very low yields.

Figure 3.1 shows ${ }^{13} \mathrm{C}$ NMR spectrum of compound 13 . The full and unambiguous assignments of ${ }^{13} \mathrm{C}$ NMR resonances were achieved by using combined applications of standard NMR spectroscopic techniques such as APT, COSY, TOCSY, NOESY, gs-HSQC, and gs-HMBC.The presence of signals of carbons of $C \equiv$ CTMS and $\mathrm{C} \equiv C$ TMS at 93.9 and 92.3 ppm proves the formation of the desired product. In the 'aliphatic' part, there is a signal of TMS group carbons at 0.17 ppm . In the 'aromatic' part, there are signals of phenyl ring carbons (119.1 (Ph C-2,6), 129.9 (Ph C-4), 131.7 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), and 139.1 ( $\mathrm{Ph} \mathrm{C}-1$ )) and signals of pyrazole ring carbons (98.2 (C-4), 126.8 (C-5), and 163.8 (C-3).


Figure 3.1 ${ }^{13} \mathrm{C}$ NMR spectrum of compound 13

### 3.1.2. Cyclisation of the $\mathbf{2 H}$-furo[2,3-c]pyrazoles from corresponding 4-alkynyl-pyrazol-3-ols

It is known that the compounds bearing alkynyl and hydroxy groups in adjacent positions can undergo cyclisation reaction ${ }^{70}$. The most recent approaches have focused on the synthesis of benzofurans from 2-alkynyl phenols. In most published literature, benzo[b]furans were obtained via tandem Sonogashira coupling/5-endo-dig cyclisation by applying palladium catalyst or copper salts ${ }^{70 c}$. Demera et al. ${ }^{70 \mathrm{c}}$ presented a novel base-promoted synthesis of benzo[b]furans with good yields; this synthetic approach is palladium/copper free and avoids applying expensive and air sensitive reagents ${ }^{70 \mathrm{c}}$.

Having prepared a series of hydroxy-ethynyl substrates 4-13, the optimal conditions for the subsequent cyclization reaction were investigated by using pyrazole 4 as the model compound (Table 3.2).

Table 3.2 Optimization of reaction conditions for construction of furopyrazole $\mathbf{1 5}$

| Entry | Base |  <br> 4 <br> Catalyst | 15 <br> Temperature, ${ }^{\circ} \mathbf{C}$ | Time, h | Yield, \% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ | - | 60 | 24 | no product |
| 2 | $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ | - | 120 | 96 | traces |
| 3 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | - | 120 | 96 | 47\% |
| 4 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | $\left[\left(\mathrm{Ph}_{3} \mathrm{P}\right) \mathrm{Au}\right] \mathrm{Cl}(10$ | 120 | 14 | 69\% |
| 5 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | AgOTf (10 mol\%) | 120 | 14 | 92\% |
| 6 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | AgOTf (10 mol\%) | 80 | 96 | 15\% |
| 7 | - | AgOTf (1 eq.) | 120 | 96 | traces |
| 8 | $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ | AgOTf (10 mol\%) | 120 | 96 | 69\% |

First, the desired cyclization of pyrazole 4 to $2 H$-furo[2,3-c]pyrazole 15 was attempted by using the synthetic protocol, involving $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ in dry DMF at $60^{\circ} \mathrm{C}$, which was previously successfully used for the preparation of benzo[b]furans by the cyclisation of $o$-alkynylphenols ${ }^{74 \mathrm{c}}$. Unfortunately, no addition of the hydroxy group across the carbon-carbon triple bond was observed even after heating the reaction mixture in the analogues reaction conditions for 24 hours (Table 3.2, Entry 1). The low reactivity of the pyrazole hydroxy functionality in comparison with that of $o$ alkynylphenols may be explained by the structural specificity of pyrazol-3-ols, whereas they could exist in two tautomeric forms, namely the OH -form and the $\mathrm{NH}-$ form ${ }^{71}$. The formation via the 5-endo-dig cyclization of some amount of compound 15 was detected by LC/MS measurements only when the reaction mixture contained
$\mathrm{Cs}_{2} \mathrm{CO}_{3}$ as a base and was heated at $120{ }^{\circ} \mathrm{C}$ for 4 days (Table 3.2, Entry 2). Surprisingly, when $\mathrm{K}_{2} \mathrm{CO}_{3}$ was used instead of $\mathrm{Cs}_{2} \mathrm{CO}_{3}$, the desired 2-phenyl-2H-furo[2,3$c$ pyrazole was obtained in a significantly higher yield (47\%, Table 3.2, Entry 3). Some of the most effective catalysts for the electrophilic activation of alkynes under homogeneous conditions are gold(I) ${ }^{72}$ and silver(I) ${ }^{73}$ salts or complexes, and a broad range of versatile synthetic methods have been developed for the construction of car-bon-heteroatom bonds by using these types of catalysts. For example, the gold (I) catalyst $\left[\left(\mathrm{Ph}_{3} \mathrm{PAu}\right)_{3} \mathrm{O}\right] \mathrm{BF}_{4}$ was applied in the regioselective intramolecular cyclization of alkynols to construct bicyclic ethers ${ }^{74}$, while the silver(I) catalyst AgOTf efficiently catalyzed the intramolecular cyclization of phenoxyethynyl diols into 2,3-unsaturated lactones ${ }^{75}$. In this case, adding $10 \mathrm{~mol} \%$ of chloro(triphenylphosphine)gold(I) improved the yield of product 15 to $69 \%$ (Table 3.2, Entry 4). Even better results were obtained when AgOTf was used as a catalyst. In this case, the target product $\mathbf{1 5}$ was obtained with an excellent $92 \%$ yield (Table 3.2, Entry 5).

It is important to note that the reaction temperature had a significant effect on the yield of the product, and when the temperature was lowered to $80^{\circ} \mathrm{C}$, the yield of the product did not exceed $15 \%$ (Table 3.2, Entry 6). The presence of the base plays a crucial role in the cyclization described herein, and the transformation of pyrazole 4 to pyrazole $\mathbf{1 5}$ did not occur in the presence of only the catalyst and no base (Table 3.2, Entry 7).

Finally, the reaction optimization experiments showed that the $\mathrm{Cs}_{2} \mathrm{CO}_{3} / \mathrm{AgOTf}$ system does not offer any advantages for this cyclization compared to the $\mathrm{K}_{2} \mathrm{CO}_{3} / \mathrm{AgOTf}$ system (Table 3.2, Entry 8).


4-13


15-23

Scheme 3.2 Cyclization of hydroxy-alkynyl substrates to 2,5-disubstituted 2 H -furo[2,3-c]pyrazoles

With the identified optimized conditions for the 5-endo-dig cyclization reaction, the scope of this transformation for the preparation of several 2,5-disubstituted 2 H -furo[2,3-c]pyrazoles was explored (Scheme 3.2). For substrates 4-13, the reactions were complete after 14 hours at $120^{\circ} \mathrm{C}$, and the products $\mathbf{1 5}-\mathbf{2 3}$ were generated in fair to excellent yields (Table 3.3).
Table 3.3 Synthesized various 2-phenyl-2H-furo[2,3-c]pyrazole derivatives

|  | $\mathbf{1 5}$ | $\mathbf{1 6}$ | $\mathbf{1 7}$ | $\mathbf{1 8}$ | $\mathbf{1 9}$ | $\mathbf{2 0}$ | $\mathbf{2 1}$ | $\mathbf{2 2}$ | $\mathbf{2 3}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{R}$ | Ph | $4^{\prime}-\mathrm{MePh}$ | $4^{\prime}-\mathrm{EtPh}$ | $4^{\prime}$-FPh | $4^{\prime}$-OEtPh | 3-Thienyl | $n$-Bu | $n$-Pent | cyclopropyl |
| Yield, \% | 92 | 87 | 92 | 86 | 81 | 62 | 94 | 82 | 90 |

However, the cyclization was not accomplished when pyrazole 13 ( $\mathrm{R}=\mathrm{TMS}$ ) was used as a precursor; the decomposition of compound $\mathbf{1 3}$ was observed.

The structure assignment of compounds $\mathbf{1 5 - 2 3}$ was determined by multinuclear NMR and IR spectroscopy and high-resolution mass spectrometry (HRMS) data.


Figure $3.2{ }^{1} \mathrm{H}$ NMR spectrum of compound 20
Figure 3.2 shows ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{2 0}$. The presence of a singlet signal of $4-\mathrm{H}$ proton at 6.54 ppm proves the formation of desired furopyrazole product. In the 'aromatic' part, there are signals of phenyl ring protons (multiplets of Ph 4H , $\mathrm{Ph} 3,5-\mathrm{H}$, and $\mathrm{Ph} 2,6-\mathrm{H}$ at $7.26-7.28,7.44-7.47$, and $7.71-7.72 \mathrm{ppm}$, respectively), thienyl ring protons (7.37-7.39 (Th 4,5-H) and 7.66-7.67 (Th 2-H) ppm), and a pyrazole ring proton ( $7.74(3-\mathrm{H}) \mathrm{ppm})$.

To sum up, it has been demonstrated a new, four-step synthetic route to $2 H$ furo[ $2,3-c]$ pyrazoles, starting from commercially available 1 -phenyl-3-pyrazolidinone. The oxidation of the latter compound with an aqueous acidic $\mathrm{FeCl}_{3}$ solution and iodination of the intermediate compound with iodine in DMF smoothly afforded 1-phenyl-4-iodopyrazol-3-ol, which underwent a Pd-catalyzed coupling with terminal alkynes to give the corresponding 4 -alkynyl-3-hydroxy-1-phenyl- 1 H -pyrazoles. The desired 5 -endo-dig cyclization leading to the formation of the 2 H -furo[2,3-c]pyrazole ring system can be easily achieved by the heating of the aforementioned hydroxyalkynyl substrates with base in DMF in the presence of a gold(I) or silver(I) catalyst.

### 3.2. Synthesis of substituted $\mathbf{2 H}$-pyrazolo[4,3-c]pyridines

Metwally and Deeb ${ }^{41}$ have recently reported that pyridine nucleaus containing compounds are significantly attractive to scientists due to their employment in drug discovery. Considering high therapeutic characterization of the pyridine-incorporated drugs, medicinal chemists have a worthwhile task to synthesize novel chemotherapeutic agents. Among numerous biologically active annelated pyrazole derivatives, synthetically demanding $2 H$-pyrazolo[4,3-c]pyridines are relatively understudied ${ }^{6}$.

In recent research, it has been demonstrated that $2 H$-pyrazolo[4,3-c]pyridines can as well be easily accessed from 3-hydroxy-1-phenyl-1H-pyrazole-4-carbaldehyde and the corresponding ethanone via intermediate triflates, making use of a Sonogashira cross-coupling and dry ammonia induced cyclization reactions ${ }^{6,77}$. Therefore, in this study, it was further examined the applicability of this synthetic approach and prepared a library of various $2 H$-pyrazolo[4,3-c]pyridines, primarily varying by the substituents at the $2-, 4-$, and 6 -positions in order to assess their biological activity and formulate a possible structure-activity relationships ${ }^{6}$.

### 3.2.1. Synthesis of starting materials

In a series of recent publications, it has been demonstrated that pyrazole-4carbaldehydes, carrying an alkynyl function adjacent to the formyl moiety, are valuable starting materials for the construction of condensed pyrazole systems such as di-pyrazolo[1,5-a:4,3-c]pyridines ${ }^{76 a}$, $2 H$ - and $3 H$-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazoles, $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b’]dipyridines, 13,13a-dihydro$3 H$-pyrazolo[4',3':3,4]pyrido[1,2-a]pyrimidi-nes ${ }^{76 \mathrm{~b}}$, and 2,6-diphenyl-2H-pyra-zolo[4,3-c] pyridines ${ }^{6,76 c, \mathrm{~d}}$.

In this study ${ }^{16}$, there was further elaborated the synthetic potential of 1-substi-tuted- 1 H -pyrazoles that are carrying an alkynyl function adjacent to the carbonyl moiety. The synthesis of starting triflates 30, 31, 36, and $\mathbf{3 7}$ for the Sonogashira crosscoupling reaction is presented in Schemes 3.3 and 3.4. Commercially available, but otherwise undescribed 3-hydroxy-1-methyl-1 $H$-pyrazole-4-carbaldehyde 23, which was obtained by refluxing methylhydrazine with 2,3-dichlorpropionate in EtOH overnight, the precursor for the preparation of the triflate 31, was synthesized following an analogous approach, which has been previously described for its analog 3-hy-droxy-1-phenyl-1 H -pyrazole-4-carbaldehyde $\mathbf{2}^{77}$ (Scheme 3.3). In short, the Vils-meier-Haack reaction conditions were applied for the synthesis of 4-benzyloxy-1-me-thyl-1H-pyrazole $\mathbf{2 5}$, which was obtained by benzylation of the readily available starting compound $23^{76 \mathrm{c}}$ under the standard conditions with benzyl chloride, similarly to the described procedure ${ }^{78}$. Upon heating compound 25 with $\mathrm{DMF} / \mathrm{POCl}_{3}$ at $70{ }^{\circ} \mathrm{C}$ for half an hour, the target carbaldehyde 27 was formed in $60 \%$ yield (Scheme 3.3). Alternatively, the preparation of carbaldehyde 27 can as well be achieved by the oxidation of (3-(benzyloxy)-1-methyl-1H-pyrazol-4-yl)methanol with manganese dioxide ${ }^{78 c}$. The debenzylation of compound 27 was accomplished by the treatment with TFA in toluene, conditions that are typically used for the selective deprotection of $o$ benzylsalicylaldehydes ${ }^{78 \mathrm{~d}}$, furnishing the target carbaldehyde 29 in $90 \%$ yield. Having
precursor 29 in hand, the author of this dissertation further treated it with triflic anhydride in the presence of TEA in DCM ${ }^{78 \mathrm{e}}$ to obtain the corresponding triflate $\mathbf{3 1}$ in $85 \%$ yield. The same synthetic pathway was applied to obtain 3 -triflyloxy-1-phenyl-1 H -pyrazole-4-carbaldehyde 30 from 3-hydroxy-1-phenyl-1 $H$-pyrazole 2, as it has been described previously ${ }^{81}$ (Scheme 3.3) ${ }^{6}$.


Reagents and conditions: (i) $\mathrm{BnCl}, \mathrm{NaH}, \mathrm{DMF}$, argon atmosphere, $0-60^{\circ} \mathrm{C}, 1 \mathrm{~h}, 90 \%$ (for 3), $85 \%$ (for 4); (ii) $\mathrm{POCl}_{3}$, DMF, $0-60^{\circ} \mathrm{C}, 0.5 \mathrm{~h}, 85 \%$ (for $\mathbf{5}$ ), $81 \%$ (for 6); (iii) TFA, toluene, rt, $15 \mathrm{~h}, 90 \%$ (for $\mathbf{7}$ ), $85 \%$ (for $\mathbf{8}$ ); (iv) Tf $\mathrm{Tf}_{2} \mathrm{O}, \mathrm{TEA}, \mathrm{DCM}, \mathrm{rt}, 1 \mathrm{~h}, 85 \%$ (for $\mathbf{9}$ ), $83 \%$ (for $\mathbf{1 0}$ ).

Scheme 3.3 Synthesis of pyrazolecarbaldehydes 30 and 31
Triflates $\mathbf{3 6}$ and $\mathbf{3 7}$ were synthesized similarly to the previously described approach ${ }^{77}$ (Scheme 3.4). Fries rearrangement conditions ( $\mathrm{AlCl}_{3}, \mathrm{SC}_{2}$ ) were applied to 1 -phenyl- 1 H -pyrazol-3-yl acetate $\mathbf{3 2}^{79 \mathrm{a}}$ and the corresponding pivalate $\mathbf{3 3}$, which were obtained from the readily available 3-hydroxy-1-phenyl- 1 H -pyrazole $\mathbf{2}^{79 \mathrm{~b}}$ and acetic anhydride or isobutylchloride, respectively. Triflation of 1-phenyl- 1 H -pyrazol-3-ols $\mathbf{3 4}^{71}$ and $\mathbf{3 5}$ afforded 1-phenyl-1 H -pyrazol-3-yl trifluoromethanesulfonates $\mathbf{3 6}$ and $\mathbf{3 7}$ in $85 \%$ and $96 \%$ yield, respectively ${ }^{6}$.

Firstly, pyrazole $\mathbf{2}$ was acylated refluxing in acetic anhydride for half an hour to get compound $\mathbf{3 2}$ with $79 \%$ yield, whereas to obtain compound $\mathbf{3 3}$ with isolated $90 \%$ yield, 3-hydroxypyrazole $\mathbf{2}$ was reacted with isobutylchloride in DCM in the presence of pyridine. The latter compounds were used in Fries rearrangement reactions with $\mathrm{AlCl}_{3}$ in $\mathrm{CS}_{2}$ to isolate compounds $\mathbf{3 4}$ and $\mathbf{3 5}$ with $74 \%$ and $89 \%$ yields, respectively. Last, pyrazoles $\mathbf{3 4}$ and $\mathbf{3 5}$ participated in the reactions with triflate anhydride to get pseudohalogenides 36 and 37 with isolated yields $85 \%$ and $96 \%$, respectively (Scheme 3.4) ${ }^{6,77}$.


Reagents and conditions: (i) $\mathrm{Ac}_{2} \mathrm{O}, 100^{\circ} \mathrm{C}, 0.5 \mathrm{~h}$ or isobutylchloride, pyridine, $\mathrm{DCM}, \mathrm{rt}, 1 \mathrm{~h}$ (ii) $\mathrm{AlCl}_{3}, \mathrm{CS}_{2}$, reflux, 3 h ; (iii) $\mathrm{Tf}_{2} \mathrm{O}, \mathrm{TEA}, \mathrm{DCM}, \mathrm{rt}, 1 \mathrm{~h}$.

Scheme 3.4 Synthesis of ethenone 36 and propanone 37

### 3.2.2. Construction of $\mathbf{2 H}$-pyrazolo[4,3-c]pyridines from corresponding 3 -alkynylpyrazoles

The construction of the target pyrazolopyridines from 3-alkynylpyrazole-4carbaldehydes, ethanones, and propanones is represented in Scheme 3.5. The prepared triflates $\mathbf{3 0}, \mathbf{3 1}, \mathbf{3 6}$, and $\mathbf{3 7}$ were successfully coupled with various alkyl and aryl acetylenes under the standard Sonogashira cross-coupling reaction conditions $\left(\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}, \mathrm{CuI}, \mathrm{TEA}, \mathrm{DMF}\right)$ to give the corresponding 3-alkynyl-1-phenyl-1 H pyrazoles 38-55 in good to very good yields, in most cases (Scheme 3.5, Table 3.4). The couplings of 4-acetyl-1-phenyl-1H-pyrazol-3-yl trifluoromethanesulfonate (36) with phenyl-, 3-thienyl-, and $n$-pentylacetylenes gave the corresponding products in fair yields of less than $70 \%$. Notably, 1-phenyl-3-trifluoromethylsulfoxy-1 H -pyra-zole-4-carbaldehydes resulted in better yields of the Sonogashira cross-coupling reaction in comparison with the corresponding 4-ethanones. Finally, compounds 38-55, bearing the alkynyl group moiety adjacent to the carbonyl group, were treated with dry ammonia under the elevated temperature and pressure, allowing the direct formation of pyrazolo[4,3-c]pyridines 56-73 generally in very good to excellent yields (Scheme 3.5, Table 3.4) ${ }^{6,77}$.


Reagents and conditions: (i) $\mathrm{R}_{3}-\mathrm{C} \equiv \mathrm{CH}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, TEA, CuI, DMF, $60{ }^{\circ} \mathrm{C}, 1-38 \mathrm{~h}$; (ii) $\mathrm{NH}_{3}, \mathrm{MeOH}, 120^{\circ} \mathrm{C}, 15 \mathrm{~h}$.

Scheme 3.5 Synthesis of pyrazolepyridine systems 56-73
It is noteworthy that this methodology proved to be applicable to the synthesis of 2-phenyl-2H-pyrazolo[4,3-c]pyridine 63 and its 4-methyl and 4-isopropyl derivatives, 70 and 73, respectively (Scheme 3.5, Table 3.4$)^{6}$.

The synthesis of the aforementioned 63 has been accomplished before employing inconvenient highly toxic and unstable reactants, i.e., by cycloaddition of 3-pyridine, which is accessible by lead tetraacetate oxidation of 1-aminotriazolo[4,5-c]pyridine to $N$-phenylsydnone ${ }^{80}$. In contrast, the synthetic approach makes use of the convenient TMS-protected, commercially available, trimethylsilylacetylene, which efficiently undergoes Sonogashira cross-coupling with corresponding triflates 30, 36, and 37, furnishing 42,52, and 55 in 78-87\% yield. The latter, due to the convenient lability of the TMS protecting group, upon the treatment with dry ammonia under the elevated temperature and pressure, directly gives rise to the target TMS-deprotected 2-phenyl-2H-pyrazolo[4,3-c] pyridines 63, 70, and 73 in $88-95 \%$ yield (Scheme 3.5, Table 3.4) ${ }^{6}$.

Table 3.4 Synthesized Sonogashira type precursors 38-55 and pyrazolo[4,3-c]pyridines 5673

| Entry |  | 'Sonogashira' product | Yield, \% | Cyclisation product | Yield, \% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1. | 31: $\mathrm{R}^{1}=\mathrm{Me}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 38: } \mathrm{R}^{1}=\mathrm{Me}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 75 | $\begin{aligned} & \text { 56: } \mathrm{R}^{1}=\mathrm{Me}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 79 |
| 2. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 39: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 81 | $\begin{aligned} & \text { 57: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 89 |
| 3. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 40: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=3 \text {-Thienyl } \end{aligned}$ | 75 | $\begin{aligned} & \text { 58: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=3 \text {-Thienyl } \end{aligned}$ | 94 |
| 7. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 41: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\text { cyclopropyl } \end{aligned}$ | 86 | $\begin{aligned} & \text { 59: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\text { cyclopropyl } \end{aligned}$ | 98 |
| 8. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 42: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 90 | $\begin{aligned} & \mathbf{6 0}: \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 84 |
| 9. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 43: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=n \text {-Pent } \end{aligned}$ | 92 | $\begin{aligned} & \text { 61: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=n-\mathrm{Pent} \end{aligned}$ | 89 |
| 10. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 44: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OH} \end{aligned}$ | 90 | $\begin{aligned} & \text { 62: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OH} \end{aligned}$ | 86 |
| 11. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$ | $\begin{aligned} & \text { 45: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{TMS} \end{aligned}$ | 78 | $\begin{aligned} & \text { 63: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}, \\ & \mathrm{R}^{3}=\mathrm{H} \end{aligned}$ | 95 |
| 12. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 46: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 65 | $\begin{aligned} & \text { 64: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 91 |
| 13. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 47: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=3 \text {-Thienyl } \end{aligned}$ | 66 | $\begin{aligned} & \text { 65: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=3 \text {-Thienyl } \end{aligned}$ | 81 |
| 15. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 48: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\text { cyclopropyl } \end{aligned}$ | 80 | $\begin{aligned} & \text { 66: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\text { cyclopropyl } \end{aligned}$ | 80 |
| 16. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 49: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 75 | $\begin{aligned} & \text { 67: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 88 |
| 17. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \mathbf{5 0}: \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=n \text {-Pent } \end{aligned}$ | 65 | $\begin{aligned} & \text { 68: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=n \text {-Pent } \end{aligned}$ | 88 |
| 18. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 51: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OH} \end{aligned}$ | 86 | $\begin{aligned} & \text { 69: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OH} \end{aligned}$ | 86 |
| 19. | 30: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}$ | $\begin{aligned} & \text { 52: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{TMS} \end{aligned}$ | 83 | $\begin{aligned} & \text { 70: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{CH}_{3}, \\ & \mathrm{R}^{3}=\mathrm{H} \end{aligned}$ | 88 |
| 20. | 37: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}$ | $\begin{aligned} & \text { 53: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 90 | $\begin{aligned} & \text { 71: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \operatorname{Pr}_{3}, \\ & \mathrm{R}^{3}=\mathrm{Ph} \end{aligned}$ | 92 |
| 21. | 37: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}$ | $\begin{aligned} & \text { 54: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 81 | $\begin{aligned} & \text { 72: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \operatorname{Pr}_{3}, \\ & \mathrm{R}^{3}=n-\mathrm{Bu} \end{aligned}$ | 84 |
| 22. | 37: $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}$ | $\begin{aligned} & \text { 55: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \mathrm{Pr}, \\ & \mathrm{R}^{3}=\mathrm{TMS} \end{aligned}$ | 87 | $\begin{aligned} & \text { 73: } \mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=i \operatorname{Pr}_{3}, \\ & \mathrm{R}^{3}=\mathrm{H} \end{aligned}$ | 92 |

Figure 3.3 shows ${ }^{1} \mathrm{H}$ NMR spectra of compounds 52 and 70. It is important to mention that during the cyclization process, the trimethylsilyl group was eliminated. The presence of 7-H proton (multiplet) and 6-H proton (doublet) signals at 7.46-7.48,
and 8.20 ppm in the spectrum of compound 70 proves the formation of the desired pyrazolopyridine 70. In the 'aliphatic' part of compound 52 spectrum, there are signals of trimethylsiliyl group protons at 0.30 ppm , which is absence in the spectrum of compound 70 and methyl group protons at 2.66 ppm (in the spectrum of compound 70, the signal of methyl group protons is observed at 2.85 ppm ). In the 'aromathic' part of compound 52 spectrum, there are signals of phenyl ring protons (multiplets of $4-\mathrm{H}$, $3,5-\mathrm{H}$, and $2,6-\mathrm{H}$ ) at $7.36-7.38,7.46-7.48$, and $7.70-7.71 \mathrm{ppm}$, respectively; moreover, there is a signal of pyrazole proton at 5 -position at 8.40 ppm . Meanwhile, in the 'aromathic' part of compound 70 spectrum, there are signals of phenyl ring protons (multiplets of $4-\mathrm{H}, 3,5-\mathrm{H}$, and $2,6-\mathrm{H}$ at $7.46-7.48,7.55-7.58$, and $7.91-7.93 \mathrm{ppm}$, respectively), and there is a singlet signal of pyrazole proton at 3-position at 8.58 ppm . The complete and unambiguous assignments of the ${ }^{1} \mathrm{H}$ resonances were achieved by using standard Bruker software in conjunction with standard NMR spectroscopic techniques, such as DEPT, COSY, TOCSY, NOESY, gs-HSQC, and gs-HMBC.


Figure $3.3{ }^{1} \mathrm{H}$ NMR spectra of compounds 52 and 70
3.2.3. Evaluation of biological activity of synthesized pyrazolo[4,3-c]pyridines

All synthesized 2H-pyrazolo[4,3-c]pyridines were sent to the Palacký University in Olomouc, the Czech Republic, for screening for anticancer activity against two cell lines, i.e., K-562 and MCF-7.

### 3.2.3.1. Anticancer activity in vitro

The above mentioned synthesized pyrazole compounds 56-74 were evaluated for their cytotoxicity against two human cancer cell lines, i.e., K-562 (chronic myeloid leukemia cells) and MCF-7 (breast cancer cells). In most cases, it was noticed that the majority of pyrazolopyridines has exhibited moderate cytotoxicity with $\mathrm{GI}_{50}$ values in the micromolar range (Table 3.5). Notably, the most effective compounds contained phenyl group at the 2-position with no substituent at the 4-position and either an aryl or alkyl at the 6-position (compounds 57, 58, 60, 61). It is noteworthy that the replacing of phenyl substituent at 2-position (compound 57) with a methyl group resulted in a complete loss of the activity (compound 56). The increasing of bulkiness of the substituents at the 4 -position reduced the cytotoxicity of derivatives, i.e., the introducing of methyl (compounds $\mathbf{6 4}, \mathbf{6 8}$ ) and isopropyl (compound 71) substituents resulted in a gradual decrease in potency of the compounds compared to the unsubstituted derivatives (compounds $\mathbf{5 7}, \mathbf{6 1}$ ). The replacement of the substituents at the 6 -position enabled the fine-tuning of cytotoxicity of the compounds. Surprisingly, lack of the substituent at this position or introduction of a polar ethylhydroxy group resulted in completely inactive derivatives (compounds $62,63,69,70$ ), while various other substituents were relatively well tolerated. The most effective substituents at 6 -position were alkyl chains and aromatic rings. The derivatives, bearing a phenyl ring at the 7position, exhibited good cytotoxic values with $\mathrm{GI}_{50}$ values for compound $\mathbf{5 7}$, reaching $3.4 \mu \mathrm{M}$ for $\mathrm{K}-562$ and $4.8 \mu \mathrm{M}$ for MCF-7 cells (Table 3.5).

Table 3.5 Synthesized Sonogashira type precursors 38-55 and pyrazolo[4,3-c]pyridines 5673

| Structure | Compound no | $\mathbf{R}^{1}$ | $\mathbf{R}^{2}$ | $\mathbf{R}^{3}$ | GI ${ }_{50}$ ( $\mu \mathrm{M}$ )* |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | K-562 | MCF-7 |
|  | 56 | Me | H | Ph | >100 | >100 |
|  | 57 | Ph | H | Ph | 3.4 | 4.8 |
|  | 58 | Ph | H | 3-thienyl | 8.3 | 9.8 |
|  | 59 | Ph | H | cyclopropyl | 60.5 | 69.3 |
|  | 60 | Ph | H | $n$-Bu | 6.7 | 13.3 |
|  | 61 | Ph | H | $n$-Pent | 2.1 | 11.8 |
|  | 62 | Ph | H | ethylhydroxy | $>100$ | $>100$ |
|  | 63 | Ph | H | H | >100 | >100 |
|  | 64 | Ph | $\mathrm{CH}_{3}$ | Ph | 6.3 | 11.0 |
|  | 65 | Ph | $\mathrm{CH}_{3}$ | 3-thienyl | 10.0 | 22.2 |
|  | 66 | Ph | $\mathrm{CH}_{3}$ | cyclopropyl | 25.9 | 63.5 |
|  | 67 | Ph | $\mathrm{CH}_{3}$ | $n$-Bu | 8.5 | 11.1 |
|  | 68 | Ph | $\mathrm{CH}_{3}$ | $n$-Pent | 6.9 | 17.0 |
|  | 69 | Ph | $\mathrm{CH}_{3}$ | ethylhydroxy | $>100$ | $>100$ |
|  | 70 | Ph | $\mathrm{CH}_{3}$ | H | >100 | >100 |
|  | 71 | Ph | $i \mathrm{Pr}$ | Ph | 36.8 | 91.0 |
|  | 72 | Ph | $i \operatorname{Pr}$ | $n$-Bu | 12.9 | 33.3 |
|  | 73 | Ph | $i \operatorname{Pr}$ | H | >100 | 98.8 |

[^0]Unfortunately, the attempts to replace the phenyl ring with a slightly less aromatic thiophene or aliphatic cyclopropyl rings resulted in less active derivatives; phenyl derivatives 57 and 64 were twofold more potent than analogous thiophen-3-yl derivatives 58 and $\mathbf{6 5}^{6}$ (Table 3.5).

Interestingly, the replacing of the cyclopropyl ring with linear aliphatic substituents resulted in the increased cytotoxicity of the compounds. Thus, among the prepared derivatives, compound $\mathbf{6 1}$, bearing a pentyl substituent at the 6-position, a phenyl ring at the 2-position, and lacking a substituent at the 4-position, proved to be the most cytotoxic, reaching $2.1 \mu \mathrm{M} \mathrm{GI}_{50}$ value for $\mathrm{K}-562$ cells (Table 3.5) ${ }^{6}$.

### 3.2.3.2. Effect on the cell cycle and apoptosis

All compounds displaying a $\mathrm{GI}_{50}$ lower than $80 \mu \mathrm{M}$ in at least one cell line underwent cell-cycle investigation in K-562 and MCF-7 cell lines to gain preliminary information about their mechanism of action (Table 3.6). The cells were treated with tested compounds for 24 h at $10 \mu \mathrm{M}$ concentrations. Tubuline-interfering agent nocodazole and cyclin-dependent kinase 4 inhibitor palbociclib were used as positive controls, causing dominant mitotic and G1 arrest, respectively.
Table 3.6 Cell cycle analysis in K-562 and MCF-7 cells treated with active 2-phenyl-2H-py-razolo[4,3-c]pyridines at a single dose of $10 \mu \mathrm{M}$

| Compound | K-562 cell cycle phases (\%) |  |  |  | MCF-7 cell cycle phases (\%) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | subG1 | $\mathbf{G 1}$ | $\mathbf{S}$ | $\mathbf{G 2} / \mathbf{M}$ | subG1 | G1 | $\mathbf{S}$ | $\mathbf{G 2 / M}$ |
| Untreated | 15.0 | 34.6 | 50.3 | 15.0 | 7.1 | 60.7 | 34.1 | 5.3 |
| $\mathbf{5 7}$ | 43.0 | 19.8 | 5.5 | 74.7 | 14.3 | 56.8 | 26.2 | 17.0 |
| $\mathbf{5 8}$ | 34.2 | 21.7 | 37.3 | 41.1 | 24.2 | 39.5 | 29.1 | 31.4 |
| $\mathbf{5 9}$ | 19.6 | 32.3 | 52.2 | 15.5 | 7.5 | 64.7 | 29.4 | 5.9 |
| $\mathbf{6 0}$ | 37.7 | 24.8 | 12.8 | 62.3 | 31.1 | 36.0 | 30.3 | 33.7 |
| $\mathbf{6 1}$ | 35.4 | 14.8 | 3.3 | 82.0 | 50.3 | 18.1 | 26.3 | 55.6 |
| $\mathbf{6 3}$ | 46.2 | 13.7 | 36.1 | 50.2 | 11.9 | 65.5 | 20.5 | 14.1 |
| $\mathbf{6 4}$ | 45.7 | 26.2 | 29.2 | 44.6 | 22.7 | 37.2 | 31.2 | 31.6 |
| $\mathbf{6 5}$ | 15.3 | 34.7 | 45.8 | 19.6 | 9.0 | 63.6 | 30.2 | 6.3 |
| $\mathbf{6 6}$ | 40.8 | 28.2 | 30.5 | 41.4 | 19.3 | 48.5 | 24.0 | 27.5 |
| $\mathbf{6 7}$ | 28.3 | 21.8 | 28.3 | 49.9 | 31.1 | 50.2 | 18.1 | 31.7 |
| $\mathbf{7 0}$ | 21.3 | 35.1 | 46.6 | 18.3 | 6.7 | 62.6 | 30.6 | 6.9 |
| $\mathbf{7 1}$ | 47.1 | 26.7 | 44.2 | 29.0 | 7.8 | 66.1 | 24.5 | 9.4 |
| nocodazole $(25$ | 49.4 | 3.6 | 26.8 | 69.7 | 32.1 | 14.4 | 49.9 | 35.7 |
| palbociclib | 35.0 | 54.0 | 30.1 | 15.9 | 7.5 | 90.0 | 5.5 | 4.6 |

The majority of compounds exhibited a clear effect on the cell cycle in both cell lines; the increased cell populations in G2/M phases and corresponding decreases in G1 and S phase populations were observed (Table 3.6). The highest percentages of a G2/M population were found in cultures treated with compounds 57, 60, 61, 64, and 68, all of which displayed the strongest cytotoxicities as well (Table 3.6). In addition, a substantial increase of sub-G1 populations was observed in cultures treated with the most potent compounds $57,58,60,61,64,65,67$, and 72 , indicating ongoing apop-
tosis. Sub-G1 populations have been usually higher in K-562 cells, probably as a consequence of their higher sensitivity to novel compounds, which was observed in the cytotoxicity assays (Table 3.6$)^{6}$.

b
K-562


Figure 3.4 Cell cycle arrest in K-562 cells treated with compound $\mathbf{6 1}$ for 24 h. (a) DNA histograms of cells treated with different doses of the compound 61. (b) Phosphorylation of histone H 3 at serine-10 in cells treated with $5 \mu \mathrm{M}$ dose of compound 61. Nocodazole was used as a positive control. ${ }^{10}$ Due to the strong cytotoxicity of $\mathbf{6 1}$ in the K-562 cell line, it was sought to identify the type of cell death that occurred.

Compound 61 with the strongest effect on the cell cycle was further assayed at several concentrations (Figure 3.4). DNA histograms revealed a clear dose-dependent arrest in G2/M phases. In order to discriminate between arrest in G2 and M, there was quantified the phosphorylation of histone H 3 at serine-10, a common mitotic marker. Flow cytometric analysis revealed a strong accumulation of cells with phosphorylated histone, confirming the arrest in mitosis (Figure 3.4) ${ }^{6}$.


Figure 3.5 Induction of apoptosis in cells treated with different doses of compound $\mathbf{6 1}$ for 24 h . (a) Caspase- $3 / 7$ activity was measured in lysates prepared from treated K-562 cells by using the fluorogenic substrate Ac-DEVD-AMC and normalized against untreated control cell lysates. (b) Immunoblotting analysis of apoptosis-related proteins in treated K-562 and MCF-7 cells. Doxorubucin (DOX, $1 \mu \mathrm{M}$ ) was used as a control. Actin levels were detected to verify equal loading ${ }^{6}$.

The caspase activation in treated cells was measured by an enzymatic assay, using fluorescently labeled peptide substrate Ac-DEVD-AMC of caspases 3/7 (Figure 3.5) that revealed clear dose-dependent responses in the micromolar range. In parallel,
the lysates of treated K-562 and MCF-7 cells were subjected to immunoblotting. The analysis revealed a dose-dependent increase in the 89 kDa fragment of PARP-1, a known caspase substrate (Figure 3.5$)^{6}$. The author as well detected phosphorylation of histone H2AX at Ser-139 ( $\gamma \mathrm{H} 2 \mathrm{AX}$ ), a modification required for DNA fragmentation during apoptosis ${ }^{81}$.

There has been observed the appearance of a slowly migrating, phosphorylated form of Bcl-2 connected with the G2/M arrest of the cell cycle and apoptosis. In summary, the analyses confirmed that compound 61 can activate apoptotic machinery in a dose-dependent manner ${ }^{6}$.

In order to describe the mechanism of antimitotic action of the prepared compounds, there was explored the CDK1/cyclin B , a well-known mitotic regulator, as a possible target, especially due to the structural similarity to $2,3,4,6$-substituted pyra-zolo[4,3-c]pyridines which have been identified as kinase inhibitors ${ }^{82}$. The synthesized compounds 56-74 were screened for their inhibitory activity against CDK1, but the assay did not reveal any inhibition (data not shown). The further profiling of compound 61 against other mitotic kinases, including DAPK3, CDK1, CHK1, NIM1, AURKA, NDR1, NEK2/4/6, PIM2, PLK2/3, TTK, and WEE1, did not point to any of these kinases as a target of compound $\mathbf{6 1}$ (data not shown). This finding was expected, given that these compounds lack an H -bond donor-acceptor motif which is common in most kinase inhibitors, including the previously mentioned pyrazolo[4,3c]pyridines ${ }^{6,77}$.

To sum up, there was developed an efficient synthetic approach to obtain variously substituted 2 H -pyrazolo[4,3-c]pyridines, employing Sonogashira cross-coupling and a subsequent substituent-tolerant annulation reaction in the presence of ammonia. Consequently, a group of substituted $2 H$-pyrazolo[4,3-c]pyridines was evaluated for the cytotoxicity against K-562 and MCF-7 cancer cell lines; they exhibited anticancer activity in vitro through the arresting cell cycle in mitosis and induction of apoptosis; however, the mechanism of cellular action remains unclear.

### 3.3. Synthesis of substituted benzopyrano[2,3-c]pyrazol-4(2H)-ones

It is known that benzopyrano and pyrazole compounds have a broad spectrum of biological activities ${ }^{42-44}$. Particularly, the heterocyclic compounds, containing pyrazole and benzofuran nuclei, are significantly attractive to scientists for numerous applications and evaluations of biological activities ${ }^{59-62}$. Due to this reason, it was decided to synthesize a series of novel benzopyrano[2,3-c]pyrazol-4(2H)-ones and screen their biological activities. 2-hydroxy aromatic ketones are applicable synthons to synthesize oxygen-containing heterocyle systems such as benzofuranone, chromanone, benzoxazole, and dibenzooxazepine; moreover, these heterocycles can be employed as building blocks for drugs such as celiprolol, acebutolol, and propafenone ${ }^{83}$. Holzer et al. reported a simple and straightforward synthesis of fluorosubstituted 3-methyl-1-phenylchromeno[2,3-c]pyrazol-4(1H)-one from the corresponding 4-aroylpyrazol-5-ols, using $\mathrm{NaH} / \mathrm{DMF}$ or $\mathrm{K}_{2} \mathrm{CO}_{3} / \mathrm{MeCN}$ system ${ }^{84}$. The author of this dissertation has decided to examine this methodology and to synthesize the substituted chromeno[2,3-c]pyrazol-4(2H)-ones (Scheme 3.6).

First of all, as starting materials were chosen 3-hydroxy-1-phenyl-1H-pyrazole 2 and 3-hydroxy-1-methyl-1H-pyrazole (23). These synthons were transformed to acylated pyrazoles $\mathbf{7 5 - 8 9}$ by using aroyl chlorides or aroyl acids. $O$-acylation with aroyl chlorides was performed in the presence of TEA in chloroform at room temperature to obtain the desired products ${ }^{85}$. Meanwhile, the treating of pyrazoles 2 or $\mathbf{2 3}$ with aroyl acids was more complicated (Table 3.7). 3-hydroxypyrazole $\mathbf{2}$ and 2-bromo-4-methylbenzoic acid were chosen as a model system to determine optimal reaction conditions. Firstly, the author tried to perform this acylation under the above mentioned reaction conditions, using TEA as a base and chloroform as a solvent; however, no desired product was detected (Table 3.7, Entry 1) ${ }^{85}$. Later, the esterification was tested in the presence of pyridine: chloroform was used as a solvent; unfortunately, the reaction did not work (Table 3.7, Entry 2) ${ }^{86}$. Phakhodee et al. published an investigation on aryl esterification mediated by the $\mathrm{Ph}_{3} \mathrm{P}-\mathrm{I}_{2} / \mathrm{Et}_{3} \mathrm{~N}$ system; the author decided to examine this methodology, but the desired product was not obtained (Table 3.7, Entry 4) ${ }^{87}$.

Table 3.7 Study of reaction conditions for $O$-acylation by using aroyl acids

|  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | Base | Solvent | Additives | Temperature | Yield, \% |
| 1 | TEA | Chloroform | - | rt | No |
| 2 | pyridine | Toluene | - | $110{ }^{\circ} \mathrm{C}$ | No |
| 3 | $\mathrm{K}_{2} \mathrm{CO}_{3}{ }^{88}$ | DMF | Cu | rt - > $120^{\circ} \mathrm{C}$ | No |
| 4 | TEA | DCM | $\mathrm{I}_{2}, \mathrm{PPh}_{3}$ | Rt | No |
| 5 | DMAP | DCM | DCC | $0^{\circ} \mathrm{C}->\mathrm{rt}$ | 80 |

In 2012, Kwon et al. reported a synthetic route, involving DCC coupling in order to transform benzoic acids to esters ${ }^{89}$. This DCC coupling was examined under Kwon proposed reaction condition by treating 3-hydroxypyrazole with 2-bromo-4methylbenzoic acid. The desired product was obtained with $80 \%$ yield (Table 3.7, Entry 5). Under this protocol, firstly, DCC was added to the solution of benzoic acid in DCM, and the mixture was stirred for 1 hour at $0^{\circ} \mathrm{C}$ temperature in order to make acid active for the esterifiction. After that, 3-hydroxyprazole and DMAP was added to the reaction mixture, and it was stirred at $0{ }^{\circ} \mathrm{C}$ temperature for 2 hours. Later, the stirring was continued for 24 hours at room temperature. Under these conditions, the compounds $\mathbf{7 6}, \mathbf{7 8}, \mathbf{8 0}, \mathbf{8 1}, \mathbf{8 5}$, and $\mathbf{8 7}$ were obtained with isolated $71-81 \%$ yield (Scheme 3.6, Table 3.9).


2, $\mathrm{R}^{1}=\mathrm{Ph}$
23, $\mathrm{R}^{1}=\mathrm{Me}$

75-83, $\mathrm{R}^{1}=\mathrm{Ph}, 71-89 \%$
$\mathbf{8 4 - 8 9}, \mathrm{R}^{1}=\mathrm{Me}, 70-85 \%$

Scheme 3.6 $O$-acylation of pyrazoles 2 and 23 with aroyl acids or chlorides


Figure 3.6 ${ }^{1} \mathrm{H}$ NMR spectra of compound 78
Figure 3.6 shows ${ }^{1} \mathrm{H}$ NMR spectrum of compound 78. In the 'aliphatic' part, there is a singlet signal of methyl group protons at 3.86 ppm . In the 'aromatic' part, there are signals of phenyl ring protons (multiplets of $\mathrm{Ph} 4-\mathrm{H}, 5-\mathrm{H}$, and $6-\mathrm{H}$ at $7.40-$ $7.44,7.18-7.21$, and $7.86-7.88 \mathrm{ppm}$, respectively) and signals of pyrazole ring protons (singlet of $5-\mathrm{H}$ at 7.32 ppm and doublet of $4-\mathrm{H}$ ar $6.23-6.24 \mathrm{ppm}$ ).


Scheme 3.7 Fries rearrangement reaction of pyrazoles 75-89

The next step in this synthetic approach was to prepare suitable compounds for the cyclizations of benzopyrano[2,3-c]pyrazol-4(2H)-ones (Scheme 3.7). Pyrazole esters were treated with excess of $\mathrm{AlCl}_{3}$ in carbon disulfide at $50^{\circ} \mathrm{C}$ temperature for 3-12 hours (Scheme 3.7). Using Fries rearrangement reaction conditions, the compounds 90-100 were synthesized in fair to very good yields (15-84\%) (Scheme 3.7, Table 3.9).

It is worth pointing out that 4-aroylpyrazol-3-ols, which contained other hetaroyl substituent than phenyl ring, did not accomplish Fries rearrangement reaction to afford the desired products, and the decomposition of 4-aroylpyrazol-3-ols to 3hydroxypyrazole was observed (Table 3.8, Entry 1). Nevertheless, it was decided to test other reaction conditions in order to get the desired compounds and avoid decomposition. 4-aroylpyrazol-3-ol $\mathbf{8 1}$ was chosen as a model compound for this investigation (Table 3.8, Entry 2-5).

Table 3.8 Study of reaction conditions for Fries rearrangement by using 4-hetaroylpyrazol-3-ol


| Entry | Base | Solvent | Temperature | Yield, \% |
| :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathrm{AlCl}_{3}$ | $\mathrm{CS}_{2}$ | $50^{\circ} \mathrm{C}$ | Decomposition |
| 2 | $\mathrm{BF}_{3} * \mathrm{Et}_{2} \mathrm{O}$ | - | $130^{\circ} \mathrm{C}$ | Decomposition |
| 3 | $\mathrm{BF}_{3} * \mathrm{Et}_{2} \mathrm{O}$ | Toluene | $130^{\circ} \mathrm{C}$ | Decomposition |
| 4 | Zn powder | DMF | $120^{\circ} \mathrm{C}$ | Decomposition |
| 5 | TfOH | ACN | rt | Decomposition |

In 2000, Jessica et al. published a one-step synthesis of monomethyl- and monoethyl- derivatives of acetylhydroquinones from boron trifluoride ethyl or methyl etherate complexes and hydroquinone diesters ${ }^{90}$. However, upon treatment with $\mathrm{BF}_{3}$ etherate, the expected 4 -aroylpyrazol-3-ol was not isolated; the starting material decomposed to 3-hydroxypyrazole (Table 3.8, Entry 2). $\mathrm{BF}_{3}$ etherate suitability to obtain the desired compound was as well tested by using solvent free conditions (Table 3.8, Entry 3). In 2004, Paul and Gupta reported that zinc powder by using DMF as a solvent significantly catalyzed the selective Fries rearrangement of acetylated phenols under the microwave/conventional heating, and the selective migration of the acyl group was observed with good yields (64-80\%) (Table 3.8) ${ }^{91}$. These proposed reaction conditions were tested (Table 3.8, Entry 4); unfortunately, the starting compound 81 decomposed to 3-hydroxypyrazole. Further, the synthetic route published by Murashige et al. was tried, but the desired product was not detected: the starting material $\mathbf{8 1}$ decomposed to 3-hydroxypyrazole 1 (Table 3.8, Entry 5) ${ }^{92}$.

Fries rearrangement was not accomplished with compounds 81-83 and 89. It has been noticed that during Fries rearrangement, the phenyl group at 1-position of the pyrazole ring influenced better results.




Figure $3.7{ }^{1} \mathrm{H}$ NMR spectra of compound 99
Figure 3.7 shows ${ }^{1} \mathrm{H}$ NMR spectrum of compound 99 . The absence of pyrazole $4-\mathrm{H}$ proton signal at $6.23-6.24 \mathrm{ppm}$, which was in the above showed Figure 3.6 of compound 87, proves the ester transformation to 4 -aroylpyrazol-3-ol 99. In the 'aliphatic' part, there is the same signal of methyl group protons at 3.80 ppm . In the 'aromatic' part, there are signals of phenyl ring protons (multiplets of $5-\mathrm{H}$ and $4,6-\mathrm{H}$ at $7.34-7.40 \mathrm{ppm}$ and $7.21-7.24 \mathrm{ppm}$, respectively) and a signal of pyrazole ring proton, doublet of $5-\mathrm{H}$, at 7.52 ppm .


Scheme 3.8 Construction of benzopyrano[2,3-c]pyrazol-4(2H)-ones 101-110
Having successfully prepared 4-aroylpyrazol-3-ols $\mathbf{9 0 - 1 0 0}$, their ability to participate in cyclization reaction to form benzopyrano[2,3-c]pyrazol-4(2H)-ones was
examined (Scheme 3.8$)^{84}$. This was briefly demonstrated by treating compounds $\mathbf{9 0}$ 100, bearing keto- and a hydroxy group at adjacent positions with $\mathrm{K}_{2} \mathrm{CO}_{3}$, where DMF was used as a solvent, for 18 hours at $120^{\circ} \mathrm{C}$ temperature (Scheme 3.8) ${ }^{84 \mathrm{~b}}$. In order to increase the yield of this reaction, NaH was employed as a base instead of $\mathrm{K}_{2} \mathrm{CO}_{3}$, and DMF changed to ACN; however ${ }^{84 \mathrm{~b}}$, the desired result was not achieved: the product 101 was isolated with $35 \%$ yield. The direct formation of benzopyrano[2,3-c]pyrazol$4(2 H)$-ones $\mathbf{1 0 1} \mathbf{- 1 1 0}$ was observed in very good yields ( $71-91 \%$, Scheme 3.8, Table 3.9).

Table 3.9 Structures and isolated yields of benzopyrano[2,3-c]pyrazol-4( $2 H$ )-ones and their precursors

| Entry | $\begin{aligned} & \mathrm{RCOCl} / \\ & \mathrm{RCOOH} \end{aligned}$ | Acylation product | Yield, \% | Product after Fries rearrangement reaction | Yield, \% | Cyclisation product | Yield, \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  |  | 86 |  <br> 90 | 81 |  | 90 |
| 2 |  |  <br> 76 | 82 |  | 75 |  | 75 |
| 3 |  |  <br> 77 | 81 |  | 77 |  <br> 103 | 91 |
| 4 |  |  | 88 |  | 84 |  | 88 |


| Entry | $\mathrm{RCOCl} /$ <br> RCOOH | $\underset{\text { uct }}{\text { Acylation prod- }}$ | Yield, \% | Product after Fries rearrangement reaction | Yield, \% | Cyclisation product | Yield, \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 5 |  |  | 80 |  | 83 |  | 89 |
| 6 |  |  | 84 |  | 80 | decomposition | - |
| 7 |  |  | 71 | decomposi |  |  |  |
| 8 |  |  | 82 | decomposi | on | - |  |
| 9 |  |  | 89 | decomposi | ion |  |  |
| 10 |  |  | 85 |  | 25 |  | 80 |


| Entry | $\begin{aligned} & \mathrm{RCOCl} / \\ & \mathrm{RCOOH} \end{aligned}$ | Acylation product | Yield, \% | Product after Fries rearrangement reaction | $\begin{gathered} \text { Yield, } \\ \% \end{gathered}$ | Cyclisation product | $\begin{gathered} \text { Yield, } \\ \% \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 11 |  |  <br> 85 | 83 |  | 69 |  | 93 |
| 12 |  |  | 80 |  | 80 |  | 90 |
| 13 |  |  <br> 87 | 85 |  | 65 |  | 76 |
| 14 |  |  <br> 88 | 85 |  | 74 |  | 71 |
| 15 |  |  <br> 89 | 80 | decomposition |  | - |  |

Presumably, due to the nitro group at 4-position in the benzoic acid, the cyclisation was not accomplished with compound 95 (Table 3.9).

Figure 3.8 shows ${ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}$ HMBC spectra of compound $\mathbf{1 1 0}$. There are two signals of $\mathrm{N}-1$ and $\mathrm{N}-2$ at -109.9 and -185.0 ppm , respectively. ${ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}$ HMBC spectra shows correlation between $\mathrm{CH}_{3}$ protons and $\mathrm{N}-2$ nitrogen atom and correlation between $3-\mathrm{H}$
proton and N-2 nitrogen atom. Moreover, in this spectra, there is a correlation between $\mathrm{CH}_{3}$ protons and $\mathrm{N}-1$ nitrogen atom.


Figure $3.8{ }^{15} \mathrm{~N},{ }^{1} \mathrm{H}$ HMBC spectra of compound $\mathbf{1 1 0}$
To sum up, as a result of this research, there was demonstrated a new three-step synthetic approach to chromeno[2,3-c]pyrazol-4(2H)-ones, starting from commercially available 1-phenylpyrazolidin-3-one. The desired cyclization product can be easily achieved by the heating of corresponding hydroxy-aroyl substrates with base in DMF.

### 3.4. Synthesis of pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazoles and related new ring systems by tandem cyclisation of vic-alkynylpyrazole-4-carbaldehydes with (het)aryl-1,2-diamines

Benzimidazole derivatives are associated with various types of pharmacokinetic and pharmacodynamic properties. For instance, the most prominent benzimidazole compound in nature is $N$-riosyldimethylbenzimidazole, which serves as an axial ligand for cobalt in vitamin $\mathrm{B} 12^{93}$. Ramesh Babu et al. reported that numerous pyr-ido[1,2-a]benzimidazoles have interesting biological properties like antiimflammatory, analgesic, antimicrobial, antiviral, and antineoplastic activities. Moreover, some of these compounds as well possess the fluorescent properties and were found to be useful in synthetic fibres ${ }^{94}$. It is known that compounds that are bearing ethynyl- and aldehyde groups in adjacent positions are important starting materials to synthesize the condensed heterocyclic compounds ${ }^{95}$. In addition, o-phenylendiamines are significant reagents to get fused $N$-heterocyclic building blocks, which are ubiquitous in
natural products and play a significant role in the pharmaceutical and agrochemical industries ${ }^{76 b}$.

In the continuation of this program, which is devoted to exploiting the synthetic potential of these building blocks, the author presents a simple method for the synthesis of 2 H - and 3 H -pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazole, $3 H$-pyrazolo[4,3$c$ ]imidazo[1,2-a:5,4-b']dipyridines, and 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyr-ido[1,2-a]perimi-dine derivatives. These compounds represent previously unknown tetracyclic and pentacyclic $N$-containing heterocyclic systems ${ }^{76 b}$.

### 3.4.1. Construction of the $\mathbf{3 H}$-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazoles

The synthesis strategy for the construction of the $3 H$-pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyr-ido[1,2-a]benzimidazole ring system was based on the tandem cyclisation of vic-al-kynylpyrazole-4-carbaldehydes with benzene-1,2-diamines, as outlined in Scheme 1. Recently, either the addition of 2-arylbenzimidazoles to alkynyl bromides followed by Pd-catalyzed intramolecular C-H vinylation or Rh-catalyzed intramolecular oxidative cross-coupling of 1 -styrylbenzimidazoles has been employed for the construction of the benzimidazo $[2,1-a]$ isoquinoline ring system, which is the benzo analogue of the aforementioned ring system ${ }^{96}$. However, since the commonly used method for obtaining benzimidazo[2,1-a]isoquinoline derivatives consists of reacting 2-alkynylbenzaldehydes with 1,2 -phenylenediamines under different reaction conditions, several variants such as microwave-accelerated tandem processes, silver(I) catalyzed tandem reactions in water, and iodocyclisations have been applied ${ }^{766}$.

The synthesis strategy for the construction of the $3 H$-pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a$ ]benzimidazole ring system was based on the tandem cyclisation of vic-al-kynylpyrazole-4-carbaldehydes with benzene-1,2-diamines.

### 3.4.1.1. The synthesis of the starting $\mathbf{5}$-alkynylpyrazole-4-carbaldehydes

The synthesis of the starting 5 -alkynylpyrazole-4-carbaldehydes 113-114 was accomplished via Sonogashira coupling of the corresponding 5-chloropyrazole-4carbaldehyde 112 and phenylacetylene or 1-hexine (Scheme 3.9). The intermediate compound 5-chloropyrazole-4-carbaldehyde $\mathbf{1 1 2}$ was obtained from pyrazole $\mathbf{1 1 1}$ by Vilsmeier-Haack reaction by using DMF and $\mathrm{POCl}_{3}\left(\right.$ Scheme 3.9) ${ }^{76 \mathrm{~d}}$.


Reagents and conditions: (i) $\mathrm{POCl}_{3}$, DMF, $-10-65^{\circ} \mathrm{C}, 1 \mathrm{~h}$; (ii) $\mathrm{R}-\mathrm{C} \equiv \mathrm{CH}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}, \mathrm{TEA}$, CuI, DMF, $75{ }^{\circ} \mathrm{C}, 1-12 \mathrm{~h}$;

Scheme 3.9 Synthesis of starting materials 113 and 114

The figure 3.9 shows ${ }^{13} \mathrm{C}$ NMR spectrum of compound $114(\mathrm{R}=n \mathrm{Bu})$. The presence of signals of $C \equiv \mathrm{CnBu}$ and $\mathrm{C} \equiv C n \mathrm{Bu}$ carbons at 70.4 and 97.2 ppm , respectively, proves the formation of the desired product. In the 'aliphatic' part, there are signals of $n \mathrm{Bu}$ group carbons at $13.7,19.3,22.2,30.4 \mathrm{ppm}$, respectively. In the 'aromatic' part, there are signals of phenyl ring carbons (119.9 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 128.3 ( $\mathrm{Ph} \mathrm{C}-4$ ), $129.8(\mathrm{Ph}$ $\mathrm{C}-3,5)$, and 138.9 ( $\mathrm{Ph} \mathrm{C}-1$ )) and signals of pyrazole ring carbons ( 125.7 (C-4), 128.3 (C-5), $139.3(\mathrm{C}-3))$ and $184.8(\mathrm{C}=\mathrm{O})$.

| $\begin{gathered} \stackrel{\infty}{\oplus} \\ \stackrel{\oplus}{\oplus} \\ \end{gathered}$ |  |  | $\stackrel{\text { a }}{\text { ¢ }}$ | \% | ¢ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |



Figure 3.9 ${ }^{13} \mathrm{C}$ NMR spectra of compound 114
3.4.1.2. Tandem cyclisations of the $3 H$-pyrazolo $\left[4^{\prime}, 3 ': 3,4\right]$ pyrido $[1,2-$ a]benzimidazoles from the corresponding 5-alkynylpyrazole-4-carbaldehydes

The synthesis strategy for the construction of the $3 H$-pyrazolo[4',3':3,4]pyr-ido[1,2-a] benzimidazole ring system was started from the investigation of optimal conditions for the cyclisation, where starting material $\mathbf{1 1 3}$ was chosen as a model compound. In order to find the optimal conditions for the tandem cyclisation, the starting material 113 was first reacted with benzene-1,2-diamine in DMSO under metalfree conditions or in the presence of a CuI catalyst with either conventional heating or microwave irradiation (Table 3.9, Entry 1,2) ${ }^{97}$. However, in all cases, the reaction gave a complex mixture of products, containing only the traces of desired cyclisation product 115. In contrast, the changing of reaction solvent to DMF dramatically enhanced the yield of compound $\mathbf{1 1 5}$ to $90 \%$ without any need for the CuI catalyst (Table 3.9, Entry 3, Scheme 3.10). The reaction of $\mathbf{1 1 4}$ with benzene-1,2-diamine under analogous conditions afforded the compound 116 in $79 \%$ yield (Scheme 3.10). DMSO may have been an inadequate solvent under the applied reaction conditions due to its
oxidizing properties ${ }^{98 a}$, whereas one of the starting materials, namely, benzene-1,2diamine, is very sensitive to the oxidation ${ }^{103 b, 76 \mathrm{~b}}$.


Scheme 3.10 Construction of pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazoles 115-116
Table 3.9 Optimization of tandem cyclisation conditions

| Entry | Solvent | Additives | Temperature | Yield, $\%$ |
| :---: | :--- | :--- | :--- | :--- |
| 1 | DMSO | - | $120^{\circ} \mathrm{C}^{*}$ | Traces |
| 2 | DMSO | CuI | $120^{\circ} \mathrm{C}^{*}$ | Traces |
| 3 | DMF | - | $120^{\circ} \mathrm{C}$ | $90 \%$ |

*The reaction was employed by microwave irradiation
With the optimal conditions at hand, the scope of the reaction was assessed by reacting pyrazole-4-carbaldehydes 113-114 with 4-methyl-, 4-chloro-, and 4-nitro-benzene-1,2-diamines (Scheme 3.11). In principle, the use of 'asymmetric' benzene-1,2-diamines can lead to the regioisomeric reaction products (117a-f and 118a-f, respectively, Scheme 3.11). The results obtained with precursor $\mathbf{1 1 3}$ demonstrated that the use of benzene-1,2-diamines with either electron-donating or electron-withdrawing groups in most cases gave good to acceptable yields of the target tetracycles 117ac, whereas the isomeric compounds 118a-c were not isolated by using this method (Scheme 3.11).


Scheme 3.11 Construction of pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazole derivatives 117 118

However, the reaction of compound 114 with 4-methylbenzene-1,2-diamine provided regioisomers $\mathbf{1 1 7 d}$ and $\mathbf{1 1 8 d}$ (ratio 1:0.55) as an inseparable mixture of $70 \%$ of total yield (Scheme 3.11) ${ }^{76 \mathrm{~b}}$. Surprisingly, in the case of the reaction of 4-chloro-benzene-1,2-diamine with pyrazole 114, the corresponding 9-chloro derivative 117 e was obtained only in $60 \%$ yield, whereas the employment of 4-nitrobenzene-1,2-diamine gave the 8 -nitro derivative $\mathbf{1 1 8 f}$ as the sole isolated product in $65 \%$ yield. Only trace amounts of the other regioisomers (118e and 117f, respectively) could be detected in the reaction mixture (Scheme 3.11) ${ }^{76 b}$.

### 3.4.1.3. Iodocyclisation of the $\mathbf{5}$-alkynylpyrazole-4-carbaldehyde

The study of Ouyang et al. presented that the cyclisation and halogenation could be achieved by one-pot reaction using o-phenylendiamines and benzenaldehydes, bearing alkinyl group in the $2^{\text {nd }}$ position (Scheme 3.12) ${ }^{95 \mathrm{a}}$.


Scheme 3.12. Iodocyclisation of the 5-alkynylpyrazole-4-carbaldehyde 113 in one pot reaction

The authors claimed that the iodocyclisation technology is an efficient method to perform copper-promoted electrophilic tandem cyclisation to form fused benzimidazoles with the halo group which are important substrates for drug discovery (Scheme 3.12).


Scheme 3.13 Study of compound 113 iodocyclisation
The synthetic route, proposed by Ouyang et al., was performed with precursor 113 (Scheme 3.13) in DMF in the presence of copper iodide and iodine; however, the desired product was not obtained. Later, it was decided to change the solvent from DMF to DMSO, but the reaction did not work. In order to develop iodocyclisation with starting material 113, the amount of copper iodide was increased from $5 \mathrm{~mol} \%$ to $20 \mathrm{~mol} \%$; unfortunately, the desired halogenated product was not achieved (Scheme 3.13).

### 3.4.2. Functionalization of $\mathbf{3 H}$-pyrazolo $\left[4^{\prime}, 3 ': 3,4\right]$ pyrido[1,2-a]benzimi-

## dazoles

The author as well explored the further functionalization of tetracycle $\mathbf{1 1 5}$ (Table 3.10). In the research of Ouyang et al., they demonstrated the functionalization of cyclized benzimidazole ${ }^{94 a}$ by transforming it to halogenated product, which could be suitable for Pd cross-coupling reactions (Table 3.10).

Table 3.10 Study of halogenation reaction conditions



Entry Solvent Additives Time, h Temperature, Yield, \% Product

| 1 | DMF | $\mathrm{I}_{2}$, CuI, | 48 | 120 |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | DMF | $\mathrm{I}_{2}$, KOH | 24 | rt | 0 | $\mathbf{1 1 9 a}$ |
| 3 | DMF | NIS | 48 | rt |  |  |
| 4 | DMF | NBS | 48 | rt | 66 | $\mathbf{1 1 9 b}$ |

It was decided to examine this protocol to the benzimidazole systems. Firstly, precursor $\mathbf{1 1 5}$ was examined in iodination reaction by varying different reagents and temperatures; unfortunately, no successful result was achieved (Table 3.10, Entry 13). Later, the treatment of pyrazole $\mathbf{1 1 5}$ with NBS in DMF at room temperature gave the bromo derivative 119b (Table 3.10, Entry 4) ${ }^{76 \mathrm{~b}}$.


Figure 3.10 ${ }^{1} \mathrm{H}$ NMR spectra of compounds 115 and 119b

The formation of halogenated compound was confirmed by MS where two peaks $\left(\mathrm{M}^{+}\right.$and $\left.[\mathrm{M}+2]^{+}\right)$were observed. The attachment of the bromine atom at position 8 was fully shown by the careful NMR spectroscopic analysis (Figure 3.10).

Figure 3.10 shows ${ }^{1} \mathrm{H}$ NMR spectra of starting compound 115 and brominated product $\mathbf{1 1 9 b}$. The absence of $8-\mathrm{H}$ proton signal at 6.94 ppm in the spectrum of compound 119b, which was in the above shown spectra of compound 115, proves that hydrogen atom was replaced by the bromine. There are signals of $\mathrm{CH}_{3}$ group protons in both compounds spectra at 3.06 ppm (compound 115) and 3.04 ppm (compound 119b). There are signals of $7-\mathrm{H}$ and $4-\mathrm{H}$ protons at 6.36 ppm and 6.93 ppm (compound 115) and 6.41 ppm and 6.95 ppm (compound 119b), respectively. In the 'aromatic' part, there are signals of $9-\mathrm{H}$ proton and protons of monosubstituted phenyl rings linked to pyrazolo and C-5 atom at 7.40-7.70 ppm. Moreover, there is a signal of 10H proton at 7.96 ppm (compound 115) and 7.78 ppm (compound 119b).

The ability of the bromine atom in $\mathbf{1 1 9 b}$ to participate in palladium-catalyzed cross-coupling reactions was proven by the Suzuki-type reaction of $\mathbf{1 1 9 b}$ with phenylboronic acid, which afforded compound $\mathbf{1 2 0}$ in $57 \%$ yield. This coupling was carried out at $100{ }^{\circ} \mathrm{C}$ under microwave irradiation in EtOH by using $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ as the catalyst and aq $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ as a base (Scheme 3.14). Microwave-assisted Sonogashira reaction conditions $\left(\mathrm{Pd}_{( }\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}, \mathrm{CuI}\right.$, and triethylamine at $130^{\circ} \mathrm{C}$ and 100 W for 10 min ) were applied to the cross-coupling of $\mathbf{1 1 9 b}$ with phenylacetylene. The reaction proceeded smoothly to afford compound 121 in $59 \%$ yield (Scheme 3.14$)^{76 \mathrm{~b}}$.


Reagents and conditions: (i) NBS, DMF, rt 24 h; (ii) 'Suzuki' $\mathrm{PhB}(\mathrm{OH})_{2}, \mathrm{Cs}_{2} \mathrm{CO}_{3}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, $\mathrm{EtOH}, \mathrm{MW}, 10{ }^{\circ} \mathrm{C}, 50 \mathrm{~W}, 10 \mathrm{~min}$ (for 120); (iii) 'Sonogashira' phenylacetylene, $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, TEA, CuI, DMF, MW, $130^{\circ} \mathrm{C}, 100 \mathrm{~W}, 10 \mathrm{~min}$ (for 121).

Scheme 3.14 Synthesis and cross-coupling reactions of compound 119b
To sum up, the ability to functionalize pyrazolo pyrido benzimidazoles by employing halogenation and various cross coupling reactions was investigated.

### 3.4.3. Construction of the $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridines

In continuation of this study, it was decided to evaluate 2,3-diaminopyridine activity for tandem cyclisation reaction with pyrazole-4-carbdehydes 113-114. When 2,3-diaminopyridine was used as the diamine component in the reaction with precursors 113-114, the reaction provided the $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridine derivatives 122-123 (Scheme 3.15) ${ }^{76 \mathrm{~b}}$.

In principle, the formation of two isomeric structures occurred in this reaction due to the different position of the pyridine nitrogen in the ring system plane. However, only regioisomers $\mathbf{1 2 2} \mathbf{- 1 2 3}$ were isolated from the complex reaction mixture with low yields of $35 \%$ and $40 \%$, respectively (Scheme 3.15) ${ }^{76 \mathrm{~b}}$. In order to increase the yield of this reaction, copper iodide was employed as a catalyst; however, the desired result was not achieved.


Scheme 3.15 Construction of the $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridine ring system

### 3.4.4. Construction of the $13,13 a-d i h y d r o-3 H$-pyrazolo[4',3':3,4]pyri-do[1,2-a]perimidine ring system

It is known that pyrido[1,2-a]pyrimidines play a significant role in medicinal chemistry due to their numerous pharmacological activities such as antidepressant, gastrointestinal protective, neurotropic, anticancer activities, and stress protecting. This scaffold is found in some of marketed drugs, for instance: pemirolast, pirenperone, and barmastine (Figure 3.11) ${ }^{99}$. Due to the above mentioned reasons, it was decided to synthesize 13,13a-dihydro-3H-pyrazo-lo[4',3':3,4]pyrido[1,2-a]perimidine ring system by using naphtalendiamine.

Pirenperone

Barmastine

Figure 3.11 Pyrido[1,2- $a$ ]pyrimidine scaffold containing drugs
The reaction of 113-114 with 1,8-naphthalenediamine under the same reaction conditions as described above (heating in DMF) yielded no product of 124-125 type. A similar phenomenon has been observed by Nagarayan and coworkers ${ }^{100}$ when they explored the tandem cyclisation of 2-(alkynyl)carbazole-3-carbaldehydes with different diamines. Thus, the author of this dissertation tried to adapt the copper-catalyzed annulation reaction conditions that were recently demonstrated by Tokimizu et al. ${ }^{101}$ in the synthesis of highly fused perimidines. Nevertheless, the heating of pyrazole 113 with 1.8 equiv of 1,8 -naphthalenediamine in DMF at $120^{\circ} \mathrm{C}$ in the presence of 10 mol \% of CuI under conventional conditions gave a complex reaction mixture with no obvious formation of the desired product 124. The author of this dissertation was
pleased to find that microwave-assisted heating $\left(140^{\circ} \mathrm{C}, 150 \mathrm{~W}, 40 \mathrm{~min}\right)$ provided the pyrazolo[4',3':3,4]pyrido[1,2-a]perimidine compound 124 in $45 \%$ yield (Scheme 3.16).


Scheme 3.16 Construction of the 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyrido[1,2-a]perimidines

When the same reaction conditions were applied with precursor 114 , the accordant congener $\mathbf{1 2 5}$ was formed in $44 \%$ isolated yield (Scheme 3.16). Next, it was chosen to employ copper-catalyzed annulation proposed by Tokimizu et al. ${ }^{101}$, though pyrazole 113 was conventionally heated in DMF with 1,8-naphtalendiamine in the presence of $10 \mathrm{~mol} \%$ of copper iodide; nevertheless, no desired result was achieved. Later, it was presumed that microwave-assisted heating could accomplish this reaction ${ }^{76 \mathrm{~b}}$ (Scheme 3.16).

### 3.4.5. Construction of the $\mathbf{2 H}$-pyrazolo $\left[4^{\prime}, 3 \mathbf{3}^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimida-

 zolesConsidering the aforementioned results, the reaction between carbaldehydes 39, 42, and benzene-1,2-diamines was as well explored (Scheme 3.17). Again, the reaction of both precursors $\mathbf{3 9}, \mathbf{4 2}$ with benzene-1,2-diamine provided the corresponding $2 H$-pyrazolo[4',3':3,4]pyrido[1,2-a] compounds $\mathbf{1 2 6} \mathbf{- 1 2 7}$ in acceptable yields (Scheme 3.17) ${ }^{76 \mathrm{~b}}$.


Scheme 3.17 Construction of the $2 H$-pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido[1,2- $a$ ] compounds 126-127

When precursors 39, 42 reacted with 4-methylbenzene-1,2-diamine, the reaction gave regioisomers (128a, 129a from 39 and 128c, 129c from 42) as inseparable mixtures (Scheme 3.18). Compounds 128a and 129a were formed in the ratio of 1:0.62, whereas their congeners 128c and 129c were formed in the ratio of 1:0.25 (Scheme 3.18). Additionally, the formation of regioisomeric products was observed in the course of the reaction between pyrazole 39 and 4-chlorobenzene-1,2-diamine (Scheme 3.18). In this case, the author of the dissertation was lucky to separate and fully characterize each of the isomers. 9-Chloro-2,5-diphenyl-2H-pyra-zolo[4',3':3,4]pyrido[1,2-a]benzimidazole (128b) was obtained in $40 \%$ yield, whereas its 8 -substituted analogue 129b was formed in $30 \%$ yield (Scheme 3.18) ${ }^{76 \mathrm{~b}}$. In the case of the reaction of pyrazole 42 with 4 -chlorobenzene-1,2-diamine, the regioisomer 128d was isolated as a sole product in $57 \%$ yield (Scheme 3.18). Notably, the 3-alkynylpyrazoles $\mathbf{3 9}, 42$ required longer heating in DMF ( 48 h ) than the 5-alkynylpyrazoles 113-114 in the tandem cyclisation reaction with benzene-1,2-diamines ${ }^{76 \mathrm{~b}}$.


Scheme 3.18 Construction of the $2 H$-pyrazolo[ $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a$ ] compounds 128-129
In conclusion, there has been developed a synthetic approach for tandem cyclisation via pyrazole compounds, bearing alkynyl and aldehyde groups in neighboring positions. It was demonstrated that the heating of 5-alkynyl- or 3-alkynylpyrazole-4- carbaldehydes with the appropriate $o$-phenylendiamines without any catalyst in DMF provided 2 H - and $3 H$-pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazoles. It should be noted that 3-alkynylpyrazoles 39 and $\mathbf{4 2}$ cyclisation was performed for longer period ( 48 h ) than 5-alkynylpyrazoles $\mathbf{1 1 3 - 1 1 4}$ cyclisation ( 24 h ) with benzene-1,2-diamines. Further, the treating of 5-alkynylpyrazole-4-carbaldehyde with pyridine-2,3diamine led to obtain 1-methyl-3-phenyl-3H-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridines, though the conducting of reaction with naphthalene-1,8-diamine provided 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyrido[1,2-a]perimidines. Notwithstanding, promoted by copper iodide under microwave heating, the latter cyclisation proceeded ${ }^{76 \mathrm{~b}}$.

### 3.4.6. Single-crystal X-ray diffraction analysis

Suitable crystals of $\mathbf{1 1 8 f}$ and $\mathbf{1 2 7}$ for X-ray diffraction analysis were obtained from acetonitrile; the molecular structure and crystallographic numbering is shown in Figure 3.12. The selected geometric parameters are given in the Tables 3.11 and $3.12^{76 \mathrm{~b}}$.

Molecule 118 f consists of almost planar 3H-pyrazolo[4',3':3,4]pyrido[1,2$a$ ]benzimidazole moiety with attached phenyl ring, which is turned to the plane of the heterocyclic unit for c.a. $20^{\circ}$ (Figure 3.12, Table 3.11). The side chain butyl carbons $C(17), C(18)$, and $C(19)$ are in the plane of the heterocyclic unit, and only $C(20)$ is out of this plane ${ }^{76 \mathrm{~b}}$.
Table 3.11 Selected geometric parameters of $\mathbf{1 1 8 f}\left(\AA^{\circ},^{\circ}\right)$

| $\mathrm{N} 1-\mathrm{C} 12$ | $1.322(27)$ | N7-N8 | $1.379(23)$ |
| :---: | :---: | :---: | :---: |
| $\mathrm{N} 1-\mathrm{C} 2$ | $1.380(28)$ | $\mathrm{N} 8-\mathrm{C} 9$ | $1.323(28)$ |
| $\mathrm{N} 4-\mathrm{C} 12$ | $1.407(25)$ | $\mathrm{N} 7-\mathrm{C} 11$ | $1.364(26)$ |
| $\mathrm{N} 4-\mathrm{C} 3$ | $1.407(27)$ | $\mathrm{N} 7-\mathrm{C} 21$ | $1.426(28)$ |
| N1-C12-N4 | $113.80(18)$ | $\mathrm{C} 9-\mathrm{N} 8-\mathrm{N} 7$ | $106.63(16)$ |
| $\mathrm{C} 3-\mathrm{N} 4-\mathrm{C} 12-\mathrm{N} 1$ | $0.44(22)$ | $\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 11-\mathrm{C} 6$ | $-179.02(19)$ |
| $\mathrm{C} 18-\mathrm{C} 17-\mathrm{C} 5-\mathrm{C} 6$ | $-2.69(29)$ | $\mathrm{C} 11-\mathrm{N} 7-\mathrm{C} 21-\mathrm{C} 22$ | $-21.27(0.33)$ |

The bond lengths and bond angles within the imidazole ring of $\mathbf{1 1 8 f}$ are similar to those of pyrido[1,2-a]benzimidazole derivatives, while bond lengths and bond angles in the pyrazole ring are comparable with the values found by the analysis of 1phenylpyrazole derivatives (Table 3.11) ${ }^{76 \mathrm{~b}}$.

The $2 H$-pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazole unit of $\mathbf{1 2 7}$ is planar and is as well almost co-planar with the phenyl ring (Table 3.12).

Table 3.12 Selected geometric parameters of $\mathbf{1 2 7}\left(\AA^{\circ},{ }^{\circ}\right)$

| N1-C12 | $1.316(17)$ | N7-N8 | $1.371(16)$ |
| :---: | :---: | :---: | :---: |
| N1-C2 | $1.386(19)$ | N7-C11 | $1.336(19)$ |
| N4-C12 | $1.406(18)$ | N8-C9 | $1.357(18)$ |
| N4-C3 | $1.408(18)$ | N8-C21 | $1.428(19)$ |
| N1-C12-N4 | $113.84(13)$ | C9-N8-N7 | $112.90(12)$ |
| C3-N4-C12-C10 | $-179.61(12)$ | N7-C11-C10-C12 | $-179.98(13)$ |
| C6-C5-C17-C18 | $2.33(20)$ | C9-N8-C21-C22 | $176.98(14)$ |

Moreover, all carbon atoms of the side butyl chain are located in the same plane as the phenyl ring and heterocyclic ring system atoms. The bond lengths within the planar molecule of $\mathbf{1 2 7}$ are similar to those of $\mathbf{1 2 7}$ (Figure 3.12, Table 3.12) ${ }^{76 \mathrm{~b}}$.



Figure 3.12 ORTEP drawing of $3 H$-pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazole $\mathbf{1 1 8 f}$ and ORTEP drawing of $2 H$-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazole 127

### 3.4.7. Optical investigations

The electronic absorption spectra of the selected 3 H pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ benzimidazoles (115-116, 117a-c, 119b, and $\mathbf{1 2 0}$ 121), $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b’]dipyridines (122-123), 2 H pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles (126-127 and 128b, d), and 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyrido[1,2-a]perimidine (124) were recorded in THF (Table 3.13).

Table 3.13 Absorption ( $\lambda_{\mathrm{abs}}$ and $\varepsilon$ ) and fluorescence ( $\lambda_{\mathrm{em}}$ and quantum yield, $\Phi_{\mathrm{f}}$ ) parameters for compounds 115-116, 117a-c, 119b, 120-124, 126-127, and 128b, d in THF*

| Entry | Compound | $\lambda_{\text {abs }}(\mathrm{nm})$ | $\underset{\substack{\varepsilon \times 10^{3},\left(\mathbf{d m}^{3} \mathbf{m o l}^{-1} \mathbf{c m}^{-} \\{ }^{1}\right)}}{ }$ | $\begin{aligned} & \lambda_{\text {em }}{ }^{*} \\ & (\mathbf{n m}) \end{aligned}$ | Stokes shift (nm) | $\Phi_{f}(\%)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 115 | $\begin{array}{r} 209 \\ 283 \\ \hline \end{array}$ | $\begin{aligned} & 31.21 \\ & 43.63 \\ & \hline \end{aligned}$ | 458 | 175 | 45 |
| 2 | 116 | $\begin{aligned} & 208 \\ & 273 \\ & 281 \end{aligned}$ | $\begin{aligned} & 28.86 \\ & 57.23 \\ & 61.30 \end{aligned}$ | 397 | 116 | 13 |
| 3 | 117a | $\begin{aligned} & 208 \\ & 285 \end{aligned}$ | $\begin{aligned} & 24.03 \\ & 34,84 \end{aligned}$ | 464 | 179 | 49 |
| 4 | 117b | $\begin{aligned} & 208 \\ & 286 \\ & \hline \end{aligned}$ | $\begin{aligned} & 27.81 \\ & 38.08 \\ & \hline \end{aligned}$ | 446 | 160 | 27 |
| 5 | 117c | $\begin{aligned} & 209 \\ & 283 \\ & 337 \end{aligned}$ | $\begin{aligned} & 31.48 \\ & 42.19 \\ & 13.41 \end{aligned}$ | 488 | 151 | 0.1 |
| 6 | 119b | $\begin{aligned} & 208 \\ & 278 \\ & 286 \\ & \hline \end{aligned}$ | $\begin{aligned} & 14.59 \\ & 15.62 \\ & 16.33 \end{aligned}$ | 456 | 170 | 10 |
| 7 | 120 | $\begin{aligned} & 200 \\ & 209 \\ & 295 \\ & \hline \end{aligned}$ | $\begin{aligned} & 27.50 \\ & 52.17 \\ & 72.97 \end{aligned}$ | 478 | 183 | 28 |
| 8 | 121 | $\begin{aligned} & 209 \\ & 302 \\ & \hline \end{aligned}$ | $\begin{array}{r} 47.19 \\ 59.22 \\ \hline \end{array}$ | 473 | 171 | 28 |


| Entry | Compound | $\lambda_{\text {abs }}(\mathrm{nm})$ | $\begin{gathered} \varepsilon^{\times \times 10^{3}}, \\ \left(\mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-}\right. \\ 1) \end{gathered}$ | $\begin{aligned} & \lambda_{\text {em* }} \\ & (\mathbf{n m}) \end{aligned}$ | Stokes shift (nm) | $\Phi_{f}(\%)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 9 | 122 | $\begin{aligned} & 209 \\ & 287 \end{aligned}$ | $\begin{aligned} & 32.04 \\ & 43.46 \end{aligned}$ | 438 | 151 | 27 |
| 10 | 123 | $\begin{aligned} & 209 \\ & 284 \\ & 311 \\ & \hline \end{aligned}$ | $\begin{aligned} & 20.01 \\ & 42.30 \\ & 11.05 \end{aligned}$ | 380 | 96 | 25 |
| 11 | 124 | $\begin{aligned} & 209 \\ & 380 \\ & \hline \end{aligned}$ | $\begin{aligned} & 57.52 \\ & 19.51 \\ & \hline \end{aligned}$ | 456 | 76 | 0.1 |
| 12 | 126 | $\begin{aligned} & 200 \\ & 242 \\ & 339 \end{aligned}$ | $\begin{aligned} & 28.62 \\ & 36.37 \\ & 20.43 \end{aligned}$ | 389 | 50 | 38 |
| 13 | 127 | $\begin{aligned} & 241 \\ & 281 \\ & 339 \\ & \hline \end{aligned}$ | $\begin{aligned} & 41.59 \\ & 35.78 \\ & 23.27 \end{aligned}$ | 386 | 47 | 58 |
| 14 | 128b | $\begin{aligned} & 245 \\ & 289 \\ & 336 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 30.65 \\ & 32.20 \\ & 18.26 \\ & \hline \end{aligned}$ | 385 | 49 | 45 |
| 15 | 128d | $\begin{aligned} & 244 \\ & 284 \\ & 337 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 45.80 \\ & 41.02 \\ & 27.40 \\ & \hline \end{aligned}$ | 382 | 45 | 51 |

* $\lambda_{\text {ex }}=310 \mathrm{~nm}$.

No significant differences were observed between the main absorption bands of 3H-pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a] benzimidazoles (Table 3.13, Entries 1-6 and 9$11)^{76 \mathrm{~b}}$. The $3 H$-pyrazolo $[4,3-c]$ imidazo[1,2-a:5,4- $\left.b^{\prime}\right]$ dipyridines (Table 3.13), except the nitro-compound 117c, showed similar results. Pyrazole 117c exhibited a significant bathochromic shift of the near ultraviolet band compared to the rest compounds and showed an absorption maximum at $337 \mathrm{~nm}^{76 b}$.

The electronic spectra of the $2 H$-pyrazolo[4',3':3,4]pyrido[1,2a] benzimidazoles (Table 3.13, Entries 14-17) possessed a near ultraviolet band with an absorption maximum at approximately 340 nm in all cases, whereas the substituents attached at various positions of the heterocyclic ring system had a negligible influence on the position of the absorption bands. The fluorescence emission profile of the aforementioned fused heterocycles was obtained in THF (Table 3.13) and revealed a strong influence of molecular structure on the character of the fluorescence emission. The solutions of those $3 H$-pyrazolo[4',3':3,4]pyrido[1,2a]benzimidazoles, which possessed 5-phenyl substitutent, exhibited a strong fluorescence with an emission maximum ( $\lambda_{\mathrm{em}}$ ) in the range from 446 (117b, Entry 4) to $488 \mathrm{~nm}(\mathbf{1 1 7 c}$, Entry 5) and large Stokes shifts of c.a. 170 nm . However, 5-butyl substituted compound $\mathbf{1 1 6}$ showed a significant hypsochromic shift (c.a. 60 nm ) to compare with 5-phenyl one 115 (Entries 2 and 3, respectively). A similar feature was found in the case of the $3 H$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridines $\mathbf{1 2 2}(\mathrm{R}$ $=\mathrm{Ph})$ and $\mathbf{1 2 3}(\mathrm{R}=n \mathrm{Bu})$, which spectra contained the emission maximum $\left(\chi_{\mathrm{em}}\right)$ at 438 and 380 nm , respectively. The spectrum of the brominated compound $\mathbf{1 1 9 b}$ revealed $\chi_{\text {em }}$ at 456 nm (Entry 8), similar to that of the starting substrate 115 (Entry 1), but the
further substitution of bromine by a phenyl ring (compound 120) or a 2-phenylethenyl moiety (compound 121) initiated a significant bathochromic shift to $\chi_{\mathrm{em}}$ at 478 and 473 nm , respectively ${ }^{76 \mathrm{~b}}$.

For all the $2 H$-pyrazolo[ $\left.4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles (126-127, 128b, d), the fluorescence spectra displayed the emission maximum ( $\chi_{\text {em }}$ ) ranging from 382 to 385 nm and Stokes shifts of c.a. 50 nm . By contrast, the electronic and spatial properties of the substituents had only a limited influence on the character of the fluorescence spectra. The solution that contain the 13,13a-dihydro-3Hpyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a]perimidine 124 exhibited $\chi_{\mathrm{em}}$ at $456 \mathrm{~nm}^{76 \mathrm{~b}}$.

The fluorescence quantum yield ( $\Phi f$ ) of the solutions was estimated by the integrating sphere method. It appeared that the fluorescence quantum yield was sensitive to the structure of the compounds. For the 5-phenyl-benzimidazole 115, the observed $\Phi_{\mathrm{f}}$ value was c.a. $45 \%$, whereas for its 5-butyl analogue $\mathbf{1 1 6}, \Phi_{\mathrm{f}}$ was fixed slightly above $13 \%$. It is known that the direct attachment of halogens, especially heavy ones, or a nitro group to the core of an aromatic compound can significantly reduce the fluorescence quantum yield $\left(\Phi_{\mathrm{f}}\right)^{96}$. Indeed, in the case of the 8 -bromobenzimidazole 119b, $\Phi_{\mathrm{f}}$ fell to $9.7 \%$, whereas for the 8 -nitro compound $117 \mathbf{c}, \Phi_{\mathrm{f}}$ dropped to a negligible value of $0.1 \%$. The 13,13 a-dihydro- 3 H pyrazolo $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a]$ perimidine (133) emitted only negligible fluorescence $(\Phi f=0.1 \%)$. The values of $\Phi_{\mathrm{f}}$ for the 5-phenyl- and 5-butyl-2Hpyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2-a]benzimidazoles (126-127) were $38 \%$ and $58 \%$, respectively. The attachment of a chlorine atom at the 9-position of the heterocyclic core did not significantly change the fluorescence quantum yield $\left(\Phi_{\mathrm{f}}\right)$, which remained at high values of $49 \%$ and $45 \%$ for compounds 128b and 128d, respectively ${ }^{76 b}$.

In conclusion, it was demonstrated that the 2 H - and 3 H pyrazolo[ $\left.4^{\prime}, 3 ': 3,4\right]$ pyrido[1,2- $a$ ]benzimidazole ring systems could be obtained without any catalyst by heating 5-alkynyl- or 3-alkynylpyrazole-4-carbaldehydes with the appropriate benzene-1,2-diamines in DMF. The reaction of 5-alkynylpyrazole-4carbaldehydes with pyridine-2,3-diamine leads to the production of 1-methyl-3-phenyl-3H-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b']dipyridines, whereas the use of naphthalene-1,8-diamine leads to 13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyrido[1,2a]perimidines. However, in the latter case, CuI catalysis and microwave heating is necessary. The pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazole fluorophores were characterized by their good quantum yields and large Stokes shifts. Therefore, the derivatives of these heterocycles can be considered for use as attractive, tailor-made emitters in the applications as energetic substances and markers ${ }^{76 \mathrm{~b}}$.

## 4. EXPERIMENTAL PART

### 4.1. Chemistry

### 4.1.1. Instrumentation

The melting points were determined on a Reichert-Kofler hot-stage microscope or in capillary tubes on capillary melting point apparatus Electrothermal MEL$\mathrm{TEMP}^{\circledR}$ and are uncorrected. Mass spectra were obtained on a Shimadzu LCMS 2020 Single Quadrupole Liquid Chromatograph Mass Spectrometer. IR spectra in KBr pellets were recorded on a Bruker Tensor 27 spectrometer and are reported in the frequency of absorption ( $\mathrm{cm}^{-1}$ ) or on a Bruker Vertex v70 FTIR spectrometer equipped with a diamond ATR accessory. HRMS spectra were recorded with a Bruker micrO-TOF-QIII spectrometer. ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, and ${ }^{15} \mathrm{~N}$ NMR spectra were recorded from $\mathrm{CDCl}_{3}$ solutions at $25^{\circ} \mathrm{C}$ on either a Bruker Avance III 400 instrument (400 MHz for ${ }^{1} \mathrm{H}, 100 \mathrm{MHz}$ for ${ }^{13} \mathrm{C}, 40 \mathrm{MHz}$ for ${ }^{15} \mathrm{~N}$ ) by using a directly detecting BBFO probe or on a Bruker Avance III 700 instrument ( 700 MHz for ${ }^{1} \mathrm{H}, 176 \mathrm{MHz}$ for ${ }^{13} \mathrm{C}$ ) equipped with a $5 \mathrm{~mm} \mathrm{TCI}{ }^{1} \mathrm{H}^{13} \mathrm{C}^{15} \mathrm{~N} / \mathrm{D}$ z-gradient cryoprobe. The solvent (residual) signals were used as internal standards and were related to TMS with $\delta 7.26 \mathrm{ppm}\left({ }^{1} \mathrm{H}\right)$ and $\delta 77.00 \mathrm{ppm}\left({ }^{13} \mathrm{C}\right) .{ }^{15} \mathrm{~N}$ NMR spectra were referenced against neat, external nitromethane. The full and unambiguous assignments of ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$, and ${ }^{15} \mathrm{~N}$ NMR resonances were achieved by using combined applications of standard NMR spectroscopic techniques such as APT, COSY, TOCSY, NOESY, gs-HSQC, and gs-HMBC. Diffraction data were collected on Bruker-Nonius KappaCCD diffractometer at room temperature and as well at $100{ }^{\circ} \mathrm{C}$. The crystal structures were solved by using known programs. The UV/vis spectra were recorded on a Perkin Elmer Lambda 35 UV/vis spectrometer. The fluorescence spectra were recorded on an FL920 fluorescence spectrometer from Edinburgh Instruments. The PL quantum yields were measured from dilute THF solutions by an absolute method by using the Edinburgh Instruments integrating sphere that was excited with a Xe lamp. Optical densities of the sample solutions were ensured to be below 0.1 to avoid reabsorption effects. All the optical measurements were performed at rt under ambient conditions. For chromatographic separations, silica gel 60 (230-400 mesh, Merck) was used.

### 4.1.2. Materials

All chemicals were purchased from Sigma-Aldrich and Fluorochem, and used as received without further purification. Organic solvents were purified and dried by the standard methods ${ }^{102}$.

### 4.1.2.1. General procedure for the preparation of 4-alkynyl-3-hydroxy-1-phenyl-1H-pyrazoles 4-13 by the Sonogashira-type cross-coupling reaction.

In to the solution of 4-iodo-1-phenyl-1H-pyrazol-3-ol (4) (1 mmol) in dry DMF $(2 \mathrm{~mL})$ under an argon atmosphere were added TEA $(0.7 \mathrm{~mL}, 5 \mathrm{mmol})$, the appropriate ethyne ( 1.5 mmol ), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(70 \mathrm{mg}, 0.1 \mathrm{mmol})$, and $\mathrm{CuI}(36 \mathrm{mg}, 0.2 \mathrm{mmol})$. The mixture was stirred under an argon atmosphere at $58{ }^{\circ} \mathrm{C}$ temperature for the given
time, and then it was diluted with water and extracted with ethyl acetate. The combined organic layers were washed with brine and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and the solvent was evaporated. The residue was purified by flash chromatography ( $\mathrm{SiO}_{2}$, eluent: ethyl acetate/ $n$-hexane, $1: 6, \mathrm{v} / \mathrm{v}$ ) to yield compounds 5-13.

### 4.1.2.1.1. 4-[(4-Methylphenyl)ethynyl]-1-phenyl-1H-pyrazol-3-ol (5)

The reaction mixture was stirred for 12 hours. White solid, yield $206 \mathrm{mg}, 75 \%, \mathrm{mp}$
 $168-169{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3125, 3069, 3029 ( $\left.\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right), 2923\left(\mathrm{CH}_{\text {aliph }}\right), 2217(\mathrm{C} \equiv \mathrm{C}), 1597$, 1532, 1504, 1314, 1208 (C=C, C-N, C-O), 815, 755, 688 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.36\left(\mathrm{CH}_{3}\right), 7.14-7.15(\mathrm{~m}, 2 \mathrm{H}$, CPh 3,5-H), 7.29-7.31 (m, 1H, NPh 4-H), 7.42-7.43 (m, $2 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}), 7.48-7.50(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.53-7.54$ (m, 2H, NPh 2,6-H), $7.87(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.74(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{OH})$. ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.6\left(\mathrm{CH}_{3}\right), 77.8$ $(C H \equiv \mathrm{CPh}), 92.6(\mathrm{C}-4), 92.8(\mathrm{C} \equiv \mathrm{CPh}), 119.2(\mathrm{NPh} \mathrm{C}-2,6)$, 120.4 (CPh C-1), 126.8 (NPh C-4), 129.2 (CPh C-3,5), 130.0 (NPh C-3,5), 131.1 (C5), 131.6 (CPh C-2,6), 138.3 (CPh C-4), 139.1 (NPh C-1), 163.8 (C-3). MS m/z (\%): $275\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 297.0998, found 297.0998.

### 4.1.2.1.2. 4-[(4-Ethylphenyl)ethynyl]-1-phenyl-1H-pyrazol-3-ol (6)



The reaction mixture was stirred for 3 hours. Yellowish solid, yield $225 \mathrm{mg}, 78 \%$, mp $164-165^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3060,3029\left(\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right)$, 2964, $2928\left(\mathrm{CH}_{\text {aliph }}\right), 2220$ $(\mathrm{C} \equiv \mathrm{C}), 1597,1540,1503,1414,1208(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O})$, 830, 751, $678(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 1.25(\mathrm{t}$, $\left.{ }^{3} J=7.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.66\left(\mathrm{q},{ }^{3} \mathrm{~J}=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.18-$ 7.18 (m, 2H, CPh 3,5-H), 7.29-7.31 (m, 1H, NPh 4-H), 7.45$7.50(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.53-7.54(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh}$ 2,6-H), $7.87(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.91(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \operatorname{ppm} 15.5\left(\mathrm{CH}_{3}\right), 28.9\left(\mathrm{CH}_{2}\right), 77.8(C \equiv \mathrm{CPh}), 92.6(\mathrm{C}-4), 92.8$ ( $\mathrm{C} \equiv \mathrm{CPh}$ ), 119.2 ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 120.7 (CPh C-1), 126.8 ( $\mathrm{NPh} \mathrm{C}-4$ ), 128.0 (CPh C-3,5), 130.0 (NPh C-3,5), 131.2 (C-5), 131.6 (CPh C-2,6), 139.0 (CPh C-4), 144.6 (NPh C1), 163.9 (C-3). MS m/z (\%): $289\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{ONa}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 311.1155, found 311.1155 .

### 4.1.2.1.3. 4-[(4-Fluorophenyl)ethynyl]-1-phenyl-1H-pyrazol-3-ol (7)

The reaction mixture was stirred for 12 hours. White solid, yield $203 \mathrm{mg}, 73 \%, \mathrm{mp}$ $197-198^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3067, $3052\left(\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right), 2218(\mathrm{C} \equiv \mathrm{C})$, 1597, 1536, 1501, 1212 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}$ ), 831, 754, $679(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 7.03\left(\mathrm{t},{ }^{3} J(4 \mathrm{FPh} 3,5-\mathrm{H}\right.$,

$2,6-\mathrm{H})=8.5 \mathrm{~Hz}, 2 \mathrm{H}, 4 \mathrm{FPh} 3,5-\mathrm{H}), 7.30-7.32$ (m, 1H, NPh 4H), 7.48-7.52 (m, 4H, NPh 3,5-H, 4FPh 2,6-H), 7.53-7.54 (m, 2H, NPh 2,6-H), 7.88 (s, 1H, 5-H), 11.11 (br s, 1H, OH). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}}$ ppm 79.2 ( $C \equiv \mathrm{C} 4 \mathrm{FPh}$ ), 91.6 ( $\mathrm{C} \equiv C 4 \mathrm{FPh}$ ), 92.2 (C-4), 115.7 ( ${ }^{2} J=22.1 \mathrm{~Hz}, 4 \mathrm{FPh} \mathrm{C}-3,5$ ), 119.2 (NPh C-2,6), 119.6 ( ${ }^{4} \mathrm{~J}=3.8 \mathrm{~Hz}, 4 \mathrm{FPh} \mathrm{C}-1$ ), 126.6 (NPh C-4), 130.0 ( $\mathrm{NPh} \mathrm{C}-3,5$ ), 131.1 (C-5), 133.5 ( $^{3} \mathrm{~J}=8.9 \mathrm{~Hz}, 4 \mathrm{FPh}$ $\mathrm{C}-2,6$ ), 139.1 ( $\mathrm{NPh} \mathrm{C}-1$ ), 162.6 ( ${ }^{1} J=248.5 \mathrm{~Hz}, 4 \mathrm{FPh} \mathrm{C}-4$ ), 163.6 (C-3). MS m/z (\%): $279\left([\mathrm{M}+\mathrm{H}]^{+}, 00\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{11} \mathrm{FN}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 301.0748, found 301.0748 .
4.1.2.1.4. 4-[(4-Etoxyphenyl)ethynyl]-1-phenyl-1H-pyrazol-3-ol (8)


The reaction mixture was stirred for 12 hours. White solid, yield $240 \mathrm{mg}, 79 \%, \mathrm{mp} 194-195^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3056\left(\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right), 2977,2923\left(\mathrm{CH}_{\text {aliph }}\right), 2134(\mathrm{C} \equiv \mathrm{C})$, 1597, 1502, 1473, 1245, 1173, 1043 (C=C, C-N, C-O), 824, 754, $678\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.42\left(\mathrm{t},{ }^{3} J\left(\mathrm{CH}_{3}, \mathrm{CH}_{2}\right)=7.0\right.$ $\left.\mathrm{Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.05\left(\mathrm{q},{ }^{3}{ }^{3}\left(\mathrm{CH}_{2}, \mathrm{CH}_{3}\right)=7.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.85-$ 6.86 (m, 2H, CPh 3,5-H), 7.28-7.30 (m, 1H, NPh 4-H), 7.457.46 (m, 2H, CPh 2,6-H), 7.47-7.49 (m, 2H, NPh 3,5-H), $7.53-7.55$ (m, 2H, NPh 2,6-H), 7.86 (s, 1H, 5-H), 10.74 (br s, $1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.9\left(\mathrm{CH}_{3}\right)$, $63.6\left(\mathrm{CH}_{2}\right)$, $76.8(\mathrm{C} \equiv \mathrm{CPh}), 92.6(\mathrm{C}-4), 92.7(\mathrm{C} \equiv \mathrm{CPh}), 114.6(\mathrm{CPh} \mathrm{C-3,5)}$,115.4 (CPh C-1), 119.0 (NPh C-2,6), 126.7 (NPh C-4), 129.9 (NPh C-3,5), 130.7 (C-5), 133.2 (CPh C-2,6), 139.2 (NPh C-1), 159.1 (CPh C-4), 163.5 (C-3). MS m/z (\%): 305 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 327.1104, found 327.1105 .

### 4.1.2.1.5. 1-Phenyl-4-[(thiophen-3-yl)ethynyl]-1H-pyrazol-3-ol (9)

The reaction mixture was stirred for 2 hours. Brown solid, yield $197 \mathrm{mg}, 74 \%$, mp
 204.5-205.5 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3127, 3109, 3069, 3059, 3046, 3004 ( $\mathrm{OH}, \mathrm{CH}_{\text {arom }}$ ), $2220(\mathrm{C} \equiv \mathrm{C}), 1595,1533$, 1503, 1397, 1305, 1254, 1206, 1060 (C=C, C-N, C-O), 751, 681 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 $\left.\mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right): \delta_{\mathrm{H}} \mathrm{ppm} 7.21\left(\mathrm{dd},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=5.0 \mathrm{~Hz},{ }^{4} J(4-\right.$ $\mathrm{H}, 2-\mathrm{H})=1.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Th} 4-\mathrm{H}), 7.23-7.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.44-$ 7.47 (m, 1H, Ph $3,5-\mathrm{H}), 7.62\left(\mathrm{dd},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=4.9 \mathrm{~Hz},{ }^{4} J(5-\right.$ $\mathrm{H}, 2-\mathrm{H})=2.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Th} 5-\mathrm{H}), 7.71-7.72(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H})$, $7.79\left(\mathrm{dd},{ }^{4} J(2-\mathrm{H}, 5-\mathrm{H})=3.0 \mathrm{~Hz},{ }^{4} J(2-\mathrm{H}, 4-\mathrm{H})=1.2 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Th $2-$ H), $8.62(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.07(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz, DMSO- $\mathrm{d}_{6}$ ): $\delta_{\mathrm{C}} \mathrm{ppm}$ 79.6 ( $C=\mathrm{CTh}$ ), $87.0(\mathrm{C} \equiv C T h), 91.3$ (C-4), 117.2 (Ph C-2,6), 121.8 (Th C-3), 125.8 (Ph C-4), 126.8 (Th C-5), 128.9 (Th C-2), 129.5 (Ph C-3,5, Th C-4), 130.7 (C-5),
139.1 (Ph C-1), 162.7 (C-3). MS m/z (\%): 267 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{OS}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 267.0587, found 267.0584.

### 4.1.2.1.6. 4-(Hex-1-yn-1-yl)-1-phenyl-1H-pyrazol-3-ol (10)

The reaction mixture was stirred for 24 hours. Yellow solid, yield $204 \mathrm{mg}, 85 \%, \mathrm{mp}$
 $140-141.5^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3120, 3070, 3049, 2958, 2931 ( $\mathrm{OH}, \mathrm{CH}_{\text {arom }}, \mathrm{CH}_{\text {aliph }}$ ), 1597, 1587 1526, 1504, 1415, 1307, 1234, 1210, 1062 (C=C, C-N, C-O), 756, 678 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}}$ ppm $0.94\left(\mathrm{t},{ }^{3} J=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right.$ ), 1.47 ( sext, ${ }^{3} J=7.4 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.59 (quin, ${ }^{3} \mathrm{~J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), $2.42\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 7.26-7.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, $7.44-7.46(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.48-7.50(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.74(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.65$ (s, 1H, OH). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 19.5\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $31.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $69.1\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, $92.9(\mathrm{C}-4), 93.8$ ( $\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}$ ), 119.0 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 126.5 ( $\mathrm{Ph} \mathrm{C}-1$ ), 129.9 ( $\mathrm{Ph} \mathrm{C-3,5)}$,130.9 (C-5), 139.2 (Ph C-4), $163.9(\mathrm{C}-3) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 241\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 241.1335, found 241.1332.

### 4.1.2.1.7.4-(Hept-1-yn-1-yl)-1-phenyl-1H-pyrazol-3-ol (11)

The reaction mixture was stirred for 12 hours. Brown solid, yield $198 \mathrm{mg}, 78 \%$, mp
 $110-111{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3131, $3057(\mathrm{OH}$, $\left.\mathrm{CH}_{\text {arom }}\right), 2947,2922\left(\mathrm{CH}_{\text {aliph }}\right), 2163(\mathrm{C} \equiv \mathrm{C}), 1595,1529,1499$ ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}$ ), 761, 691 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.91\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.3 \mathrm{~Hz}\right.$, $3 \mathrm{H}, \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}$ ), 1.35 ( sext, ${ }^{3} \mathrm{~J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.42 (quin, ${ }^{3} J=7.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 1.61 (quin, ${ }^{3} \mathrm{~J}=7.3 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), $2.42\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right), 7.26-$ 7.28 (m, 1H, Ph 4-H), 7.44-7.46 (m, 2H, Ph 3,5-H), 7.49-7.50 (m, 2H, Ph 2,6-H), $7.74(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.41$ (br s, $1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.2$ $\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right)$, $19.8\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right)$, $22.4\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $28.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 31.3$ $\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 69.1\left(\mathrm{C} \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right), 92.8(\mathrm{C}-4), 93.9\left(\mathrm{C} \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right), 119.0(\mathrm{Ph} \mathrm{C}-2,6)$, 126.5 ( $\mathrm{Ph} \mathrm{C}-4$ ), 129.9 ( $\mathrm{Ph} \mathrm{C-3,5)}$,130.9 (C-5), 139.2 ( $\mathrm{Ph} \mathrm{C}-4$ ), 163.8 (C-3). MS m/z (\%): $255\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 277.1311, found 277.1310 .
4.1.2.1.8. 4-(Cyclopropylethynyl)-1-phenyl-1H-pyrazol-3-ol (12)


The reaction mixture was stirred for 2 hours. White solid, yield $143 \mathrm{mg}, 64 \%, \mathrm{mp} 185.5-187^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3122, 3074, $3051\left(\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right) 2999,2954,2920\left(\mathrm{CH}_{\text {aliph }}\right), 1596$, 1526, 1504, 1420, 1308, 1242, 1210, 1063(C=C, C-N, C-O), 758, $678\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $(700 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.79-0.81\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 0.82-0.86(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2}\right), 1.45-1.49\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 7.26-7.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.44-7.48 (m, 4H, Ph 2,3,5,6-H), $7.72(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.45(\mathrm{~s}, 1 \mathrm{H}$,
$\mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 0.6\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 8.8\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 63.8$ $\left(C \equiv \mathrm{CC}_{3} \mathrm{H}_{5}\right), 92.2(\mathrm{C}-4), 96.2\left(\mathrm{C} \equiv \mathrm{CC}_{3} \mathrm{H}_{5}\right), 118.5(\mathrm{Ph} \mathrm{C-2,6)}, 126.0(\mathrm{Ph} \mathrm{C}-4), 129.4(\mathrm{Ph}$ $\mathrm{C}-3,5), 130.5$ (C-5), 138.6 (Ph C-1), 163.4 (C-3). MS m/z (\%): 225 ([M+H] ${ }^{+}, 100$ ). HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 225.1022, found 225.1021.

### 4.1.2.1.9. 1-Phenyl-4-((trimethylsilyl)ethynyl)-1H-pyrazol-3-ol (13)

The reaction mixture was stirred for 24 hours. White solid, 64 mg , yield $50 \%$, mp

$175-176{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3073, $3059(\mathrm{OH}$, $\left.\mathrm{CH}_{\text {arom }}\right), 2956,2926\left(\mathrm{CH}_{\text {aliph }}\right), 2155(\mathrm{C} \equiv \mathrm{C}), 1600,1586,1530$, 1507, 1247, 1208, 1201, $1064\left(\mathrm{CH}_{3}-\mathrm{Si}, \mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}\right), 868$, 841, 812, 758, $692\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.24$ (s, 9H, TMS), 7.26-7.30 (m, 1H, Ph 4-H), 7.44-7.47 (m, 2H, Ph 3,5-H), 7.49-7.51 (m, $2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.83(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 0.4$ (TMS) 92.5, 94.1 ( $C \equiv$ CTMS, C-4), 98.3 ( $\mathrm{C} \equiv C \mathrm{TMS}$ ), 119.3 (Ph C-2,6), 127.0 (Ph C-4), 130.1 ( $\mathrm{Ph} \mathrm{C-3,5)}$,131.8 (C-5), 139.2 (Ph C-1), 164.0 (C-3). MS m/z (\%): $257\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{OSiNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 279.0951, found 279.0951.

### 4.1.2.2. General procedure for the cyclization of 4-alkynyl-3-hydroxy-1-phenyl-1H-pyrazoles 4-13.

$\operatorname{AgOTf}(13 \mathrm{mg}, 0.05 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(110 \mathrm{mg}, 1 \mathrm{mmol})$ were added into the solution of appropriate pyrazoles $\mathbf{4 - 1 3}(0.5 \mathrm{mmol})$ in absolute DMF $(1 \mathrm{ml})$. The mixture was stirred at $120^{\circ} \mathrm{C}$ temperature for 14 hours, diluted with water, and the extraction was done with ethyl acetate. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$; the solvent was evaporated. The residue was purified by flash chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $\left.1: 6, \mathrm{v} / \mathrm{v}\right)$ to yield compounds 15-23.

### 4.1.2.2.1. 2,5-Diphenyl-2H-furo[2,3-c]pyrazole (15)



Yellowish solid, yield $112 \mathrm{mg}, 86 \%$, mp $193.5-194.3^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\max }, \mathrm{cm}^{-1}\right): 3139,3062,3045\left(\mathrm{CH}_{\text {arom }}\right), 1622,1596,1587,1511,1481,1454$, $1434,1388,1315,1271,1198,1154,1073,1035,1004$ (C=C, C-N, C-OC), 748, 722, $685\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR (700 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.67(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.19-7.22(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.25-$ 7.28 (m, 1H, CPh 4-H), 7.35-7.37 (m, 2H, CPh 3,5-H), 7.38-7.40 (m, 2H, NPh 3,5-H), 7.65-7.66 (m, 2H, NPh 2,6-H), 7.69 (s, 1H, 3-H), 7.71-7.72 $(\mathrm{m}, 2 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 96.5(\mathrm{C}-4), 113.8$ (C-3a), 116.3 (C-3), 119.2 (NPh C-2,6), 124.6 (CPh C-2,6), 126.3 (NPh C-4), 128.6 (CPh C-4), 128.9 (CPh C-3,5), 129.6 (NPh C-3,5), 130.7 (CPh C-1), 141.1 (NPh C1), 159.3 (C-5), 167.0 (C-6a). MS m/z (\%): 261 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{NaO}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 283.0842, found 283.0842.

### 4.1.2.2.2. 5-(4-Methylphenyl)-2-phenyl-2H-furo[2,3-c]pyrazole (16)

White solid, yield $119 \mathrm{mg}, 87 \%, \mathrm{mp} 211-212^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\mathrm{v}_{\text {max }}, \mathrm{cm}^{-1}$ ): 3045,
 3020 ( $\mathrm{CH}_{\text {arom }}$ ), 2956, $2917\left(\mathrm{CH}_{\text {aliph }}\right), 1694,1597,1500,1441,1388$ (C=C, $\mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 796,746,684(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.39\left(\mathrm{CH}_{3}\right), 6.68(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H})$, 7.23-7.25 (m, 2H, 5-Ph 3,5-H), 7.26-7.28 (m, 1H, NPh 4-H), 7.45-7.47 (m, 2H, NPh 3,5-H), 7.67-7.69 (m, 2H, 5-Ph 2,6-H), 7.71-7.73 (m, 2H, NPh $2,6-\mathrm{H}), 7.74(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.5\left(\mathrm{CH}_{3}\right)$, 95.6 (C-4), 113.8 (C-3a), 116.1 (C-3), 119.1 (NPh C-2,6), 124.6 (5-Ph C2,6), 126.2 (NPh C-4), 128.0 ( $5-\mathrm{Ph} \mathrm{C}-1$ ), 129.6 ( $\mathrm{NPh} \mathrm{C}-3,5$ ), 129.6 ( $5-\mathrm{Ph} \mathrm{C}-$ 3,5), 138.7 (5-Ph C-4), 141.1 (NPh C-1), 159.6 (C-5), 168.9 (C-6a). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-169.9(\mathrm{~N}-2),-127.9(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%):$ $275\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{NaO}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 297.0998, found 297.0999.

### 4.1.2.2.3. 5-(4-Ethylphenyl)-2-phenyl-2H-furo[2,3-c]pyrazole (17)



White solid, yield $133 \mathrm{mg}, 92 \%, \mathrm{mp} 180-181^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3060\left(\mathrm{CH}_{\text {arom }}\right), 2965,2921\left(\mathrm{CH}_{\text {aliph }}\right), 1618,1597,1501,1480,1447$, 1386 (C=C, C-N, C-O-C), 834, 753, 687 (CH=CH of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.25\left(\mathrm{t},{ }^{3} J=7.6 \mathrm{~Hz}\right.$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.67 (q, ${ }^{3} J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ), 6.66 ( $\left.\mathrm{s}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.24-7.26(\mathrm{~m}$, 3H, 5-Ph 3,5-H, NPh 4-H), 7.43-7.45 (m, 2H, NPh 3,5-H), 7.68-7.71 (m, $4 \mathrm{H}, 5-\mathrm{Ph} 2,6-\mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 7.72$ (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{c}} \mathrm{ppm} 15.6\left(\mathrm{CH}_{3}\right), 28.9\left(\mathrm{CH}_{2}\right), 95.6(\mathrm{C}-4), 113.8(\mathrm{C}-3 \mathrm{a}), 116.1(\mathrm{C}-$ 3), 119.1 ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 124.7 ( $5-\mathrm{Ph} \mathrm{C}-2,6$ ), 126.2 (NPh C-4), 128.2 ( $5-\mathrm{Ph} \mathrm{C}-$ 1), 128.4 ( $5-\mathrm{Ph} \mathrm{C}-3,5$ ), 129.6 ( $\mathrm{NPh} \mathrm{C}-3,5$ ), 141.1 ( $\mathrm{NPh} \mathrm{C}-1$ ), 145.1 ( $5-\mathrm{Ph} \mathrm{C-}$ 4), 159.6 (C-5), 168.9 (C-6a). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-170.0$ (N-2), $-128.0(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 289\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{ONa}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 311.1155, found 311.1156.

### 4.1.2.2.4. 5-(4-Fluorophenyl)-2-phenyl-2H-furo[2,3-c]pyrazole (18)



White solid, yield $120 \mathrm{mg}, 86 \%, \mathrm{mp} 220-221^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\mathrm{cm}^{-1}$ ): 3066, 3047 ( $\mathrm{CH}_{\text {arom }}$ ), 1596, 1498, 1447, 1385, 1223 (C=C, C-N, C-O-C), 843, 750, 687 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.67(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.13\left(\mathrm{t},{ }^{3} J(4 \mathrm{FPh} 3,5-\mathrm{H}\right.$, $2,6-\mathrm{H})=8.5 \mathrm{~Hz}, 2 \mathrm{H}, 4 \mathrm{FPh} 3,5-\mathrm{H}$ ), 7.27-7.29 (m, 1H, NPh 4-H), 7.45-7.48 (m, 2H, NPh 3,5-H), 7.71-7.73 (m, 2H, NPh 2,6-H), 7.75-7.77 (m, 3H, $4 \mathrm{FPh} 2,6-\mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 96.2$ (C-4), 113.7 (C-3a), 116.1 (C-3), 116.2 ( $4 \mathrm{FPh} \mathrm{C}-3,5,{ }^{2} J 61.1 \mathrm{~Hz}$ ), 119.2 (NPh C-2,6), 126.4 (NPh C-4), 126.5 ( $4 \mathrm{FPh} \mathrm{C}-2,6,{ }^{3} J=8.1 \mathrm{~Hz}$ ), 127.0 ( $4 \mathrm{FPh} \mathrm{C}-1,{ }^{4} J=2.7$ Hz ), 129.6 ( $\mathrm{NPh} \mathrm{C}-3,5$ ), 141.1 (NPh C-1), 158.3 (C-5), 162.9 ( $4 \mathrm{FPh} \mathrm{C}-4$, $\left.{ }^{1} J=248.6 \mathrm{~Hz}\right), 168.9(\mathrm{C}-6 \mathrm{a}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-169.4$ (N-2), -127.5 (N-1). MS m/z (\%): 279 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{11} \mathrm{FN}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 301.0748, found 301.0749.

### 4.1.2.2.5. 5-(4-Ethoxyphenyl)-2-phenyl-2H-furo[2,3-c]pyrazole (19)



White solid, yield $123 \mathrm{mg}, 81 \%, \mathrm{mp} 190-191^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3063$, $3046\left(\mathrm{CH}_{\text {arom }}\right)$, 2978, $2924\left(\mathrm{CH}_{\text {aliph }}\right), 1597,1501,1387,1249$, 1239 (C=C, C-N, C-O-C). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.44$ (t, $\left.{ }^{3} J=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.08\left(\mathrm{q},{ }^{3} \mathrm{~J}=7.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.58(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 6.94-$ 6.96 (m, 2H, 5-Ph 3,5-H), 7.25-7.27 (m, 1H, NPh 4-H), 7.44-7.47 (m, 2H, NPh 3,5-H), 7.69-7.72 (m, 4H, NPh 2,6-H, 5-Ph 2,6-H), 7.72 (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.9\left(\mathrm{CH}_{3}\right), 63.7\left(\mathrm{CH}_{2}\right), 94.5(\mathrm{C}-4)$, 113.9 (C-3a), 114.9 (5-Ph C-3,5), 115.8 (C-3), 119.1 (NPh C-2,6), 123.4 (5-Ph C-1), 126.09 (NPh C-4), 126.13 (5-Ph C-2,6), 129.6 (NPh C-3,5), 141.2 (NPh C-1), 159.46 (5-Ph C-4), 159.54 (C-5), 168.9 (C-6a). ${ }^{15}$ N NMR (71 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-170.6(\mathrm{~N}-2),-128.2(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 304$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 327.1104, found 327.1104.

### 4.1.2.2.6. 2-Phenyl-5-(thiophen-3-yl)-2H-furo[2,3-c]pyrazole (20)



Yellowish solid, yield $83 \mathrm{mg}, 62 \%, \mathrm{mp} 191-193{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3139,3128,3110,3097,3065,3045\left(\mathrm{CH}_{\text {arom }}\right), 1594,1587,1510,1497$, 1480, 1463, 1441, 1386, 1328, 1239, 1218, 1177, 1075, 1036, 1026 (C=C, $\mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 777,684\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.54(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.26-7.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, $7.37-7.39(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Th} \mathrm{H}), 7.44-7.47$ (m, 2H, Ph 3,5-H), 7.66 (dd, ${ }^{4} J(2-\mathrm{H}, 5-$ $\mathrm{H})=2.8 \mathrm{~Hz},{ }^{4} J(2-\mathrm{H}, 4-\mathrm{H})=1.4 \mathrm{~Hz}, 1 \mathrm{H}$, Th $\left.2-\mathrm{H}\right), 7.71-7.72(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-$ $\mathrm{H}), 7.74(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 96.1$ (C-4), 113.6 (C-3a), 116.3 (C-3), 119.1 (Ph C-2,6), 121.3 (Th C-2), 124.7 (Th C), 126.2 (Ph C-4), 126.7 (Th C), 129.6 (Ph C-3,5), 132.3 (Th C-3), 141.1 (Ph C-1), 156.1 (C-5), $168.7(\mathrm{C}-6 \mathrm{a}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 267\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{OS}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 267.0587, found 267.0589.

### 4.1.2.2.7. 5-Butyl-2-phenyl-2H-furo[2,3-c]pyrazole (21)

Brown solid, yield $54 \mathrm{mg}, 93 \%$, $\mathrm{mp} 88.5-90^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$,
 $\left.\mathrm{cm}^{-1}\right): 3141,3126,3074,3050\left(\mathrm{CH}_{\text {arom }}\right), 2968,2954,2928,2857\left(\mathrm{CH}_{\text {aliph }}\right)$, $1595,1577,1506,1460,1447,1438,1382,1320,1231,1202,1120,1073$, 1035 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C})$, 753, $691(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.95\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.4 \mathrm{~Hz}, 3 \mathrm{H}\right.$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.42 (sext, ${ }^{3} J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.70 (quin, ${ }^{3} J=7.6$ $\mathrm{Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), $2.69\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.06(\mathrm{~s}, 1 \mathrm{H}, 4-$ $\mathrm{H}), 7.22-7.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.41-7.43(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.61(\mathrm{~s}, 1 \mathrm{H}$, $3-\mathrm{H}), 7.67-7.68(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,4-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8$ $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 28.7\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 29.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 96.7(\mathrm{C}-4)$, 113.0 (C-3a), 115.1 (C-3), 118.9 (Ph C-2,6), 125.8 (Ph C-4), 129.4 (Ph C-3,5), 141.2 (Ph C-1), 163.1 (C-5), 168.1 (C-6a). MS m/z (\%): 263 ([M+Na] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{NaO}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 263.1155, found 263.1152 .

### 4.1.2.2.8. 5-Pentyl-2-phenyl-2H-furo[2,3-c]pyrazole (22)



White solid, yield $104 \mathrm{mg}, 82 \%$, $\mathrm{mp} 77-78^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3068,3048\left(\mathrm{CH}_{\text {arom }}\right)$, 2952, $2926\left(\mathrm{CH}_{\text {aliph }}\right), 1599,1580,1508,1448$, 1388 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}$ ), 727, $686(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 0.94\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.1 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right)$, 1.37-1.42 (m, 4H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{C}_{2} H_{4} \mathrm{CH}_{3}$ ), 1.75 (quin, ${ }^{3} \mathrm{~J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), $2.71\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right), 6.09(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.25-$ 7.28 (m, 1H, Ph 4-H), 7.44-7.46 (m, 2H, Ph 3,5-H), 7.65 (s, 1H, 3-H), 7.70-7.71 (m, 2H, Ph 2,4-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.1$ $\left(\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{CH}_{3}\right)$, $22.5\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $27.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $29.1\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right)$, 31.4 $\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 96.8(\mathrm{C}-4), 113.1(\mathrm{C}-3 \mathrm{a}), 115.2(\mathrm{C}-3), 119.0(\mathrm{Ph} \mathrm{C}-2,6), 125.9(\mathrm{Ph}$ C-4), 129.5 (Ph C-3,5), 141.3 (Ph C-1), 163.2 (C-5), 168.9 (C-6a). ${ }^{15} \mathrm{~N}$ NMR ( 71 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-172.0(\mathrm{~N}-2),-128.3(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 255\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}$ (ESI) for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 277.1311, found 277.1311.

### 4.1.2.2.9. 5-Cyclopropyl-2-phenyl-2H-furo[2,3-c]pyrazole (23)



Brown solid, yield $101 \mathrm{mg}, 90 \%$, $\mathrm{mp} 98.5-100{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3149,3135,3089,3027,3007\left(\mathrm{CH}_{\text {arom }}\right), 2955,2917,2850\left(\mathrm{CH}_{\text {aliph }}\right)$, 1595, 1585, 1507, 1460, 1443, 1390, 1374, 1296, 1230, 1213, 1122, 1045, 1035, (C=C, C-N, C-O-C), 757, $689(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.94-0.97\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right)$, 1.92-1.96 (m, 1H, CH( $\left.\left.\mathrm{CH}_{2}\right)_{2}\right), 6.06(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.21-7.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-$ H), 7.40-7.43 (m, 2H, Ph 3,5-H), $7.59(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}), 7.65-7.66(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph}$ 2,6-H). ${ }^{13} \mathrm{C}$ NMR (176 MHz, $\left.\mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{C}} \mathrm{ppm} 6.9\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 9.9$ $\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 95.4(\mathrm{C}-4), 113.1(\mathrm{C}-3 \mathrm{a}), 114.9(\mathrm{C}-3), 118.8$ (Ph C-2,6), 125.7 (Ph C-4), 129.4 (Ph C-3,5), 141.1 (Ph C-1), 163.7 (C-5), 168.3 (C-6a). MS m/z (\%): 225 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 225.1022, found 225.1022.

### 4.1.2.3. 3-(Benzyloxy)-1-methyl-1H-pyrazole (25) ${ }^{103}$.



A solution of 3-hydroxy-1-methyl-1H-pyrazole (23) (710 mg, 7.2 mmol ) in dry DMF ( 20 mL ) was cooled to $0^{\circ} \mathrm{C}$ temperature under inert atmosphere, and NaH ( $60 \%$ dispersion in mineral oil, $290 \mathrm{mg}, 7.2$ mmol ) was added portion wise. After stirring mixture for 15 min , benzyl chloride ( $0.82 \mathrm{~mL}, 7.2 \mathrm{mmol}$ ) was added drop wise. The mixture was stirred at $60^{\circ} \mathrm{C}$ temperature for 1 hour, then poured into water and extracted with ethyl acetate. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtrated, and the solvent was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $\left.1: 7, \mathrm{v} / \mathrm{v}\right)$ to give pure compound 25 as a brown liquid, yield $1205 \mathrm{mg}, 90 \%$. IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3118, 3089, $3064,3032,3008\left(\mathrm{CH}_{\text {arom }}\right)$, $2931\left(\mathrm{CH}_{\text {aliph }}\right), 1537,1491,1429,1360,1223,1052,1018$ ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 731,696,658,457\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.75\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 5.20\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.66(\mathrm{~d}$, $\left.{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.4 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.13\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.4 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 7.30-7.33(\mathrm{~m}$, 1H, Ph 4-H), 7.36-7.39 (m, 2H, Ph 3,5-H), 7.45-7.48 (m, 2H, Ph 2,6-H). ${ }^{13} \mathrm{C}$ NMR
( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 39.0\left(\mathrm{CH}_{3}\right), 70.9\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 90.5(\mathrm{C}-4), 127.8(\mathrm{Ph} \mathrm{C-2,6)}$, 128.0 ( $\mathrm{Ph} \mathrm{C}-4$ ), 128.5 ( $\mathrm{Ph} \mathrm{C-3.5)}$,131.4 (C-5), 137.3 (Ph C-1), 163.3 (C-3). MS m/z (\%): $189\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 211.0842, found 211.0842 .
4.1.2.4. 3-(Benzyloxy)-1-methyl-1H-pyrazole-4-carbaldehyde (27). Phosphorus oxychloride ( $1.59 \mathrm{~mL}, 17 \mathrm{mmol}$ ) was added dropwise to DMF $(1.32 \mathrm{~mL}, 17$
 mmol ) at $-10{ }^{\circ} \mathrm{C}$ temperature. Then, pyrazole 25 ( $800 \mathrm{mg}, 4.25$ mmol ) was added to the Vilsmeier-Haack complex, and the reaction mixture was heated at $70{ }^{\circ} \mathrm{C}$ temperature for 12 hours. After the neutralization with $10 \%$ aq $\mathrm{NaHCO}_{3}$ solution, the precipitate was filtered off and recrystallized from DCM to give pure compound 27 as a white solid, yield $775 \mathrm{mg}, 85 \%, \mathrm{mp} 54-56^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3121,3099,3045\left(\mathrm{CH}_{\text {arom }}\right), 2992,2947\left(\mathrm{CH}_{\text {aliph }}\right), 1666(\mathrm{C}=\mathrm{O}), 1588$, 1577, 1541, 1510, 1315, 1181 (C=C, C-N, C-O-C), 894, 883, 706 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.77\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 5.32(\mathrm{~s}$, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 7.32-7.36 (m, 1H, Ph 4-H), 7.36-7.40 (m, 2H, Ph 3,5-H), 7.46-7.47 (m, $2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.69(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.76(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\text {C }} \mathrm{ppm} 39.8\left(\mathrm{CH}_{3}\right), 71.0\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 109.6(\mathrm{C}-4), 128.0(\mathrm{Ph} \mathrm{C}-2,6), 128.3(\mathrm{Ph} \mathrm{C}-4)$, 128.6 (Ph C-3.5), 133.4 (C-5), 136.4 ( $\mathrm{Ph} \mathrm{C}-1$ ), 163.3 (C-3), 183.1 (CHO). MS m/z (\%): $217\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 239.0791, found 239.0791 .
4.1.2.5. 3-(Benzyloxy)-1-methyl-1H-pyrazole-4-carbaldehyde (29).

|  |
| :---: |

Into the solution of pyrazole $27(570 \mathrm{mg}, 2.6 \mathrm{mmol})$ in toluene ( 10 $\mathrm{mL})$, TFA $(10 \mathrm{~mL})$ was added. The mixture was stirred at room temperature for 48 hours. Toluene and TFA were evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 1, \mathrm{v} / \mathrm{v}$ ) to give pure compound 29 as a colorless solid, yield $295 \mathrm{mg}, 90 \%$, $\mathrm{mp} 201-202^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3291(\mathrm{OH})$, $3099\left(\mathrm{CH}_{\text {arom }}\right)$, 2992, $2947\left(\mathrm{CH}_{\text {aliph }}\right), 1666(\mathrm{C}=\mathrm{O}), 1541,1510,1315,1181(\mathrm{C}=\mathrm{C}, \mathrm{C}-$ N). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , DMSO- $d_{6}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.67\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 8.04(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.60$ (s, 1H, CHO), $10.90(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( $\left.176 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta_{\mathrm{C}} \mathrm{ppm} 38.9\left(\mathrm{CH}_{3}\right)$, 108.4 (C-4), $134.0(\mathrm{C}-5), 161.9(\mathrm{C}-3), 182.6(\mathrm{CHO}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 127\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 149.0322, found 149.0321.

### 4.1.2.6. 4-Formyl-1-methyl-1H-pyrazol-3-yl trifluoromethanesulfonate

 (31). Pyrazole 29 ( $250 \mathrm{mg}, 2 \mathrm{mmol}$ ), trifluormethansulfonic anhydride ( $1 \mathrm{~mL}, 6$ $\mathrm{mmol})$, and TEA ( $1 \mathrm{~mL}, 7.2 \mathrm{mmol}$ ) were dissolved in DCM ( 20 mL ), and the mixture was stirred at room temperature for 1 hour. The reaction mixture was poured into water and extracted with ethyl acetate. The combined organic layers were washed with brine and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$; the solvent was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 4, \mathrm{v} / \mathrm{v}$ ) to give pyrazole 31 as a brown solid, yield $435 \mathrm{mg}, 85 \%, \mathrm{mp} 54-55^{\circ} \mathrm{C}$ (ethyl acetate $)$. IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3091,3043,3017\left(\mathrm{CH}_{\text {arom }}\right), 2958,2918\left(\mathrm{CH}_{\text {aliph }}\right), 1675(\mathrm{C}=\mathrm{O})$,

1552, 1429, 1205, 1135, 884, 875, 794, 744, 601, 511 (C-C, C-N, C-F). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \operatorname{ppm} 3.93\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.87(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.80(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO})$. ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 40.7\left(\mathrm{CH}_{3}\right), 113.6(\mathrm{C}-4), 116.0,117.8,119.6$, $121.5\left(\mathrm{CF}_{3}{ }^{1} \mathrm{~J}=321.2 \mathrm{~Hz}\right), 134.7(\mathrm{C}-5), 151.2(\mathrm{C}-3), 181.0(\mathrm{CHO}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 259$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{SNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 280.9814, found 280.9814.

### 4.1.2.7. 1-Phenyl-1H-pyrazol-3-yl 2-methylpropanoate (33).



Into the solution of 3-hydroxy-1-phenyl-1H-pyrazole (2) ${ }^{75}$ ( $470 \mathrm{mg}, 2.9$ $\mathrm{mmol})$ in DCM $(10 \mathrm{~mL})$, pyridine $(2.3 \mathrm{~mL})$ and isobutyrylchloride ( 0.33 $\mathrm{mL}, 3.2 \mathrm{mmol}$ ) were added. The mixture was stirred at room temperature for 30 min , poured into water, and extracted with DCM . The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtrated, and the solvent was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 4$, v/v) to give pure compound 33 as a brown liquid, yield $601 \mathrm{mg}, 90 \%$. IR $\left(v_{\max }, \mathrm{cm}^{-1}\right): 3134,3064,3050$ $\left(\mathrm{CH}_{\text {arom }}\right), 2977,2938,2912\left(\mathrm{CH}_{\text {aliph }}\right), 1761(\mathrm{C}=\mathrm{O}), 1599,1532,1453,1391,1124$, 1113, $1092(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 752,688(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 1.36\left(\mathrm{~d},{ }^{3} \mathrm{~J}\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)=7.1 \mathrm{~Hz}, 6 \mathrm{H}\right.$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 2.85-2.90\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 6.41\left(\mathrm{~d},{ }^{3} \mathrm{~J}(4-\mathrm{H}, 5-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right)$, 7.28-7.30 (m, 1H, Ph 4-H), 7.44-7.46 (m, 2H, Ph 3,5-H), 7.64-7.66 (m, 2H, Ph 2,6H), $7.86\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 18.8$ $\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 34.1\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 98.8(\mathrm{C}-4), 118.7(\mathrm{Ph} \mathrm{C-2,6}), 126.5(\mathrm{C}-5), 127.7(\mathrm{Ph}$ C-4), 129.4 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 139.7 ( $\mathrm{Ph} \mathrm{C}-1$ ), 156.8 (C-3), 174.2 (O-C=O). MS m/z (\%): $231\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 253.0947, found 253.0947 .

### 4.1.2.8. 1-(3-Hydroxy-1-phenyl-1H-pyrazol-4-yl)-2-methylpropan-1-one

 (35). Into the solution of $\mathrm{AlCl}_{3}(2.453 \mathrm{~g}, 18.4 \mathrm{mmol})$ in $\mathrm{CS}_{2}(8 \mathrm{~mL})$, a solution of pyrazole $33(350 \mathrm{mg}, 1.5 \mathrm{mmol})$ in $\mathrm{CS}_{2}(28 \mathrm{~mL})$ was added drop wise at $0^{\circ} \mathrm{C}$ temperature. The mixture was stirred at 55 ${ }^{\circ} \mathrm{C}$ temperature for 3 hours. After the neutralization with ice-cold water ( 33 mL ) and $6 \mathrm{~N} \mathrm{HCl}(15 \mathrm{~mL})$, the precipitate was filtered off and purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 4, \mathrm{v} / \mathrm{v}$ ) to give pure compound 35 as a white solid, yield $310 \mathrm{mg}, 89 \%$, $\mathrm{mp} 97-9{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3309(\mathrm{OH}), 3122$, 3068, $3052\left(\mathrm{CH}_{\text {arom }}\right)$, 2965, $2932\left(\mathrm{CH}_{\text {aliph }}\right), 1656(\mathrm{C}=\mathrm{O}), 1559,1530,1508,1459$, 1319, 1233, $1217(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 747,685,671(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.07\left(\mathrm{~d},{ }^{3} \mathrm{~J}=6.9 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.33-3.36$ $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 7.29-7.31(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.47-7.49(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.80-$ $7.81(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.87(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 11.12(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 18.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 36.9\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 109.4(\mathrm{C}-4), 118.1(\mathrm{Ph} \mathrm{C-2,6})$, 126.5 (Ph C-4), 129.6 (Ph C-3,5), 131.4 (C-5), 138.9 (Ph C-1), 161.4 (C-3), 198.6 $(\mathrm{C}=\mathrm{O}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 231\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$ calcd 253.0947, found 253.0947.
4.1.2.9. 4-(2-Methylpropanoyl)-1-phenyl-1H-pyrazol-3-yl trifluoromethanesulfonate (37). This compound was synthesized in analogy to compound 29 from
 pyrazole $35(252 \mathrm{mg}, 2 \mathrm{mmol})$. Pyrazole 37 was obtained as a white solid, yield $495 \mathrm{mg}, 96 \%$, mp $96-97{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3094,3061\left(\mathrm{CH}_{\text {arom }}\right), 2980,2937,2918\left(\mathrm{CH}_{\text {aliph }}\right), 1555,1452$, 1427, 1231, 1213, 1201, 1138 (C=C, C-N, C-F), 973, 881, 760, 740, 607, 599, $505\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.23-1.24\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 3.15-3.18 (m, 1H, CH(CH3 $\left.)_{2}\right), 7.39-7.41(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.49-$ $7.51(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.66-7.67(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.35(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (176 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}}$ ppm $18.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 38.8\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 113.6(\mathrm{C}-4), 116.0$, $117.8,119.5(\mathrm{Ph} \mathrm{C}-2,6), 119.6,121.5\left(\mathrm{CF}_{3}{ }^{1} J=322.1 \mathrm{~Hz}\right), 128.5$ ( $\mathrm{Ph} \mathrm{C}-4$ ), 129.9 ( Ph C-3,5), 130.8 (C-5), 138.5 (Ph C-1), 151.9 (C-3), 196.6 (C=O). MS m/z (\%): 363 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{SNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 385.0440, found 385.0440.

### 4.1.2.10. General procedure for the preparation of 4-alkynyl-1-phenyl-1H-pyrazole-4-carbaldehydes, ethanones, and propanones by Sonogashira crosscoupling reaction.

Into the solution of appropriate pyrazoles $\mathbf{3 0}, \mathbf{3 1}, \mathbf{3 6}$, or $\mathbf{3 7}(0.5 \mathrm{mmol})$ in dry DMF ( 1 mL ) under the argon atmosphere TEA ( $4.0 \mathrm{~mL}, 2.5 \mathrm{mmol}$ ), appropriate acetylene $(0.75 \mathrm{mmol}), \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(35 \mathrm{mg}, 0.05 \mathrm{mmol})$, and $\mathrm{CuI}(18 \mathrm{mg}, 0.1 \mathrm{mmol})$ were added. The mixture was stirred for the given time under argon atmosphere at $70{ }^{\circ} \mathrm{C}$ temperature, diluted with water, and the extraction was done with ethyl acetate. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and the solvent was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 8, \mathrm{v} / \mathrm{v}$ ) to yield compounds 38-55.
4.1.2.10.1. 1-Methyl-3-(phenylethynyl)-1H-pyrazole-4-carbaldehyde (38).


The reaction mixture was stirred for 8 hours. Brown amorphous solid, 105 mg , yield $75 \%, \mathrm{mp} 123-124{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3100, 3057, $3042\left(\mathrm{CH}_{\text {arom }}\right), 2954,2925\left(\mathrm{CH}_{\text {aliph }}\right), 2221(\mathrm{C} \equiv \mathrm{C}), 1681$ (CHO), 1598, 1560, 1533, 1497, 1441, 1181 (C=C, C-N), 909, 880, 775, 755, $690\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR (700 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.85\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.24-7.27(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph} 3,4,5-$ $\mathrm{H}), 7.46-7.47(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.90(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.88(\mathrm{~s}, 1 \mathrm{H}$, CHO). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 40.1\left(\mathrm{CH}_{3}\right), 79.0$ $(C \equiv \mathrm{CPh}), 94.1(\mathrm{C} \equiv C \mathrm{Ph}), 113.6(\mathrm{C}-3), 122.0(\mathrm{Ph} \mathrm{C}-1), 125.0(\mathrm{C}-4), 128.6$ ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 134.8 (C-5), 184.3 (CHO). MS m/z (\%): 211 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{OSNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 233.0685, found 233.0685.
4.1.2.10.2. 1-Phenyl-3-[(thiophen-3-yl)ethynyl]-1H-pyrazole-4-carbaldehyde (40). The reaction mixture was stirred for 8 hours. White solid, yield 105 mg ,
 $75 \%, \mathrm{mp} 123-124{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3126, 3036 $\left(\mathrm{CH}_{\text {arom }}\right), 2229(\mathrm{C} \equiv \mathrm{C}), 1676(\mathrm{C}=\mathrm{O}), 1596,1561,1462,1409,1297$, $1125(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 909,873,815,621,598,512(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 7.28(\mathrm{~d}$, ${ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=4.9 \mathrm{~Hz}, 1 \mathrm{H}$, Th $\left.4-\mathrm{H}\right), 7.34\left(\mathrm{dd},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=4.9 \mathrm{~Hz}\right.$, ${ }^{4} J(5-\mathrm{H}, 2-\mathrm{H})=2.9 \mathrm{~Hz}, 1 \mathrm{H}$, Th $\left.5-\mathrm{H}\right), 7.40-7.42(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.68$ $\left(\mathrm{d},{ }^{4} \mathrm{~J}(2-\mathrm{H}, 5-\mathrm{H})=2.9 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Th $\left.2-\mathrm{H}\right), 7.50-7.52(\mathrm{~m}, 2 \mathrm{H}$, Ph 3,5H), 7.74-7.76 (m, 2H, Ph 2,6-H), $8.44(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 10.09(\mathrm{~s}, 1 \mathrm{H}$, CHO). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 78.6(C \equiv \mathrm{CTh}), 90.3$ ( $\mathrm{C} \equiv C \mathrm{Th}$ ), 120.0 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 120.9 (Th C-3), 125.91 (C-4), 125.95 (Th C-5), 128.5 (Ph C-4), 128.7 (C-5), 129.9 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 130.0 (Th C-4), 130.8 (Th C-2), 138.7 (C-3), 138.9 ( $\mathrm{Ph} \mathrm{C}-1$ ), 182.7 (CHO). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-158.2(\mathrm{~N}-1)$, $-72.0(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 279\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{OSNa}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 301.0406, found 301.0406.

### 4.1.2.10.3. 3-(Cyclopropylethynyl)-1-phenyl-1H-pyrazole-4-carbaldehyde

(41). The reaction mixture was stirred for 1 hour. White solid, yield $103 \mathrm{mg}, 86 \%, \mathrm{mp}$
 $90-91^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3121,3065,3014\left(\mathrm{CH}_{\text {arom }}\right)$, 2881, $2782\left(\mathrm{CH}_{\text {aliph }}\right), 2232(\mathrm{C} \equiv \mathrm{C}), 1677(\mathrm{C}=\mathrm{O}), 1599,1530,1504$, 1361, 1226 (C=C, C-N), 865, 849, 801, 757, 704, 685, $506(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ 0.92-0.98 (m, 4H, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 1.52-1.57\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 7.37-$ 7.41 (m, 1H, Ph 4-H), 7.47-7.52 (m, 2H, Ph 3,5-H), 7.70-7.73 (m, $2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.37(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.97(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO}) .{ }^{13} \mathrm{C}$ NMR (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 0.4\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right)$, $9.1\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 65.5$ $\left(C \equiv \mathrm{CCH}\left(\mathrm{CH}_{2}\right)_{2}, 100.2\left(\mathrm{C} \equiv \mathrm{CCH}\left(\mathrm{CH}_{2}\right)_{2}\right), 119.9(\mathrm{Ph} \mathrm{C-2,6}), 125.9(\mathrm{C}-4), 128.3(\mathrm{Ph} \mathrm{C}-\right.$ 4), 128.4 (C-5), 129.8 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 138.9 (C-3), 139.3 ( $\mathrm{Ph} \mathrm{C}-1$ ), $184.8(\mathrm{CHO}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-157.6(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS m/z (\%): 237 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{15} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 259.0842, found 259.0844.
4.1.2.10.4. 3-(1-Hexyn-1-yl)-1-phenyl-1H-pyrazole-4-carbaldehyde (42).


The reaction mixture was stirred for 2 hours. Brown liquid, yield $113 \mathrm{mg}, 90 \%$. IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-}\right): 3046\left(\mathrm{CH}_{\text {arom }}\right), 2961,2935\left(\mathrm{CH}_{\text {aliph }}\right)$, 1678, 1529, 1507, 1228 ( $\mathrm{CHO}, \mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 758,685(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene) $\mathrm{cm}^{-1}$. ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}}$ ppm $0.98\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.48-1.53(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.63-1.69 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), $2.53\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.1\right.$ $\mathrm{Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}, 7.37-7.39(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.47-7.49(\mathrm{~m}, 2 \mathrm{H}$, Ph 3,5-H), 7.70-7.72 (m, 2H, Ph 2,6-H), 8.40 (s, 1H, 5-H), 10.01 (s, $1 \mathrm{H}, \mathrm{CHO}) .{ }^{13} \mathrm{C}$ NMR (176 MHz, $\left.\mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{C}} \mathrm{ppm} 13.7\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 19.3\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 30.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 70.4\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, $97.2\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right), 119.9$ (Ph C-2,6), 125.7 (C-4), 128.28 (C-5), 128.34 (Ph C-4), 129.8 (Ph C-3,5), 138.8 (Ph $\mathrm{C}-1), 139.3(\mathrm{C}-3), 184.9(\mathrm{CHO}) .{ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-159.1(\mathrm{~N}-1)$,
$-71.8(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 253\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$ calcd 253.1335, found 235.1337.
4.1.2.10.5. 3-(Hept-1-yn-1-yl)-1-phenyl-1H-pyrazole-4-carbaldehyde (43).


The reaction mixture was stirred for 4 hours. Brown liquid, yield $128 \mathrm{mg}, 92 \%$. IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3124, $3053\left(\mathrm{CH}_{\text {arom }}\right), 2956,2931$, $2860\left(\mathrm{CH}_{\text {aliph }}\right), 2243(\mathrm{C}=\mathrm{C}), 1684(\mathrm{C}=\mathrm{O}), 1599,1531,1504,1464$, 1398, $1296(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 956,757,689,509(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.89-0.93$ ( $\mathrm{m}, 3 \mathrm{H}, \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}$ ), 1.34-1.39 (m, $2 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.44-1.48 (m, 2H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 1.65-1.69 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), 2.48$2.51\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right), 7.36-7.39(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.46-7.49(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.69-7.72(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.38(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO})$. ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.0\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right), 19.5\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right), 22.2$ $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $28.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $31.1\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $70.3\left(\mathrm{C} \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right)$, 97.2 $\left(\mathrm{C} \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right), 119.8(\mathrm{Ph} \mathrm{C-2,6}), 125.7(\mathrm{C}-4), 128.2(\mathrm{Ph} \mathrm{C}-4), 128.3(\mathrm{C}-5), 129.7(\mathrm{Ph}$ $\mathrm{C}-3,5), 138.8$ (Ph C-1), 139.3 (C-3), 184.8 (CHO). $\left.{ }^{15} \mathrm{~N} \mathrm{NMR} \mathrm{(71} \mathrm{MHz} \mathrm{CDCl},\right)_{3}$ ): $\delta_{\mathrm{N}}$ ppm $-158.6(\mathrm{~N}-1),-72.4(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 267\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 289.1306, found 289.1311.
4.1.2.10.6. 3-(4-Hydroxybut-1-yn-1-yl)-1-phenyl-1H-pyrazole-4-carbaldehyde (44). The reaction mixture was stirred for 4 hours. White solid, 108 mg , yield
 $90 \%, \mathrm{mp} 111-112{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3454(\mathrm{OH})$, 3126, $3065\left(\mathrm{CH}_{\text {arom }}\right), 2929\left(\mathrm{CH}_{\text {aliph }}\right), 2240(\mathrm{C} \equiv \mathrm{C}), 1679(\mathrm{C}=\mathrm{O})$, 1598, 1531, 1504, 1462, 1402, 1362, 1315, 1226, 1051 (C=C, $\mathrm{C}-\mathrm{N}), 866,812,796,755,685(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.74-2.76$ (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ ), 3.87-3.88 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ ), 7.35-7.37 $(\mathrm{m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.44-7.47(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.66-7.68(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.37$ (s, 1H, 5-H), 9.95 (CHO). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm}$ 24.1 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$, $60.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$, $72.4\left(C \equiv \mathrm{CC}_{2} \mathrm{CH}_{4} \mathrm{OH}\right)$, $94.0\left(\mathrm{C} \equiv \mathrm{CC}_{2} \mathrm{H}_{4} \mathrm{OH}\right)$, 119.8 (NPh C-2,6), 125.8 (C-4), 128.4 (Ph C-4), 129.5 (Ph C-3,5), 129.8 (C-5), 138.1 (C-3), 138.7 (Ph C-1), $184.8(\mathrm{CHO}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 241$ ([M+H $\left.{ }^{+}, 100\right), 263\left([\mathrm{M}+\mathrm{Na}]^{+}\right.$, 30). HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 263.0792, found 263.0791 .

### 4.1.2.10.7. 1-Phenyl-3-[(trimethylsilyl)ethynyl]-1H-pyrazole-4-carbalde-

 hyde (45). The reaction mixture was stirred for 10 hours. White solid, yield 105 mg , $78 \%$, mp 97-98 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3128, 3067 $\left(\mathrm{CH}_{\text {arom }}\right), 2960,2854\left(\mathrm{CH}_{\text {aliph }}\right), 2167(\mathrm{C} \equiv \mathrm{C}), 1678(\mathrm{C}=\mathrm{O}), 1598$, 1531, 1503, 1363, 1251, 1223 (C=C, C-N), 956, 838, 751, 726, 683 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.30(\mathrm{~s}, 9 \mathrm{H}, \mathrm{TMS}), 7.37-7.41(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.47-7.51 (m, 2H, Ph 3,5-H), 7.70-7.72 (m, 2H, Ph 2,6-H), 8.39 (s, $1 \mathrm{H}, 5-\mathrm{H}), 10.01(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}}$ ppm 0.2 (TMS), 93.6 ( $\mathrm{C} \equiv C \mathrm{TMS}$ ), 102.0 ( $C \equiv \mathrm{CTMS}$ ), 120.1 ( $\mathrm{Ph} \mathrm{C}-$

2,6), 126.2 (C-4), 128.3 (C-5), 128.5 (Ph C-4), 129.9 (Ph C-3.5), 138.7 (C-3), 138.8 ( $\mathrm{Ph} \mathrm{C}-1$ ), 184.7 (CHO). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-157(\mathrm{~N}-1)$, $\mathrm{N}-2$ was not found. MS m/z (\%): 269 ([M+H] $\left.{ }^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{OSiNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$ calcd 291.0924, found 291.0928.

### 4.1.2.10.8. 1-\{1-Phenyl-3-[(thiophen-3-yl)ethynyl]-1H-pyrazol-4-yl\}ethan-

 1-one (47). The reaction mixture was stirred for 8 hours. White solid, yield 97 mg , 66\%, mp 116-117 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3117, 3054, $3022\left(\mathrm{CH}_{\text {arom }}\right), 2996,2955,2917\left(\mathrm{CH}_{\text {aliph }}\right), 2232(\mathrm{C} \equiv \mathrm{C})$, 1655 (C=O), 1597, 1514, 1508, 1351, 1264, 1232, 1058 (C=C, C-N), 872, 774, 761, 708, 691, 680, 620, $503(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.73\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.28\left(\mathrm{~d},{ }^{3} \mathrm{~J}(4-\mathrm{H}, 5-\mathrm{H})=5.2 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Th 4-H), 7.35-7.37 (m, 1H, Th 5-H), 7.39-7.42 (m, 1H, Ph 4H), $7.50-7.52(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.68\left(\mathrm{~d},{ }^{4} \mathrm{~J}(2-\mathrm{H}, 5-\mathrm{H})=2.9\right.$ $\mathrm{Hz}, 1 \mathrm{H}$, Th 2-H), 7.75-7.77 (m, 2H, Ph 2,6-H), 8.47 (s, 1H, 5-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 29.1\left(\mathrm{CH}_{3}\right), 80.8$ ( $C \equiv \mathrm{CTh}$ ), 90.1 ( $\mathrm{C} \equiv C \mathrm{Th}$ ), 119.7 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 121.1 (Th C-3), 125.8 (Th C-5), 126.9 (C4), 128.1 (Ph C-4), 129.7 (Th C-4), 129.7 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 129.9 (C-5), 130.2 (Th C-2), $136.2(\mathrm{C}-3), 138.8(\mathrm{Ph} \mathrm{C}-1), 192.2(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}$ $-160.4(\mathrm{~N}-1),-72.5(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 293\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{OSiNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 315.0563, found 315.0561.
4.1.2.10.9. 1-[3-(Cyclopropylethynyl)-1-phenyl-1H-pyrazol-4-yl]ethan-1one (48). The reaction mixture was stirred for 1 hour. White solid, yield $100 \mathrm{mg}, 80 \%$,
 $\mathrm{mp} 127-128{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\mathrm{v}_{\text {max }}, \mathrm{cm}^{-1}$ ): 3129, 3078 $\left(\mathrm{CH}_{\text {arom }}\right), 2993,2956,2923\left(\mathrm{CH}_{\text {aliph }}\right), 2233(\mathrm{C} \equiv \mathrm{C}), 1670(\mathrm{C}=\mathrm{O})$, 1599, 1521, 1448, 1362, 1260, 1241, 1221 (C=C, C-N), 977, 940, 863, 751, 705, $683(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.90-0.96(\mathrm{~m}, 4 \mathrm{H}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 1.53-1.57\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right)$, $2.62\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 7.34-7.36 (m, 1H, Ph 4-H), 7.45-7.47 (m, 2H, Ph 3,5-H), 7.69 ( $\mathrm{m}, 2 \mathrm{H}$, Ph $2,6-\mathrm{H}$ ), 8.38 ( $\mathrm{s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 0.4\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 8.8\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{2}\right), 29.1\left(\mathrm{CH}_{3}\right), 67.8\left(\mathrm{C} \equiv \mathrm{CCH}\left(\mathrm{CH}_{2}\right)_{2}\right)$, $99.9\left(\mathrm{C} \equiv \mathrm{CCH}\left(\mathrm{CH}_{2}\right)_{2}\right), 119.8$ (Ph C-2,6), 126.9 (C-4), 128.0 (Ph C-4), 129.7 (C-5), 129.8 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 136.7 (C-3), 139.0 ( $\mathrm{Ph} \mathrm{C}-1$ ), 192.6 (C=O). ${ }^{15} \mathrm{~N}$ NMR ( 71 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-160.2(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS m/z (\%): $251\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 273.0998,
 found 273.0998 .
4.1.2.10.10. 1-[3-(Hex-1-yn-1-yl)-1-phenyl-1H-pyra-zol-4-yl]ethan-1-one (49). The reaction mixture was stirred for 1 hour. White solid, yield $100 \mathrm{mg}, 80 \%$, mp 127-128 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3129, $3078\left(\mathrm{CH}_{\text {arom }}\right), 2993,2956,2923$ $\left(\mathrm{CH}_{\text {aliph }}\right), 2233(\mathrm{C} \equiv \mathrm{C}), 1670(\mathrm{C}=\mathrm{O}), 1599,1521,1448,1362$, 1260, 1241, 1221 (C=C, C-N), 977, 940, 863, 751, 705, 683 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.94(\mathrm{t}$,
$\left.{ }^{3} J=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.48-1.51\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.62-1.66(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.49-2.51 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), $2.51\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{3}\right), 7.32-7.34(\mathrm{~m}, 1 \mathrm{H}$, Ph 4-H), 7.43-7.46 (m, 2H, Ph 3,5-H), 7.69-7.70 (m, 2H, Ph 2,6-H), 8.38 (s, 1H, 5H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.7\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 19.4\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 22.2$ $\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $29.0\left(\mathrm{CH}_{3}\right)$, $30.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $72.7\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, $96.9\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, 119.7 (Ph C-2,6), 126.7 (C-4), 128.0 (Ph C-4), 129.7 (C-5), 129.8 (Ph C-3,5), 136.7 (C-3), 138.9 (Ph C-1), 192.5 (C=O). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-160.1$ (N1), $-70.8(\mathrm{~N}-2)$. MS m/z (\%): $267\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{ONa}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 289.1311, found 289.1312 .

### 4.1.2.10.11. 1-[3-(Hept-1-yn-1-yl)-1-phenyl-1H-pyrazol-4-yl]ethan-1-one

(50). The reaction mixture was stirred for 4 hours. Brown liquid, yield $91 \mathrm{mg}, 65 \%$.

|  |  |
| :---: | :---: | IR $\left(v_{\max }, \mathrm{cm}^{-1}\right): 3130,3064\left(\mathrm{CH}_{\text {arom }}\right), 2957,2928,2860$ $\left(\mathrm{CH}_{\text {aliph }}\right), 2240(\mathrm{C} \equiv \mathrm{C}), 1666(\mathrm{C}=\mathrm{O}), 1522,1457,1362,1262$, 1245, 1217 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}$ ), 751, 706, $685(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 0.91$ (t, ${ }^{3} \mathrm{~J}=7.3 \mathrm{~Hz}, \quad 3 \mathrm{H}, \quad \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}$ ), $1.33-1.38 \quad(\mathrm{~m}, \quad 2 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.43-1.47 (m, 2H, C2 $\mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 1.64-1.69 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), $2.50\left(\mathrm{t},{ }^{3} \mathrm{~J}=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right.$ ), $2.63\left(\mathrm{CH}_{3}\right), 7.31-7.36(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.44-7.46(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H}), 7.69-7.70(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.39(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.0$ $\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right)$, $19.7\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}\right), 22.3\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 28.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $29.0\left(\mathrm{CH}_{3}\right)$, $31.3\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 72.7\left(C \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right), 97.0\left(\mathrm{C} \equiv \mathrm{CC}_{5} \mathrm{H}_{11}\right), 119.7(\mathrm{Ph} \mathrm{C}-2,6), 126.7$ (C-4), 128.0 (Ph C-4), 128.6 (C-5), 129.7 (Ph C-3,5), 136.7 (C-3), 138.9 (Ph C-1), $192.5(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-161.8(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS m/z (\%): $281\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 303.1468 , found 303.1468 .

### 4.1.2.10.12. 1-[3-(4-Hydroxybut-1-yn-1-yl)-1-phenyl-1H-pyrazol-4-

yl]ethan-1-one (51). The reaction mixture was stirred for 4 hours. White solid, yield
 $109 \mathrm{mg}, 86 \%, \mathrm{mp} 121-122{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3130,3068,3026\left(\mathrm{CH}_{\text {arom }}\right), 2995,2961,2945\left(\mathrm{CH}_{\text {aliph }}\right)$, $2240(\mathrm{C} \equiv \mathrm{C}), 1666(\mathrm{C}=\mathrm{O}), 1521,1460,1350,1262,1249$, 1217 (C=C, C-N), 755, 731, 685, 577, $468(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.59\left(\mathrm{CH}_{3}\right), 2.75-2.77\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 3.89-$ 3.91 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{OH}$ ), $7.36-7.38(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.47-7.49 (m, 2H, Ph 3,5-H), 7.69-7.71 (m, 2H, Ph 2,6H), $8.36(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 24.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$, 28.8 ( $\left.\mathrm{CH}_{3}\right), 60.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 74.7\left(\mathrm{C} \equiv \mathrm{CC}_{2} \mathrm{H}_{4} \mathrm{OH}\right), 93.7\left(\mathrm{C} \equiv \mathrm{CC}_{2} \mathrm{H}_{4} \mathrm{OH}\right), 119.7(\mathrm{NPh}$ C-2,6), 126.4 (C-4), 128.2 (Ph C-4), 129.8 (Ph C-3,5), 130.1 (C-5), 136.3 (C-3), 138.9 ( $\mathrm{Ph} \mathrm{C}-1$ ), $192.0(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-160.4(\mathrm{~N}-1),-70.5(\mathrm{~N}-$ 2). MS m/z (\%): $255\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 277.0947, found 277.0947.
4.1.2.10.13. 1-\{1-Phenyl-3-[(trimethylsilyl)ethynyl]-1H-pyrazol-4-yl\}ethan-1-one (52). The reaction mixture was stirred for 12 hours. White solid, yield 117 mg ,
 $83 \%, \mathrm{mp} 89-90^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3129, 3065 $\left(\mathrm{CH}_{\text {arom }}\right), 2959,2899\left(\mathrm{CH}_{\text {aliph }}\right), 2167(\mathrm{C} \equiv \mathrm{C}), 1666(\mathrm{C}=\mathrm{O})$, 1523, 1507, 1442, 1363, 1261, 1206 (C=C, C-N), 858, 847, 755, $690\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.30$ (s, 9H, TMS), 2.66 (s, 3H, $\left.\mathrm{CH}_{3}\right), 7.36-7.38(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.46-7.48(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-$ $\mathrm{H}), 7.70-7.71(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.40(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 0.3$ (TMS), $29.1\left(\mathrm{CH}_{3}\right), 96.3$ ( $C \equiv$ CTMS $), 101.9$ ( $\mathrm{C} \equiv C T M S$ ), 119.9 (Ph C-2,6), 127.5 (C-4), 128.2 (C-5), 129.8 ( Ph C-4), 129.9 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 135.9 (C-3), 138.9 ( $\mathrm{Ph} \mathrm{C-1)}$,192.5 (C=O). ${ }^{15} \mathrm{~N} \mathrm{NMR}(71 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-158.7(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS m/z (\%): $283\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{OSiNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 305.1081, found 305.1081.
4.1.2.10.14. 2-Methyl-1-[1-phenyl-3-(phenylethynyl)-1H-pyrazol-4-yl]pro-pan-1-one (53). The reaction mixture was stirred for 1 hour. White solid, yield 141
 $\mathrm{mg}, 90 \%, \mathrm{mp} 105-100^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3121, $3061\left(\mathrm{CH}_{\text {arom }}\right), 2968,2869\left(\mathrm{CH}_{\text {aliph }}\right), 2265(\mathrm{C} \equiv \mathrm{C}), 1664$ (C=O), 1521, 1443, 1350, 1227 (C=C, C-N), 989, 875, 751, 685 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( 700 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 1.29\left(\mathrm{~d},{ }^{3} \mathrm{~J}=6.9 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 3.72-3.76 (m, 1H, CH(CH3 $)_{2}$ ), 7.37-7.40 (m, 4H, NPh 4-H, CPh 3,4,5-H), 7.48-7.50 (m, 2H, NPh 3,5-H), 7.61-7.62 (m, $2 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}), 7.75-7.76$ (m, 2H, NPh 2,6-H), 8.47 (s, 1H, $5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 18.9\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 38.0\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 81.3$ $(C \equiv \mathrm{CPh})$, $94.0(\mathrm{C} \equiv \mathrm{CPh}), 119.9$ (NPh C-2,6), 122.2 ( $\mathrm{CPh} \mathrm{C}-1$ ), 125.8 (C-4), 128.1 (NPh C-4), 128.6 (CPh C-3,5), 129.3 (CPh C-4), 129.8 (NPh C-3,5), 130.6 (C-5), 131.9 (CPh C-2,6), 135.7 (C-3), 139.0 ( $\mathrm{NPh} \mathrm{C}-1$ ), 198.9 (C=O). ${ }^{15} \mathrm{~N}$ NMR ( 71 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-158.6(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS m/z (\%): $315\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 337.1311, found 337.1311.

### 4.1.2.10.15. 1-[3-(Hex-1-yn-1-yl)-1-phenyl-1H-pyrazol-4-yl]-2-methylpro-

 pan-1-one (54). The reaction mixture was stirred for 1 hour. Colorless liquid, yield $119 \mathrm{mg}, 81 \%$. IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3122,3056\left(\mathrm{CH}_{\text {arom }}\right), 2961$, 2929, 2871, $2855\left(\mathrm{CH}_{\text {aliph }}\right), 2237(\mathrm{C} \equiv \mathrm{C}), 1665(\mathrm{C}=\mathrm{O}), 1519$, 1436, 1352, 1228 (C=C, C-N), 959, 879, 857, 756, 686 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.93-0.95\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.22\left(\mathrm{~d},{ }^{3} \mathrm{~J}=7.0\right.$ $\left.\mathrm{Hz}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.47-1.52\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.62-$ $1.66\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 2.49-2.51\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, 3.64-3.70 (m, 1H, CH(CH3 $\left.)_{2}\right), 7.31-7.36(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.44-7.48(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh}$ 3,5-H), 7.68-7.72 (m, 2H, NPh 2,6-H), $8.41(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.7\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right)$, $18.8\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $19.4\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right) \quad 22.2$ $\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $30.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $37.6\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $72.6\left(C \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, 96.0
$\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, 119.7 ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 125.3 (C-4), 127.9 (NPh C-4), 129.6 (NPh C-3.5), 130.3 (C-5), 136.1 (C-3), 139.0 (NPh C-1), 199.2 (C=O). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-160.5(\mathrm{~N}-1),-72.2(\mathrm{~N}-2)$. MS m/z (\%): 295 ( $\left.[\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left(\left[\mathrm{M}+\mathrm{Na}+\mathrm{H}_{2} \mathrm{O}\right]^{+}\right)$calcd 335.1730, found 335.1723.

### 4.1.2.10.16. 2-Methyl-1-\{1-phenyl-3-[(trimethylsilyl)ethynyl]-1H-pyrazol-

 4-yl\}propan-1-one (55). The reaction mixture was stirred for 12 hours. White solid, yield $135 \mathrm{mg}, 87 \%, \mathrm{mp} 91-92^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3124,3060\left(\mathrm{CH}_{\text {arom }}\right), 2987,2971,2933,2903$ $\left(\mathrm{CH}_{\text {aliph }}\right)$, $2166(\mathrm{C} \equiv \mathrm{C}), 1665(\mathrm{C}=\mathrm{O}), 1518,1435,1354$, 1247, 1226 (C=C, C-N), 873, 848, 757, $705(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}}$ ppm 0.30 (s, $9 \mathrm{H}, \mathrm{TMS}$ ), $1.20-1.26$ ( $\left.\mathrm{m}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 3.72-3.76 (m, 1H, CH(CH3 $)_{2}$ ), 7.36-7.39 (m, 1H, Ph 4-H), 7.45-7.49 (m, 2H, Ph 3,5-H), 7.71-7.72 (m, 2H, Ph 2,6-H), $8.42(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 0.3(\mathrm{TMS}), 18.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $37.7\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $96.1\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right)$, $100.9\left(\mathrm{C} \equiv \mathrm{CC}_{4} \mathrm{H}_{9}\right), 119.9(\mathrm{Ph} \mathrm{C-2,6)} ,126.2(\mathrm{C}-4)$, 128.2 (Ph C-4), 129.7 (Ph C-3,5), 130.5 (C-5), 135.3 (C-3), 139.0 (Ph C-1), 199.2 $(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-158.9(\mathrm{~N}-1), \mathrm{N}-2$ was not found. MS $\mathrm{m} / \mathrm{z}(\%): 311\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{ONa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 333.1394, found 333.1394.4.

### 4.1.2.11. General procedure for the cyclization of 3-ethynyl-1-phenyl-1H-pyrazole-4-carbaldehydes (38-45), ethanones (46-52), and propanones (53-55).

A solution of compound $38-55(0.5 \mathrm{mmol})$ in dry ammonia and methanol $\left(\mathrm{NH}_{3} / \mathrm{MeOH} 2 \mathrm{M}, 8 \mathrm{~mL}\right)$ was heated at $120^{\circ} \mathrm{C}$ temperature for 15 hours in a steel reactor. The solvent was evaporated, and the crude was purified by column chromatography ( $\mathrm{SiO}_{2}$, eluent: ethyl acetate $/ n$-hexane, $1: 4, \mathrm{v} / \mathrm{v}$ ) to yield compounds 56-73.
4.1.2.11.1. 2-Methyl-6-phenyl-2H-pyrazolo[4,3-c]pyridine (56). White solid, yield $83 \mathrm{mg}, 79 \%, \mathrm{mp} 159-160^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\nu_{\text {max }}, \mathrm{cm}^{-1}$ ): 3106, 3062, 3030
 ( $\mathrm{CH}_{\text {arom }}$ ), 2997, $2943\left(\mathrm{CH}_{\text {aliph }}\right), 1615,1471,1367,1240,1153(\mathrm{C}=\mathrm{C}, \mathrm{C}-$ N), $926,858,831,755,688,677(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene) . ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 4.27\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.38-7.40(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}$ ), 7.47-7.50 (m, 2H, Ph 3,5-H), 7.93 (s, 1H, 7-H), 8.05-8.09 (m, 2H, Ph 2,6-H), 8.08 (s, 1H, 3-H), 9.28 (s, 1H, 4-H). ${ }^{13} \mathrm{C}$ NMR (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 40.9\left(\mathrm{CH}_{3}\right), 107.0(\mathrm{C}-7), 119.4(\mathrm{C}-3 \mathrm{a}), 125.1(\mathrm{C}-$ 3), 127.2 (Ph C-2,6), 128.4 (Ph C-4), 128.8 (Ph C-3,5), 140.3 (Ph C-1), 146.3 (C-4), 151.1 (C-6), 151.6 (C-7a). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}}$ ppm -160.6 (N-2), -97.6 (N-1), -87.7 (N-5). MS m/z (\%): $210\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 210.1027, found 210.1026.
4.1.2.11.2. 2-Phenyl-6-(thiophen-3-yl)-2H-pyrazolo[4,3-c]pyridine
 White solid, yield $130 \mathrm{mg}, 94 \%$, mp $129-130{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3132, 3113, 3064, $3041\left(\mathrm{CH}_{\text {arom }}\right), 1506,1362,1254,1202$, 1039 (C=C, C-N), 862, 803, 760, 750, 687 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.42-7.43(\mathrm{~m}, 1 \mathrm{H}$, Th 4-H), 7.47-7.49 (m, 1H, Ph 4-H), 7.56-7.59 (m, 2H, Ph 3,5-H), 7.70$7.71 \mathrm{~m}, 1 \mathrm{H}$, Th $5-\mathrm{H}), 7.91(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.92-7.94(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H})$, $8.00-8.01(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Th} 2-\mathrm{H}), 8.58\left(\mathrm{~d},{ }^{4} J(3-\mathrm{H}, 4-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}\right), 9.28$ $\left(\mathrm{d},{ }^{4} J(4-\mathrm{H}, 3-\mathrm{H})=1.2 \mathrm{~Hz} 1 \mathrm{H}, 4-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm}$ 106.4 (C-7), 119.8 (C-3a), 121.4 (Ph C-2,6), 122.1 (C-3), 123.3 (Th C2), 126.2 (Th C-5), 126.4 (Th C-4), 129.0 (Ph C-4), 130.0 (Ph C-3,5), 140.1 (Ph C-1), 142.7 (Th C-3), 147.4 (C-4), 147.6 (C-6), 151.7 (C-7a). ${ }^{15}$ N NMR (71 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-145.1(\mathrm{~N}-2),-99.7(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 278\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{~S}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 278.0746, found 278.0747.
4.1.2.11.3. 6-Cyclopropyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (59). White
 solid, yield $115 \mathrm{mg}, 98 \%, \mathrm{mp} 143-144^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3116,3090,3036,3006\left(\mathrm{CH}_{\text {arom }}\right), 2923,2850\left(\mathrm{CH}_{\text {aliph }}\right), 1597,1497$, 1377, 1318, 1201, 1053, 1038 (C=C, C-N), 769, 761, 691 (CH=CH of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 0.99-$ $1.01\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 1.05-1.07\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 2.14-2.18$ (m, 2H, $\left.\mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 7.40(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.42-7.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.52-7.54 (m, 2H, Ph 3,5-H), 7.87-7.89 (m, 2H, Ph 2,6-H), $8.52(\mathrm{~s}, 1 \mathrm{H}$, $3-\mathrm{H}), 9.12(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 8.9$ $\left(\mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 17.5\left(\mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 106.6(\mathrm{C}-7), 119.3(\mathrm{C}-3 \mathrm{a}), 121.4(\mathrm{Ph}$ C-2,6), 122.0 (C-3), 128.8 (Ph C-4), 129.8 (Ph C-3,5), 140.1 ( $\mathrm{Ph} \mathrm{C}-1$ ), 146.9 (C-4), 151.6 (C-7a), 155.7 (C-6). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-145.3$ (N-2), -91.0 (N-5), N-1 was not found. MS m/z (\%): 236 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 236.1182, found 236.1184.
4.1.2.11.4. 6-Butyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (60). Brown amorphous solid, yield $105 \mathrm{mg}, 84 \%$. IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3138,3013\left(\mathrm{CH}_{\text {arom }}\right)$, 2949, 2930, $2867\left(\mathrm{CH}_{\text {aliph }}\right), 1507,1468,1372,1320,1208,1053,1037$ $(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 763,751,689\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.95-0.97\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.40-$ $1.45\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.76-1.80\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 2.88-$ $2.90\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 7.40(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.43-7.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.53-7.55 (m, 2H, Ph 3,5-H), 7.88-7.89 (m, 2H, Ph 2,6-H), 8.52 (s, 1H, $3-\mathrm{H}), 9.18(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.1$ $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right)$, $22.6\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 32.0\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 38.3\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 108.3(\mathrm{C}-7)$, 119.2 (C-3a), 121.4 (Ph C-2,6), 121.8 (C-3), 128.8 (Ph C-4), 129.9 (Ph C-3,5), 140.2 (Ph C-1), $147.0(\mathrm{C}-4), 151.7$ (C-7a), $155.7(\mathrm{C}-6) .{ }^{15} \mathrm{~N} \mathrm{NMR}\left(71 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}$ -145.5 ( $\mathrm{N}-2$ ), -78.9 (N-5), $\mathrm{N}-1$ was not found. MS m/z (\%): 252 ([M+H], 100 ). HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 252.1495, found 252.1497.
4.1.2.11.5. 6-Pentyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (61). Brown amor-
 phous solid, yield $118 \mathrm{mg}, 89 \%$. IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3122, 3063, 3040 $\left(\mathrm{CH}_{\text {arom }}\right), 2949,2925,2856\left(\mathrm{CH}_{\text {aliph }}\right), 1466,1324,1206,1037$ (C=C, CN), 862, 759, 750, 684 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.89-0.90\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right), 1.37-1.39(\mathrm{~m}$, $\left.4 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{3}\right), 1.78-1.81\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} H_{7}\right), 2.86-2.89(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{3}$ ), $7.40(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.43-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.52-7.55$ (m, 2H, Ph 3,5-H), 7.88-7.89 (m, 2H, Ph 2,6-H), 8.52 ( $\mathrm{s}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 9.18 (s, 1H, 4-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm} 14.2\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right)$, $22.7\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $29.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 31.7\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 38.6\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{11}\right)$, 108.3 (C-7), 119.1 (C-3a), 121.4 (Ph C-2,6), 121.8 (C-3), 128.8 (Ph C-4), 129.8 (Ph $\mathrm{C}-3,5), 140.2$ (Ph C-1), 146.9 (C-6), 151.7 (C-4), 155.8 (C-7a). ${ }^{15} \mathrm{~N}$ NMR ( 71 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-145.3(\mathrm{~N}-2),-79.8(\mathrm{~N}-5), \mathrm{N}-1$ was not found. $\mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 266$ ( $\left.[\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 266.1652, found 266.1652.
4.1.2.11.6. 2-(2-Phenyl-2H-pyrazolo[4,3-c]pyridin-6-yl)ethan-1-ol (62).
$\mathrm{HOC}_{2} \mathrm{H}_{4}$

Brown solid, yield $86 \%, 103 \mathrm{mg}, \mathrm{mp} 215-216^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3187(\mathrm{OH}), 3108,3067,3022\left(\mathrm{CH}_{\text {arom }}\right), 2961,2935$, $2916\left(\mathrm{CH}_{\text {aliph }}\right), 1503,1325,1227,1213,1056(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 765,752$, 677 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 3.11-3.12\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 4.05-4.06(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ ), 7.45-7.48 (m, 2H, H-7, Ph 4-H), 7.55-7.58 (m, 2H, Ph 3,5-H), 7.89-7.90 (m, 2H, Ph 2,6-H), 8.58 (s, 1H, 3-H), 9.16 (s, $1 \mathrm{H}, 4-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 39.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$, $62.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 109.5(\mathrm{C}-7), 119.4(\mathrm{C}-3 \mathrm{a}), 121.5$ (Ph C-2,6), 122.3 (C-3), 129.1 (Ph C-4), 129.9 (Ph C-3,5), 140.0 (Ph C-1), 151.3 (C-4), 153.6 (C-7a). ${ }^{15}$ N NMR (71 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-144.3$ (N-2), -100.5 (N-1), -88.3 (N-5). MS m/z (\%): 240 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{O}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 240.1131, found 240.1131 .
4.1.2.11.7. 2-Phenyl-2H-pyrazolo[4,3-c]pyridine (63). White solid, yield 193 $\mathrm{mg}, 95 \%, \mathrm{mp} 132^{\circ} \mathrm{C}$ (ethyl acetate),(lit mp 132-133 ${ }^{\circ} \mathrm{C}^{10}$ ). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right)$ : $3129,3094,3064\left(\mathrm{CH}_{\text {arom }}\right), 1614,1490,1466,1365,1349,1315,1210,1178$ (C=C, C-N), 916, 827, 760, 750, 684, 587, 508, 431 (CH=CH of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm} 7.45-7.48(\mathrm{~m}, 1 \mathrm{H}$, Ph 4-H), 7.54-7.56 (m, 2H, Ph 3,5-H), 7.62-7.64 (m, 1H, 7-H), 7.90-7.91 (m, 2H, Ph 2,6-H), 8.31-8.32 (m, 1H, 6-H), $8.63(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}), 9.27(\mathrm{~s}, 1 \mathrm{H}$, 4-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm} 111.8$ (C-7), 120.5 (C-3a), 121.6 (Ph C-2,6), 122.4 (C-3), 129.1 (Ph C-4), 129.9 (Ph C-3,5), 140.0 (Ph C-1), 142.1 (C-6), 147.5 (C-4), $150.4(\mathrm{C}-7 \mathrm{a}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.4(\mathrm{~N}-2)$, $-99.6(\mathrm{~N}-1),-89.8(\mathrm{~N}-5) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 196\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{3}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 196.0869, found 196.0869.
4.1.2.11.8. 4-Methyl-2-phenyl-6-(thiophen-3-yl)-2H-pyrazolo[4,3-c]pyridine (65). White solid, yield $118 \mathrm{mg}, 81 \%, \mathrm{mp} 92-93{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3129,3104,3072\left(\mathrm{CH}_{\text {arom }}\right), 2982$, 2948, $2911\left(\mathrm{CH}_{\text {aliph }}\right), 1546$, 1507, 1374, 1202, 1045 (C=C, C-N), 849, 796, 763, 744, 690 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.39-7.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Th} 4-\mathrm{H})$, 7.43-7.45 (m, 1H, Ph 4-H), 7.53-7.55 (m, 2H, Ph 3,5-H), 7.677.68 (m, 1H, Th 5-H), 7.72 (s, 1H, 7-H), 7.90-7.91 (m, 2H, Ph 2,6H), 8.01-8.02 (m, 1H, Th 2-H), 8.49 (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( 176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 23.3\left(\mathrm{CH}_{3}\right), 104.3(\mathrm{C}-7), 119.7(\mathrm{C}-3 \mathrm{a})$, 121.2 (Ph C-2,6), 121.9 (C-3), 123.2 (Th C-2), 126.2 (Th C-4,5), 128.7 (Ph C-4), 129.8 (Ph C-3,5), 140.1 (Ph C-1), 142.8 (Th C-3), 147.3 (C-4), $152.0(\mathrm{C}-7 \mathrm{a}), 156.2(\mathrm{C}-6) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.4$ (N-2), -99.6 (N-1), -89.8 (N-5). MS m/z (\%): 292 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{~S}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 292.0903, found 292.0903.

### 4.1.2.11.9. 6-Cyclopropyl-4-methyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine

(66). White solid, yield $99 \mathrm{mg}, 80 \%, \mathrm{mp} 85-86^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\mathrm{v}_{\text {max }}, \mathrm{cm}^{-1}$ ):
 3129, 3078, $3044\left(\mathrm{CH}_{\text {arom }}\right), 2996,2911,2850\left(\mathrm{CH}_{\text {aliph }}\right), 1500,1401$, 1300, 1202, 1071, 1041 (C=C, C-N), 821, 760, 750, 685, 538 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.95-0.98\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 0.99-1.01(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CHCH}_{2}$ ), 2.14-2.16 (m, 2H, CH $\mathrm{CHCH}_{2}$ ), $2.75\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 7.16 (s, 1H, 7-H), 7.40-7.42 (m, 1H, Ph 4-H), 7.50-7.53 (m, 2H, Ph 3,5-H), 7.86-7.87 (m, 2H, Ph 2,6-H), 8.44 (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm} 8.7\left(\mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 17.6$ $\left(\mathrm{CH}_{2} \mathrm{CHCH}_{2}\right), 23.0\left(\mathrm{CH}_{3}\right), 103.6(\mathrm{C}-7), 119.4(\mathrm{C}-3 \mathrm{a}), 121.2(\mathrm{Ph} \mathrm{C-2,6)}, 121.7(\mathrm{C}-3)$, 128.5 (Ph C-4), 129.8 (Ph C-3,5), 140.2 (Ph C-1), 152.0 (C-7a), 155.6 (C-4), 155.9 (C-6). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.9(\mathrm{~N}-2),-90.4(\mathrm{~N}-5), \mathrm{N}-1$ was not found. MS m/z (\%): $250\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 250.1339, found 250.1337 .
4.1.2.11.10. 6-Butyl-4-methyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (67).


Brown liquid, yield $116 \mathrm{mg}, 88 \%$. IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): $3070\left(\mathrm{CH}_{\text {arom }}\right)$, 2955, 2928, $2870\left(\mathrm{CH}_{\text {aliph }}\right), 1618,1597,1544,1510,1374,1165$, 1043 (C=C, C-N), 758, 688 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.94-0.97(\mathrm{~m}, 3 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.40-1.45 (m, 2H, C2 $\mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.71-1.79 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), $2.81\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.81-2.86\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, 7.24 (s, 1H, 7-H), 7.42-7.44 (m, 1H, H-4), 7.52-7.54 (m, 2H, Ph $3,5-\mathrm{H}), 7.88-7.89(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.49(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.2\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right)$, $22.6\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 23.1\left(\mathrm{CH}_{3}\right), 32.1$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 38.3\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 106.0(\mathrm{C}-7), 119.2(\mathrm{C}-3 \mathrm{a}), 121.3(\mathrm{Ph} \mathrm{C}-2,6), 121.8$ (C-3), 128.6 (Ph C-4), 129.8 (Ph C-3.5), 140.2 (Ph C-1), 152.0 (C-7a), 155.5 (C-4), 155.6 (C-6). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.6$ (N-2), $-85.2(\mathrm{~N}-5), \mathrm{N}-1$ was
not found. MS m/z (\%): $266\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$ calcd 266.1652, found 266.1654.
4.1.2.11.11. 4-Methyl-6-pentyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (68).
 Brown liquid, yield $123 \mathrm{mg}, 88 \%$. IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3100, 3071 $\left(\mathrm{CH}_{\text {arom }}\right), 2954,2928,2870,2857\left(\mathrm{CH}_{\text {aliph }}\right), 1619,1597,1544$, $1510,1374,1309,1043(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 758,688(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.82-$ $0.91\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{4} \mathrm{H}_{11} \mathrm{CH}_{3}\right), 1.35-1.41\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{3}\right), 1.76-$ $1.80\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 2.82\left(\mathrm{CH}_{3}\right), 2.84-2.86(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{9}$ ), 7.24 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-7$ ), $7.42-7.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.53-$ 7.55 (m, 2H, Ph 3,5-H), 7.88-7.89 (m, 2H, Ph 2,6-H), $8.50(\mathrm{~s}, 1 \mathrm{H}$, 3-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.2\left(\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{CH}_{3}\right), 22.8\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $23.0\left(\mathrm{CH}_{3}\right), 29.6\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 31.7\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 38.4\left(\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{11}\right)$, $106.1(\mathrm{C}-$ 7), 119.2 (C-3a), 121.3 (Ph C-2,6), 121.9 (C-3), 128.7 (Ph C-4), 129.8 (Ph C-3,5), 140.2 (Ph C-1), 152.0 (C-7a), 155.4 (C-4), 155.6 (C-6). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-148.9(\mathrm{~N}-2),-85.0(\mathrm{~N}-5), \mathrm{N}-1$ was not found. MS m/z (\%): $280\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, 100). HRMS (ESI) for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 280.1808, found 280.1808 .
4.1.2.11.12. 2-(4-Methyl-2-phenyl-2H-pyrazolo[4,3-c]pyridin-6-yl)ethan-1ol (69). White solid, yield $109 \mathrm{mg}, 86 \%$, mp $126-127^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$
 $\left.{ }^{1}\right): 3227(\mathrm{OH}), 3149,3072,3018\left(\mathrm{CH}_{\text {arom }}\right), 2945,2919,2870$ $\left(\mathrm{CH}_{\text {aliph }}\right), 1622,1501,1389,1374,1306,1212,1164,1053$ (C=C, C-N), 844, 756, 741, 682, $571(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 2.74$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $3.03\left(\mathrm{t},{ }^{3} \mathrm{~J}=5.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 4.00-4.02(\mathrm{~m}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 7.22(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.40-7.42(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-7, \mathrm{Ph}$ 4-H), 7.49-7.51 (m, 2H, Ph 3,5-H), 7.84-7.85 (m, 2H, Ph 2,6$\mathrm{H}), 8.48(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm}$ $23.0\left(\mathrm{CH}_{3}\right), 39.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right) 62.7\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right), 107.0(\mathrm{C}-7), 119.2(\mathrm{C}-3 \mathrm{a}), 121.2$ (Ph C-2,6), 122.0 (C-3), 128.7 ( $\mathrm{Ph} \mathrm{C}-4$ ), 129.8 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 140.0 ( $\mathrm{Ph} \mathrm{C}-1$ ), 151.5 (C7a), 153.4 (C-6), 155.7 (C-4). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-146.8(\mathrm{~N}-2)$, -92.3 (N-5), N-1 was not found. MS m/z (\%): 254 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{3} \mathrm{O}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 254.1288, found 254.1287.
4.1.2.11.13. 4-Methyl-2-phenyl-2H-pyrazolo[4,3-c]pyridine (70). Brown
 amorphous solid, yield $92 \mathrm{mg}, 88 \%$. IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3066\left(\mathrm{CH}_{\text {arom }}\right)$, $2995\left(\mathrm{CH}_{\text {aliph }}\right), 1596,1498,1416,1370,1345,1237,1190,1031(\mathrm{C}=\mathrm{C}$, C-N), 807, 756, 746, 687, 638, $529(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.85$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 7.46-7.48 (m, 2H, 7-H, Ph 4-H), 7.55-7.58 (m, 2H, Ph 3,5-H), 7.91$7.93(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.20\left(\mathrm{~d},{ }^{3} J(6-\mathrm{H}, 7-\mathrm{H})=6.4 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right), 8.58$ (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 23.0\left(\mathrm{CH}_{3}\right), 109.5$ (C-7), 120.6 (C-3a), 121.5 (Ph C-2,6), 122.0 (C-3), 128.9 ( $\mathrm{Ph} \mathrm{C}-4$ ), 129.9 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 140.1 (Ph C-1), 142.1 (C-6), 150.9 (C-4), 156.4 (C-7a). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ):
$\delta_{\mathrm{N}} \mathrm{ppm}-146.9(\mathrm{~N}-2),-98.48(\mathrm{~N}-1),-91.0(\mathrm{~N}-5) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 210\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 210.1026, found 210.1026.
4.1.2.11.14. 2,6-Diphenyl-4-(propan-2-yl)-2H-pyrazolo[4,3-c]pyridine (71).
 White solid, yield $143 \mathrm{mg}, 92 \%, \mathrm{mp} 97-98^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3126,3064,3023\left(\mathrm{CH}_{\text {arom }}\right), 2960,2924,2898$ $\left(\mathrm{CH}_{\text {aliph }}\right), 1600,1510,1466,1399,1314,1277,1203,1046$ (C=C, C-N), 767, 760, 756, 696, 684, $635(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.54$ (d, ${ }^{3} J=7.0$ $\left.\mathrm{Hz}, 6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.52-3.56\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 7.38-7.40(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{CPh} 4-\mathrm{H}), 7.44-7.47(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.48-7.52(\mathrm{~m}, 2 \mathrm{H}$, CPh 3,5-H), $7.55-7.58$ (m, 2H, NPh 3,5-H), 7.89 (s, 1H, 7-H), 7.93-7.96 (m, 2H, NPh 2,6-H), 8.18-8.19 (m, 2H, CPh 2,6-H), 8.57 (s, 1H, 3-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.9\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 35.8\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 104.5(\mathrm{C}-7)$, 118.0 (C-3), 121.2 (NPh C-2,6), 127.0 (CPh C-2,6), 128.2 (CPh C-4), 128.56 (CPh C3,5), 128.62 (NPh C-4), 129.7 ( $\mathrm{NPh} \mathrm{C}-3,5$ ), 140.1 (CPh C-1), 140.4 ( $\mathrm{NPh} \mathrm{C-1)}$, (C-6), $152.8(\mathrm{C}-7 \mathrm{a}), 164.1(\mathrm{C}-4) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-148.9(\mathrm{~N}-2)$, $-86.5(\mathrm{~N}-1),-90.8(\mathrm{~N}-5) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 314\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{3}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 314.1652, found 314.1652.
4.1.2.11.15. 6-Butyl-2-phenyl-4-(propan-2-yl)-2H-pyrazolo[4,3-c]pyridine
(72). Brown liquid, yield $123 \mathrm{mg}, 84 \%$. IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3111,3065,3047\left(\mathrm{CH}_{\text {arom }}\right)$,
 2959, 2929 ( $\mathrm{CH}_{\text {aliph }}$ ), 1615, 1598, 1541, 1510, 1467, 1376, 1306, 1275, 1211, 1196, 1161, 1044 (C=C, C-N), 848, 757, 688 ( $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.95-0.97\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right.$ ), 1.40-1.44 (m, $2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.44-1.47 (m, 6H, CH( $\left.\mathrm{CH}_{3}\right)_{2}$ ), 1.75-1.80 (m, 2H, CH ${ }_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.85-2.90 (m, 2H, $\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), 3.41$3.45\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 7.23(\mathrm{~s}, 1 \mathrm{H}, 7-\mathrm{H}), 7.42-7.46(\mathrm{~m}, 1 \mathrm{H}$, H-7), 7.53-7.54 (m, 2H, Ph 3,5-H), 7.89-7.90 (m, 2H, Ph 2,6H ), $8.53(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 14.2\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.1$ $\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 22.6\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 31.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 36.3\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 38.3$ $\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 105.8$ (C-7), 116.9 (C-3a), 121.3 (C-3), 121.4 ( $\mathrm{Ph} \mathrm{C-2,6)}$,128.5 (Ph C-4), 129.8 (Ph C-3,5), 140.3 (Ph C-1), 152.8 (C-7a), 155.4 (C-6), 164.1 (C-4). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.4(\mathrm{~N}-2),-99.6(\mathrm{~N}-1),-90.8(\mathrm{~N}-5) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 294$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 294.1965, found 294.1965.
4.1.2.11.16. 2-Diphenyl-4-(propan-2-yl)-2H-pyrazolo[4,3-c]pyridine (73).
 Brown liquid, yield $109 \mathrm{mg}, 92 \%$. IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): $3061\left(\mathrm{CH}_{\text {arom }}\right)$, 2966, 2927, $2869\left(\mathrm{CH}_{\text {aliph }}\right), 1609,1510,1498,1369,1276,1238$, 1193, 1026 (C=C, C-N), 806, 757, 688 (CH=CH of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.47\left(\mathrm{~d},{ }^{3} \mathrm{~J}=7.1 \mathrm{~Hz}\right.$, $\left.6 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.46-3.50\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 7.44-7.47(\mathrm{~m}, 2 \mathrm{H}$, 7-H, Ph 4-H,), 7.54-7.56 (m, 2H, Ph 3,5-H), 7.90-7.92 (m, 2H, Ph $2,6-\mathrm{H}), 8.25\left(\mathrm{~d},{ }^{3} \mathrm{~J}(6-\mathrm{H}, 7-\mathrm{H})=6.4 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right), 8.60(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \operatorname{ppm} 21.8\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 35.4\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 109.4(\mathrm{C}-$
7), 118.7 (C-3a), 121.4 (Ph C-2,6), 121.5 (C-3), 128.7 (Ph C-4), 129.7 (Ph C-3,5), 140.0 (C-6), 142.1 (Ph C-1), 151.2 (C-7a), $165.0(\mathrm{C}-4) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-147.4(\mathrm{~N}-2),-100.6(\mathrm{~N}-1),-95.3(\mathrm{~N}-5) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 238\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{3}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 238.1339, found 238.1339.

### 4.1.2.12. General procedure for the preparation of $\boldsymbol{o}$-acylated-1-phenyl-1Hpyrazoles by using aroyl chlorides.

Into the solution of corresponding 3-hydroxy-1H-pyrazole 1 or $23(0.5 \mathrm{mmol})$ in chloroform ( 5 mL ), appropriate aroyl chloride $(0,5 \mathrm{mmol})$, triethylamine ( 1 mmol ) were added. The mixture was stirred for 30 min at room temperature; later, it was diluted with water $(10 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 10 \mathrm{~mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 7$, v/v) to provide compounds $\mathbf{7 5}, 77,79,81$, 84-85.
4.1.2.12.1. 1-Phenyl-1H-pyrazol-3-yl 2-chlorobenzoate (75). White solid,
 yield $129 \mathrm{mg}, 86 \%, \operatorname{mp} 52-53^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3153, $3132\left(\mathrm{CH}_{\text {arom }}\right), 1756(\mathrm{C}=\mathrm{O}), 1599,1535,1457,13891236$, 1168 (C-O-C, $\mathrm{C}=\mathrm{C}, \mathrm{C}=\mathrm{N}), 756,746,703,680(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ mono-, and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.56\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.3 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.28-7.30(\mathrm{~m}, 1 \mathrm{H}$, NPh 4-H), 7.39-7.42 (m, 1H, CPh 3-H), 7.44-7.46 (m, 2H, NPh 3,5-H), 7.50-7.54 (m, 2H, CPh 4,5-H), 7.66-7.67 (m, 2H, NPh $2,6-\mathrm{H}), 7.91\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.3 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 8.16\left(\mathrm{~d},{ }^{3} J(\mathrm{CPh} 6-\right.$ $\mathrm{H}, 5-\mathrm{H})=7.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.2$ (C-4), 119.0 (NPh C-2,6), 126.7, 126.8 (CPh C-5, NPh C-4), 128.0 (C-5), 129.6 (NPh C3,5), 131.6 (CPh C-3), 132.5 (CPh C-4), 133.7 (CPh C-6), 135.1 (CPh C-2), 139.9 (NPh C-1), $156.2(\mathrm{C}-3), 162.2(\mathrm{C}=\mathrm{O}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 299\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{ClO}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 321.0401, found 321.0402
4.1.2.12.2. 1-Phenyl-1H-pyrazol-3-yl 2,4-dichlorobenzoate (77). White solid,
 yield $135 \mathrm{mg}, 81 \%, \operatorname{mp} 85-86^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3177,3135,3095,3067,3029\left(\mathrm{CH}_{\text {arom }}\right), 1752(\mathrm{C}=\mathrm{O}), 1601$, $1585,1538,1474,1456,1389,1278,1264,1233,1096,1057$, 1046, 1005 (C-O-C, C=C, C-N), 865, 818, 760, 747, 683, 672, $476\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.53\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.5 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 4-\mathrm{H}), 7.27-7.29(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.37$ (d, ${ }^{3} \mathrm{~J}(\mathrm{CPh} 5-\mathrm{H}, 6-$ $\mathrm{H})=9.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 7.43-7.45(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.54$ (s, 1H, CPh 3-H), 7.64-7.66 (m, 2H, NPh 2,6-H), $7.90\left(\mathrm{~d},{ }^{3} J(5-\right.$ $\mathrm{H}, 4-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.12\left(\mathrm{~d},{ }^{3} J(\mathrm{CPh} 6-\mathrm{H}, 5-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 6-\mathrm{H}\right) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.1(\mathrm{C}-4), 118.9$ (NPh C-2,6), 126.4 (CPh C-1), 126.7 (NPh C-4), 127.3 (CPh C-5), 128.0 (C-5), 131.5 (CPh C-3), 133.5 (CPh C-6), 136.3 (CPh C-2), 139.6 (CPh C-4), 139.7 (NPh C-1), 156.3 (C-3), 161.3 (C=O). ${ }^{15} \mathrm{~N}$

NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.6(\mathrm{~N}-2),-103.9(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 332\left([\mathrm{M}]^{+}\right.$, 100), 334 ([M+2] ${ }^{+}$, 60). HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{C}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 355,0012, found 355,0011 .
4.1.2.12.3. 1-Phenyl-1H-pyrazol-3-yl 2-chloro-4-fluorobenzoate (79). White
 solid, yield $126 \mathrm{mg}, 80 \%, \mathrm{mp} 83-84^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3069,3044\left(\mathrm{CH}_{\text {arom }}\right), 1750(\mathrm{C}=\mathrm{O}), 1600,1577,1489$, 1456, 1387, 1298, 1232, 1004 (C-O-C, C=C, C-N, C-F), 913, $877,765,752,685,600(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ $6.54\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.10-7.13(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh}$ 5-H), 7.26-7.31 (m, 2H, NPh 4-H, CPh, 3-H), 7.44-7.46 (m, 2H, NPh 3,5-H), 7.65-7.67 (m, 2H, NPh 2,6-H), 7.90 (d, ${ }^{3} \mathrm{~J}(5-\mathrm{H}, 4-$ $\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.24\left(\mathrm{dd},{ }^{3} J(\mathrm{CPh} 6-\mathrm{H}, 5-\mathrm{H})=8.8 \mathrm{~Hz},{ }^{4} J(\mathrm{CPh}\right.$ $6-\mathrm{H}, 3-\mathrm{H})=6.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.2$ (C-4), $114.39,114.51\left({ }^{2} J=21.5 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-5\right), 119.0$ ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 119.19 , 119.33 ( ${ }^{2} J=25.1$ $\mathrm{Hz}, \mathrm{CPh} \mathrm{C}-3$ ), $124.28,124.30\left({ }^{4} \mathrm{~J}=3.3 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-1\right), 126.8$ (NPh C-4), 128.0 (C-5), 129.6 (NPh C-3,5), $134.69,134.74\left({ }^{3} \mathrm{~J}=10.0 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-6\right)$, 137.31, 137.38 ( ${ }^{3} \mathrm{~J}=10.9$ $\mathrm{Hz}, \mathrm{CPh} \mathrm{C}-2), 139.8$ (NPh C-1), 156.4 (C-3), 161.23 (C=O), 164.15, $165.62\left({ }^{1} \mathrm{~J}=258.7\right.$ $\mathrm{Hz}, \mathrm{CPh} \mathrm{C}-4) .{ }^{15} \mathrm{~N}$ NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.3(\mathrm{~N}-2),-104.0(\mathrm{~N}-1) . \mathrm{MS}$ $\mathrm{m} / \mathrm{z}(\%): 317\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{ClFN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 339.0307, found 339.0307.
4.1.2.12.4. 1-Phenyl-1H-pyrazol-3-yl 2-chloropyridine-3-carboxylate (82).
 Yellow solid, yield $123 \mathrm{mg}, 82 \%, \mathrm{mp} 105-106^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3059\left(\mathrm{CH}_{\text {arom }}\right), 1764(\mathrm{C}=\mathrm{O}), 1596,1587,1537$, $1464,1410,1276,1249,1219,1079,1074,1133,1001(\mathrm{C}=\mathrm{C}, \mathrm{C}-$ $\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 782,766,757,724,618,523,489,421(\mathrm{C}-\mathrm{Cl}$, $\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \operatorname{ppm} 6.49\left(\mathrm{~d},{ }^{3} \mathrm{~J}(4-\mathrm{H}, 5-\mathrm{H})=2.7 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.21-$ $7.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.34\left(\mathrm{dd},{ }^{3} J(\mathrm{Pyr} 5-\mathrm{H}, 4-\mathrm{H})=7.8 \mathrm{~Hz},{ }^{3} J(\mathrm{Pyr}\right.$ $5-\mathrm{H}, 6-\mathrm{H})=4.8 \mathrm{~Hz}, 1 \mathrm{H}$, Pyr $5-\mathrm{H}), 7.37-7.39(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H})$, $7.59-7.60(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.87\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.7 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $5-\mathrm{H}), 8.40\left({ }^{3} J(\operatorname{Pyr} 6-\mathrm{H}, 5-\mathrm{H})=7.7 \mathrm{~Hz},{ }^{4} J(\operatorname{Pyr} 6-\mathrm{H}, 4-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \operatorname{Pyr} 6-\mathrm{H}\right), 8.52$ $\left({ }^{3} J(\right.$ Pyr $4-\mathrm{H}, 5-\mathrm{H})=4.8 \mathrm{~Hz},{ }^{4} J(\mathrm{Pyr} 4-\mathrm{H}, 6-\mathrm{H})=4.8 \mathrm{~Hz}, 1 \mathrm{H}$, Pyr $\left.4-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR (176 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 98.9$ (C-4), 118.6 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 122.2 (Pyr C-5), 124.9 (Pyr C1), 126.6 ( $\mathrm{Ph} \mathrm{C}-4$ ), 128.0 (C-5), 129.4 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 139.5 (Ph C-1), 141.1 (Pyr C-6), 150.9 (Pyr C-2), 152.7 (Pyr C-4), 155.9 (C-3), 160.9 (C=O). MS m/z (\%): 300 $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{10} \mathrm{ClN}_{3} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 322.0354, found 322.0353 .
4.1.2.12.5. 1-Phenyl-1H-pyrazol-3-yl 3-chlorothiophene-2-carboxylate (83). White solid, yield $136 \mathrm{mg}, 89 \%$, $\mathrm{mp} 132-133{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3078 ( $\mathrm{CH}_{\text {arom }}$ ), 1739 ( $\mathrm{C}=\mathrm{O}$ ), 1513, 1458, 1414, 1393, 1361, 1264, 1248, 1066 (C=C, C-N, $\mathrm{C}-\mathrm{O}-\mathrm{C}), 913,759,750,683,639,609,501(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of monosubstituted

benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.51$ (d, ${ }^{3} J(4-\mathrm{H}, 5-$ $\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}), 7.10\left(\mathrm{~d},{ }^{3} \mathrm{~J}(\mathrm{Th} 4-\mathrm{H}, 5-\mathrm{H})=5.3 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Th 4H), 7.28-7.30 (m, 1H, NPh 4-H), 7.44-7.7.46 (m, 2H, NPh 3,5-H), $7.61\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=5.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Th} 5-\mathrm{H}\right), 7.65-7.67(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh}$ $2,6-\mathrm{H}), 7.89\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR (176 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.2$ (C-4), 119.0 ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 123.9 (Th C1), 126.7 ( $\mathrm{Ph} \mathrm{C}-4$ ), 128.0 (C-5), 129.6 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 130.7 ( $\mathrm{Th} \mathrm{C}-3$ ), 132.2 (Th C-4), 134.2 (Th C-2), 156.1 (C-3), 157.6 (C=O). ${ }^{15} \mathrm{~N}$ NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.6(\mathrm{~N}-2),-103.6(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 304$ ( $[\mathrm{M}]^{+}$, 100), 306 ( $[\mathrm{M}+2]^{+}, 30$ ). HRMS (ESI) for $\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{SNa}\left([\mathrm{M}+\mathrm{Na}]^{+}\right) 326.9965$, found 326.9966 .
4.1.2.12.6. 1-Methyl-1H-pyrazol-3-yl 2-chlorobenzoate (84). Brown liquid,
 yield $100 \mathrm{mg}, 85 \%$. IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3126\left(\mathrm{CH}_{\text {arom }}\right), 2944\left(\mathrm{CH}_{\text {aliph }}\right)$, 1758 ( $\mathrm{C}=\mathrm{O}$ ), 1591, 1528, 1479, 1463, 1432, 1406, 1282, 1243, 1101, 1033 (C-O-C, C=C, C-N), 865, 745, 714, 657, 615, 475 $\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.83\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.23\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\right.$ $\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}), 7.30\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.4 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right)$, 7.33-7.36 (m, 1H, Ph 3-H), 7.45-7.49 (m, 2H, Ph 4,5-H), 8.07 $\left(\mathrm{dd},{ }^{3} J(6-\mathrm{H}, 5-\mathrm{H})=7.8 \mathrm{~Hz},{ }^{3} J(6-\mathrm{H}, 4-\mathrm{H})=1.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ph} 6-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 39.4\left(\mathrm{CH}_{3}\right), 96.6(\mathrm{C}-4), 126.8(\mathrm{Ph} \mathrm{C-3}), 128.4(\mathrm{Ph}$ C-1), 131.4 (C-5), 131.5 (Ph C-4), 133.5 (Ph C-6), 132.3 ( $\mathrm{Ph} \mathrm{C}-5$ ), 134.8 ( $\mathrm{Ph} \mathrm{C}-2$ ), $154.6(\mathrm{C}-3), 162.4(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-196.6(\mathrm{~N}-2),-98.7$ (N-1). MS m/z (\%): $237\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$ calcd 259.0245, found 259.0245.
4.1.2.12.7. 1-Methyl-1H-pyrazol-3-yl 2,4-dichlorobenzoate (86). Brown
 solid, yield $98 \mathrm{mg}, 72 \%, \mathrm{mp} 72-73{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3120,3099,3089\left(\mathrm{CH}_{\text {arom }}\right)$, 2984, $2949\left(\mathrm{CH}_{\text {aliph }}\right), 1759$ (C=O), 1650, 1585, 1525, 1477, 1465, 1436, 1408, 1374, 1289, 1276, 1236, 1214, 1156, 1093, 1030 (C-O-C, C=C, C-N), 869, 837, 762, 741, 677, 654, 537, 488, $468(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $(700 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 3.85\left(\mathrm{~s}, \mathrm{CH}_{3}\right), 6.24\left(\mathrm{~d},{ }^{3} \mathrm{~J}(4-\mathrm{H}, 5-\mathrm{H})=2.3 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 4-\mathrm{H}), 7.31\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.2 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 7.35(\mathrm{dd}$, $\left.{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(5-\mathrm{H}, 3-\mathrm{H})=1.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ph} 5-\mathrm{H}\right), 7.52(\mathrm{~d}$, $\left.{ }^{4} J(3-\mathrm{H}, 5-\mathrm{H})=1.9 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}\right), 8.06\left(\mathrm{~d},{ }^{3} J(6-\mathrm{H}, 5-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 39.5\left(\mathrm{CH}_{3}\right), 96.6(\mathrm{C}-4), 126.7(\mathrm{Ph} \mathrm{C}-1), 127.3(\mathrm{Ph} \mathrm{C-5}), 131.4$ (C-5), 131.5 (Ph C-3), 133.4 (Ph C-6), 136.1 (Ph C-2), 139.5 (Ph C-4), 154.5 (C-3), $161.6(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-196.3(\mathrm{~N}-2),-98.8(\mathrm{~N}-1) . \mathrm{MS}$ $\mathrm{m} / \mathrm{z}(\%): 271\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 271.0036, found 271.0035 .
4.1.2.12.8. 1-Methyl-1H-pyrazol-3-yl 2-chloro-4-fluorobenzoate (88).
 Brown solid, yield $107 \mathrm{mg}, 84 \%$, $\mathrm{mp} 87-88^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3062,3038,3002\left(\mathrm{CH}_{\text {arom }}\right), 2983,2956\left(\mathrm{CH}_{\text {aliph }}\right)$, 1745 (C=O), 1599, 1577, 1524, 1491, 1468, 1432, 1391, 1298, 1234, 1208, 1094, 1032 (C-O-C, C=C, C-N, C-F), 901, 832, $772,760,683,597,484(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ $3.84\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.22-6.23(\mathrm{~m}, 1 \mathrm{H}, 4-\mathrm{H}), 7.06-7.09(\mathrm{~m}, 1 \mathrm{H}$, Ph $3-\mathrm{H}), 7.24\left(\mathrm{dd},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(5-\mathrm{H}, 3-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Ph 5-H), $7.31\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.2 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 8.14-8.17(\mathrm{~m}$, $1 \mathrm{H}, 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 39.4\left(\mathrm{CH}_{3}\right), 96.6(\mathrm{C}-4), 114.3,114.4$ $\left({ }^{2} J=21.5 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-5\right), 119.0,119.2\left({ }^{2} J=24.8 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-3\right), 124.49,124.51\left({ }^{4} J=3.5 \mathrm{~Hz}\right.$, Ph C-1), 131.4 (C-5), 134.5, 134.6 ( $\left.{ }^{3} J=9.7 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-6\right), 137.05,137.11$ ( ${ }^{3} \mathrm{~J}=10.9 \mathrm{~Hz}$, Ph C-2), 154.5 (C-3), 161.5 (C=O), 164.0, 165.5 ( $\left.{ }^{1} J=257.6 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-4\right) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-196.3(\mathrm{~N}-2),-98.6(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 255\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, 100). HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{ClFN}_{2} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 255.0331, found 255.0330 .
4.1.2.12.9. 1-Methyl-1H-pyrazol-3-yl 2-chloropyridine-3-carboxylate (89).
 Brown solid, yield $95 \mathrm{mg}, 80 \%, \mathrm{mp} 78-79{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3075,3043\left(\mathrm{CH}_{\text {arom }}\right), 2930\left(\mathrm{CH}_{\text {aliph }}\right), 1755(\mathrm{C}=\mathrm{O})$, $1168,1579,1563,1531,1464,1433,1406,1272,1242,1228$, 1219, 1130, 1071, 1039 (C=C, C-N, C-O-C), 782, 766, 757, 724, $618,523,489,421(\mathrm{C}-\mathrm{Cl}) .{ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ $3.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.49\left(\mathrm{~d},{ }^{3} \mathrm{~J}(4-\mathrm{H}, 5-\mathrm{H})=2.7 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.21-$ $7.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.34\left(\mathrm{dd},{ }^{3} J(\mathrm{Pyr} 5-\mathrm{H}, 4-\mathrm{H})=7.8 \mathrm{~Hz},{ }^{3} J(\mathrm{Pyr}\right.$ $5-\mathrm{H}, 6-\mathrm{H})=4.8 \mathrm{~Hz}, 1 \mathrm{H}, \operatorname{Pyr} 5-\mathrm{H}), 7.37-7.39(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 3,5-\mathrm{H})$, $7.59-7.60(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 7.87\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.7 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $5-\mathrm{H}), 8.40\left({ }^{3} J(\operatorname{Pyr} 6-\mathrm{H}, 5-\mathrm{H})=7.7 \mathrm{~Hz},{ }^{4} J(\operatorname{Pyr} 6-\mathrm{H}, 4-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \operatorname{Pyr} 6-\mathrm{H}\right), 8.52$ $\left({ }^{3} J(\mathrm{Pyr} 4-\mathrm{H}, 5-\mathrm{H})=4.8 \mathrm{~Hz},{ }^{4} J(\mathrm{Pyr} 4-\mathrm{H}, 6-\mathrm{H})=4.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 4-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 39.4\left(\mathrm{CH}_{3}\right), 96.4(\mathrm{C}-4), 122.3(\mathrm{Ph} \mathrm{C}-5), 125.4(\mathrm{Ph} \mathrm{C-1)}$, 131.5 (C-5), 141.1 (Ph C-6), 150.9 ( $\mathrm{Ph} \mathrm{C}-2$ ), 152.7 ( $\mathrm{Ph} \mathrm{C}-4$ ), $154.3(\mathrm{C}-3), 161.4(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-195.7(\mathrm{~N}-2),-99.1(\mathrm{~N}-1),-70.7(\operatorname{Pyr} \mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}$ (\%): $238\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{ClN}_{3} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 238.0378, found 238.0377 .

### 4.1.2.13. General procedure for the preparation of $\boldsymbol{o}$-acylated-1-phenyl-1Hpyrazoles by using aroyl acids.

Into the solution of corresponding aroyl acid ( 0.5 mmol ) in dichlormethane ( 5 mL ), DCC ( $0,53 \mathrm{mmol}$ ) was added at $0^{\circ} \mathrm{C}$ temperature, and the reaction mixture was stirred for 1 hour. Later, 3-hydroxypyrazole $(0,05 \mathrm{mmol})$ and DMAP were added, and the reaction was stirred for 2 hours at $0{ }^{\circ} \mathrm{C}$ temperature. After 2 hours of stirring, the reaction mixture was stirred for 12 hours at room temperature. After completion of the reaction as indicated by TLC, the mixture was quenched with water ( 10 mL ) and extracted with dichormethane $(3 \times 10 \mathrm{~mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The
obtained residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 6$, v/v) to provide the desired products $\mathbf{7 6}, \mathbf{7 8}, \mathbf{8 0}, \mathbf{8 1}, \mathbf{8 5}, 87$.
4.1.2.13.1. 1-Phenyl-1H-pyrazol-3-yl 2-bromo-4-methylbenzoate (76).


White solid, yield $146 \mathrm{mg}, 82 \%, \mathrm{mp} 64-65^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3073, 3062, $3038\left(\mathrm{CH}_{\text {arom }}\right)$, 2961, 2919 $\left(\mathrm{CH}_{\text {aliph }}\right), 1748(\mathrm{C}=\mathrm{O}), 1599,1535,1503,1450,1388,1278$, 1240, 1208, 1102, 1055 (C-O-C, C=C, C-N), 757, 743, 682 ( $\mathrm{C}-\mathrm{Br}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.41\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.55(\mathrm{~d}$, $\left.{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.24\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.0 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 7.28-7.30(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.44-7.46(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.58(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CPh} 3-\mathrm{H}), 7.66-7.67(\mathrm{~m}, 2 \mathrm{H}$, NPh 2,6-H), $7.90\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 8.08\left(\mathrm{~d},{ }^{3} J(6-\mathrm{H}, 5-\mathrm{H})=8.0 \mathrm{~Hz}, 1 \mathrm{H}\right.$, CPh 6-H). ${ }^{13} \mathrm{C}$ NMR (176 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.4\left(\mathrm{CH}_{3}\right), 99.2(\mathrm{C}-4), 118.9(\mathrm{NPh}$ C-2,6), 123.3 ( $\mathrm{CPh} \mathrm{C}-2$ ), 126.7 ( $\mathrm{NPh} \mathrm{C}-4$ ), 126.8 ( $\mathrm{CPh} \mathrm{C}-1$ ), 127.9 (C-5), 128.3 ( CPh C-5), 129.6 (NPh C-3,5), 132.7 (CPh C-6), 135.7 (CPh C-3), 139.9 (NPh C-1), 145.1 (CPh C-4), $156.6(\mathrm{C}-3), 162.5(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.9$ (N-2), -105.0 (N-1). MS m/z (\%): 356 ([M] $\left.{ }^{+}, 100\right), 358\left([\mathrm{M}+2]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{BrN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 379.0053, found 379.0053 .
4.1.2.13.2. 1-Phenyl-1H-pyrazol-3-yl 2,3-difluorobenzoate (78). White solid,
 yield $132 \mathrm{mg}, 88 \%, \mathrm{mp} 75-76^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3095, $3110\left(\mathrm{CH}_{\text {arom }}\right), 1747(\mathrm{C}=\mathrm{O}), 1604,1539,1492,1457,1398$, 1236, 1140, 1172, 1139, 1075 (C-O-C, C=C, C=N, C-F), 840, 749, 719, $689\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.53\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.6 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 4-\mathrm{H}), 7.20-7.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 7.30-7.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh}$ 4-H), 7.44-7.48 (m, 3H, NPh 3,5-H, CPh 4-H), 7.65-7.68 (m, 2H, NPh 2,6-H), $7.91\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 7.92-7.94$ (m, 1H, CPh 6-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.2$ (C4), 119.0 (NPh C-2,6), 119.53, $119.58\left(^{2,3} J=6.2 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-1\right)$, 122.52, 122.54, 122.69, $122.71\left({ }^{2} J=17.5 \mathrm{~Hz},{ }^{3} J=1.5 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-4\right)$, 124.08, 124.13, $124.14,124.19\left({ }^{3} J=6.6 \mathrm{~Hz},{ }^{4} \mathrm{~J}=1.8 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-5\right), 126.8$ ( $\mathrm{NPh} \mathrm{C}-4$ ), $127.33,127.37$ ( ${ }^{4,5} J=3.8 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-6$ ), 128.1 (C-5), 139.8 (NPh C-1), 150.22, 150.30, 150.54, 150.61, $\left.151.72,151.80,151.96,152.03^{1} J=250 \mathrm{~Hz},{ }^{2} J=12.7 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-2,3\right)$, 156.23 (C-3), $160.35,160.39,160,39,160.43\left({ }^{3,4} J=3.6 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 301\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~F}_{2} \mathrm{~N}_{2} \mathrm{NaO}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 323.0603, found 323.0602.
4.1.2.13.3. 1-Phenyl-1H-pyrazol-3-yl 2-chloro-4-nitrobenzoate (80). Brown
 solid, yield $144 \mathrm{mg}, 84 \%, \mathrm{mp} 117-118^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\max }, \mathrm{cm}^{-1}\right): 3093\left(\mathrm{CH}_{\text {arom }}\right), 1756(\mathrm{C}=\mathrm{O}), 1598,1526,1502$, 1450, 1396, 1347, 1262, 1236 (C-O-C, C=C, C=N, C-NO ${ }_{2}$ ), 851, 776, 758, 731, $688(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ mono-, and trisubstituted benzenes). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ $6.56\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.30-7.32(\mathrm{~m}, 1 \mathrm{H}$, NPh 4-H), 7.45-7.47 (m, 2H, NPh 3,5-H), 7.65-7.66 (m, 2H, NPh 2,6-H), $7.92\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 8.22(\mathrm{dd}$, ${ }^{3} J(\mathrm{CPh} 5-\mathrm{H}, 6-\mathrm{H})=8.6 \mathrm{~Hz},{ }^{4} J(\mathrm{CPh} 5-\mathrm{H}, 3-\mathrm{H})=2.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh}$ $5-\mathrm{H}), 8.28\left(\mathrm{~d},{ }^{3} J(\mathrm{CPh} 6-\mathrm{H}, 5-\mathrm{H})=8.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 6-\mathrm{H}\right), 8.38\left(\mathrm{~d},{ }^{4} J(\mathrm{CPh} 3-\mathrm{H}, 5-\mathrm{H})=2.2\right.$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{CPh} 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 98.9$ (C-4), $119.0(\mathrm{NPh} \mathrm{C}-$ 2,6), 121.7 (CPh C-5), 126.5 (CPh C-3), 127.0 ( $\mathrm{NPh} \mathrm{C}-4$ ), 128.2 (C-5), 129.7 ( NPh C-3,5), 133.1 (CPh C-6), 133.9 (CPh C-1), 136.1 (CPh C-2), 139.7 (NPh C-1), 150.1 (CPh C-4), $155.9(\mathrm{C}-3), 160.8(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-177.7(\mathrm{~N}-$ 2), $-103.6(\mathrm{~N}-1),-17.7\left(\mathrm{NO}_{2}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 344\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{Cl} \mathrm{N}_{3} \mathrm{O}_{4} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 366.0252, found 366.0251.
4.1.2.13.4. 1-Phenyl-1H-pyrazol-3-yl 2,6-dichloropyridine-3-carboxylate
 (81). White solid, yield $119 \mathrm{mg}, 71 \%$, $\mathrm{mp} 131-132^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3174,3078\left(\mathrm{CH}_{\text {arom }}\right), 1751(\mathrm{C}=\mathrm{O})$, $1598,1526,1502,1450,1396,1340,1267,1259,1246,1145$, 1135, 1061 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}=\mathrm{N}, \mathrm{C}-\mathrm{O}-\mathrm{C}), 995,871,773,755,686,667$, 616, 562, $504\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ monosubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 6.55\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.6 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 4-\mathrm{H}), 7.30-7.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.43-7.48(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph}$ $3,5-\mathrm{H}, \operatorname{Pyr} 5-\mathrm{H}), 7.64-7.67$ (m, 2H, Ph 2,6-H), 7.91 (d, ${ }^{3} J(5-$ $\mathrm{H}, 4-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.46\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.6 \mathrm{~Hz}, 1 \mathrm{H}\right.$, Pyr 6-H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 99.0(\mathrm{C}-4), 119.0(\mathrm{Ph} \mathrm{C-2,6)} 123.1$, (Pyr C-5), 127.7 (Pyr C-1), 127.0 (Ph C-4), 128.2 (C-5), 129.7 (Ph C-3,5), 139.7 (Ph C-1), 143.3 (Pyr C-6), 151.1 (Pyr C-2), 154.2 (Pyr C-4), 156.0 (C-3), 160.4 (C=O). MS m/z (\%): 333 ([M+H] $\left.{ }^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{15} \mathrm{H}_{9} \mathrm{ClN}_{3} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 355.9964 , found 355.9965 .

### 4.1.2.13.5. 1-Methyl-1H-pyrazol-3-yl 2-bromo-4-methylbenzoate

(85).


White solid, yield $123 \mathrm{mg}, 83 \%$, $\mathrm{mp} 57-58^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3039,3025\left(\mathrm{CH}_{\text {arom }}\right), 2922\left(\mathrm{CH}_{\text {aliph }}\right), 1748$ (C=O), 1602, 1529, 1488, 1458, 1432, 1409, 1270, 1230, 1209, $1156,1098,1082,1053,1032(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 876,870,829,762$, 752, 680, 665, $456(\mathrm{C}-\mathrm{Br}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.39(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{PhCH}_{3}\right), 3.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.27\left(\mathrm{~d},{ }^{3} J(4-\mathrm{H}, 5-\mathrm{H})=2.3 \mathrm{~Hz}, \mathrm{C}-4\right)$, $7.20-7.22(\mathrm{~m}, 1 \mathrm{H}$, Ph $5-\mathrm{H}), 7.32\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 4-\mathrm{H})=2.1 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $5-\mathrm{H}), 7.55(\mathrm{~s}, 1 \mathrm{H}, \mathrm{Ph} 3-\mathrm{H}), 8.01-8.02(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.3\left(\mathrm{PhCH}_{3}\right), 39.4\left(\mathrm{NCH}_{3}\right), 96.7(\mathrm{C}-4), 123.1(\mathrm{Ph}$

C-2), 127.0 (Ph C-1), 128.2 (Ph C-5), 131.4 (C-5), 132.5 (Ph C-6), 135.6 (Ph C-3), 144.9 ( $\mathrm{Ph} \mathrm{C}-4$ ), $154.7(\mathrm{C}-3), 162.7(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}$ -197.1 (N-2), -100.5 (N-1). MS m/z (\%): 294 ([M] $\left.{ }^{+}, 100\right), 296\left([M+2]^{+}, 98\right)$. HRMS (ESI) for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{BrN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 316.9896, found 316.9896.
4.1.2.13.6. 1-Methyl-1H-pyrazol-3-yl 2,3-difluorobenzoate (87). Yellow
 solid, yield $93 \mathrm{mg}, 78 \%, \mathrm{mp} 67-68^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3087\left(\mathrm{CH}_{\text {arom }}\right), 2952\left(\mathrm{CH}_{\text {aliph }}\right), 1756,1752(\mathrm{C}=\mathrm{O}), 1592$, $1528,1488,1473,1467,1407,1282,1274,1271,1248,1237$, 1214, 1122 (C-O-C, C=C, C=N, C-F), 950, 819, 753, 747, 608, 567, 479 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.23-6.24$ $(\mathrm{m}, 1 \mathrm{H}, 4-\mathrm{H}), 7.18-7.22(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 5-\mathrm{H}), 7.32\left(\mathrm{~d},{ }^{3} \mathrm{~J}(5-\mathrm{H}, 4-\right.$ $\mathrm{H})=1.8 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.39-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh} 4-\mathrm{H}), 7.85-7.88(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{Ph} 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 39.5\left(\mathrm{CH}_{3}\right)$, 96.6 (C-4), $119.69,119.72\left(^{2,3} J=5.3 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-1\right), 122.37,122.47$ $\left({ }^{2,3} J=17.6 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-4\right), 124.03,124.06,124.10\left({ }^{3,4} J=7.0 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-5\right), 127.25,127.27$ $\left.{ }^{4,5} J=3.5 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-6\right), 131.4$ (C-5), 150.19, 150.27, 150.57, 150.64, 151.69, 151.77, $151.99,152.06\left({ }^{1} J=250 \mathrm{~Hz},{ }^{2} J=12.3 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-3\right), 154.5(\mathrm{C}-3), 160.55,160.57,160.59$ $\left({ }^{3,4} J=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right) .{ }^{15} \mathrm{~N}$ NMR (71 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-195.8(\mathrm{~N}-2),-98.6(\mathrm{~N}-1)$. MS m/z (\%): $239\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~F}_{2} \mathrm{~N}_{2} \mathrm{NaO}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 261.0446, found 261.0447.

### 4.1.2.14. General procedure for the preparation of 3-hydroxypyrazole ketones by using Fries rearrangement reaction.

Into the solution of corresponding 3-hydroxy-1H-pyrazole $\mathbf{7 5 - 8 6}$ ( 0.5 mmol ) in carbon disulfide ( 7 mL ), the appropriate solution of $\mathrm{AlCl}_{3}(6 \mathrm{mmol})$ in carbon disulfide ( 2 ml ) was added. The mixture was stirred for 5 hours at $50^{\circ} \mathrm{C}$ temperature; later, it was diluted with ice water $(15 \mathrm{ml})$ and $6 \mathrm{~N} \mathrm{HCl}(7 \mathrm{ml})$. The obtained residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 4$, $\mathrm{v} / \mathrm{v}$ ) to provide to yield compounds $\mathbf{9 0} \mathbf{- 1 0 1}$.
4.1.2.14.1. (2-Chlorophenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4-yl)methanone (90). White solid, yield $121 \mathrm{mg}, 81 \%, \mathrm{mp} 158-159^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$,
 $\left.\mathrm{cm}^{-1}\right): 3129(\mathrm{OH}), 3072\left(\mathrm{CH}_{\text {arom }}\right), 1629(\mathrm{C}=\mathrm{O}), 1591,1471,1454$, 1435, 1338, 1314, 1203, 1060 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}$ ), 928, 752, 714, 690, 667, $651\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.30-7.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H})$, 7.39-7.43 (m, 3H, CPh 5-H, NPh C-3,5), 7.45-7.48 (m, 1H, CPh 4-H), 7.49-7.51 (m, 2H, CPh C-3,6), 7.62-7.63 (m, 2H, NPh 2,6$\mathrm{H}), 7.85(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 9.16(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 108.5(\mathrm{C}-4), 119.3(\mathrm{NPh} \mathrm{C-2,6)}, 127.1$ (CPh C-5), 127.7 ( $\mathrm{NPh} \mathrm{C}-4$ ), 129.0 (CPh C-6), 129.7 (NPh C-3,5), 130.1 (C5), 130.8 ( $\mathrm{CPh} \mathrm{C}-3$ ), 131.0 (CPh C-2), 132.0 ( $\mathrm{CPh} \mathrm{C}-4$ ), 137.6 ( $\mathrm{CPh} \mathrm{C}-1$ ), 138.9 ( NPh $\mathrm{C}-1), 164.7(\mathrm{C}-3), 190.9(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.7(\mathrm{~N}-2)$,
-117.6 (N-1). MS m/z (\%): $299\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{ClN}_{2} \mathrm{NaO}_{2}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 321.0401, found 321.0401
4.1.2.14.2. (2-Bromo-4-methylphenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4yl)methanone (91). White solid, yield $134 \mathrm{mg}, 75 \%, \mathrm{mp} 139-140^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): 3129, $3114(\mathrm{OH}), 3056\left(\mathrm{CH}_{\text {arom }}\right), 2956,2924$

|  |
| :---: | $\left(\mathrm{CH}_{\text {aliph }}\right), 1762(\mathrm{C}=\mathrm{O}), 1624,1598,1558,1530,1462,1398,1340$, 1317, 1266, 1242, 1204, 1042 (C=C, C-N), 751, 688, 668, 603 (C$\mathrm{Br}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR (700 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 2.42\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.24-7.26(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh}$ 5-H), 7.31-7.33 (m, 1H, NPh 4-H), 7.38-7.39 (m, 1H, CPh 6-H), $7.42-7.45$ (m, 2H, NPh 3,5-H), 7.54 (s, 1H, CPh 3-H), 7.63-7.65 (m, 2H, NPh 2,6-H), $7.85(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 21.3\left(\mathrm{CH}_{3}\right), 108.5(\mathrm{C}-7), 119.25(\mathrm{CPh} \mathrm{C}-2), 119.33$ ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 127.7 (NPh C-4), 128.3 (CPh C-5), 129.0 (CPh C-6), 129.7 (NPh C-3,5), 130.2 (C-5), 134.6 (CPh C-3), 136.8 (CPh C-1), 139.0 (Ph C-1), 143.0 (CPh C-4), 164.9 (C-3), 191.8 (C=O). MS m/z (\%): 357 ([M] ${ }^{+}, 100$ ), 359 $\left([\mathrm{M}+2]^{+}, 100\right)$. HRMS $(\mathrm{ESI})$ for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{BrN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 379.0053, found 379.0053.

4.1.2.14.3. (2,4-Dichlorophenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4-yl)methanone (92). White solid, yield $135 \mathrm{mg}, 81 \%, \mathrm{mp} 149-150{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR
 $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3477(\mathrm{OH}), 3095,3067,3044,3029\left(\mathrm{CH}_{\text {arom }}\right), 1752$ (C=O), 1601, 1585, 1538, 1502, 1474, 1456, 1402, 1389, 1374, 1278, 1264, 1233, 1096, 1046, $1005(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 865,818,760$, $747,683,672,476,406(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.33-7.35(\mathrm{~m}, 1 \mathrm{H}$, NPh 4-H), $7.40\left(\mathrm{~d},{ }^{3} J(\mathrm{CPh} 5-\mathrm{H}, 6-\mathrm{H})=8.2 \mathrm{~Hz},{ }^{4} J(\mathrm{CPh} 5-\mathrm{H}, 3-\mathrm{H})=1.9\right.$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 7.43-7.47$ (m, 3H, NPh 3,5-H, CPh 6-H), 7.54 $\left(\mathrm{d},{ }^{4} J(3-\mathrm{H}, 5-\mathrm{H})=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 3-\mathrm{H}\right), 7.63-7.64(\mathrm{~m}, 2 \mathrm{H}$, NPh 2,6H), $7.85(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 108.5$ (C-4), 119.4 (NPh C-2,6), 127.6 (CPh C-5), 127.9 (NPh C-4), 129.7 (CPh C-6, NPh C-3,5), 130.0 (C-5), 130.8 (CPh C-3), 132.2, 136.1, 137.6 (CPh C$1,2,4), 164.7(\mathrm{C}-3), 189.8(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N} \operatorname{NMR}\left(71 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-180.8(\mathrm{~N}-2)$, $-118.6(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 332$ ([M] $\left.{ }^{+}, 100\right), 334$ ( $[\mathrm{M}+2]^{+}, 70$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 355.0012 , found 355.0011 .

[^1]
found 323.0602.
$\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 108.4$ (C-4), 119.5 ( $\mathrm{NPh} \mathrm{C}-2,6$ ), 120.84, $120.94\left({ }^{2,3} J=17.4 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-4\right), 124.98,125.00\left({ }^{4,5} \mathrm{~J}=3.9 \mathrm{~Hz}, \mathrm{CPh}\right.$ C-6), $125.11,125.14,125.15,125.17\left({ }^{3} J=6.3 \mathrm{~Hz},{ }^{4} J=4.6 \mathrm{~Hz}, \mathrm{CPh}\right.$ C-5), 127.9 (NPh C-4), $128.49,128.55\left({ }^{2,3} J=11.1 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-1\right)$, $129.75,129.80(J=8.9 \mathrm{~Hz}, \mathrm{C}-5)$, 129.80 (NPh C-3,5), 147.21, $147.29,148.66,148.74$ ( $\left.{ }^{1} J=255.1 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-2\right), 150.12$, 150.19 , $151.54,151.62$ ( ${ }^{1} J=251.3 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-3$ ), 165.14 (C-3), 186.75, $186.76\left(^{2,3} \mathrm{~J}=2.9 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 301\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~F}_{2} \mathrm{~N}_{2} \mathrm{NaO}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 323.0603,
4.1.2.14.5. (2-Chloro-4-fluorophenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4yl)methanone (94). White solid, yield $142 \mathrm{mg}, 90 \%$, mp $143-144{ }^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3203(\mathrm{OH}), 3088,3064\left(\mathrm{CH}_{\text {arom }}\right), 1625,1599,1582$ ( $\mathrm{C}=\mathrm{O}$ ), 1476, 1424, 1337, 1313, 1263, 1216, 1199, 1048 (C-F, $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 937,887,873,824,780,759,697,687,669,618,608$, $491\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.12-7.14(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H})$, 7.28-7.28 (m, 1H, CPh 3-H), 7.32-7.35 (m, 1H, NPh 4-H), 7.437.46 (m, 2H, NPh 3,5-H), 7.52-7.54 (m, 1H, CPh 6-H), 7.64-7.65 (m, 2H, NPh 2,6-H), $7.86(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 108.5(\mathrm{C}-4), 114.54,114.66\left({ }^{2} J=21.5 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-\right.$ 5), $118.43,118.57\left({ }^{2} J=24.8 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-3\right), 119.4$ (NPh C-2,6), 127.9 (NPh C-4), 129.7 (NPh C-3,5), 129.9 (C-5), $130.71,130.76$ ( ${ }^{3} J=9.2 \mathrm{~Hz}, \mathrm{CPh}$ C6), 132.74, $132.80\left({ }^{3} J=10.7 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-2\right), 134.02,134.04\left({ }^{4} J=3.8 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-1\right)$, 138.9 ( $\mathrm{NPh} \mathrm{C}-1$ ), $162.98,164.34$ ( ${ }^{1} J=255.1 \mathrm{~Hz}, \mathrm{CPh} \mathrm{C}-4$ ), $164.8(\mathrm{C}-3), 190.0(\mathrm{C}=\mathrm{O})$. ${ }^{15} \mathrm{~N}$ NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-178.5(\mathrm{~N}-2),-116.8(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 317$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{ClFN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 339.0307, found 339.0307.
4.1.2.14.6. (2-Chloro-4-nitrophenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4yl)methanone (95). Yellow solid, yield $137 \mathrm{mg}, 80 \%, \mathrm{mp} 208-20{ }^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): $3382(\mathrm{OH}), 3088,3072,3022\left(\mathrm{CH}_{\text {arom }}\right), 1636$ ( $\mathrm{C}=\mathrm{O}$ ), 1595, 1578, 1523, 1509, 1492, 1470, 1445, 1348, 1340, 1333, 1316, 1229, 1215, 1161, 1115, 1045 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}=\mathrm{N}, \mathrm{C}-\mathrm{NO}_{2}$ ), 934, 877, $758,741,668,616(\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.31-7.35(\mathrm{~m}$, 1 H , NPh 4-H), 7.49-7.51 (m, 2H, NPh 3,5-H), 7.77-7.82 (m, 3H, NPh 2,6-H, CPh 6-H), $8.28\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.4 \mathrm{~Hz},{ }^{4} J(5-\mathrm{H}, 3-\right.$ $\mathrm{H})=2.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 8.40\left(\mathrm{~d},{ }^{4} \mathrm{~J}(3-\mathrm{H}, 5-\mathrm{H})=2.1 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}\right)$, 11.44 (br s, $1 \mathrm{H}, \mathrm{OH}$ ). ${ }^{13} \mathrm{C}$ NMR ( 176 MHz, DMSO- $d_{6}$ ): $\delta_{\mathrm{C}} \mathrm{ppm}$ 110.1 (C-4), 118.8 (NPh C-2,6), 123.0, 125.1, 127.4 (NPh C-4H), 130.0 (C-5, NPh C-3,5), 131.0, 133.4 (CPh C-1,2), 139.0 (NPh C-1), 145.9, 148.8, 162.1 (C-3), $185.5(\mathrm{C}=\mathrm{O}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 344\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{ClN}_{3} \mathrm{NaO}_{4}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 366.0251, found 366.0252.
4.1.2.14.7. (2-Chlorophenyl)(3-hydroxy-1-methyl-1H-pyrazol-4-yl)methanone (96). Brown solid, 100 mg , yield $85 \%, \mathrm{mp} 96-97^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$,
 $\left.\mathrm{cm}^{-1}\right): 3486(\mathrm{OH}), 3080,3051,3040,3024\left(\mathrm{CH}_{\text {arom }}\right), 2955,2924$ $\left(\mathrm{CH}_{\text {aliph }}\right), 1749$ (C=O), 1598, 1567, 1531, 1502, 1451, 1417, 1393, 1379, 1339, 1265, 1257, 1243, 1133, 1059, 1025 (C=C, C-N), 869, $814,772,749,710,684,668,562,504(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of monoand disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm}$ $3.75\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.31(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 7.34-7.37(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H})$, 7.41-7.44 (m, 2H, Ph 3,5-H), 7.46-7.48 (m, 1H, Ph 6-H). ${ }^{13}$ C NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 39.8\left(\mathrm{CH}_{3}\right), 106.7(\mathrm{C}-4), 126.9(\mathrm{Ph} \mathrm{C}-$ 5), 128.9 (Ph C-6), 130.7 (Ph C-3), 130.9 (Ph C-2), 131.7 (Ph C-4), 134.0 (C-5), 137.8 ( $\mathrm{Ph} \mathrm{C}-1$ ), $164.4(\mathrm{C}-3), 190.4(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR $\left(71 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{N}} \mathrm{ppm}-195.6(\mathrm{~N}-$ 2), $-113.9(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 237\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{Na}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 259.0245 , found 259.0245 .

### 4.1.2.14.8. (2-Bromo-4-methylphenyl)(3-hydroxy-1-methyl-1H-pyrazol-4-

 yl)methanone (97). White solid, 102 mg , yield $69 \%, \mathrm{mp} 152-153^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.\mathrm{v}_{\text {max }}, \mathrm{cm}^{-1}\right): 3252\left(\mathrm{OH}, \mathrm{CH}_{\text {arom }}\right)$, $2941\left(\mathrm{CH}_{\text {aliph }}\right), 1754(\mathrm{C}=\mathrm{O})$, 1644, 1618, 1600, 1565, 1536, 1492, 1477, 1413, 1341, 1175, 1152, 1078, 1045 (C=C, C-N), 846, 831, 797, 774, 721, 607, 464 ( $\mathrm{C}-\mathrm{Br}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.34$ (s, $3 \mathrm{H}, \mathrm{CCH}_{3}$ ), 3.70 ( $\mathrm{s}, 3 \mathrm{H}$, $\mathrm{NCH}_{3}$ ), 7.14-7.16 (m, 1H, Ph 5-H), 7.26-7.27 (m, 1H, Ph 6-H), $7.30(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 7.44\left(\mathrm{~d},{ }^{3} J(3-\mathrm{H}, 5-\mathrm{H})=1.6 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 21.1\left(\mathrm{CCH}_{3}\right), 39.6\left(\mathrm{NCH}_{3}\right), 106.4(\mathrm{C}-$ 4), 119.0 ( $\mathrm{Ph} \mathrm{C}-2$ ), 128.0 ( $\mathrm{Ph} \mathrm{C}-5$ ), 128.7 ( $\mathrm{Ph} \mathrm{C-6)}$,134.2 (C-5), 134.3 (Ph C-3), 136.8 (Ph C-1), 142.5 (Ph C-4), 164.2 (C-3), 191.0 (C=O). ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-196.3(\mathrm{~N}-2),-113.5(\mathrm{~N}-1),-95.3(\mathrm{~N}-$ 5). MS m/z (\%): 294 ([M] $\left.{ }^{+}, 100\right), 296$ ( $[\mathrm{M}+2]^{+}, 100$ ). HRMS (ESI) for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{BrN}_{3} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 316.9896, found 316.9897.
4.1.2 14.9. (2,4-Dichlorophenyl)(3-hydroxy-1-methyl-1H-pyrazol-4-yl)methanone (98). White solid, yield $108 \mathrm{mg}, 80 \%, \mathrm{mp} 184-185{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR
 $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3262(\mathrm{OH}), 3071,3025\left(\mathrm{CH}_{\text {arom }}\right), 1635(\mathrm{C}=\mathrm{O}), 1565$, $1536,1503,1462,1429,1415,1340,1291,1172,1159,1103,1077$, 1054 (C=C, C-N), 926, 865, 827, 721, 669, 641, 567, 534, 453, 422 ( $\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz, DMSO- $d_{6}$ ): $\delta_{\text {н }} \operatorname{ppm} 3.64\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.41\left(\mathrm{~d},{ }^{3} \mathrm{~J}(6-\mathrm{H}, 5-\right.$ $\mathrm{H})=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ph} 6-\mathrm{H}), 7.48-7.49(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 5-\mathrm{H}), 7.68\left(\mathrm{~d},{ }^{4} J(3-\right.$ $\mathrm{H}, 5-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ph} 3-\mathrm{H}$ ), 7.88 (s, 1H, 5-H). ${ }^{13} \mathrm{C}$ NMR ( 176 MHz , DMSO- $d_{6}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 38.9\left(\mathrm{CH}_{3}\right)$, 107.2 (C-4), 127.5 ( $\mathrm{Ph} \mathrm{C}-5$ ), 129.5 (Ph C-3), 129.8 (Ph C-6), 130.7 (Ph C-2), 134.6 (Ph C-1), 136.4 (C-5), 139.0 (Ph C-4) 161.1 (C-3), 185.4 (C=O). ${ }^{15} \mathrm{~N}$ NMR
(71 MHz, DMSO- $d_{6}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-193.4(\mathrm{~N}-2),-110.8(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 271\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, 100). HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 292.9855, found 292.9855.
4.1.2.14.10. (2,3-Difluorophenyl)(3-hydroxy-1-methyl-1H-pyrazol-4-yl)methanone (99). White solid, yield $101 \mathrm{mg}, 85 \%, \mathrm{mp} 164-165{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR

|  |
| :---: | $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3279(\mathrm{OH}), 3037\left(\mathrm{CH}_{\text {arom }}\right), 2925\left(\mathrm{CH}_{\text {aliph }}\right), 1642(\mathrm{C}=\mathrm{O})$, $1564,1543,1512,1472,1428,1348,1266,1201,1169,1077$ (C-F, $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 993,869,856,793,773,751,721,649,615,602,548$, $515,457\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.80\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.20-7.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph}$ 5-H), 7.34-7.36 (m, 1H, Ph 4-H), 7.37-7.40 (m, 1H, Ph 6-H), 7.52 (d, $J=3.1 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 9.39$ (br s, $1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \operatorname{ppm} 39.9\left(\mathrm{CH}_{3}\right), 106.6(\mathrm{C}-4), 120.51,120.61\left(^{2,3} \mathrm{~J}=17.2\right.$ $\mathrm{Hz}, \mathrm{Ph} \mathrm{C}-4), 124.87,124.90\left({ }^{4,5} J=3.8 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-6\right), 124.95,124.98$, $124.99,125.02\left({ }^{3} J=6.4 \mathrm{~Hz},{ }^{4} J=4.6 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-5\right), 128.71,128.77\left({ }^{2,3} J\right.$ $=11.4 \mathrm{~Hz}$, Ph C-1), 133.62, $133.67\left({ }^{5} \mathrm{~J}=8.2 \mathrm{~Hz}, \mathrm{C}-5\right), 147.09,147.17,148.54,148.61$ $\left({ }^{1} J=254.5 \mathrm{~Hz},{ }^{2} J=13.9 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-2\right), 150.05,150.12,151.48,151.55\left({ }^{1} J=250.9 \mathrm{~Hz}\right.$, ${ }^{2} J=13.0 \mathrm{~Hz}$, Ph C-3), $164.8(\mathrm{C}-3), 186.13,186.14\left(J={ }^{3,4} 1.8 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right) .{ }^{15} \mathrm{~N}$ NMR (71 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-194.6(\mathrm{~N}-2),-113.1(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 239\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~F}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 261.0446, found 261.0446.

4.1.2.14.11. (2-Chloro-4-fluorophenyl)(3-hydroxy-1-methyl-1H-pyrazol-4yl)methanone (100). White solid, yield $94 \mathrm{mg}, 74 \%, \mathrm{mp} 169-170^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3264(\mathrm{OH}), 3082,3032\left(\mathrm{CH}_{\text {arom }}\right), 2956,2927$ $\left(\mathrm{CH}_{\text {aliph }}\right), 1635(\mathrm{C}=\mathrm{O}), 1600,1560,1542,1506,1486,1426,1409$, $1379,1342,1288,1278,1260,1207,1169,1157,1073,1046$ (CF, C=C, C-N), 934, 888, 855, 817, 797, 769, 685, 646, 609, 599, $564,429\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz, DMSO- $d_{6}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.65\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.27-7.30$ m 1H, Ph 5-H), 7.46-7.48 (m, 1H, Ph 6-H), 7.51-7.52 (m, 1H, Ph $3-\mathrm{H}), 7.87(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 10.62(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , DMSO- $\left.d_{6}\right): \delta_{\mathrm{C}} \mathrm{ppm} 38.8\left(\mathrm{CH}_{3}\right), 107.2(\mathrm{C}-4), 114.28,114.40$ ( $\left.{ }^{2} J=21.4 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-5\right), 116.88,117.03\left({ }^{2} J=25.2 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-3\right)$, 130.14, $130.19\left({ }^{3} J=9.3 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-6\right), 130.71,130.77\left({ }^{3} J=10.8 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-2\right), 136.3$ (C5), 136.67, $136.69\left({ }^{4} J=3.5 \mathrm{~Hz}, \mathrm{Ph} \mathrm{C}-1\right), 161.0(\mathrm{C}-3), 161.36,162.78\left({ }^{1} J=249.9 \mathrm{~Hz}\right.$, Ph C-4), 185.4 (C=O). MS m/z (\%): 255 ( $[\mathrm{M}+\mathrm{H}]^{+}, 100$ ). HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{FClN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 277.0151, found 277.0151.

### 4.1.2.15. General procedure for the preparation of benzopyrano[2,3-c]pyrazol-4(2H)-ones 101-110.

Into the solution of corresponding pyrazole $\mathbf{9 0}-100(0.5 \mathrm{mmol})$ in DMF ( 5 mL ), potassium carbonate ( 1 mmol ) was added. The mixture was stirred overnight at 120 ${ }^{\circ} \mathrm{C}$ temperature; later, it was diluted with water ( 10 mL ) and extracted with ethyl acetate $(3 \times 10 \mathrm{~mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified
by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $\left.1: 8, \mathrm{v} / \mathrm{v}\right)$ to provide to yield compounds 101-110.
4.1.2.15.1. 2-Phenyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one (101). White
 solid, yield $118 \mathrm{mg}, 90 \%, \mathrm{mp} 204-205^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3096,3069,3044\left(\mathrm{CH}_{\text {arom }}\right), 1664(\mathrm{C}=\mathrm{O}), 1600,1585,1572,1492$, 1467, 1452, 1432, 1318, 1293, 1217, 1161, 1101, 1052, 1024 (C-O$\mathrm{C}, \mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 955,931,905,880,764,744,708,684,664,530,503$, 493, 471, $435\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.38-7.42$ (m, 2H, 6-H, Ph 4-H), 7.52-7.55 (m, 3H, 8-H, Ph 3,5-H), 7.70-7.72 (m, 1H, 7-H), 7.79-7.80 $(\mathrm{m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.32\left(\mathrm{dd},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=7.9 \mathrm{~Hz},{ }^{4} J(5-\mathrm{H}, 7-\mathrm{H})=1.7 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 5-\mathrm{H}), 8.58(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (176 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 108.7(\mathrm{C}-3 \mathrm{a}), 118.8$ (C-8), 119.9 (Ph C-2,6), 122.5 (C-4a), 124.3 (C-6), 125.4 (C-3), 127.0 (C-5), 134.7 (C-7), $156.0(\mathrm{C}-8 \mathrm{a}), 162.3(\mathrm{C}-9 \mathrm{a}), 174.8(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR (71 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}$ $-170.5(\mathrm{~N}-2),-116.6(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 263\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{NaO}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 285.0634, found 285.0636.

### 4.1.2.15.2. 7-Methyl-2-phenyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

 (102). White solid, yield $104 \mathrm{mg}, 75 \%, \mathrm{mp} 235-236{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$ $\left.{ }^{1}\right): 3109,3066\left(\mathrm{CH}_{\text {arom }}\right), 2996,2922\left(\mathrm{CH}_{\text {aliph }}\right), 1648(\mathrm{C}=\mathrm{O}), 1618$, $1600,1575,1499,1466,1436,1405,1287,1246,1233,1211,1182$, $1145,1113(\mathrm{C}-\mathrm{O}-\mathrm{C}, \mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 953,905,860,771,750,710,682$, $663,545,481\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR (700 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.18-7.20(\mathrm{~m}, 1 \mathrm{H}, 6-$ H), $7.32(\mathrm{~s}, 1 \mathrm{H}, 8-\mathrm{H}), 7.39-7.41(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.51-7.53(\mathrm{~m}, 2 \mathrm{H}$, Ph 3,5-H), 7.77-7.80 (m, 2H, Ph 2,6-H), $8.19\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.0\right.$ $\mathrm{Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.56(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $22.0\left(\mathrm{CH}_{3}\right), 108.8(\mathrm{C}-3 \mathrm{a}), 118.2(\mathrm{C}-8), 119.8(\mathrm{Ph} \mathrm{C}-2,6), 120.2(\mathrm{C}-$ 4a), 125.3 (C-3), 125.7 (C-6), 126.8 (C-5), 128.3 (Ph C-4), 129.9 (Ph $\mathrm{C}-3,5), 139.2$ ( $\mathrm{Ph} \mathrm{C}-1$ ), 146.3 (C-7), 156.2 (C-8a), 162.4 (C-9a), $174.8(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-170.8(\mathrm{~N}-2),-116.8(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 277$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS $(\mathrm{ESI})$ for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 277.0792, found 277.0791

### 4.1.2.15.3. 7-Chloro-2-phenyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

 (103). White solid, yield $135 \mathrm{mg}, 91 \%$, mp 209-210 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3110,3069,3055,3043,3030\left(\mathrm{CH}_{\text {arom }}\right), 1671$ (C=O), 1601, 1583, 1561, 1509, 1467, 1441, 1411, 1282, 1214, 1203, 1079 (C-O-C, C=C, C-N), 962, 939, 919, 905, 869, 830, 767, 745, $726,703,682,659,619,590,541,499,476,444(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}}$ ppm $7.36\left(\mathrm{dd},{ }^{3} J(6-\mathrm{H}, 5-\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(6-\mathrm{H}, 8-\mathrm{H})=1.9 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right)$, 7.41-7.44 (m, 1H, Ph 4-H), 7.52-7.56 (m, 3H, 8-H, Ph 3,5-H), 7.77$7.79(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ph} 2,6-\mathrm{H}), 8.25\left(\mathrm{~d},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}\right), 8.56$ (s, $1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 108.6$ (C-3a),
118.4 (C-8), 119.9 (Ph C-2,6), 121.2 (C-4a), 125.1 (C-6), 125.6 (C-3), 128.3 (C-5), 128.6 (Ph C-4), 130.0 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 139.1 ( $\mathrm{Ph} \mathrm{C}-1$ ), 140.7 (C-7), 156.2 (C-8a), 162.2 (C-9a), $173.9(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-169.9(\mathrm{~N}-2),-115.6(\mathrm{~N}-$ 1). MS m/z (\%): $297\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{16} \mathrm{H}_{9} \mathrm{ClN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$ calcd 319.0245 , found 319.0245 .
4.1.2.15.4. 8-Fluoro-2-phenyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one (104). Yellow solid, yield $123 \mathrm{mg}, 88 \%, \mathrm{mp} 255-256^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-}$
 $\left.{ }^{1}\right): 3104,3022,3003\left(\mathrm{CH}_{\text {arom }}\right), 1654(\mathrm{C}=\mathrm{O}), 1617,1600,1510$, 1483, 1441, 1434, 1300, 1286, 1264, 1213, 1183, 1157, 1065, 1052, 1021(C-O-C, C-F, C=C, C-N), 945, 911, 896, 866, 851, 818, 753, $699,683,659,616,600,508,498,456(\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 7.30-$ $7.35(\mathrm{~m}, 1 \mathrm{H}, 6-\mathrm{H}), 7.41-7.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ph} 4-\mathrm{H}), 7.49-7.57(\mathrm{~m}, 3 \mathrm{H}$, 7-H, Ph 3,5-H), 7.79-7.83 (m, 2H, Ph 2,6-H), 8.09 (d, ${ }^{3} J(5-\mathrm{H}, 6-$ $\mathrm{H})=7.9 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.59(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 108.6$ (C-3a), 120.0 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), $120.8,120.9$ ( ${ }^{2} J=17.6 \mathrm{~Hz}, \mathrm{C}-7$ ), 121.96, 121.98 ( $\left.{ }^{4} J=3.8 \mathrm{~Hz}, \mathrm{C}-5\right), 123.75,123.79\left({ }^{3} J=6.5 \mathrm{~Hz}, \mathrm{C}-6\right), 124.6$ (C-4a), 125.6 (C-3), 128.6 ( $\mathrm{Ph} \mathrm{C}-4$ ), 130.0 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 139.1 ( $\mathrm{Ph} \mathrm{C}-1$ ), $144.70,144.76$ ( ${ }^{2} J=11.7 \mathrm{~Hz}$, C-8a), 150.74, 152.18 ( $\left.{ }^{1} J=253.3 \mathrm{~Hz}, \mathrm{C}-8\right), 161.8$ (C-9a), $173.91,173.93$ ( ${ }^{4} \mathrm{~J}=2.8 \mathrm{~Hz}$, $\mathrm{C}=\mathrm{O}) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 281\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{16} \mathrm{H}_{9} \mathrm{FN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$ calcd 303.0540, found 303.0540 .

### 4.1.2.15.5. 7-Fluoro-2-phenyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

(105). White solid, yield $125 \mathrm{mg}, 89 \%, \mathrm{mp} 254-255^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-}$
 $\left.{ }^{1}\right): 3097,3060\left(\mathrm{CH}_{\text {arom }}\right), 1662(\mathrm{C}=\mathrm{O}), 1590,1576,1508,1471,1442$, 1421, 1300, 1280, 1249, 1237, 1145, 1097, 1049 (C-F, C-O-C, C=C, C-N), 977, 961, 931, 913, 873, 824, 784, 769, 753, 717, 707, 685, $660,619,505,490,476,452(\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.10-7.13(\mathrm{~m}, 1 \mathrm{H}, 6-$ H), 7.21-7.24 (m, 1H, 8-H), 7.41-7.44 (m, 1H, Ph 4-H), 7.52-7.55 (m, 2H, Ph 3,5-H), 7.77-7.79 (m, 2H, Ph 2,6-H), 8.32-8.35 (m, 1H, $5-\mathrm{H}), 8.56(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 105.23$, $105.40\left({ }^{2} J=25.9 \mathrm{~Hz}, \mathrm{C}-8\right), 108.6(\mathrm{C}-3 \mathrm{a}), 112.73,112.86\left({ }^{2} J=22.3 \mathrm{~Hz}\right.$, C-6), 119.40, 119.42 ( ${ }^{4} J=2.6 \mathrm{~Hz}, \mathrm{C}-4 \mathrm{a}$ ), 119.9 ( $\mathrm{Ph} \mathrm{C}-2,6$ ), 125.6 (C-3), 128.5 ( $\mathrm{Ph} \mathrm{C}-$ 4), 129.42, 129.48 ( ${ }^{3} J=10.8 \mathrm{~Hz}, \mathrm{C}-5$ ), 130.0 ( $\mathrm{Ph} \mathrm{C}-3,5$ ), 139.1 ( $\mathrm{Ph} \mathrm{C}-1$ ), 157.22, 157.29 ( $\left.{ }^{3} J=13.3 \mathrm{~Hz}, \mathrm{C}-8 \mathrm{a}\right), 162.4(\mathrm{C}-9 \mathrm{a}), 165.64,167.09\left({ }^{1} \mathrm{~J}=255.2 \mathrm{~Hz}, \mathrm{C}-7\right), 173.8$ (C=O). ${ }^{15} \mathrm{~N}$ NMR (71 MHz, $\mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-170.5(\mathrm{~N}-2),-116.1(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}$ (\%): $281\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{16} \mathrm{H}_{9} \mathrm{FN}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 303.0540, found 303.0540 .
4.1.2.15.6. 2-Methyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one (106). White solid, yield $80 \mathrm{mg}, 80 \%$, $\mathrm{mp} 167-168^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3092, 3022 $\left(\mathrm{CH}_{\text {arom }}\right), 2998,2941\left(\mathrm{CH}_{\text {aliph }}\right), 1656(\mathrm{C}=\mathrm{O}), 1605,1565,1489,1454,1424,1320$, 1283, 1214, 1185, 1169, 1144, 1104, 1077 (C-O-C, C=C, C-N), 996, 983, 947, 878,

$863,760,726,710,679,656,625,610,532,478,437(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}}$ pp $3.99\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.30-7.33(\mathrm{~m}, 1 \mathrm{H}, 6-\mathrm{H}), 7.44-7.46(\mathrm{~m}, 1 \mathrm{H}, 8-$ H), 7.62-7.64 (m, 1H, 7-H), $8.24\left(\mathrm{dd},{ }^{3} J(5-\mathrm{H}, 6-\mathrm{H})=7.9 \mathrm{~Hz},{ }^{4} J(5-\mathrm{H}, 7-\right.$ $\mathrm{H})=1.4 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 8.03(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{C}} \mathrm{ppm} 40.4\left(\mathrm{CH}_{3}\right), 107.0(\mathrm{C}-3 \mathrm{a}), 118.1(\mathrm{C}-8), 122.35(\mathrm{C}-4 \mathrm{a}), 123.9$ (C-6), 126.7 (C-5), 129.1 (C-3), 134.3 (C-7), 155.7 (C-8a), 161.5 (C9a), $174.4(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N} \operatorname{NMR}\left(71 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-170.4(\mathrm{~N}-2)$, $-115.8(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 201\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 223.0478, found 223.0479.
4.1.2.15.7. 2,7-Dimethyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one (107).


White solid, yield $100 \mathrm{mg}, 93 \%, \mathrm{mp} 210-211^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3096,3002\left(\mathrm{CH}_{\text {arom }}\right), 2945,2925,2919\left(\mathrm{CH}_{\text {aliph }}\right), 1667$, 1661 (C=O), 1626, 1577, 1567, 1499, 1470, 1438, 1414, 1317, 1245, 1173, 1112 (C-O-C, C=C, C-N), 866, 829, 772, 763, 723, 709, 580, $541,473\left(\mathrm{CH}=\mathrm{CH}\right.$ of trisubstituted benzene). ${ }^{1} \mathrm{H}$ NMR ( 700 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 2.48\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CCH}_{3}\right), 4.01\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{NCH}_{3}\right), 7.16(\mathrm{~d}$, $\left.{ }^{3} J(6-\mathrm{H}, 5-\mathrm{H})=8.0 \mathrm{~Hz},{ }^{3} J(6-\mathrm{H}, 8-\mathrm{H})=1.5 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right), 7.28(\mathrm{~s}, 1 \mathrm{H}, 8-$ $\mathrm{H}), 8.00(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}), 8.14-8.15(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 22.0\left(\mathrm{CCH}_{3}\right), 40.5\left(\mathrm{NCH}_{3}\right), 107.2(\mathrm{C}-3 \mathrm{a}), 118.1(\mathrm{C}-$ 8), 120.2 (C-4a), 125.4 (C-6), 126.7 (C-5), 129.0 (C-3), 145.9 (C-7), 156.1 (C-8a), 161.7 (C-9a), $174.6(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-187.0$ (N-2), $-112.0(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 215\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 237.0364, found 237.0364.

### 4.1.2.15.8. 7-Chloro-2-methyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

 (108). White solid, yield $105 \mathrm{mg}, 90 \%, \mathrm{mp} 259-260^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-}$
$\left.{ }^{1}\right): 3098,3073,3055,3025\left(\mathrm{CH}_{\text {arom }}\right), 2955,2925\left(\mathrm{CH}_{\text {aliph }}\right), 1662$ (C=O), 1603, 1578, 1558, 1483, 1458, 1435, 1417, 1398, 1314, 1289, 1272, 1205, 1183, 1163, 1105, 1081 (C-O-C, C=C, C-N), 989, 970, 957, 914, 892, 862, 820, 764, 744, 707, 675, 633, 610, 588, 535, 449, $438\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 4.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.32$ (dd, ${ }^{3} J(6-\mathrm{H}, 5-$ $\left.\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(6-\mathrm{H}, 8-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}\right), 7.51\left(\mathrm{~d},{ }^{4} J(8-\mathrm{H}, 6-\mathrm{H})=2.0\right.$ $\mathrm{Hz} 1 \mathrm{H}, 8-\mathrm{H}), 8.01(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}), 8.21\left(\mathrm{~d},{ }^{3} \mathrm{~J}(5-\mathrm{H}, 6-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\right.$ H). ${ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm} 40.7\left(\mathrm{CH}_{3}\right), 107.1(\mathrm{C}-3 \mathrm{a})$, 118.3 (C-8), 121.2 (C-4a), 124.9 (C-6), 128.2 (C-5), 129.3 (C-3), 140.3 (C-7), 156.0 (C-8a), $161.5(\mathrm{C}-9 \mathrm{a}), 173.6(\mathrm{C}=\mathrm{O}) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-185.6(\mathrm{~N}-$ 2), $-110.6(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 235\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{ClO}_{2} \mathrm{~N}_{2} \mathrm{Na}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 257.0088 , found 257.0089.

### 4.1.2.15.9. 8-Fluoro-2-methyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

 (109). White solid, yield $83 \mathrm{mg}, 76 \%, \mathrm{mp} 235-236^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3087\left(\mathrm{CH}_{\text {arom }}\right), 2952,2931\left(\mathrm{CH}_{\text {aliph }}\right), 1662(\mathrm{C}=\mathrm{O}), 1623,1617,1587,1568,1486$,
$1445,1431,1312,1269,1262,1218,1180,1161,1128,1067,1020$ (C-F, C-O-C, C=C, C-N), 973, 905, 879, 849, 751, 707, 688, 648, $614,531,424\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 4.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.27-7.31$ (m, $1 \mathrm{H}, 6-\mathrm{H}), 7.46-7.49(\mathrm{~m}, 1 \mathrm{H}, 7-\mathrm{H}), 8.04-8.05(\mathrm{~m}, 2 \mathrm{H}, 3,5-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 40.7\left(\mathrm{CH}_{3}\right), 107.1(\mathrm{C}-3 \mathrm{a})$, $120.45,120.55\left({ }^{2} J=17.6 \mathrm{~Hz}, \mathrm{C}-7\right), 121.84,121.86\left({ }^{4} J=3.5 \mathrm{~Hz}, \mathrm{C}-\right.$ 5), $123.50,123.53$ ( $\left.{ }^{3} J=5.3 \mathrm{~Hz}, \mathrm{C}-6\right), 124.6$ (C-4a), 129.3 (C-3), $144.54,144.61\left({ }^{2} J=12.3 \mathrm{~Hz}, \mathrm{C}-8 \mathrm{a}\right), 150.69,152.12\left({ }^{1} J=251.7 \mathrm{~Hz}, \mathrm{C}-8\right), 161.1$ (C-9a), $173.64,173.66\left({ }^{4} J=3.4 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right) .{ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-196.9(\mathrm{~N}-2)$, $-109.9(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 219\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{FO}_{2} \mathrm{~N}_{2} \mathrm{Na}$ $\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 241.0384, found 241.0383.

### 4.1.2.15.10. 7-Fluoro-2-methyl[1]benzopyrano[2,3-c]pyrazol-4(2H)-one

 (110). White solid, yield $78 \mathrm{mg}, 71 \%, \mathrm{mp} 239-240^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): 3099, 3067, $3044\left(\mathrm{CH}_{\text {arom }}\right), 2954,2918\left(\mathrm{CH}_{\text {aliph }}\right), 1661(\mathrm{C}=\mathrm{O}), 1622$, 1602, 1569, 1495, 1472, 1438, 1410, 1315, 1254, 1222, 1163, 1146, $1100,1080(\mathrm{C}-\mathrm{F}, \mathrm{C}-\mathrm{O}-\mathrm{C}, \mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 995,974,947,919,880,859$, $819,783,763,728,707,674,636,607,483,460,416(\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}}$ ppm $4.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.06-7.10(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{H}),, 7.17-7.19(\mathrm{~m}, 1 \mathrm{H}, 8-$ $\mathrm{H}), 8.01(\mathrm{~s}, 1 \mathrm{H}, 3-\mathrm{H}), 8.28-8.31(\mathrm{~m}, 1 \mathrm{H}, 6-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 40.6\left(\mathrm{CH}_{3}\right), 105.1,105.2\left({ }^{2} J=25.9 \mathrm{~Hz}, \mathrm{C}-8\right), 107.0(\mathrm{C}-$ 3a), $112.5,112.6\left({ }^{2} J=22.6 \mathrm{~Hz}, \mathrm{C}-6\right), 119.40,119.41\left({ }^{4} J=2.5 \mathrm{~Hz}, \mathrm{C}-4 \mathrm{a}\right)$, 129.2 (C-3), 129.27, 129.33 ( ${ }^{3} \mathrm{~J}=10.6 \mathrm{~Hz}, \mathrm{C}-5$ ), $157.03,157.11$ ( $\left.{ }^{3} J=13.4 \mathrm{~Hz}, \mathrm{C}-8 \mathrm{a}\right), 161.7$ (C-9a), $165.4,166.9\left({ }^{1} J=255.0 \mathrm{~Hz}, \mathrm{C}-7\right), 173.54(\mathrm{C}=\mathrm{O})$. ${ }^{15} \mathrm{~N}$ NMR ( $71 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-187.0(\mathrm{~N}-2),-111.7(\mathrm{~N}-1) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 219$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{FO}_{2} \mathrm{~N}_{2} \mathrm{Na}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$calcd 241.0384, found 241.0386.

### 4.1.2.16. Typical experimental procedure for the synthesis of tetracycles

 115-118.Corresponding alkyne $\mathbf{1 1 3}$ or $\mathbf{1 1 4}(0.5 \mathrm{mmol})$ and the appropriate $o$-aryldiamine ( 0.6 $\mathrm{mmol})$ were dissolved in absolute DMF ( 2 mL ), and the mixture was stirred at $120^{\circ} \mathrm{C}$ temperature for 24 hours. After the completion of reaction as indicated by TLC, the mixture was quenched with water $(10 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 10$ $\mathrm{mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $\left.1: 4, \mathrm{v} / \mathrm{v}\right)$ to provide the desired product.
4.1.2.16.1. 1-Methyl-3,5-diphenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazole (115). White solid, yield $168 \mathrm{mg}, 90 \%, \operatorname{mp} 231^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3045\left(\mathrm{CH}_{\text {arom }}\right), 2923\left(\mathrm{CH}_{\text {aliph }}\right), 1653,1596,1509,1495,1450,1371,1267(\mathrm{C}=\mathrm{C}$, $\mathrm{C}-\mathrm{N}), 761,737,710,700\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR

( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\text {н }} \mathrm{ppm} 3.06$ (s, 3H, CH3), 6.34-6.38 (m, $1 \mathrm{H}, 7-\mathrm{H}), 6.91-6.97(\mathrm{~m}, 1 \mathrm{H}, 8-\mathrm{H}), 6.93(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.34-7.39$ $(\mathrm{m}, 1 \mathrm{H}, 9-\mathrm{H}), 7.37-7.43(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.51-7.56(\mathrm{~m}, 2 \mathrm{H}$, NPh 3,5-H), 7.55-7.59 (m, 2H, CPh 2,6-H), 7.57-7.62 (m, 2H, CPh 3,5-H), 7.61-7.66 (m, 1H, CPh 4-H), 7.68-7.72 (m, 2H, NPh 2,6-H), 7.95-7.98 (m, 1H, 10-H). ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(\mathrm{CH}_{3}\right), 99.4(\mathrm{C}-4), 109.7(\mathrm{C}-11 \mathrm{~b}), 113.9$ (C-7), 119.4 (C-10), 120.6 (C-8), 123.3 (NPh C-2,6), 124.3 (C9), 127.6 ( $\mathrm{NPh} \mathrm{C}-4$ ), 129.1 (CPh C-3,5), 129.2 (CPh C-2,6), 129.6 (NPh C-3,5), 129.8 (C-6a), 130.2 (CPh C-4), 134.6 (CPh C-1), 138.7 (C-3a), 139.2 (NPh C-1), 140.2 (C5), 145.0 (C-10a), 145.8 (C-11a), $146.6(\mathrm{C}-1) .{ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}$ -205.9 (N-6), -183.3 (N-3), -73.5 (N-2). MS m/z (\%): 375 ([M+H] $\left.{ }^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 375.1604, found 375.1604.
4.1.2.16.2. 5-Butyl-1-methyl-3-phenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2a]benzimidazole (116). White solid, yield $140 \mathrm{mg}, 79 \%, \mathrm{mp} 181^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): $3048\left(\mathrm{CH}_{\text {arom }}\right), 2968\left(\mathrm{CH}_{\text {aliph }}\right), 1656,1596$, 1506, $1454(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 763,744,695(\mathrm{CH}=\mathrm{CH}$ of monoand disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 1.02\left(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.52-1.63(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.78-1.88 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.99 ( s , $\left.3 \mathrm{H}, 1-\mathrm{CH}_{3}\right), 3.24-3.29\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.79(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H})$, 7.28-7.34 (m, 1H, 8-H), 7.40-7.45 (m, 1H, NPh 4-H), 7.45$7.50(\mathrm{~m}, 1 \mathrm{H}, 9-\mathrm{H}), 7.54-7.60(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.67-7.70$ (m, 2H, NPh 2,6-H), 7.89-7.93 (m, 1H, 7-H), 7.98-8.02 (m, 1H, 10-H). ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(1-\mathrm{CH}_{3}\right), 13.9\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.3\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.4$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.8\left(\mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 96.5(\mathrm{C}-4), 109.0(\mathrm{C}-11 \mathrm{~b}), 114.0(\mathrm{C}-7), 119.6(\mathrm{C}-$ 10), 121.0 (C-8), 123.2 (NPh C-2,6), 124.2 (C-9), 127.4 (NPh C-4), 129.6 (NPh C3,5), 129.8 (C-6a), 138.7 (C-3a), 139.3 (NPh C-1), 142.2 (C-5), 145.3 (C-10a), 146.1 (C-11a), $146.4(\mathrm{C}-1) .{ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-204.9(\mathrm{~N}-6),-184.8(\mathrm{~N}-$ 3), $-75.1(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 355\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{23} \mathrm{H}_{23} \mathrm{~N}_{4}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 355.1917, found 355.1917.
4.1.2.16.3. 1,9-Dimethyl-3,5-diphenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2-
a]benzimidazole (117a). White solid, yield $146 \mathrm{mg}, 75 \%, \mathrm{mp} 223^{\circ} \mathrm{C}$ (ethyl acetate).
 IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3055\left(\mathrm{CH}_{\text {arom }}\right), 2922\left(\mathrm{CH}_{\text {aliph }}\right), 1655,1597$, 1507, 1372, $1275(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 760,704(\mathrm{C}=\mathrm{C}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm}$ $2.46\left(\mathrm{~s}, 3 \mathrm{H}, 9-\mathrm{CH}_{3}\right), 3.04\left(\mathrm{~s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}\right), 6.22\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\right.$ $\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}), 6.76\left(\mathrm{dd},{ }^{3} \mathrm{~J}(8-\mathrm{H}, 7-\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(8-\right.$ $\mathrm{H}, 10-\mathrm{H})=1.3 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}), 6.91(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.36-6.41(\mathrm{~m}$, 1 H , NPh 4-H), 7.50-7.55 (m, 2H, NPh 3,5-H), 7.55-7.58 (m, $2 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}), 7.56-7.60(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CPh} 3,5-\mathrm{H}), 7.61-7.66$ (m, 1H, CPh 4-H), 7.67-7.71 (m, 2H, NPh 2,6-H) 7.74 (d, $\left.{ }^{4} J(10-\mathrm{H}, 8-\mathrm{H})=1.3 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$,
$\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(1-\mathrm{CH}_{3}\right), 21.5\left(9-\mathrm{CH}_{3}\right), 99.0(\mathrm{C}-4), 109.7(\mathrm{C}-11 \mathrm{~b}), 113.3(\mathrm{C}-7)$, 119.1 (C-10), 122.1 (C-8), 123.3 (NPh C-2,6), 127.5 (NPh C-4), 127.8 (C-6a), 129.0 (CPh C-2,6), 129.2 (CPh C-3,5), 129.6 (NPh C-3,5), 130.1 (CPh C-4), 134.1 (C-9), 134.7 (CPh C-1), 138.6 (C-3a), 139.2 (NPh C-1), 140.1 (C-5), 145.4 (C-10a), 145.8 (C-11a), $146.5(\mathrm{C}-1) .{ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-206.1(\mathrm{~N}-6),-183.6(\mathrm{~N}-$ 3), -73.7 (N-2). MS m/z (\%): $389\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, 100). HRMS (ESI) for $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{4}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 389.1761, found 389.1760.

### 4.1.2.16.4. 9-Chloro-1-methyl-3,5-diphenyl-3H-pyrazolo[4',3':3,4]pyrido

 [1,2-a]benzimidazole (117b). White solid, yield $157 \mathrm{mg}, 77 \%, \mathrm{mp} 257^{\circ} \mathrm{C}$ (ethyl ac- etate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3055\left(\mathrm{CH}_{\text {arom }}\right), 2923\left(\mathrm{CH}_{\text {aliph }}\right), 1652$, 1595, 1507, 1431, 1372 (C=C, C-N), 810, 761, 731, 704 (C$\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 3.03\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.23\left(\mathrm{~d},{ }^{3} J(7-\right.$ $\mathrm{H}, 8-\mathrm{H})=8.9 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}), 6.88\left(\mathrm{dd},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=8.9 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $8-\mathrm{H}), 6.97(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.38-7.43(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.51-$ 7.56 (m, 2H, NPh 3,5-H), 7.55-7.58 (m, 2H, CPh 2,6-H), 7.587.63 (m, 2H, CPh 3,5-H), 7.62-7.66 (m, 1H, CPh 4-H), 7.65$7.71(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 7.90-7.92(\mathrm{~m}, 1 \mathrm{H}, 10-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(\mathrm{CH}_{3}\right), 99.9(\mathrm{C}-4), 109.4(\mathrm{C}-$ 11 b ), 114.5 (C-7), 118.6 (C-10), 121.0 (C-8), 123.4 (NPh C-2,6), 127.8 (NPh C-4), 128.3 (C-6a), 129.2 (CPh C-2,6), 129.3 (CPh C-3,5), 129.6 (NPh C-3,5), 130.0 (C-9), 130.4 (CPh C-4), 134.1 (CPh C-1), 138.8 (C-3a), 139.0 (NPh C-1), 139.9 (C-5), 145.4 (C-10a), 146.4 (C-11a), 146.7 (C-1). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-206.5(\mathrm{~N}-$ 6), -182.8 (N-3), -72.0 (N-2). MS m/z (\%): 411 ([M+2] $\left.{ }^{+}, 30\right), 409\left(\mathrm{M}^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{25} \mathrm{H}_{18} \mathrm{ClN}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 409.1215, found 409.1212.
4.1.2.16.5. 1-Methyl-9-nitro-3,5-diphenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2a]benzimidazole (117c). Yellow solid, yield $100 \mathrm{mg}, 48 \%, \mathrm{mp} 296^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3058\left(\mathrm{CH}_{\text {arom }}\right), 2930\left(\mathrm{CH}_{\text {aliph }}\right), 1655,1596$, 1511, 1438, 1340 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{C}-\mathrm{NO}_{2}$ ), 817, 760, 720, 707, $696\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.37$ $\left(\mathrm{d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=9.2 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 7.03(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.41-$ 7.46 (m, 1H, NPh 4-H), 7.53-7.58 (m, 2H, NPh 3,5-H), 7.56-7.59 (m, 2H, CPh 2,6-H), 7.62-7.66 (m, 2H, CPh 3,5H), 7.67-7.71 (m, 2H, NPh 2,6-H), 7.68-7.73 (m, 1H, CPh $4-\mathrm{H}), 7.82\left(\mathrm{dd},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=9.2 \mathrm{~Hz},{ }^{4} J(8-\mathrm{H}, 10-\mathrm{H})=2.3 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 8-\mathrm{H}), 8.78\left(\mathrm{~d},{ }^{4} J(10-\mathrm{H}, 8-\mathrm{H})=2.3 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}}$ ppm $13.7\left(\mathrm{CH}_{3}\right), 101.0(\mathrm{C}-4), 109.4(\mathrm{C}-11 \mathrm{~b}), 113.7(\mathrm{C}-7), 115.2(\mathrm{C}-10), 115.7(\mathrm{C}-8)$, 123.5 (NPh C-2,6), 128.0 (NPh C-4), 129.2 (CPh C-2,6), 129.5 (CPh C-3,5), 129.7 (NPh C-3,5), 130.7 (CPh C-4), 133.7 (C-6a), 133.8 (CPh C-1), 138.7 (C-3a), 138.8 (NPh C-1), 139.9 (C-5), 144.6 (C-9), 144.8 (C-10a), 147.0 (C-1), 148.4 (C-11a). ${ }^{15} \mathrm{~N}$ NMR (40 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-205.4(\mathrm{~N}-6),-181.6(\mathrm{~N}-3),-71.1(\mathrm{~N}-2),-11.6$
$\left(\mathrm{NO}_{2}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 420\left([\mathrm{M}+1]^{+}, 100\right) . \mathrm{HRMS}(\mathrm{ESI})$ for $\mathrm{C}_{25} \mathrm{H}_{18} \mathrm{~N}_{5} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 420.1455 , found 420.1458 .
4.1.2.16.6. 5-Butyl-1,9-dimethyl-3-phenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2$a$ ]benzimidazole (117d) and 5-Butyl-1,8-dimethyl-3-phenyl-3Hpyrazolo[4', 3':3,4]py-rido[1,2-a]benzimidazole (118d). White solids, yield 129 mg ,
 $70 \%$. Compounds were obtained as an inseparable mixture in ratio 1:0.55 (117d:118d).
4.1.2.16.6.1. 117d: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.01(\mathrm{t}$, $J=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), $1.51-1.63$ ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.77-1.87 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.55 ( $\mathrm{s}, 3 \mathrm{H}$, $\left.9-\mathrm{CH}_{3}\right), 2.98\left(\mathrm{~s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}\right), 3.21-$ 3.27 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), 6.77 ( $\mathrm{s}, 1 \mathrm{H}$, $4-\mathrm{H}), 7.12\left(\mathrm{~d},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=8.3 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}\right), 7.39-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.53-7.60$ (m, 2H, NPh 3,5-H), 7.66-7.70 (m, 2H, NPh 2,6-H), $7.76\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.4 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $7-\mathrm{H}), 7.79(\mathrm{~s}, 1 \mathrm{H}, 10-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(1-\mathrm{CH}_{3}\right), 13.9$ $\left(\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 21.5\left(9-\mathrm{CH}_{3}\right), 22.3\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.8\left(\mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{7}\right)$, 96.1 (C-4), 109.1 (C-11b), 113.4 (C-7), 119.5 (C-10), 122.5 (C-8), 123.2 (NPh C-2,6), 127.3 (NPh C-4), 127.8 (C-6a), 129.6 (NPh C-3,5), 134.0 (C-9), 138.7 (C-3a), 139.3 (NPh C-1), 142.1 (C-5), 145.7 (C-10a), 146.2 (C-11a), 146.4 (C-1). ${ }^{15} \mathrm{~N}$ NMR (40 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-205.2(\mathrm{~N}-6),-185.0(\mathrm{~N}-3),-75.4(\mathrm{~N}-2)$.
4.1.2.16.6.2. 118d: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.03(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.52-1.64 (m, 2H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.78-1.88 (m, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 2.57$ ( $\mathrm{s}, 3 \mathrm{H}, 9-\mathrm{CH}_{3}$ ), $2.98\left(\mathrm{~s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}\right), 3.26\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 6.77(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H})$, $7.30\left(\mathrm{~d},{ }^{3} J(9-\mathrm{H}, 10-\mathrm{H})=8.4 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 7.39-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.53-7.60(\mathrm{~m}$, 2H, NPh 3,5-H), 7.67-7.71 (m, 2H, NPh 2,6-H), 7.70 (s, 1H, 7-H), 7.88 (d, ${ }^{3} J(10-\mathrm{H}, 9-$ $\mathrm{H})=8.3 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(1-\mathrm{CH}_{3}\right), 13.8$ $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.2\left(8-\mathrm{CH}_{3}\right), 22.3\left(\mathrm{C}_{2} \mathrm{H}_{4} C H_{2} \mathrm{CH}_{3}\right), 29.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.8\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, 96.4 (C-4), 109.1 (C-11b), 114.0 (C-7), 119.1 (C-10), 123.2 (NPh C-2,6), 125.6 (C9), 127.4 (NPh C-4), 129.6 (NPh C-3,5), 130.0 (C-6a), 130.7 (C-8), 138.6 (C-3a), 139.3 (NPh C-1), 142.2 (C-5), 143.3 (C-10a), 145.9 (C-11a), 146.3 (C-1). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-205.2(\mathrm{~N}-6),-185.0(\mathrm{~N}-3),-75.4(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 369$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. Calcd. for $\mathrm{C}_{24} \mathrm{H}_{24} \mathrm{~N}_{4}$ (368.47): C, 78.23; H, 6.57; N, 15.21. Found: C, 77.84; H, 6.57; N, 15.17.
4.1.2.16.7. 5-Butyl-9-chloro-1-methyl-3-phenyl-3H-pyrazolo[4',3':3,4]pyri-do[1,2-a]benzimidazole (117e). White solid, yield $117 \mathrm{mg}, 60 \%$, mp $166^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3064\left(\mathrm{CH}_{\text {arom }}\right), 2963\left(\mathrm{CH}_{\text {aliph }}\right), 1658,1597,1505,1453,1422$, $1274(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 920,804,761,696(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes) ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.02\left(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right.$ ), 1.51-1.62 (m, 2H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.74-1.84 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.94 ( $\mathrm{s}, 3 \mathrm{H}, 1-$ CH3), 3.15-3.21 (m, 2H, $\left.\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.80(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.20\left(\mathrm{dd},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=8.9 \mathrm{~Hz}\right.$,

$\left.{ }^{4} J(8-\mathrm{H}, 10-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}\right), 7.40-7.46(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-$ H), 7.54-7.60 (m, 2H, NPh 3,5-H), 7.64-7.69 (m, 2H, NPh $2,6-\mathrm{H}), 7.73\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.9 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 7.90(\mathrm{~d}$, $\left.{ }^{4} J(10-\mathrm{H}, 8-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.7\left(1-\mathrm{CH}_{3}\right), 13.9\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.3$ $\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $29.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $33.6\left(\mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{7}\right), 96.8$ (C-4), 108.9 (C-11b), 114.5 (C-7), 119.1 (C-10), 121.1 (C8), 123.2 (NPh C-2,6), 127.5 (NPh C-4), 128.3 (C-6a), 129.6 (NPh C-3,5), 129.7 (C-9), 138.8 (C-3a), 139.1 (NPh C-1), 141.9 (C-5), 146.2 (C-10a), 146.4 (C-1), 147.0 (C-11a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-205.5(\mathrm{~N}-6),-184.4(\mathrm{~N}-3),-74.0(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 391$ ( $\left.[\mathrm{M}+2]^{+}, 39\right), 389$ $\left(\mathrm{M}^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{ClN}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 389.1533, found 389.1536.

### 4.1.2.16.8. 5-Butyl-1-methyl-8-nitro-3-phenyl-3H-pyrazolo[4',3':3,4]pyri-do[1,2-a]benzimidazo-le (118f). White solid, yield $130 \mathrm{mg}, 65 \%$, mp $237^{\circ} \mathrm{C}$ (ethyl

 acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3073\left(\mathrm{CH}_{\text {arom }}\right), 2964\left(\mathrm{CH}_{\text {aliph }}\right)$, 1658, 1594, 1505, 1333, 1278 ( $\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}, \mathrm{NO}_{2}$ ), 763, 733, $696(\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.06(\mathrm{t}$, $\left.J=7.3 \mathrm{~Hz}, \quad 3 \mathrm{H}, \quad \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), \quad 1.62-1.72 \quad(\mathrm{~m}, \quad 2 \mathrm{H}$, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $1.84-1.93\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 2.98$ ( $\mathrm{s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}$ ), 3.33-3.39 (m, 2H, $\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), 6.98 ( s , $1 \mathrm{H}, 4-\mathrm{H}), 7.44-7.50(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.57-7.63(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.66-7.70(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 7.95\left(\mathrm{~d},{ }^{3} J(10-\mathrm{H}, 9-\mathrm{H})=9.0 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $10-\mathrm{H}) 8.39\left(\mathrm{dd},{ }^{3} J(9-\mathrm{H}, 10-\mathrm{H})=9.0 \mathrm{~Hz},{ }^{4} J(9-\mathrm{H}, 7-\mathrm{H})=1.9 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 8.91\left(\mathrm{~d},{ }^{4} J(7-\right.$ $\mathrm{H}, 9-\mathrm{H})=1.9 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.6\left(1-\mathrm{CH}_{3}\right), 13.8$ $\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right)$, $22.3\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $29.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $33.7\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right)$, $98.1(\mathrm{C}-4)$, 109.1 (C-11b), 111.1 (C-7), 118.9 (C-10), 120.2 (C-9), 123.4 (NPh C-2,6), 128.0 ( NPh C-4), 128.7 (C-6a), 129.7 (NPh C-3,5), 138.9 (NPh C-1), 139.4 (C-3a), 141.4 (C-8), 142.2 (C-5), 146.8 (C-1), 146.8 (C-11a), 149.9 (C-10a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-204.1(\mathrm{~N}-6),-183.4(\mathrm{~N}-3),-71.5(\mathrm{~N}-2),-11.8\left(\mathrm{NO}_{2}\right) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 400$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{~N}_{5} \mathrm{O}_{2}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 400.1768, found 400.1766.

### 4.1.2.16.9. Bromination of pyrazole 124 and subsequent cross-coupling reactions

### 4.1.2.16.9.1. 8-Bromo-1-methyl-3,5-diphenyl-3H-pyrazolo[4',3':3,4]pyri-do[1,2-a]benzimidazole (119b).

Compound 115 (1.1 g, 4 mmol ) was dissolved in DMF ( 5 mL ), and N bromosuccinimide ( $720 \mathrm{mg}, 4 \mathrm{mmol}$ ) was added. After stirring for 24 hours at room temperature, water was added; the produced solid was filtered off with suction and subsequently recrystalized from chloroform. White solid, yield $1.20 \mathrm{~g}, 66 \%$, mp 283 ${ }^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3054\left(\mathrm{CH}_{\text {arom }}\right), 2963\left(\mathrm{CH}_{\text {aliph }}\right), 1652,1595,1453$, 1262, 1096, 1071, $1023(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 802,761,707(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 3.05\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.41$

$\left(\mathrm{d},{ }^{4} J(7-\mathrm{H}, 9-\mathrm{H})=1.4 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 6.99(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.38-$ 7.43 (m, 1H, NPh 4-H), 7.46 (dd, ${ }^{3} J(9-\mathrm{H}, 10-\mathrm{H})=8.6 \mathrm{~Hz}$, $\left.{ }^{4} J(7-\mathrm{H}, 9-\mathrm{H})=1.5 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 7.51-7.56(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh}$ $3,5-\mathrm{H}), 7.54-7.58$ (m, 2H, CPh 2,6-H), 7.60-7.65 (m, 2H, CPh 3,5-H), 7.68-7.71 (m, 3H, CPh 4-H, NPh 2,6-H), 7.84 $\left(\mathrm{d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.4 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.7\left(\mathrm{CH}_{3}\right), 100.1(\mathrm{C}-4), 109.5(\mathrm{C}-11 \mathrm{~b})$, 113.6 (C-8), 117.0 (C-7), 120.1 (C-10), 123.4 (NPh C-2,6), 127.7 (C-9), 127.8 (NPh C-4), 129.1 (CPh C-2,6), 129.3 (CPh C-3,5), 129.7 (NPh C3,5), 130.7 (C-6a), 130.5 (CPh C-4), 133.9 (CPh C-1), 138.7 (C-3a), 139.0 (NPh C1), 139.9 (C-5), 143.0 (C-10a), 146.1 (C-11a), 146.7 (C-1). ${ }^{15} \mathrm{~N}$ NMR ( 40 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-206.6(\mathrm{~N}-6),-182.7(\mathrm{~N}-3),-71.7(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 455\left([\mathrm{M}+2]^{+}\right.$, 89), $453\left(\mathrm{M}^{+}, 89\right)$. HRMS (ESI) for $\mathrm{C}_{25} \mathrm{H}_{18} \mathrm{BrN}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 453.0709, found 453.0711 .

### 4.1.2.16.9.2. 1-Methyl-3,5,8-triphenyl-3H-pyrazolo[4',3':3,4]pyrido[1,2-

 a]benzimidazole (120). Into the solution of pyrazole $\mathbf{1 1 9 b}(227 \mathrm{mg}, 0.5 \mathrm{mmol})$ in $\mathrm{EtOH}(2.5 \mathrm{~mL})$ under argon atmosphere, phenylboronic acid ( $79 \mathrm{mg}, 0.65 \mathrm{mmol}$ ), 1 M aqueous $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ solution $(1 \mathrm{~mL})$, and $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(40 \mathrm{mg}, 7 \mathrm{~mol} \%)$ were added, and the reaction mixture was irradiated $(50 \mathrm{~W})$ at $100^{\circ} \mathrm{C}$ temperature for 10 min . After the completion of reaction as indicated by TLC, the mixture was quenched with water $(10 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 10 \mathrm{~mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{NaSO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified by column chromatography ( $\mathrm{SiO}_{2}$, eluent: ethyl acetate $/ n$-hexane, $1: 3, \mathrm{v} / \mathrm{v}$ ). White solid, yield $129 \mathrm{mg}, 57 \%, \mathrm{mp} 241{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3056\left(\mathrm{CH}_{\text {arom }}\right), 2923\left(\mathrm{CH}_{\text {aliph }}\right), 1651,1596,1510,1495,1468$, $1375(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 762,745,705\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 3.06\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.56\left(\mathrm{~d},{ }^{4} J(7-\mathrm{H}, 9-\mathrm{H})=1.4 \mathrm{~Hz}, 1 \mathrm{H}\right.$, 7-H), 6.98 ( $\mathrm{s}, 1 \mathrm{H}, 4-\mathrm{H}$ ), 7.23-7.28 (m, 1H, 8-CPh 4-H), 7.29-7.33 (m, 2H, 8-CPh 2,6H), 7.32-7.37 (m, 2H, 8-CPh 3,5-H), 7.38-7.43 (m, 1H, NPh 4-H), 7.52-7.57 (m, 2H, NPh 3,5-H), 7.61-7.64 (m, 2H, 5-CPh 2,6-H), 7.62-7.66 (m, 2H, 5-CPh 3,5-H), 7.64$7.66(\mathrm{~m}, 1 \mathrm{H}, 9-\mathrm{H}), 7.64-7.68(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{CPh} 4-\mathrm{H}), 7.70-7.73(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H})$, $7.99\left(\mathrm{~d},{ }^{3} J(10-\mathrm{H}, 9-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.8$ $\left(\mathrm{CH}_{3}\right), 99.3(\mathrm{C}-4), 109.8(\mathrm{C}-11 \mathrm{~b}), 112.4(\mathrm{C}-7), 113.7(\mathrm{C}-8), 119.3(\mathrm{C}-10), 123.4(\mathrm{NPh}$ C-2,6), 123.8 (C-9), 126.6 (8-CPh C-4), 126.9 (8-CPh C-2,6), 127.4 (NPh C-4), 128.7 (8-CPh C-3,5), 129.2 (5-CPh C-3,5), 129.3 (5-CPh C-2,6), 129.6 (NPh C-3,5), 130.2 (5-CPh C-4), 130.4 (C-6a), 134.6 (5-CPh C-1), 138.7 (C-3a), 139.1 (NPh C-1), 140.2 (C-5), 141.3 (8-CPh C-1), 144.6 (C-10a), 146.3 (C-11a), 146.6 (C-1). ${ }^{15} \mathrm{~N}$ NMR (40 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-205.7(\mathrm{~N}-6),-183.0(\mathrm{~N}-3),-73.2(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 451$ $\left([\mathrm{M}+1]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{31} \mathrm{H}_{23} \mathrm{~N}_{4}\left([\mathrm{M}+1]^{+}\right)$calcd 451.1917, found 451.1920.
4.1.2.16.9.3. 1-Methyl-3,5-diphenyl-8-(phenylethynyl)-3H-pyrazolo[4',3': 3,4]pyrido[1,2-a $]$ benz-imidazole (121). Into the solution of 119b ( $227 \mathrm{mg}, 0.5$ mmol ) in dry DMF ( 2.5 mL ) under argon atmosphere triethylamine ( $0.11 \mathrm{~mL}, 0.75$
 $\mathrm{mmol})$, phenylacetylene ( $0.08 \mathrm{~mL}, 0.75 \mathrm{mmol}$ ), $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(35 \mathrm{mg}, 0.05 \mathrm{mmol})$, and $\mathrm{CuI}(10 \mathrm{mg}$, 0.05 mmol ) were added, and the reaction mixture was irradiated $(100 \mathrm{~W})$ at $130{ }^{\circ} \mathrm{C}$ temperature for 10 min . After the completion of reaction as indicated by TLC, the mixture was quenched with water $(10 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 10$ mL ). The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$, eluent: ethyl acetate $/ n$-hexane, $1: 5, \mathrm{v} / \mathrm{v})$. White solid, yield $140 \mathrm{mg}, 59 \%, \mathrm{mp} 245^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-1}$ ): $3057\left(\mathrm{CH}_{\text {arom }}\right), 2922\left(\mathrm{CH}_{\text {aliph }}\right), 2210(\mathrm{C} \equiv \mathrm{C}), 1651,1596,1510,1493,1375(\mathrm{C}=\mathrm{C}, \mathrm{C}-$ $\mathrm{N}), 808,763,703\left(\mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 3.05\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 6.51-6.52(\mathrm{~m}, 1 \mathrm{H}, 7-\mathrm{H}), 6.97(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H})$, $7.29-7.33(\mathrm{~m}, 1 \mathrm{H}, \mathrm{C} \equiv \mathrm{CPh} 4-\mathrm{H}), 7.31-7.35(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C} \equiv \mathrm{CPh} 3,5-\mathrm{H}), 7.39-7.44(\mathrm{~m}, 1 \mathrm{H}$, NPh 4-H), 7.42-7.46 (m, 2H, C $\equiv \mathrm{CPh} 2,6-\mathrm{H}), 7.52-7.55(\mathrm{~m}, 1 \mathrm{H}, 9-\mathrm{H}), 7.52-7.57(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.57-7.60(\mathrm{~m}, 2 \mathrm{H}, 5-\mathrm{CPh} 2,6-\mathrm{H}), 7.61-7.66(\mathrm{~m}, 2 \mathrm{H}, 5-\mathrm{CPh} 3,5-\mathrm{H})$, $7.65-7.70(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{CPh} 4-\mathrm{H}), 7.68-7.72(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 7.89\left(\mathrm{~d},{ }^{3} \mathrm{~J}(10-\mathrm{H}, 9-\right.$ $\mathrm{H})=8.4 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(\mathrm{CH}_{3}\right), 88.0$ $(\mathrm{C} \equiv C \mathrm{Ph}), 90.3(\mathrm{C} \equiv \mathrm{CPh}), 99.8(\mathrm{C}-4), 109.7(\mathrm{C}-11 \mathrm{~b}), 115.0(\mathrm{C}-8), 117.4(\mathrm{C}-7), 119.2$ (C-10), 123.4 (NPh C-2,6), 123.5 (C =CPh C-1), 127.7 (NPh C-4), 127.9 (C-9), 128.1 (C $\equiv \mathrm{CPh} \mathrm{C}-4$ ), 128.3 (C $\equiv \mathrm{CPh} \mathrm{C}-3,5$ ), 129.1 (5-CPh C-2,6), 129.3 (5-CPh C-3,5), 129.6 (NPh C-3,5), 129.7 (C-6a), 130.3 (5-CPh C-4), 131.4 (C =CPh C-2,6), 134.2 (5-CPh C-1), 138.8 (C-3a), 139.1 (NPh C-1), 140.2 (C-5), 145.3 (C-10a), 146.7 (C-1), 146.8 (C-11a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-206.3(\mathrm{~N}-6),-182.9(\mathrm{~N}-3),-72.7(\mathrm{~N}-$ 2). MS m/z (\%): $475\left([\mathrm{M}+1]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{33} \mathrm{H}_{23} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 475.1917, found 475.1922.

### 4.1.2.16.10. Preparation of $3 \boldsymbol{H}$-pyrazolo[4,3-c]imidazo[1,2-a:5,4-b’]dipyridines

4.1.2.16.10.1. 1-Methyl-3,5-diphenyl-3H-pyrazolo[4,3-c]imidazo[1,2-a:5,4$b^{\prime}$ ']dipyridine (122).


Compound 122 was synthesized from 113 and pyridine-2,4diamine in the same manner as described for the preparation of compounds 115-118, except that the reaction period was 72 hours. White solid, yield $66 \mathrm{mg}, 35 \%$, mp $227{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3044\left(\mathrm{CH}_{\text {arom }}\right), 2959\left(\mathrm{CH}_{\text {aliph }}\right), 1654$, $1598,1510,1399(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 770,756,713,699(\mathrm{CH}=\mathrm{CH}$ of monosubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}}$ ppm 3.04 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 6.97 ( s, 1H, 4-H), 7.32 (dd, ${ }^{3} J(9-\mathrm{H}, 10-$ $\left.\mathrm{H})=8.1 \mathrm{~Hz},{ }^{3} J(9-\mathrm{H}, 8-\mathrm{H})=4.7 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 7.39-7.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.48-7.54$ (m, 2H, CPh 3,5-H), 7.52-7.58 (m, 3H, CPh 4-H, NPh 3,5-H), 7.59-7.63 (m, 2H, CPh

2,6-H), 7.68-7.72 (m, 2H, NPh 2,6-H), 8.11 (dd, ${ }^{3} J(8-H, 9-H)=4.7 \mathrm{~Hz},{ }^{4} J(8-\mathrm{H}, 10-$ $\mathrm{H})=1.5 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}), 8.21\left(\mathrm{dd},{ }^{3} J(10-\mathrm{H}, 9-\mathrm{H})=8.1 \mathrm{~Hz},{ }^{4} J(10-\mathrm{H}, 8-\mathrm{H})=1.5 \mathrm{~Hz}, 1 \mathrm{H}, 10-\right.$ H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(\mathrm{CH}_{3}\right), 100.9(\mathrm{C}-4), 109.5(\mathrm{C}-11 \mathrm{~b})$, 120.1 (C-9), 126.3 (C-10), 123.5 (NPh C-2,6), 127.8 (NPh C-4), 127.8 (CPh C-3.5), 129.5 (CPh C-2,6), 129.7 (NPh C-3,5), 129.4 (CPh C-4), 134.4 (CPh C-1), 137.1 (C10a), 139.0 (NPh C-1), 139.3 (C-3a), 140.5 (C-5), 141.3 (C-8), 144.1 (C-6a), 146.4 (C-11a), $146.7(\mathrm{C}-1) .{ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-206.5(\mathrm{~N}-6),-182.3$ ( $\mathrm{N}-$ 3), -104.3 ( $\mathrm{N}-7$ ), $-72.5(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 376\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{~N}_{5}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 376.1557 , found 376.1559 .

4.1.2.16.10.2. 5-Butyl-1-methyl-3-phenyl-3H-pyrazolo[4,3-c]imidazo[1,2$a: 5, \mathbf{4}-\boldsymbol{b}^{\prime}$ ]dipyridine (123). This compound was synthesized in analogy to $\mathbf{1 2 2}$, except
 that $\mathbf{1 1 4}$ was used as the educt. White solid, yield 71 mg , $40 \%, \mathrm{mp} 161{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\mathrm{v}_{\text {max }}, \mathrm{cm}^{-1}$ ): 3031 $\left(\mathrm{CH}_{\text {arom }}\right), 2966\left(\mathrm{CH}_{\text {aliph }}\right), 1655,1596,1504,1397(\mathrm{C}=\mathrm{C}, \mathrm{C}-$ $\mathrm{N}), 758,696\left(\mathrm{CH}=\mathrm{CH}\right.$ of monosubstitu-ted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 0.99(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.50-1.60 (m, 2H, C2 $\mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.76-1.86 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), 2.98 ( $\mathrm{s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}$ ), $3.68-3.74(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}$ ), $6.85(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.38-7.42(\mathrm{~m}, 1 \mathrm{H}, 9-\mathrm{H}), 7.41-$ 7.46 (m, 1H, NPh 4-H), 7.55-7.61 (m, 2H, NPh 3,5-H), 7.67-7.71 (m, 2H, NPh 2,6$\mathrm{H}), 8.19-8.23(\mathrm{~m}, 1 \mathrm{H}, 10-\mathrm{H}), 8.40-8.43(\mathrm{~m}, 1 \mathrm{H}, 8-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.8\left(1-\mathrm{CH}_{3}\right), 13.9\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.3\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 30.9\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $32.8\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 97.2(\mathrm{C}-4), 108.8(\mathrm{C}-11 \mathrm{~b}), 119.9(\mathrm{C}-9), 123.4(\mathrm{NPh} \mathrm{C}-2,6), 126.1$ (C-10), 127.6 (NPh C-4), 129.6 (NPh C-3,5), 137.1 (C-10a), 139.2 (NPh C-1), 139.6 (C-3a), 141.3 (C-8), 143.8 (C-5), 144.8 (C-6a), 146.5 (C-11a), 146.6 (C-1). ${ }^{15}$ N NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-204.9(\mathrm{~N}-6),-184.1(\mathrm{~N}-3),-104.8(\mathrm{~N}-7),-74.6(\mathrm{~N}-2)$. MS m/z (\%): $356\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{22} \mathrm{H}_{22} \mathrm{~N}_{5}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 356.1870, found 356.1872 .

### 4.1.2.16.11. Preparation of $13,13 a$-dihydro-3H-pyrazolo[4',3':3,4]pyrido [1,2-a] perimidines.

4.1.2.16.11.1. 1-Methyl-3,5-diphenyl-13,13a-dihydro-3H-pyrazolo[4',3':3,4]pyrido[1,2-a]perimidine (124).


Compound 124 was synthesized from 113 and 1,8 naphthalenediamine in the same manner as described for the preparation of compounds $\mathbf{1 1 5}$, except that $\mathrm{CuI}(10 \mathrm{mg}, 10$ $\mathrm{mol} \%$ ) was used as a catalyst, and the reaction mixture was stirred in DMF ( 2 mL ) for 40 min at $140^{\circ} \mathrm{C}$ temperature under microwave irradiation ( 150 W ). White solid, yield 96 mg , $45 \%, \mathrm{mp} 257^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}, \mathrm{cm}^{-1}$ ): $3234(\mathrm{~N}-\mathrm{H})$, $3052\left(\mathrm{CH}_{\text {arom }}\right), 2920\left(\mathrm{CH}_{\text {aliph }}\right), 1592,1562,1413,1300,1278$ (C=C, C-N), 815, 770, 760, 704 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.42\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.58$ (br s, 1H, N-H), $5.69(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 6.22\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=7.4 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 6.41(\mathrm{~s}$,
$1 \mathrm{H}, 13 \mathrm{a}-\mathrm{H}), 6.81\left(\mathrm{dd},{ }^{3} J(12-\mathrm{H}, 11-\mathrm{H})=6.5 \mathrm{~Hz},{ }^{4} J(12-\mathrm{H}, 10-\mathrm{H})=1.8 \mathrm{~Hz}, 1 \mathrm{H}, 12-\mathrm{H}\right)$, 6.85-7.02 (br m, 2H, CPh 2,6-H), 6.93-6.97 (m, 1H, 8-H), 7.06-7.14 (m, 2H, CPh 3,5-H), 7.14-7.20 (m, 1H, CPh 4-H), 7.30-7.35 (m, 1H, NPh 4-H), 7.36-7.42 (m, 2H, $10,11-\mathrm{H}), 7.40\left(\mathrm{~d},{ }^{3} J(9-\mathrm{H}, 8-\mathrm{H})=8.3 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 7.44-7.50(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H})$, 7.58-7.62 (m, 2H, NPh 2,6-H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{c}} \mathrm{ppm} 12.0\left(\mathrm{CH}_{3}\right), 66.5$ (C-13a), 91.2 (C-4), 106.6 (C-13b), 109.0 (C-12), 118.4 (C-7), 119.2 (C-10), 120.1 (C-12b), 123.0 (C-9), 123.2 (NPh C-2,6), 124.8 (C-8), 126.7 (C-11), 126.8 (NPh C4), 127.7 (CPh C-4), 127.8 (CPh C-3,5), 128.5 (CPh C-2,6), 129.3 (NPh C-3,5), 134.6 (C-9a), 137.8 (CPh C-1), 138.0 (C-3a), 138.4 (C-6a), 139.7 (NPh C-1), 141.5 (C-12a), 144.5 (C-5), 145.5 (C-1) ppm. ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-293.2(\mathrm{~N}-13)$, $-275.7(\mathrm{~N}-6),-181.2(\mathrm{~N}-3),-85.5(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 427\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. Calcd for $\mathrm{C}_{29} \mathrm{H}_{22} \mathrm{~N}_{4}$ (426.51): C, 81.66; H, 5.20; N, 13.14. Found: C, 81.73; H, 5.60; N, 13.05.
4.1.2.16.11.2. 5-Butyl-1-methyl-3-phenyl-13,13a-dihydro-3H-
pyrazolo[4',3':3,4]pyrido[1,2-a]perimidine (125). This compound was synthesized
 in analogy to $\mathbf{1 2 4}$, except that $\mathbf{1 1 4}$ was used as the educt. White solid, yield $90 \mathrm{mg}, 44 \%, \mathrm{mp} 194^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3262(\mathrm{~N}-\mathrm{H}), 3050\left(\mathrm{CH}_{\text {arom }}\right), 2951\left(\mathrm{CH}_{\text {aliph }}\right)$, 1596, 1569, 1505, 1412, 1275 (C=C, C-N), 817, 771, 694 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 0.60(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 0.83-0.95 (m, 1H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 0.98-1.10 (m, $1 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 0.98-1.16 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.16-2.25 $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 2.35\left(\mathrm{~s}, 3 \mathrm{H}, 1-\mathrm{CH}_{3}\right), 2.46-2.54(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{CH}_{2}\left(\mathrm{C}_{3} \mathrm{H}_{7}\right), 4.36(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{N}-\mathrm{H}), 5.65(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 6.14(\mathrm{~s}, 1 \mathrm{H}, 13 \mathrm{a}-\mathrm{H}), 6.67-6.72$ $(\mathrm{m}, 1 \mathrm{H}, 12-\mathrm{H}), 7.08\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=7.4 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 7.30-7.35(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H})$, $7.33-7.36(\mathrm{~m}, 2 \mathrm{H}, 10,11-\mathrm{H}), 7.37-7.42(\mathrm{~m}, 1 \mathrm{H}, 8-\mathrm{H}), 7.45-7.51(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H})$, 7.57-7.61 (m, 2H, NPh 2,6-H), 7.59-7.63 (m, 1H, 9-H). $\left.{ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{(100} \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{C}} \mathrm{ppm} 11.8\left(1-\mathrm{CH}_{3}\right), 13.6\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.0\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, $33.4\left(\mathrm{CH}_{2}\left(\mathrm{C}_{3} \mathrm{H}_{7}\right), 66.0(\mathrm{C}-13 \mathrm{a}), 88.6(\mathrm{C}-4), 106.4(\mathrm{C}-13 \mathrm{~b}), 108.5(\mathrm{C}-12), 118.0(\mathrm{C}-7)\right.$, 118.5 (C-10), 120.2 (C-12b), 123.1 (NPh C-2,6), 124.4 (C-9), 125.1 (C-8), 126.7 (NPh C-4), 126.9 (C-11), 129.2 (NPh C-3,5), 134.7 (C-9a), 138.2 (C-3a), 138.6 (C-6a), 139.8 (NPh C-1), 141.7 (C-12a), 145.0 (C-1), 146.1 (C-5). ${ }^{15} \mathrm{~N}$ NMR ( 40 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-296.0(\mathrm{~N}-13),-276.5(\mathrm{~N}-6),-182.4(\mathrm{~N}-3),-88.0(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}$ (\%): $407\left(\mathrm{M}^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{27} \mathrm{H}_{27} \mathrm{~N}_{4}\left([\mathrm{M}+1]^{+}\right)$calcd 407.2230, found 407.2233.

### 4.1.2.16.12. Typical experimental procedure for the synthesis of tetracycles 126-129.

Corresponding alkyne 39 or $42(0.5 \mathrm{mmol})$ and the appropriate $o$-aryldiamine ( 0.6 mmol ) were dissolved in absolute DMF ( 2 mL ), and the mixture was stirred at $120^{\circ} \mathrm{C}$ temperature for 48 hours. After the completion of reaction as indicated by TLC, the mixture was quenched with water $(10 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 10$ $\mathrm{mL})$. The organic layers were combined, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure. The obtained residue was purified by column
chromatography ( $\mathrm{SiO}_{2}$, eluent: ethyl acetate $/ n$-hexane, $1: 4, \mathrm{v} / \mathrm{v}$ ) to provide the desired products 126-129.
4.1.2.16.12.1. 2,5-Diphenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazole (126). White solid, yield: $90 \mathrm{mg}, 50 \%, \mathrm{mp} 241^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\max }, \mathrm{cm}^{-}$
 ${ }^{1}$ ): 3047 ( $\mathrm{CH}_{\text {arom }}$ ), 1663, 1596, 1501, 1448, 1400, 1270 (C=C, C$\mathrm{N}), 755,736,702,691(\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\text {н }}$ ppm 6.28-6.31 (m, 1H, 7H), 6.93-6.98 (m, 1H, 8-H), $6.97\left(\mathrm{~d},{ }^{5} J(4-\mathrm{H}, 1-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 4-\right.$ H), 7.30-7.36 (m, 1H, 9-H), 7.39-7.43 (m, 1H, NPh 4-H), 7.527.57 (m, 2H, NPh 3,5-H), 7.58-7.62 (m, 4H, CPh 2,3,5,6-H), 7.627.66 (m, 1H, CPh 4-H), 7.84-7.88 (m, 1H, 10-H), 7.87-7.89 (m, $2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 8.96\left(\mathrm{~d},{ }^{5} J(1-\mathrm{H}, 4-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 1-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 104.5$ (C-4), 111.3 (C-11b), 113.9 (C7), 119.3 (C-10), 120.2 (NPh C-2,6), 121.3 (C-8), 122.9 (C-1), 123.8 (C-9), 127.9 (NPh C-4), 129.0 (CPh C-2,6), 129.3 (CPh C-3,5), 129.7 (NPh C-3,5), 129.9 (CPh C4), 130.8 (C-6a), 134.8 (CPh C-1), 139.8 (NPh C-1), 139.9 (C-5), 144.3 (C-10a), 145.0 (C-3a), $148.1(\mathrm{C}-11 \mathrm{a}) .{ }^{15} \mathrm{~N}$ NMR $\left(40 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm}-210.2(\mathrm{~N}-6),-160.3(\mathrm{~N}-$ 11), $-154.3(\mathrm{~N}-2),-97.2(\mathrm{~N}-3) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 361\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 361.1448, found 361.1463.
4.1.2.16.12.2. 5-Butyl-2-phenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2-a]benzimidazole (127). White solid, yield $143 \mathrm{mg}, 84 \%, \mathrm{mp} 180^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $v_{\text {max }}$,
 $\left.\mathrm{cm}^{-1}\right): 3059\left(\mathrm{CH}_{\text {arom }}\right), 2953\left(\mathrm{CH}_{\text {aliph }}\right), 1669,1536,1504,1452$, 1402, 1274, 1211 (C=C, C-N), 758, 750, 732, 687 ( $\mathrm{CH}=\mathrm{CH}$ of mono- and disubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta_{\mathrm{H}} \mathrm{ppm} 1.04\left(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.55-1.66(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.84-1.93 (m, 2H, CH2CH2C2 $\mathrm{C}_{2}$ ), 3.26-3.32 (m, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.86\left(\mathrm{~d},{ }^{5} \mathrm{~J}(4-\mathrm{H}, 1-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.31-7.36$ (m, 1H, 8-H), 7.36-7.41 (m, 1H, NPh 4-H), 7.42-7.47 (m, 1H, 9H), 7.49-7.55 (m, 2H, NPh 3,5-H), 7.83-7.87 (m, 2H, NPh 2,6H), $7.90-7.93(\mathrm{~m}, 2 \mathrm{H}, 7,10-\mathrm{H}), 8.87\left(\mathrm{~d},{ }^{5} J(1-\mathrm{H}, 4-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}\right.$, 1-H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.9\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $29.4\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.6\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{6}\right), 101.5(\mathrm{C}-4), 110.8(\mathrm{C}-11 \mathrm{~b}), 114.0(\mathrm{C}-7), 119.6$ (C-10), 120.1 (NPh C-2,6), 121.9 (C-8), 122.7 (C-1), 123.8 (C-9), 127.7 (NPh C-4), 129.7 (NPh C-3,5), 130.8 (C-6a), 139.8 (NPh C-1), 141.7 (C-5), 144.4 (C-10a), 145.3 (C-3a), 148.3 (C-11a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-209.0(\mathrm{~N}-6),-159.9(\mathrm{~N}-$ 11), $-155.4(\mathrm{~N}-2),-97.2(\mathrm{~N}-3) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 341\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 341.1747, found 341.1762.

### 4.1.2.16.12.3. 9-Methyl-2,5-diphenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2-

 $a$ ]benzimidazole (128a) and 8-methyl-2,5-diphenyl- 2 H -pyrazolo[4',3':3,4] pyr-ido[1,2-a]benzimidazole (129a). White solids, yield $112 \mathrm{mg}, 60 \%$. Compounds were obtained as an inseparable mixture in the ratio 1:0.62 (128a:129a).4.1.2.16.12.3.1. 128a. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 2.45\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$,
 $6.15\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.5 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 6.77$ $\left(\mathrm{dd},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=8.5 \mathrm{~Hz},{ }^{4} J(8-\mathrm{H}, 10-\mathrm{H})=1.5\right.$ $\mathrm{Hz}, 1 \mathrm{H}, 8-\mathrm{H}), 6.94$ (s, 1H, 4-H), 7.38-7.43 (m, $1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.51-7.57(\mathrm{~m}, 2 \mathrm{H}$, NPh 3,5-H), 7.57-7.60 (m, 4H, CPh 2,3,5,6-H), 7.63 (br s, $1 \mathrm{H}, 10-\mathrm{H}), 7.61-7.67$ (m, 1H, CPh 4-H), 7.867.90 (m, 2H, NPh 2,6-H), 8.95 ( $\mathrm{s}, 1 \mathrm{H}, 1-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.5\left(\mathrm{CH}_{3}\right)$, 104.2 (C-4), 111.31 (C-11b), 113.4 (C-7), 112.86 (C-8), 119.0 (C-10), 120.2 ( $\mathrm{NPh} \mathrm{C}-$ 2,6), 122.83 (C-1), 127.86 (NPh C-4), 128.85 (C-6a), 128.92 (CPh C-2,6), 129.33 (CPh C-3,5), 129.7 (NPh C-3,5), 129.83 (CPh C4), 133.7 (C-9), 134.91 (CPh C-1), 139.83 (NPh C-1), 139.88 (C-5), 144.6 (C-10a), 145.0 (C-11a), 148.12 (C-3a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-210.3$ (N-6), $-154.3(\mathrm{~N}-2)$.
4.1.2.16.12.3.2. 129a. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 2.21\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 6.02 (br s, $1 \mathrm{H}, 7-\mathrm{H}), 6.94(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.15\left(\mathrm{dd},{ }^{3} J(9-\mathrm{H}, 10-\mathrm{H})=8.2 \mathrm{~Hz},{ }^{4} J(7-\mathrm{H}, 9-\right.$ $\mathrm{H})=1.3 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}), 7.38-7.43(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.51-7.57(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H})$, 7.57-7.60 (m, 4H, CPh 2,3,5,6-H), 7.61-7.67 (m, 1H, CPh 4-H), $7.73\left(\mathrm{~d}, 1 \mathrm{H},{ }^{3} J(10-\right.$ $\mathrm{H}, 9-\mathrm{H})=8.2 \mathrm{~Hz}, 10-\mathrm{H}), 7.86-7.90(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 8.93(\mathrm{~s}, 1 \mathrm{H}, 1-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 21.9\left(\mathrm{CH}_{3}\right), 104.3(\mathrm{C}-4), 111.4(\mathrm{C}-11 \mathrm{~b}), 114.0(\mathrm{C}-7)$, 118.7 (C-10), 120.2 (NPh C-2,6), 122.75 (C-1), 125.3 (C-9), 127.84 (NPh C-4), 128.80 (CPh C-4), 128.85 (CPh C-2,6), 129.39 (CPh C-3,5), 129.77 (NPh C-3,5), 130.9 (C-6a), 131.1 (C-8), 134.87 (CPh C-1), 139.83 (NPh C-1), 139.95 (C-5), 142.3 (C-10a), 144.6 (C-11a), 148.05 (C-3a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-210.3$ (N-6), -154.5 (N-2). MS m/z (\%): 375 ([M+H] ${ }^{+}$, 100). HRMS (ESI) for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{4}$ $\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 375.1604, found 375.1621.

### 4.1.2.16.12.4. 9-Chloro-2,5-diphenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2-

 a]benzimidazole (137b) and 8-chloro-2,5-diphenyl-2H-pyrazolo[4',3':3,4]pyri-do[1,2-a]benzimidazole (138b). Compounds were obtained separately by column chromatography on silica gel (Hex/EtOAc. 4:1 $\mathrm{v} / \mathrm{v})$. Yields: $40 \%$ (128b) and $30 \%$ (129b).
4.1.2.16.12.4.1. 128b. White solid, yield $79 \mathrm{mg}, 40 \%, \mathrm{mp} 185{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\text {max }}, \mathrm{cm}^{-1}\right): 3062\left(\mathrm{CH}_{\text {arom }}\right), 1665,1597,1501$, 1429, 1418, 1403, 1205 (C=C, C-N), 755, 745, 702, $685(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \operatorname{ppm} 6.16\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.9 \mathrm{~Hz}\right.$, $1 \mathrm{H}, 7-\mathrm{H}), 6.90\left(\mathrm{dd},{ }^{3} J(8-\mathrm{H}, 7-\mathrm{H})=8.9 \mathrm{~Hz},{ }^{4} J(8-\right.$ $\mathrm{H}, 10-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}), 6.99(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H})$, 7.39-7.45 (m, 1H, NPh 4-H), 7.52-7.57 (m, 2H, NPh 3,5-H), 7.57-7.62 (m, 4H, CPh $2,3,5,6-\mathrm{H}), 7.62-7.68(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh} 4-\mathrm{H}), 7.80\left(\mathrm{~d},{ }^{4} J(10-\mathrm{H}, 8-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right)$,
7.85-7.90 (m, 2H, NPh 2,6-H), $8.94(\mathrm{~s}, 1 \mathrm{H}, 1-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}}$ ppm 105.0 (C-4), 111.1 (C-11b), 114.5 (C-7), 118.8 (C-10), 120.3 (NPh C-2,6), 121.7 (C-8), 123.0 (C-1), 128.1 (NPh C-4), 129.1 (CPh C-2,6), 129.3 (CPh C-3,5), 129.39 (C-6a), 129.44 (C-9), 129.8 (NPh C-3,5), 130.1 (CPh C-4), 134.4 (CPh C-1), 139.6 (C-5), 139.7 (NPh C-1), 145.2 (C-10a), 146.0 (C-11a), 148.1 (C-3a). ${ }^{15} \mathrm{~N}$ NMR (40 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-210.3(\mathrm{~N}-6),-160.8(\mathrm{~N}-11),-154.3(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 397$ $\left([\mathrm{M}+2]^{+}, 36\right), 395\left(\mathrm{M}^{+}, 100\right)$. Calcd for $\mathrm{C}_{24} \mathrm{H}_{15} \mathrm{~N}_{4} \mathrm{Cl}(394.86): \mathrm{C}, 73.00 ; \mathrm{H}, 3.83 ; \mathrm{N}$, 14.19. Found: C, 73.33; H, 4.02; N, 14.28.
4.1.2.16.12.4.2. 129b. White solid, yield $59 \mathrm{mg}, 30 \%$, mp $229^{\circ} \mathrm{C}$ (ethyl acetate). IR ( $\left.v_{\max }, \mathrm{cm}^{-1}\right): 3059\left(\mathrm{CH}_{\text {arom }}\right), 1665,1597,1501,1455,1418,1273(\mathrm{C}=\mathrm{C}, \mathrm{C}-\mathrm{N}), 758$, $702,687\left(\mathrm{C}-\mathrm{Cl}, \mathrm{CH}=\mathrm{CH}\right.$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \operatorname{ppm} 6.19\left(\mathrm{~d},{ }^{4} J(7-\mathrm{H}, 9-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 6.98(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.28(\mathrm{dd}$, $\left.{ }^{3} J(9-\mathrm{H}, 10-\mathrm{H})=8.6 \mathrm{~Hz},{ }^{4} J(9-\mathrm{H}, 7-\mathrm{H})=2.1 \mathrm{~Hz}, 1 \mathrm{H}, 9-\mathrm{H}\right), 7.39-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H})$, $7.52-7.57(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 3,5-\mathrm{H}), 7.55-7.59(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CPh} 2,6-\mathrm{H}), 7.60-7.65(\mathrm{~m}, 2 \mathrm{H}$, CPh 3,5-H), 7.65-7.70 (m, 1H, CPh 4-H), $7.73\left(\mathrm{~d},{ }^{3} \mathrm{~J}(10-\mathrm{H}, 9-\mathrm{H})=8.6 \mathrm{~Hz}, 1 \mathrm{H}, 10-\mathrm{H}\right)$, 7.85-7.89 (m, 2H, NPh 2,6-H), $8.93(\mathrm{~s}, 1 \mathrm{H}, 1-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}}$ ppm 105.0 (C-4), 111.1 (C-11b), 114.0 (C-7), 119.8 (C-10), 120.3 (NPh C-2,6), 123.0 (C-1), 124.4 (C-9), 126.7 (C-8), 128.0 (NPh C-4), 129.2 (CPh C-3,5), 129.3 (CPh C2,6), 129.8 (NPh C-3,5), 130.2 (CPh C-4), 131.1 (C-6a), 134.2 (CPh C-1), 139.6 (C5), 139.7 (NPh C-1), 142.8 (C-10a), 145.6 (C-11a), 148.0 (C-3a). ${ }^{15} \mathrm{~N}$ NMR ( 40 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta_{\mathrm{N}} \mathrm{ppm}-210.5(\mathrm{~N}-6),-160.4(\mathrm{~N}-11),-153.6(\mathrm{~N}-2) \mathrm{ppm} . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 397$ $\left([\mathrm{M}+2]^{+}, 36\right), 395\left(\mathrm{M}^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{24} \mathrm{H}_{16} \mathrm{ClN}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 395.1058, found 395.1062.
4.1.2.16.12.5. 5-Butyl-9-methyl-2-phenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2$a$ ]benzimidazole (128c) and 5-butyl-8-methyl-2-phenyl-2H-pyrazolo[4',3':3,4] pyrido[1,2-a]benzimidazole (129c). Yield: 106 mg ( $60 \%$ ). Compounds were obtained as an inseparable mixture in the ratio $1: 0.25(\mathbf{1 2 8 c}: \mathbf{1 2 9} \mathbf{c})$.

4.1.2.16.12.5.1.128c: ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.02(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.53-1.63 (m, $2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.80-1.89 (m, 2H, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 2.52(\mathrm{~s}, 3 \mathrm{H}$, $\left.9-\mathrm{CH}_{3}\right), 3.20-3.25\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.81(\mathrm{~d}$, $\left.{ }^{5} J(4-\mathrm{H}, 1-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right), 7.13\left(\mathrm{dd},{ }^{3} J(8-\right.$ $\mathrm{H}, 7-\mathrm{H})=8.7 \mathrm{~Hz},{ }^{4} \mathrm{~J}(8-\mathrm{H}, 10-\mathrm{H})=1.3 \mathrm{~Hz}, 1 \mathrm{H}, 8-$ $\mathrm{H}), 7.34-7.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NPh} 4-\mathrm{H}), 7.48-7.53$ (m, 2H, NPh 3,5-H), 7.65-7.67 (m, 1H, 10-H), $7.75\left(\mathrm{~d},{ }^{3} J(7-\mathrm{H}, 8-\mathrm{H})=8.7 \mathrm{~Hz}, 1 \mathrm{H}, 7-\mathrm{H}\right), 7.81-$ $7.85(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NPh} 2,6-\mathrm{H}), 8.86\left(\mathrm{~d},{ }^{5} \mathrm{~J}(1-\mathrm{H}, 4-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 1-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm} 13.89\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 21.43\left(9-\mathrm{CH}_{3}\right), 22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.37$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.45\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 101.1(\mathrm{C}-4), 110.71(\mathrm{C}-11 \mathrm{~b}), 113.46(\mathrm{C}-7), 119.26$ (C-10), 120.0 (NPh C-2,6), 122.61 (C-1), 123.3 (C-8), 127.64 (NPh C-4), 128.78 (C6a), 129.6 (NPh C-3,5), 133.64 (C-9), 139.8 (NPh C-1), 141.59 (C-5), 144.64 (C-10a), 145.21 (C-3a), 148.21 (C-11a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{N}} \mathrm{ppm}-209.3(\mathrm{~N}-6)$, -160.9 (N-11), -155.4 (N-2).
4.1.2.16.12.5.2.129c: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 1.04(\mathrm{t}, J=7.4 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}$ ), 1.56-1.65 (m, 2H, $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.82-1.91 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ ), $2.55\left(\mathrm{~s}, 3 \mathrm{H}, 8-\mathrm{CH}_{3}\right), 3.24\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 6.81\left(\mathrm{~d},{ }^{5} \mathrm{~J}(4-\mathrm{H}, 1-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}\right)$, 7.24-7.26 (m, 1H, 9-H), 7.34-7.40 (m, 1H, NPh 4-H), 7.48-7.53 (m, 2H, NPh 3,5H), 7.67-7.68 (m, 1H, 7-H), 7.75-7.78 (m, 1H, 10-H), 7.81-7.85 (m, 1H, NPh 2,6H), $8.84\left(\mathrm{~d},{ }^{5} J(1-\mathrm{H}, 4-\mathrm{H})=0.8 \mathrm{~Hz}, 1 \mathrm{H}, 1-\mathrm{H}\right) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{C}} \mathrm{ppm}$ $13.88\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 21.43\left(8-\mathrm{CH}_{3}\right), 22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.44\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.50$ $\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 101.3(\mathrm{C}-4), 110.78(\mathrm{C}-11 \mathrm{~b}), 114.0(\mathrm{C}-7), 118.9(\mathrm{C}-10), 120.0(\mathrm{NPh} \mathrm{C}-$ 2,6), 122.55 (C-1), 125.27 (C-9), 127.62 (NPh C-4), 129.6 (NPh C-3,5), 130.9 (C-6a), 131.63 (C-8), 139.8 (NPh C-1), 141.67 (C-5), 142.35 (C-10a), 144.9 (C-3a), 148.14 (C-11a). MS m/z (\%): $355\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{23} \mathrm{H}_{23} \mathrm{~N}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 355.1917 , found 355.1915 .
4.1.2.16.12.6. 5-Butyl-9-chloro-2-phenyl-2H-pyrazolo[4',3':3,4]pyrido[1,2a]benzimidazole (128d). White solid, yield $107 \mathrm{mg}, 57 \%, \mathrm{mp} 227^{\circ} \mathrm{C}$ (ethyl acetate).
 IR ( $\left.v_{\text {max }}, \mathrm{cm}^{-1}\right): 3064\left(\mathrm{CH}_{\text {arom }}\right), 2960\left(\mathrm{CH}_{\text {aliph }}\right), 1671,1596$, 1499, 1427, 1209 (C=C, C-N), 798, 767, 747, 690 (C-Cl, $\mathrm{CH}=\mathrm{CH}$ of mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta_{\mathrm{H}} \mathrm{ppm} 1.05\left(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right), 1.57-$ $1.68\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.85-1.94\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right)$, 3.26-3.32 (m, 2H, CH $\left.\mathrm{C}_{2} \mathrm{H}_{7}\right), 6.93(\mathrm{~s}, 1 \mathrm{H}, 4-\mathrm{H}), 7.31\left(\mathrm{dd},{ }^{3} J(8-\right.$ $\left.\mathrm{H}, 7-\mathrm{H})=8.9 \mathrm{~Hz},{ }^{4} J(8-\mathrm{H}, 10-\mathrm{H})=2.0 \mathrm{~Hz}, 1 \mathrm{H}, 8-\mathrm{H}\right), 7.39-7.44$ (m, 1H, NPh 4-H), 7.52-7.58 (m, 2H, NPh 3,5-H), 7.82-7.86 (m, 1H, 7-H), 7.85-7.89 (m, 2H, NPh 2,6-H), 7.87-7.89 (m, $2 \mathrm{H}, 10-\mathrm{H}), 8.90(\mathrm{~s}, 1 \mathrm{H}, 1-\mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm} 13.9\left(\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{CH}_{3}\right)$, $22.2\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 29.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{2} \mathrm{H}_{5}\right), 33.5\left(\mathrm{CH}_{2} \mathrm{C}_{3} \mathrm{H}_{7}\right), 102.0(\mathrm{C}-4), 110.6(\mathrm{C}-$ 11 b ), 114.7 (C-7), 119.2 (C-10), 120.2 (NPh C-2,6), 122.1 (C-8), 122.9 (C-1), 127.9 (NPh C-4), 129.4 (C-6a), 129.5 (C-9), 129.8 (NPh C-3,5), 139.7 (NPh C-1), 141.3 (C5), 145.4 (C-10a), 146.4 (C-11a), 148.2 (C-3a). ${ }^{15} \mathrm{~N}$ NMR ( $40 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{C}} \mathrm{ppm}$ $-209.2(\mathrm{~N}-6),-154.3(\mathrm{~N}-2) . \mathrm{MS} \mathrm{m} / \mathrm{z}(\%): 377\left([\mathrm{M}+2]^{+}, 36\right), 375\left([\mathrm{M}+\mathrm{H}]^{+}, 100\right)$. HRMS (ESI) for $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{ClN}_{4}\left([\mathrm{M}+\mathrm{H}]^{+}\right)$calcd 375.1375, found 375.1371.

### 4.2. Biology

### 4.2.1. Cancer cell lines and cytotoxicity assay

Human cancer cell lines were obtained from the American Type Culture Collection and were cultivated according to the provider's instructions. In brief, MCF-7 and K-562 cell lines were maintained in DMEM medium supplemented with $10 \%$ fetal bovine serum, penicillin ( $100 \mathrm{U} / \mathrm{mL}$ ), and streptomycin $(100 \mu \mathrm{~g} / \mathrm{mL})$ at $37{ }^{\circ} \mathrm{C}$ in $5 \% \mathrm{CO}_{2}$. For the cytotoxicity assays, cells were treated in triplicate with six different doses of each compound for 72 hours. After treatment, Calcein AM solution was added for 1 hour, and the fluorescence from live cells was measured at $485 \mathrm{~nm} / 538$ nm (excitation/emission) by using a Fluoroskan Ascent microplate reader (Labsystems). The $\mathrm{GI}_{50}$ value, the drug concentration lethal to $50 \%$ of the cells, was calculated from the dose-response curves that resulted from the assays.

### 4.2.2. Flow cytometry

Asynchronous cells were seeded into a 96 well plate and, after a preincubation period, treated with the tested compounds for 24 hours at a single dose of $10 \mu \mathrm{M}$. MCF-7 cells were first washed with PBS, trypsinized, and finally treated with a solution of trypsin inhibitor $(0.1 \%)$. After the incubation, $5 \times$ staining solution $(17 \mathrm{~mm}$ trisodium citrate dihydrate, $0.5 \%$ IGEPAL® CA-630, 7.5 mm spermine tetrahydrochloride, 2.5 mm Tris; pH 7.6 containing $50 \mu \mathrm{~g} / \mathrm{mL}$ propidium iodide) was added. K562 cells were stained directly with the 5x staining solution (i.e., without trypsinization). The cells' DNA content was analyzed by flow cytometry using a 488 nm laser (BD FACS Verse with software BD FACSuite ${ }^{\mathrm{TM}}$, version 1.0.6.). Cell cycle distribution was analyzed by using ModFit LT (Verity Software House, version 4.1.7).

For the quantification of histone H 3 phosphorylation, the cells were harvested by trypsinization, washed in PBS, fixed with ice-cold $90 \%$ methanol, incubated on ice for 30 min , and washed with PBS/BSA containing $0.5 \%$ Tween-20. Then, the cells were incubated with the primary antibody raised against histone H 3 phosphorylated at Ser10 (Millipore) for 1 hour at room temperature, washed with PBS containing $1 \%$ BSA, and incubated with the secondary antibody (goat-anti-rabbit-Alexa Fluor 488, Invitrogen) for 1 hour in the dark. After washing with PBS/BSA, each sample was incubated with propidium iodide (final concentration $10 \mu \mathrm{~g} / \mathrm{mL}$ ) and RNAse A (final concentration $200 \mu \mathrm{~g} / \mathrm{mL}$ ) for 30 minutes at room temperature in the dark. Then, the cells were analyzed by flow cytometry using a 488 nm laser (BD FACS Verse with software BD FACSuite ${ }^{\mathrm{TM}}$, version 1.0.6.).

### 4.2.3. Caspase- $3 / 7$ assay

Cellular caspase- $3 / 7$ activity was measured according to the previously published procedure. ${ }^{104}$ K-562 cells were cultivated in a 96-well plate overnight. Next day, the cells were treated with increasing concentrations of compound 5 for the next 24 hours. After the incubation, 3 x caspase-3/7 assay buffer ( 150 mM HEPES pH 7.4 , $450 \mathrm{mM} \mathrm{NaCl}, 150 \mathrm{mM} \mathrm{KCl}, 30 \mathrm{mM} \mathrm{MgCl} 2,1.2 \mathrm{mM}$ EGTA, $1.5 \%$ Nonidet P40, $0.3 \%$ CHAPS, $30 \%$ sucrose, 30 mM DTT, 3 mM PMSF) containing $150 \mu \mathrm{M}$ peptide substrate Ac-DEVD-AMC (Enzo Life Sciences) was added, and after 2 hours incubation, the caspase-3/7 activity was measured by using a Fluoroskan Ascent microplate reader (Labsystems) at $346 \mathrm{~nm} / 442 \mathrm{~nm}$ (excitation/emission). The activity was normalized to an untreated control.

### 4.2.4. Immunoblotting

Immunoblotting was performed as described earlier. ${ }^{105}$ In brief, cellular lysates were prepared by harvesting cells in Laemmli sample buffer. Proteins were separated on SDS-polyacrylamide gels and electroblotted onto nitrocellulose membranes. After blocking, the membranes were incubated with specific primary antibodies overnight, washed, and incubated with peroxidase-conjugated secondary antibodies. Finally, peroxidase activity was detected with ECL+ reagents (AP Biotech) by using a CCD camera LAS-4000 (Fujifilm). Specific antibodies were purchased from Santa Cruz

Biotechnology (PARP, $\beta$-actin), Cell Signaling (Ser-139 phosphorylated H2AX, Bcl2), and Sigma Aldrich (peroxidase-labeled secondary antibodies).

## 5. THE MAIN RESULTS AND CONCLUSIONS

1.Derivatives of the 2 H -furo[2,3-c]pyrazole can be easily obtained via the silver (I) or gold (I) ion mediated cyclisation of 4-alkynyl-1-phenyl-1H-pyrazol-3-ols.
2. 3-Alkynylpyrazole-4-carbaldehydes and the corresponding alkanones are suitable precursors for the preparation of 2 H -pyrazolo[4,3-c]pyridine derivatives.
3. A simple and straightforward approach to various benzopyrano[2,3-c]pyra-zol-4 $(2 \mathrm{H})$-ones starting from easily available 4 -aroylpyrazol-3-ols is developed.
4. The reaction of 3- and 5-alkynylpyrazole-4-carbaldehydes with (het)aryl diamines results in the formation of polycylic condensed heterocyclic systems.
5. 1-Methyl-3,5-diphenyl-3H-pyrazole[4',3':3,4]pyrido[1,2-a]benzimidazole can be functionalized by employing Sonogashira- and Suzuki-type coupling reactions of the corresponding intermediate bromide.
6. 2H-Pyrazolo[4,3-c]pyridine derivatives exhibit anticancer activity against K562 and MCF-7 cancer cell lines in vitro through arresting the cell cycle in mitosis and induction of apoptosis.
7. $2 H$ - and $3 H$-pyrazole $\left[4^{\prime}, 3^{\prime}: 3,4\right]$ pyrido $[1,2-a$ ]benzimidazole derivatives are novel fluorescent organic compounds characterized by a high quantum yield $\left(\Phi_{\mathrm{f}}\right)$.

## REFERENCES

${ }^{1}$ a) CHERUKUPALLI, S., et al. An appraisal on synthetic and pharmaceutical perspectives of pyrazolo[4,3-d]pyrimidine scaffold. Bioorganic and Medicinal Chemistry, 2018, vol 26(2), 309-339 [2017-11-21]. ISSN 0968-0896. doi: 10.1016/j.bmc.2017.10.012, Science Direct; b) JACHAK, M.N., et al. Synthesis and fluorescence properties of donor-acceptor-substituted novel dipyrazolo[3,4-b:3',4'd]pyridines (DPP). Journal of Fluorescence, 2010, vol 20(3), 787-796 [2018-01-15]. ISSN 1053-0509 (print) 1573-4994 (online). doi: 10.1007/s10895-010-0622-4, PubMed, and references cited therein; c) REDDY, N.R., REDDY, G.M., REDDY, B.S., REDDY, P.P.A facile one step synthesis of 1,6-dihydro-7H-pyrazolo[4,3- $d$ ]-pyrim-idin-7-ones. Journal of Heterocyclic Chemistry, 2005, vol 42, 751-754 [2017-10-02]. ISSN 1943-5193. doi:10.1002/jhet.5570420502, Wiley Online Library.
${ }^{2}$ TANGETI, V.S., PRASAD, G.V.S., PANDA, J., VARMA, R.K. One pot multicomponent diastereoselective synthesis of novel dihydro-1H-furo[2,3-c]pyrazoles. Synthetic Communications, 2016, vol 46(10), 878-884 [2017-05-13]. ISSN 1532-2432. doi: 10.1080/00397911.2016.1174781, Taylor and Francis Online.
${ }^{3}$ TURKI, H., KAMOUN, M., LAHIANI, S., EL GHARBI, R. A simple efficient procedure for the synthesis of benzopyrano[2,3-c]pyrazoles. Journal of Heterocyclic Chemistry, 2016, vol 53(5), 1356-1362 [2018-01-10]. ISSN 1943-5193. doi: 10.1002/jhet.1759, Wiley Online Library.
${ }^{4}$ GALAL, S.A., et al. Part I: Design, synthesis and biological evaluation of novel pyrazole-benzimidazole conjugates as checkpoint kinase 2 (Chk2) inhibitors with studying their activities alone and in combination with genotoxic drugs. European Journal of Medicinal Chemistry, 2017, vol 134, 392-405 [2018-01-09]. ISSN 02235234. doi: 10.1016/j.ejmech.2017.03.090, PubMed.
${ }^{5}$ ZHANG, J., et al. An efficient and highly stereoselective synthesis of novel trifluoromethylated trans-dihydrofuro[2,3-c]pyrazoles using arsonium ylides. Tetrahedron, 2012, vol 68, 2121-2127 [2018-12-03]. ISSN 0040-4020. doi: 10.1016/j.tet.2012.01.030, Science Direct.
${ }^{6}$ MILISIUNAITE , V., et al. Synthesis and anti-mitotic activity of 2,4- or 2,6-disub-stituted- and 2,4,6-trisubstituted-2H-pyrazolo[4,3-c]pyridines. European Journal of Medicinal Chemistry, 2018, vol 150, 908-919 [2018-03-16]. ISSN 0223-5234. doi: 10.1016/j.ejmech.2018.03.037, Science Direct.
${ }^{7}$ NAYAK, M., BATCHU, H., BATRA, S. Straightforward copper-catalyzed synthesis of pyrrolopyrazoles from halogenated pyrazolecarbaldehydes. Tetrahedron Letters, 2012, vol 53, 4206-4208 [2018-02-01]. ISSN 0040-4039. doi: 10.1016/j.tetlet.2012.05.148, Science Direct.
${ }^{8}$ a) JUNGHEIM, L.N., SIGMUND, S.K. 1,3-Dipolar cycloaddition reactions of pyrazolidinium ylides with acetylenes. Synthesis of a new class of antibacterial agents. The Journal of Organic Chemistry. 1987, vol 52, 4007-4013 [2017-12-15]. ISSN 0022-3263. doi: 10.1021/jo00227a013, ACS Publications; b) INDELICATO, J.M., PASINI, C.E. The acylating potential of gamma-lactam antibacterials: Base hydrolysis of bicyclic pyrazolidinones. Journal of Medicinal Chemistry. 1988, vol 31(6),

1227-1230 [2017-12-15]. ISSN 0022-2623. doi: $10.1021 / \mathrm{jm} 00401 \mathrm{a} 026$, ACS Publications.
${ }^{9}$ FISCHER, R., et al. 4-Biphenylsubstituted pyrazolidin-3,5-dione derivatives. WO 2005016873; Chemical Abstracts Service. 2005, 142, 261530 [2018-01-31]. ISSN 2079-9292, Patentscope.
${ }^{10}$ KHIDRE, R., et al. Advances in the chemistry of pyrazolopyrazoles. Turkish Journal of Chemistry. 2013, vol 37, 1-35 [2017 10 25]. ISSN 1303-6130. doi:10.3906/kim-1204-50, Agris.
${ }^{11}$ a) GLENN, R.W., LIM, M. Keratin dyeing compounds, keratin dyeing compositions containing them, and use thereof. US 20070050923; Chemical Abstracts Service. 2007, vol 146, 322831 [2018-01-30]. PatentScope; b) VIDAL, L., MALLE, G., MONTEIL, E. Keratin fibre dye composition containing pyrazolo-azole compounds, use thereof as dye couplers, and dyeing method. WO 9735551; Chemical Abstracts Service. 1997, 127, 311355 [2018-01-30]. Espacenet.
${ }^{12}$ DECONINCK, G., SAUNIER, J.B., DESENNE, P. Composition comprenant la 2,3-diamino-6,7-dihydro-1H,5H-pyrazollo[1,2-a]pyrazol-1-one, le 4,5-diamino 1-( $\beta$ hdroxyethyl)pyrazole et le 2-chloro 6-methyl-3-amino phenol. FR 2937864; Chemical Abstracts Service. 2010, 152, 533689 [2018-01-31]. Espacenet.
${ }^{13}$ RADICS, U., MICHEL, H.J., NICLAS, H., GRABARSE, M. Stable, non-volatile nitrification inhibitors for use with ammonium and/or amide-containing fertilizers, comprising 3-oxo-3H-pyrazolo[1,2-a]pyrazol-4-ium-1-olates. DE 19958051; Chemical Abstracts Service. 2001, 135, 19121 [2018-01-30]. Espacenet.
${ }^{14}$ PALMER, C., TOWFIGHI, J., ROBERTS, R.L., HEITJAN, D.F. Allopurinol administered after inducing hypoxia-ischemia reduces brain injury in 7-day-old rats. Pe diatric Research. 1993, vol 33, 405-411 [2018-01-15]. ISSN 1530-0447. doi: 10.1203/00006450-199304000-00018, PubMed.

15 BELL, A.S., TERRETT, N.K. Pyrazolopyrimidinone Antianginal Agents. WO1993007149A1, 1993 [2018-01-30]. Espacenet.
${ }^{16}$ TAYLOR, E.C., PATEL, H., KUMAR, H. Synthesis of pyrazolo[3,4- $d$ ] pyrimidine analogues of the potent agent $N$-\{4-[2-(2-amino-4(3H)-oxo-7H-pyrrolo[2,3-d]pyrim-idin-5-yl)ethyl]benzoyl\}-L-glutamic acid (LY231514). Tetrahedron. 1992, vol 48(37), 8089-8100 [2018-12-15]. ISSN 0040-4020. doi: 10.1016/S0040-4020(01)80479-8, Science Direct.
17 JUNGHEIM, L.N., SIGMUND, S.K., FISHER, J.W., JONES, N.D., SWARTZENDRUBER, J.K. Bicyclic pyrazolidinones, steric and electronic effects on antibacterial activity. Tetrahedron Letters. 1987, vol. 28, 289-292 [2017-12-05]. ISSN 0040-4039. doi: 10.1016/S0040-4039(00)95709-5, Science Direct.
${ }^{18}$ BURGER, K., et al. Reaktionen mit Hexafluoracetonazin, XXI. [1,3-2,4]-Cycloaddition von elektronenarmen Mehrfachbindungs-Systemen an Hexafluoracetonazin -"Criss-cross"--Cycloaddition mit Acrylsäureestern. Liebigs Annalen der Chemie, 1982, vol 5, 845-852 [2018-03-25]. ISSN 0170-2041. doi: 10.1002/jlac.198219820504, Wiley Online Library.
${ }^{19}$ EL-ALALI, A., AL-KAMALI. A.S. Reactions of 1,3-dipolar aldazines and ketazines with the dipolarophile dimethyl acetylenedicarboxylate. Canadian Journal of

Chemistry, 2002, vol 80(10), 1293-1301 [2018-02-24]. ISSN 1480-3291. doi: 10.1139/v02-169, Canadian Science Publishing.
${ }^{20}$ ADIB, M., SAYAHI, M.H., AGHAALIAKBARI, B., BIJANZADEH, H.R. Reaction between isocyanides and dialkyl acetylenedicarboxylates in the presence of 2,4-dihydro-3H-pyrazol-3-ones. One-pot synthesis of highly functionalized 7-oxo-1H,7H-pyrazolo[1,2-a]pyrazoles. Tetrahedron, 2005, vol 61(16), 3963-3966 [2018-02-06]. ISSN 0040-4020. doi: 10.1016/j.tet.2005.02.050, Science Direct.
${ }^{21}$ a) POTTS, K.T., MURPHY, P.M., KUEHNLING, W.R. Cross-conjugated and pseudo-cross-conjugated mesomeric betaines. 1. Synthesis and characterization. The Journal of Organic Chemistry. 1988, 53(13), 2889-2898 [2018-02-04]. ISSN 15201690. doi: 10.1021/jo00248a002, ACS Publications; b) POTTS, K.T., KANEMASA, S., ZVILICHOVSKY, G. Reactions of (chlorocarbonyl)phenylketene. Formation of a new class of heterocyclic zwitterion. Journal of the American Chemical Society, 1980, 102(11), 3971-3972 [2018-02-04]. ISSN 1520-5126. doi: 10.1021/ja00531a059, ACS Publications.
22 a) GUO, C., et al. Discovery of pyrroloaminopyrazoles as novel PAK inhibitors. The Journal of Medicinal Chemistry, 2012, vol 55(10), 4728-4739 [2018-02-06]. ISSN 1520-4804. doi: 10.1021/jm300204j, ACS Publications; b) LI, H., et al. Identification of novel pyrrolopyrazoles as protein kinase $C \beta$ II inhibitors. Bioorganic and Medicinal Chemistry Letters, 2011, vol 21(1), 584-587 [2018-01-28]. ISSN 0960894X. doi: 10.1016/j.bmcl.2010.10.032, Science Direct; c) SHI, J., et al. Design and synthesis of 1,4,5,6-tetrahydropyrrolo[3,4-c]pyrazoles and pyrazolo[3,4-b]pyridines for Aurora-A kinase inhibitors. Bioorganic and Medicinal Chemistry Letters, 2010, vol 20(14), 4273-4278 [2018-01-20]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2010.04.083, Science Direct; d) BRASCA, M.G., et al. Optimization of 6,6-dimethyl pyrrolo[3,4-c]pyrazoles: Identification of PHA-793887, a potent CDK inhibitor suitable for intravenous dosing. Bioorganic and Medicinal Chemistry, 2010, vol 18(5), 1844-1853 [2018-01-15]. ISSN 0968-0896. doi: 10.1016/j.bmc.2010.01.042, Science Direct; e) PEVARELLO, P., et al. Bioorganic and Medicinal Chemistry Letters, 2006, vol 16 (4), 1084-1090 [2018-01-10]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2005.10.071, Science Direct; f) FANCELLI, D., et al. 1,4,5,6-tetrahydropyrrolo[3,4-c]pyrazoles: Identification of a potent aurora kinase inhibitor with a favorable antitumor kinase inhibition profile. The Journal of Medicinal Chemistry, 2006, vol 49(24), 7247-7251 [2018-01-15]. ISSN 1520-4804. doi: 10.1021/jm060897w, ACS Publications; g) FANCELLI, D., et al. Potent and selective aurora inhibitors identified by the expansion of a novel scaffold for protein kinase inhibition. The Journal of Medicinal Chemistry, 2005, 48(24), 3080-3084 [2015-0110]. ISSN 1520-4804. doi: $10.1021 / \mathrm{jm} 049076 \mathrm{~m}$, ACS Publications.
${ }^{23}$ STEEGHS, N., et al. Phase I pharmacokinetic and pharmacodynamic study of the aurora kinase inhibitor danusertib in patients with advanced or metastatic solid tumors. The Journal of Clinical Oncology, 2009, 27(30) 5094-5101 [2018-01-08]. ISSN 1527-7755. doi: 10.1200/JCO.2008.21.6655, PubMed.
${ }^{24}$ BRASCA, M.G., et al. Optimization of 6,6-dimethyl pyrrolo[3,4-c]pyrazoles: Identification of PHA-793887, a potent CDK inhibitor suitable for intravenous dosing.

Bioorganic and Medicinal Chemistry, 2010, 18(5), 1844-1853 [2018-01-18]. ISSN 0968-0896. doi: 10.1016/j.bmc.2010.01.042, PubMed.
${ }^{25}$ THERRIEN, E., et al. Discovery of bicyclic pyrazoles as class III histone deacetylase SIRT1 and SIRT2 inhibitors. Bioorganic and Medicinal Chemistry Letters, 2015, 25(12), 2514-2518 [2018-01-18]. ISSN0960-894X. doi: 10.1016/j.bmcl. 2015.04.068, Science Direct.
${ }^{26}$ SHECHTER, H., FRIEDMAN, L. Preparation of nitriles from halides and sodium cyanide. An advantageous nucleophilic displacement in dimethyl sulfoxide. The Journal of Organic Chemistry, 1960, 25(6), 877-879 [2018-01-15]. ISSN 1520-6904. doi: 10.1021/jo01076a001, ACS Publications.
${ }^{27}$ HAMILTON, H.W., ORTWINE, D.F., WORTH, D.F., BRISTOL, J.A. Synthesis and structure-activity relationships of pyrazolo[4,3-d]pyrimidin-7-ones as adenosine receptor antagonists. The Journal of Medicinal Chemistry, 1987, 30(24), 91-96 [2016-01-10]. ISSN 1520-4804. doi: 10.1021/jm00384a016, ACS Publications.
${ }^{28}$ GREGONRINI, G., LALOUE, M. Biological effects of cytokinin antagonists 7(pentylamino) and 7-(benzylamino)-3-methylpyrazolo[4,3-d]pyrimidines on suspension-cultured tobacco cells. Plant Physiology, 1980, vol 65(2), 363-367 [2018-02-10]. ISSN 1532-2548. doi: 10.1104/pp.65.2.363, PubMed.
${ }^{29}$ YUAN, J., et al. 3-Aryl pyrazolo[4,3-d]pyrimidine derivatives: Nonpeptide CRF-1 antagonists. Bioorganic and Medicinal Chemistry Letters, 2002, vol 12(16), 21332136 [2018-01-20]. ISSN 0960-894X. doi: 10.1016/S0960-894X(02)00358-X, Science Direct.
${ }^{30}$ JORDA, R., et al. Anti-leishmanial activity of disubstituted purines and related py-razolo[4,3- $d$ ]pyrimidines. Bioorganic and Medicinal Chemistry Letters, 2011, vol 21(14), 4233-4237 [2018-01-25]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2011.05.076, Science Direct.

31 TOLLEFSON, M.B., et al. 1-(2-(2,2,2-Trifluoroethoxy)ethyl-1H-pyrazolo[4,3$d]$ pyrimidines as potent phosphodiesterase 5 (PDE5) inhibitors. Bioorganic and Me dicinal Chemistry Letters, 2010, vol 20(10), 3125-3128 [2018-01-05]. ISSN 0960894X. doi: 10.1016/j.bmcl.2010.03.106, Science Direct.
${ }^{32}$ HAFEZ H.N., EL-GAZZAR ABDUL-RHMAN, B.A., AL-HUSSAIN, S.A. Bioorganic and Medicinal Chemistry Letters, 2016, vol 26(10), 2428-2433 [2018-01-05]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2016.03.117, Science Direct.
${ }^{33}$ PAL, M., et al. Novel bicyclic heterocyclic compounds, process for their preparation and compositions containing them, US20060128729 A1, 2006 [2018-01-19], Espacenet.
${ }^{34}$ BUNNAGE M. E., et al. Pyrazolo[4,3-d]pyrimidine derivatives, WO0127113 A2, 2001 [2018-01-19], Espacenet.
${ }^{35}$ XING, Y., ZUO, J., KROGSTAD, P., JUNG, M.E. Synthesis and structure-activity relationship (SAR) studies of novel pyrazolopyridine derivatives as inhibitors of enterovirus replication. Journal of Medicinal Chemistry, 2018, vol 61(4), 1688-1703 [2018-01-23]. ISSN 1520-4804. doi: 10.1021/acs.jmedchem.7b01863, ACS Publications.
${ }^{36}$ EL-GOHARY, N.S., SHAABAN, M.I. New pyrazolopyridine analogs: Synthesis, antimicrobial, antiquorum-sensing and antitumor screening. European Journal of Medicinal Chemistry, 2018, vol 152, 126-136 [2018-02-10]. ISSN 1520-4804. doi: 10.1016/j.ejmech.2018.04.025, Science Direct.
${ }^{37}$ PELIT, E. CSA-catalyzed three-component synthesis of fused polycyclic pyra-zolo[4,3-e]pyridines under ultrasonic irradiation and their antioxidant activity. Journal of the Turkish Chemical Society, Section A: Chemistry, 2017, vol 4(2), 631-648 [2018-03-15]. ISSN 2149-0120. doi: 10.18596/jotcsa.295465, DergiPark.
${ }^{38}$ SAMALA, G., BRINDHA, P., RADHIKA, D., PERUMAL, N., DHARMARAJAN, Y. Development of 3-phenyl-4,5,6,7-tetrahydro-1H-pyrazolo[4,3-c]pyridine derivatives as novel mycobacterium tuberculosis pantothenate synthestase inhibitors. European Journal of Medicinal Chemistry, 2013, vol 69, 356-364 [2018-02-18]. ISSN 1520-4804. doi: 10.1016/j.ejmech.2013.08.036, Science Direct.
${ }^{39}$ ARLAN, F.M., KHALAFY, J., MALEKI, R. One-pot three-component synthesis of a series of 4-aroyl-1,6-diaryl-3-methyl-1H-pyrazolo[3,4-b]pyridine-5-carbonitriles in the presence of aluminum oxide as a nanocatalyst. Chemistry of Heterocyclic Compounds, 2018, vol 54(1), 51-57 [2018-03-04]. ISSN 1573-8353. doi: 10.1007/s10593-018-2229-7, Springer Link.
${ }^{40}$ MILIUTINA, M., et al. A domino reaction of 3-chlorochromones with aminoheterocycles. Synthesis of pyrazolopyridines and benzofuropyridines and their optical and ecto-5'-nucleotidase inhibitory effects. Organic and Biomolecular Chemistry, 2018, vol 16(5), 717-732 [2018-03-01]. ISSN 1477-0539. doi: 10.1039/C7OB02729J, Royal Society of Chemistry.
${ }^{41}$ METWALLY, N.H., DEEB, E.A. Synthesis, anticancer assessment on human breast, liver and colon carcinoma cell lines and molecular modeling study using novel pyrazolo[4,3-c]pyridine derivatives. Bioorganic Chemistry, 2018, vol 77, 203-214 [2018-03-05]. ISSN 0045-2068. doi: 10.1016/j.bioorg.2017.12.032, Science Direct.
${ }^{42}$ ABDELRAZEK, F.M., METZ, P., METWALLY, N.H., EL-MAHROUKY, S.F. Synthesis and molluscicidal activity of new cinnoline and pyrano[2,3-c]pyrazole derivatives. Archiv der Pharmazie, 2006, vol 339(8), 456-460. ISSN 1521-4184. doi: 10.1002/ardp.200600057, PubMed.
${ }^{43}$ ISMAIL, Z.H., ALY, G.M., EL-DEGWI, M.S., HEIBA, H.I., GHORAB, M.M. Synthesis and insecticidal activity of some new pyranopyrazoles, pyrazolopyranopyrimidines, and pyrazolopyranopyridines. Egyptian Journal of Biotechnology, 2003, vol 13, 73-82 [2018-04-10]. ISSN 1110-6093.
${ }^{44}$ KUO, S.C., HUANG, L.J., NAKAMURA, H. Studies on heterocyclic compounds. 6. Synthesis and analgesic and antiinflammatory activities of 3,4-dimethylpyrano[2,3-c]pyrazol-6-one derivatives. Journal of Medicinal Chemistry, 1984, vol 27(4), 539544 [2018-03-28]. ISSN 1520-4804. doi: 10.1021/jm00370a020, ACS Publications.
${ }^{45}$ ZAKI, M.E.A., MORSY, E.M., ABDEL-MOTTI F, M., ABDEL-MEGEID, F.M.E. The behaviour of ethyl 1-acetyl-4-aryl-5-cyano-3-methyI-1,4-dihydropyrano[2,3-c]pyrazol-6-ylimidoformate towards nucleophiles. Heterocyclic Communations, 2004, vol 10(1), 97-102 [2018-04-09]. ISSN 2191-0197. doi: 10.1515/HC.2004.10.1.97, De Gruyter.
${ }^{46}$ JUNEK, H., AIGNER, H. Synthesen mit Nitrilen, XXXV: Reaktionen von Tetracyanäthylen mit Heterocyclen. Chemische Berichte, 1973, vol 106, 914-921 [2018-05-10]. ISSN 1099-0682. doi: 10.1002/cber. 19731060323 , Wiley Online Library.
${ }^{47}$ KIYANI, H., SAMIMI, H., GHORBANI, F., ESMAIELI, S. One-pot, four-component synthesis of pyrano[2,3-c]pyrazoles catalyzed by sodium benzoate in aqueous medium. Current Chemistry Letters, 2013, 2(4), 197-206 [2017-11-30]. ISSN 1927730x (Online), ISSN 1927-7296 (Print). doi: DOI: 10.5267/j.ccl.2013.07.002, Academic Journal Database.
${ }^{48}$ LI, C.-B., LI, Y.-Q., XU, D.-Z. An efficient four-component reaction for the rapid synthesis of highly functionalized pyrano[2,3-c]pyrazoles catalyzed by [Dabco$\mathrm{H}][\mathrm{AcO}]$ ionic liquid under mild condition. ChemistrySelect, 2017, vol 2, 2917-2921 [2017-10-20]. ISSN 2365-6549. doi: 10.1002/slct.201700168, Wiley Online Library.
${ }^{49}$ LAUFER, R., et al. Discovery of 4-(4-aminopyrazolo[1,5-a][1,3,5]triazin-8-yl)benzamides as novel, highly potent and selective, orally bioavailable inhibitors of Tyrosine Threonine Kinase, TTK. Bioorganic and Medicinal Chemistry Letters, 2016, vol 26(15), 3562-3566 [2017-11-20]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2016.06.021, Science Direct.
${ }^{50}$ HUTTERER, C., et al. A novel CDK7 inhibitor of the pyrazolotriazine class exerts broad-spectrum antiviral activity at nanomolar concentrations. Antimicrobial Agents and Chemotherapy, 2015, vol 59, 2062-2071 [2017-12-04]. ISSN 0066-4804 (Print), ISSN 1098-6596 (Online). doi: 10.1128/AAC.04534-14, ASM Journals.
${ }^{51}$ RIZK, H.F., IBRAHIM, S.A., EL-BORAI, M.A. Synthesis, dyeing performance on polyester fiber and antimicrobial studies of some novel pyrazolotriazine and pyrazolyl pyrazolone azo dyes. Arabian Journal of Chemistry, 2017, vol 10, S3303-S3309 [2017-10-15]. ISSN 1878-5352. doi: 10.1016/j.arabjc.2014.01.008, Science Direct.
${ }^{52}$ SHCHEGOL'KOV, E.V., et al. Synthesis, molecular docking, and biological activity of polyfluoroalkyl dihydroazolo[5,1-c][1,2,4]triazines as selective carboxylesterase inhibitors. Bioorganic and Medicinal Chemistry, 2017, vol 25(15), 3997-4007 [2017-11-29]. ISSN 0968-0896. doi: 10.1016/j.bmc.2017.05.045, Science Direct.
${ }^{53}$ NASR, T., BONDOCK, S., YOUNS, M., FAYAD, W., ZAGHARY, W. Synthesis, antitumor evaluation and microarray study of some new pyrazolo[3,4-d][1,2,3]triazine derivatives. European Journal of Medicinal Chemistry, 2017, vol 141, 603-614 [2018-01-10]. ISSN 0223-5234. doi: 10.1016/j.ejmech.2017.10.016, PubMed.
${ }^{54}$ HASSAN, A.S., MOUSTAFA, G.O., AWAD, H.M. Synthesis and in vitro anticancer activity of pyrazolo[1,5-a]pyrimidines and pyrazolo[3,4- $d][1,2,3]$ triazines. Synthetic Communications, 2017, vol 47(21), 1963-1972 [2018-01-10]. doi: 10.1080/00397911.2017.1358368, Taylor and Francis Online.
${ }^{55}$ KASABOINA, S., et al. Iodine mediated pyrazolo-quinoline derivatives as potent anti-proliferative agents. Bioorganic and Medicinal Chemistry Letters, 2018, vol 28(4), 664-667 [2018-03-26]. ISSN 0960-894X. doi: 10.1016/j.bmcl.2018.01.023, Science Direct.
${ }^{56}$ EZZATI, M., et al. The catalyst-free syntheses of pyrazolo[3,4-b]quinolin-5-one and pyrazolo $\left[4^{\prime}, 3^{\prime}: 5,6\right]$ pyrido $[2,3-d]$ pyrimidin-5,7-dione derivatives by one-pot, three-component reactions. Tetrahedron, 2017, vol 73(47), 6587-6596 [2018-01-13]. ISSN 0040-4020. doi: 10.1016/j.tet.2017.10.004, Science Direct.
${ }^{57}$ CHEN, Z., WU, J. Efficient generation of biologically active $h$-pyrazolo[5,1-a]isoquinolines via multicomponent reaction. Organic Letters, 2010, vol 12(21), 48564859 [2018-01-10]. ISSN 1523-7060 (print); 1523-7052 (online). doi: 10.1021/ol101988q, ACS Publications.
${ }^{58}$ LIU, H., LU, L., HUA, R. [Cu(maloNHC)]-catalyzed synthesis of 2-aryl pyra-zolo[5,1-a]isoquinolines by annulation of $N^{\prime}$-(2-((trimethylsilyl)ethynyl)benzylidene)hydrazides with terminal aromatic alkynes. Tetrahedron, 2017, vol 73(45), 6428-6435 [2018-01-18]. ISSN 0040-4020. doi: 10.1016/j.tet.2017.09.037, Science Direct.
${ }^{59}$ HESE, S.V., et al. Antidiabetic and allied biochemical roles of new chromeno-pyrano pyrimidine compounds: synthesis, in vitro and in silico analysis. Medicinal Chemistry Research, 2017, vol 26(4), 805-818 [2018-01-20]. ISSN 1054-2523 (print), 1554-8120 (online). doi: 10.1007/s00044-017-1794-0, Springer Link.
${ }^{60} \mathrm{LI}, \mathrm{H}$., et al. Green method for the synthesis of chromeno[2,3-c]pyrazol-4(1H)-ones through ionic liquid promoted rirected annulation of 5-(Aryloxy)-1H-pyrazole-4carbaldehydes in aqueous media. Organic Letters, 2015, vol 17(4), 932-935 [2018-02-01]. ISSN 1523-7060 (print); 1523-7052 (online). doi: 10.1021/acs.orglett.5b00033, ACS Publications.
${ }^{61}$ MORALES, P., et al. Chromenopyrazole, a versatile cannabinoid scaffold with in vivo activity in a model of multiple sclerosis. Journal of Medicinal Chemistry, 2016, vol 59(14), 6753-6771 [2018-02-05]. ISSN 0022-2623 (print); 1520-4804 (online). doi: 10.1021/acs.jmedchem.6b00397, ACS Publications.
62 BONARDI, A., et al. Structural investigations on coumarins leading to chromeno[4,3-c]pyrazol-4-ones and pyrano[4,3-c]pyrazol-4-ones: New scaffolds for the design of the tumor-associated carbonic anhydrase isoforms IX and XII. European Journal of Medicinal Chemistry, 2018, vol 146, 46-59 [2018-02-05]. doi: 10.1016/j.ejmech.2018.01.033, PubMed.
${ }^{63}$ LIM, J., et al. Enantioselective syntheses of decursinol angelate and decursin. Tetrahedron Letters, 2001, vol 42(24), 4001-4003 [2018-01-23]. ISSN 0040-4039. doi: 10.1016/S0040-4039(01)00642-6, Science Direct.
${ }^{64}$ PRESS, J.B., Novel thieno[2,3-b]- and [3,4-b]pyrans as potassium channel openers. Thiophene systems-XVII. Bioorganic and Medicinal Chemistry, 1993, vol 1(6), 423435 [2018-02-28]. ISSN 0968-0896. doi: 10.1016/S0968-0896(00)82153-7, Science Direct.
${ }^{65}$ CUMELLA, J., et al. Chromenopyrazoles: Non-psychoactive and selective CB1 cannabinoid agonists with peripheral antinociceptive properties. ChemMedChem,

2012, vol 7(3), 452-463. 0009-4374 (print); 0223-5234 (online). doi:10.1002/cmdc.201100568, PubMed.
${ }^{66}$ KHOOBI, M., et al. New tetracyclic tacrine analogs containing pyrano[2,3-c]pyrazole: Efficient synthesis, biological assessment and docking simulation study. European Journal of Medicinal Chemistry, 2014, 89, 296-303 [2018-02-22]. ISSN 00094374 (print); 0223-5234 (online). doi: 10.1016/j.ejmech.2014.10.049, PubMed.
${ }^{67}$ BONDOCK, S., KHALIFA, W., FADDA, A.A. Synthesis and antimicrobial activity of some new 4-hetarylpyrazole and furo[2,3-c]pyrazole derivatives. European Journal of Medicinal Chemistry, 2011, 46(6), 2555-2561 [2018-01-30]. ISSN 00094374 (print); 0223-5234 (online). doi: 10.1016/j.ejmech.2011.03.045, PubMed.
${ }^{68}$ KOENIG, H., GOETZ, N., KLEIN, U., ELLER, K. Process for producing $n$-substituted 3-hydroxypyrazoles, WO 9703969. Chemical Abstracts Service, 1997, 126, 199566 [2017-12-05]. Espacenet.
${ }^{69}$ ARBAČIAUSKIENĖ, E., VILKAUSKAITĖ, G., ELLER, G.A., HOLZER, W., ŠAČKUS, A. Pd-catalyzed cross-coupling reactions of halogenated 1-phenylpyrazol-3-ols and related triflates. Tetrahedron, 2009, vol 65(37), 7817-7824 [2014-12-01]. ISSN 0040-4020. doi: 10.1016/j.tet.2009.07.017, Science Direct.
${ }^{70}$ a) SUN, S.-X., et al. Highly efficient heterogeneous synthesis of benzofurans under aqueous condition. Tetrahedron, 2014, vol 70(24), 3798-3806 [2014-12-20]. ISSN 0040-4020. doi: 10.1016/j.tet.2014.04.005, Science Direct; b) ZHANG, Y., XIN, Z., XUE, J., LI, Y. Gold-catalyzed alkyne hydroxylation: Synthesis of 2-substituted benzo[b]furan compounds. Chinese Journal of Chemistry, 2008, vol 26, 1461-1464 [2015-10-25]. ISSN 1614-7065. doi: 10.1002/cjoc.200890265, Wiley Online Library; c) DAMERA, K., et al. A novel base-promoted cyclization: Synthesis of substituted benzo[b]furans. RSC Advances, 2012, vol 2(25), 9403-9405 [2016-04-15]. ISSN 2046-2069. doi: 10.1039/C2RA21302H, Royal Society of Chemistry.
${ }^{71}$ ARBAČIAUSKIENĖ, E., et al. On the tautomerism of $N$-substituted pyrazolones: 1,2-Dihydro-3H-pyrazol-3-ones versus 1H-Pyrazol-3-ols. Molecules, 2018, vol 23(1), 129 [2015-03-10]. ISSN 1420-3049. doi: 10.3390/molecules23010129, MDPI.
${ }^{72}$ DOREL, R., ECHAVARREN, A.M. Gold(I)-catalyzed activation of alkynes for the construction of molecular complexity. Chemical Reviews, 2015, vol 115(17), 90829072 [2015-02-10]. ISSN 0009-2665 (print); 1520-6890 (online). doi: 10.1021/cr500691k, PubMed.
${ }^{73}$ FANG, G., BI, X. Silver-catalysed reactions of alkynes: Recent advances. Chemical Society Reviews, 2015, vol 44, 8124-8173 [2016-10-05]. ISSN 0306-0012 (print) 1460-4744 (online). doi: 10.1039/C5CS00027K, Royal Society of Chemistry.
${ }^{74}$ KUBOTA, M., et al. Gold-catalyzed cyclization of alkyne alcohols: Regioselective construction of functionalized 6,6- and 6,7-bicyclic ethers. Chemical and Pharmaceutical Bulletin, 2016, 64(7), 845-855 [2017-04-10]. ISSN 0009-2363 (print); 13475223 (online). doi: 10.1248/cpb.c16-00204, PubMed.
${ }^{75}$ EGI, M., et al. Efficient intramolecular cyclizations of phenoxyethynyl diols into multisubstituted $\alpha, \beta$-unsaturated lactones. Organic Letters, 2013, vol 15(16), 41504153 [2018-01-10]. ISSN 1523-7060 (print), ISSN 1523-7052 (online). doi: 10.1021/ol401824v, ACS Publications.
${ }^{76}$ a) HOLZER, W., VILKAUSKAITĖ, G., ARBAČIAUSKIENĖ, E., ŠAČKUS, A. Dipyrazolo[1,5-a:4',3'-c]pyridines - a new heterocyclic system accessed via multicomponent reaction. Beilstein Journal of Organic Chemistry, 2012, vol 8, 2223-2229 [2014-10-11]. ISSN 1860-5397.doi: 10.3762/bjoc.8.251, www.beilstein-journals.org; b) MILIŠIŪNAITĖ, V., et al. Synthesis of pyrazolo[4', $\left.3^{\prime}: 3,4\right]$ pyrido[1,2a]benzimidazoles and related new ring systems by tandem cyclisation of vic-alkynylpyrazole-4-carbaldehydes with (het)aryl-1,2-diamines and investigation of their optical properties. Tetrahedron, 2015, vol 71(21), 3385-3395 [2014-09-30]. ISSN 0040-4020. doi: 10.1016/j.tet.2015.03.092, Science Direct; c) PALKA, B., et al. Synthesis of trifluoromethyl-substituted pyrazolo[4,3-c]pyridines - sequential versus multicomponent reaction approach. Beilstein Journal of Organic Chemistry, 2014, vol 10, 1759-1764 [2014-10-11]. ISSN 1860-5397. doi: 10.3762/bjoc.10.183, www.beilstein-journals.org; d) VILKAUSKAITĖ, G., ŠAČKUS, A., HOLZER, W. Sonogashira-type reactions with 5-chloro-1-phenyl-1H-pyrazole-4-carbaldehydes: A straightforward approach to pyrazolo[4,3-c]pyridines. European Journal of Organic Chemistry, 2011, 10, 5123-5133 [2014-09-28]. ISSN 1434-193X (print); 1099-0690 (online). doi: 10.1002/ejoc.201100626, Wiley online library.
${ }^{77}$ ARBAČIAUSKIENĖ, E., et al. Synthesis of 3-substituted 1-phenyl-1H-pyrazole-4-carbaldehydes and the corresponding ethanones by Pd-catalysed cross-coupling reactions. Arkivoc, 2011, vol XI, 1-21. [2014-12-15]. ISSN 1551-7012. doi: 10.3998/ark.5550190.0012.b01, ARKAT USA.
${ }^{78}$ a) HAMAGUCHI, W., et al. Synthesis, SAR study, and biological evaluation of novel quinoline derivatives as phosphodiesterase 10A inhibitors with reduced CYP3A4 inhibition. Bioorganic and Medicinal Chemistry, 2015, vol 23, 297-313 [2015-01-08]. ISSN 0968-0896. doi: 10.1016/j.bmc.2014.11.039, PubMed; b) MASUDA, N. Pyrazole compound, WO 2012133607 A1, 2012. Escpacenet; c) MAEKAWA, T., et al. 1,2-azole derivatives with hypoglycemic and hypolipidemic activity, WO 2003099793 A1, 2003. Espacenet; d) FLETCHER, S., GUNNING, P.T. Mild, efficient and rapid $O$-debenzylation of ortho-substituted phenols with trifluoroacetic acid. Tetrahedron Letters, 2008, vol 49, 4817-4819 [2014-09-30]. ISSN 0040-4039. doi: 10.1016/j.tetlet.2008.06.022, Science Direct; e) STANG, P.J., HANACK, M., SUBRAMANIAN, L.R. Perfluoroalkanesulfonic esters: Methods of preparation and applications in organic chemistry. Synthesis (Stuttg), 1982, vol 2, 85126[ 2014-10-05]. ISSN 0039-7881. doi: 10.1055/s-1982-29711, Thieme.
${ }^{79}$ a) O'BRIEN, D.F., GATES, J.W. Some reactions of 3-hydroxy-1-phenylpyrazole, Journal of Organic Chemistry, 1996, vol 31, 1538-1542 [2014-10-06]. ISSN 00223263. doi: 10.1021/jo01343a054, ACS Publications; b) KÖNIG, H., GÖTZ, N., KLEIN, U., ELLER, K. Process for producing $N$-substituted 3-hydroxypyrazoles, WO 1997003969 A1, 1997. Espacenet.
${ }^{80}$ SASAKI, T., KANEMATSU, K., UCHIDE, M. Syntheses of fused heterocycles via cycloaddition of hetaryne. Studies on heteroaromaticity. Part XLVII. Bulletin of the Chemical Society of Japan, 1971, vol 44(3), 858-959 [2014-09-27]. ISSN 1348-0634. doi: $10.1246 / \mathrm{bcsj} .44 .858$, CSJ journals.
${ }^{81}$ LU, C.R., et al. Cell apoptosis: Requirement of H2AX in DNA ladder formation, but not for the activation of caspase-3. Molecular Cell, 2006, vol 23(1), 121-132 [2015-09-20]. ISSN 1097-2765. doi: 10.1016/j.molcel.2006.05.023, PubMed.
${ }^{82}$ a) LI, S., et al. Identification of inhibitors against p90 ribosomal S6 kinase 2 (RSK2) through structure-based virtual screening with the inhibitor-constrained refined homology model. The Journal of Chemical Information and Modeling, 2011, vol 51(11), 2939-2947 [2015-10-25]. ISSN 1549-9596 (print); 1520-5142 (online). doi: 10.1021/ci2002445, PubMed; b) SMYTH, L.A. Design and evaluation of 3-aminopyrazolopyridinone kinase inhibitors inspired by the natural product indirubin. Bioorganic and Medicinal Chemistry, 2011, vol 19, 3569-3578 [2015-10-25]. ISSN 09680896. doi: 10.1016/j.bmc.2011.03.069, PubMed.
${ }^{83}$ SHAN, G., YANG, X., MA, L., RAO, Y. Pd-catalyzed C-H oxygenation with TFA/TFAA: Expedient access to oxygen-containing heterocycles and late-stage drug modification. Angewandte Chemie, 2012, vol 51, 13070-13074. ISSN 1433-7851 (print), 1521-3773 (online). doi: 10.1002/anie.201207458, Wiley Online Library.
${ }^{84}$ a) HOLZER, W., et al. Novel fluoro-substituted benzo- and benzothieno fused py-rano[2,3-c]pyrazol-4(1H)-ones. Journal of Fluorine Chemistry, 2010, vol 131(10), 1013-1024 [2015-05-30]. ISSN 0022-1139. doi: 10.1016/j.jfluchem.2010.07.007, Science Direct. b) DATTERL, B., TRÖSTNER, N., KUCHARSKI, D., HOLZER, W. Heterocyclic analogues of xanthone and xanthione. $1 H$-Pyrano[2,3- $:$ :6,5-c]dipyrazol-4(7H)-ones and thiones: Synthesis and NMR data. Molecules, 2010, vol 15(9), 61066126 [2015-05-30]. ISSN 1420-3049. doi: 10.3390/molecules15096106, PubMed.
85 a) HUBBARD, P., BRITTAIN, W.J. Mechanism of amine-catalyzed ester formation from an acid chloride and alcohol. The Journal of Organic Chemistry, 1998, vol 63(3), 677-683 [2015-06-10]. ISSN 0022-3263 (print), 1520-6904 (online). doi: 10.1021/jo9716643, ACS Publications; b) LI, Y., LIU, Y., XIONG, Y., XIONG, X. Crystal structures, vibrational spectra, and fungicidal activity of 1,5-diaryl-3-oxypyrazoles. Molecules 2014, vol 19(1), 1302-1316 [2015-06-10]. ISSN 1420-3049. doi:10.3390/molecules19011302, PubMed.
${ }^{86}$ SPIVEY, A.C., ARSENIYADIS, S. Nucleophilic catalysis by 4-(dialkylamino)pyridines revisited-the search for optimal reactivity and selectivity. Angewandte Chemie, 2004, vol 43, 5436-5441 [2015-10-20]. ISSN 1433-7851 (print), 1521-3773 (online). doi: 10.1002/anie.200460373, Wiley Online Library.
${ }^{87}$ PHAKHODEE, W., DUANGKAMOL, C., PATTARAWARAPAN, M. $\mathrm{Ph}_{3} \mathrm{P}-\mathrm{I}_{2}$ mediated aryl esterification with a mechanistic insight. Tetrahedron Letters, 2016, vol 57(19), 2087-2089 [2016-11-08]. ISSN 0040-4039. doi: 10.1016/j.tetlet.2016.03.105, Science Direct.
${ }^{88}$ WON, J.-E., et al. Effective esterification of carboxylic acids using (6-oxo-6H-pyr-idazin-1-yl)phosphoric acid diethyl ester as novel coupling agents. Tetrahedron, 2007, vol 63(51), 12720-12730 [2014-12-10]. ISSN 0040-4020. doi: 10.1016/j.tet.2007.10.011, Science Direct.
${ }^{89}$ KWON, E.M., et al. Preparation of benzoyloxy benzophenone derivatives and their inhibitory effects of icam-1 expression. Bulletin of the Korean Chemical Society, 2012, vol 33(6), 1939-1944 [2015-12-15]. ISSN 0253-2964. doi: 10.5012/bkcs.2012.33.6.1939, KOREASCIENCE.
${ }^{90}$ JESSICA, L., et al. Synthetic utility and mechanistic implications of the fries rearrangement of hydroquinone diesters in boron trifluoride complexes. The Journal of Organic Chemistry, 2000, vol 65(15), 4712-4714 [2014-12-18]. ISSN 0022-3263 (print), 1520-6904 (online). DOI: 10.1021/jo000412q, ACS Publications.
${ }^{91}$ PAUL, S., GUPTA, M. Selective fries rearrangement catalyzed by zinc powder. Synthesis, 2004, vol 11, 1789-1792 [2015-05-27]. ISSN 0039-7881. doi: 10.1055/s-2004-829152, Thieme.
${ }^{92}$ MURASHIGE, R., et al. Comparisons of $O$-acylation and Friedel-Crafts acylation of phenols and acyl chlorides and Fries rearrangement of phenyl esters in trifluoromethanesulfonic acid: Effective synthesis of optically active homotyrosines. Tetrahedron, 2011, vol 67(3), 641-649 [2015-06-05]. ISSN 0040-4020. doi: 10.1016/j.tet.2010.11.047, Science Direct.
${ }^{93}$ SALAHUDDIN, et al. Benzimidazoles: A biologically active compounds. Arabian Journal of Chemistry, 2017, vol 10(1), S157-S173 [2016-09-10]. ISSN 1878-5352. doi: 10.1016/j.arabjc.2012.07.017, Science Direct.
${ }^{94}$ RAMESHBABU, V.N.S., NARAHARIBABU, A., ANAND, V., HANUMANTHU, P. A facile synthesis of pyrido[1,2-a]benzimidazoles. Synthetic Communications, 1998, vol 28(23), 4439-4443 [2015-11-19]. ISSN 0039-7911 (print), 15322432 (online). doi: 10.1080/00397919808004479, Taylor and Francis Online.
${ }^{95}$ a) OUYANG, H.-C., TANG, R.-Y., ZHONG, P., ZHANG, X.-G., LI, J.-H. CuI/I $2^{-}$ promoted electrophilic tandem cyclization of 2-ethynylbenzaldehydes with orthobenzenediamines: Synthesis of iodoisoquinoline-fused benzimidazoles. The Journal of Organic Chemistry, 2011, vol 76(1), 223-228 [2014-09-05]. ISSN 0022-3263 (print); 1520-6904 (online). doi: 10.1021/jo102060j, ACS Publications; b) PATIL, N.T., MUTYALA, A.K., LAKSHMI, P.G., RAJU, P.V., SRIDHAR, B. Facile assembly of fused isoquinolines by gold(I)-catalyzed coupling-cyclization reactions between $o$-alkynylbenzaldehydes and aromatic amines containing tethered nucleophiles. The European Journal of Organic Chemistry, 2010, vol 10, 1999-2007 [2014-10-22]. ISSN 1434-193X (print); 1099-0690 (online). doi: 10.1002/ejoc.200901364, Wiley Online Library; c) GVOZDEV, V.D., SHAVRIN, K.N., BASKIR, E.G., EGOROV, M.P., NEFEDOV, O.M. Selective one-pot synthesis of 11 -arylmethylidene- 11 H -iso-indolo-[2,1-a]benzimidazoles and 6-arylbenzimidazo[2,1-a]isoquinolines from $o$-alkynylbenzaldehydes and o-diaminobenzenes. Mendeleev Communications, 2017, vol 27(3), 231-233 [2017-12-28]. ISSN 0959-9436. doi: 10.1016/j.mencom.2017.05.004, Science Direct; d) DYKER, G., STIRNER, W., HENKEL, G. Oxidative heterocyclization of 2-alkynylbenzaldehydes with 1,2-phenylenediamine. The European Journal of Organic Chemistry, 2000, vol 8, 1433-1441 [2014-09-15]. ISSN 1434-193X (print); 1099-0690 (online). doi: 10.1002/(SICI)1099-0690(200004)2000:8<1433::AID-EJOC1433>3.0.CO;2-7, Wiley Online Library; e) RUSTAGI, V., AGGARWAL, T., VERMA, A.K. Highly efficient Ag(I)-catalyzed regioselective tandem synthesis of diversely substituted quinoxalines and benzimidazoles in water. Green Chemistry, 2011, vol 13, 1640-1643 [2014-09-12]. ISSN 1463-9262 (print), 1463-9270 (online). doi: 10.1039/C1GC15346C, Royal Society of Chemistry.
${ }^{96}$ a) PENG, J., SHANG, G., CHEN, C., MIAO, Z., LI, B. Nucleophilic addition of benzimidazoles to alkynyl bromides/palladium-catalyzed intramolecular $\mathrm{C}-\mathrm{H}$ vinylation: Synthesis of benzo[4,5]imidazo[2,1-a]isoquinolines Journal of Organic Chemistry, 2013, vol 78(3), 1242-1248 [2014-10-08]. ISSN 0022-3263 (print); 1520-6904 (online). doi: 10.1021/jo302471z, ACS Publications; b) REDDY, V.P., IWASAKI, T., KAMBE, N. Synthesis of imidazo and benzimidazo[2,1-a]isoquinolines by rho-dium-catalyzed intramolecular double $\mathrm{C}-\mathrm{H}$ bond activation. Organic and Biomolecular Chemistry, 2013, vol 11, 2249-2253 [2014-10-05]. ISSN 1477-0520 (print); 1477-0539 (online). doi: 10.1039/C3OB27396B, Royal Society of Chemistry.
${ }^{97}$ OKAMOTO, N., SAKURAI, K., ISHIKURA, M., TAKEDA, K., YANADA, R. One-pot concise syntheses of benzimidazo[2,1-a]isoquinolines by a microwave-accelerated tandem process. Tetrahedron Letters, 2009, 50(28), 4167-4169 [2014-0928]. ISSN 0040-4039. doi: 10.1016/j.tetlet.2009.04.126, Science Direct.
${ }^{98}$ a) EPSTEIN, W.W., SWEAT, F.W. Dimethyl sulfoxide oxidations. Chemical Reviews, 1967, vol 67, 247-260 [ 2014-09-28]. ISSN 0009-2665. doi: 10.1021/cr60247a001, ACS Publications; b) MENTZEL, W., BUNGE, W. In Ullmanss Encyklopadie der technischen Chemie, 4th ed; Verlag Chemie: Weinheim, 1985, vol. 18, 269-273.
${ }^{99}$ VERMA, A.K., BISHNOI, A., FATMA, S., PARVEEN, H., SINGH, V. An easy synthetic access to Spiro derivatives containing pyrido[1,2-a]pyrimidine and quinoline scaffolds and their antimicrobial activity. ChemistrySelect, 2017, vol 2(14), 40064009 [2017-11-28]. ISSN 2365-6549. doi: 10.1002/slct.201700228, Wiley Online Library.
${ }^{100}$ CHAITANYA, T.K., PRAKASH, K.S., NAGARAJAN, R. Metal-free synthesis of benzimidazo[2,1-a]ellipticines via tandem inter and intramolecular cyclization. Tetrahedron, 2011, 67(36), 6934-6938 [2014-11-17]. ISSN 0040-4020. doi: 10.1016/j.tet.2011.06.076, Science Direct.
${ }^{101}$ TOKIMIZU, Y., et al. Direct synthesis of highly fused perimidines by copper(I)catalyzed hydroamination of 2-ethynylbenzaldehydes. Tetrahedron, 2011, 67(29), 5168-5175 [2014-09-04]. ISSN 0040-4020. doi: 10.1016/j.tet.2011.05.051, Science Direct.
${ }^{102}$ HARWOOD, L.M., MOODY, C.J. Experimental Organic Chemistry. Principles and Practice. Oxford: Blackwell Scientific, 1989.
${ }^{103}$ a) HAMAGUCHI, W., et al. Synthesis, SAR study, and biological evaluation of novel quinoline derivatives as phosphodiesterase 10A inhibitors with reduced CYP3A4 inhibition. Bioorganic and Medicinal Chemistry, 2015, vol 23, 297-313 [2014-11-29]. ISSN 0968-0896. doi: 10.1016/j.bmc.2014.11.039, Science Direct; b) MASUDA, N., et al. Pyrazole compound, WO 2012133607 A1, 2012 [2014-12-09], Espacenet.
${ }^{104}$ CARRASCO, R.A., STAMM, N.B., PATEL, B.K.R. One-step cellular caspase3/7 assay. BioTechniques, 2003, vol 34(5), 1064-1067 [2014-10-22]. ISSN 07366205 (print); 1940-9818 (online), PubMed.
${ }^{105}$ MALÍNKOVÁ, V., ŘEZNÍČKOVÁ, E., JORDA, R., GUCKÝ, T., KRYŠTOF, V. Trisubstituted purine inhibitors of $\operatorname{PDGFR} \alpha$ and their antileukemic activity in the hu${ }_{126} \mathrm{man}^{26}$ eosinophilic cell line EOL-1. Bioorganic and Medicinal Chemistry, 2017, vol

25(24), 6523-6535 [2018-03-10]. ISSN 0968-0896. doi: 10.1016/j.bmc.2017.10.032, Science Direct.

## CURRICULM VITAE

## Personal information:

Vaida Milišiūnaitė
Date of birth: 19890624
Address: Geležinio Vilko 20-47, Kaunas
Email: vaida.milisiunaite@gmail.com

## Education:

2014-2018 $\quad$| Kaunas University of Technology, Faculty of Chemical Tech- |
| :--- |
| nology, Department of Organic Chemistry. PhD in Chemistry |

2012-2014 Kaunas University of Technology, Faculty of Chemical Tech-
2013 03-06 Erasmus internship at the University of Vienna, Department of Pharmaceutical Chemistry, Division of Drug Synthesis, group of Prof. Dr. W. Holzer
2008-2012 Kaunas University of Technology, Faculty of Chemical Technology, Department of Organic Chemistry. BSc in Chemistry
2003-2008 Siauliai Didzdvaris Gymnasium
2002-2003
1996-2002
Siauliai Simonas Daukantas Gymnasium
Siauliai Lieporiai Gymnasium
Work experience:
2012 05-09
2012 09-2013 0

Chemist - analyst at Ltd "Aconitum"
Students' Research Activities supported by the Research Council of Lithuania

2013 03-06 Senior Engineer at the Institute of Synthetic Chemistry, Kaunas University of Technology
2014 01-07 Specialist of the Project at the Institute of Synthetic Chemistry, Kaunas University of Technology
2014 10-2015 05 Junior Researcher at Kaunas University of Technology, Faculty of Chemical Technology, Department of Organic Chemistry
2016 08-2017 05 Engineer of the Project at Kaunas University of Technology, Faculty of Chemical Technology, Department of Organic Chemistry
2016 12-2018 08 Junior Researcher at Kaunas University of Technology, Faculty of Chemical Technology, Department of Organic Chemistry
2017 03-2018 01 Senior Engineer at the Institute of Synthetic Chemistry, Kaunas University of Technology
Language skills: English (fluent speaking, reading, writing), Russian (interme-

## LIST UF PUBLICATIONS

## Publications in the journals inscribed into the list approved by the Information Scientific Institute (ISI)

-Milišiunnaitè, Vaida; Arbačiauskienė, Eglè; Řezníčková, Eva; Jorda, Radek; Malínková, Veronika; Žukauskaitè, Asta; Holzer, Wolfgang; Šačkus, Algirdas; Kryštof, Vladimír. Synthesis and anti-mitotic activity of 2,4- or 2,6-disubstituted- and 2,4,6-trisubstituted-2H-pyrazolo[4,3-c]pyridines // European Journal of Medicinal Chemistry. Paris : Elsevier Masson SAS. ISSN 0223-5234. eISSN 1768-3254. 2018, Vol. 150, pp. 908-919. DOI:10.1016/j.ejmech.2018.03.037.
-Milišīnaité, Vaida; Arbačiauskiené, Eglè; Bieliauskas, Aurimas; Vilkauskaitè, Gytè; Šačkus, Algirdas; Holzer, Wolfgang. Synthesis of pyrazolo[4' , $3^{\prime \prime}$ :3,4]pyrido[1,2- $a$ ]benzimidazoles and related new ring systems by tandem cyclisation of vic-alkynylpyrazole-4-carbaldehydes with (het)aryl-1,2-diamines and investigation of their optical properties // Tetrahedron. Oxford : Pergamon-Elsevier Science. ISSN 0040-4020. 2015, Vol. 71, iss. 21, pp. 3385-3395. DOI: 10.1016/j.tet.2015.03.092.

## Publications in the International and Lithuanian conference books of abstracts

-Milišīñaité, Vaida; Arbačiauskiené, Eglè; Rezničková, Eva; Jorda, Radek; Malinková, Veronika; Žukauskaitè, Asta; Holzer, Wolfgang; Kryštof, Vladimir; Šačkus, Algirdas. Synthesis of novel 2 H -pyrazolo[4,3-c]pyridines and investigation of their anti-mitotic activity // Congres SCF18, 2-4 July 2018 : book of abstracts / Societe chimique de France : [s.n.]. 2018, p. 501.
$\bullet$ ZŽukauskaitė, Asta; Milišiūnaitė, Vaida; Arbačiauskienė, Eglé; Řezníčková, Eva; Jorda, Radek; Malínková, Veronika; Holzer, Wolfgang; Kryštof, Vladimír; Šačkus, Algirdas. Synthesis and anti-mitotic activity of variously substituted 2 H -py-razolo[4,3-c]pyridines // German-Polish-Baltic Conference on Organic Chemistry: Hamburg, 15-19 May 2018: book of abstracts. Hamburg : [s.n.]. 2018. ISBN 9783000588839. p. 38.
-Milišiūnaitè, Vaida; Arbačiauskienė, Eglè; Řezníčková, Eva; Žukauskaitè, Asta; Holzer, Wolfgang; Doležal, Karel; Strnad, Miroslav; Kryštof, Vladimír; Šačkus, Algirdas. Synthesis and biological evaluation of novel 2 H -pyrazolo[4,3-c]pyridines // Bioheterocycles 2017: XVII International Conference on Heterocycles in Bioorganic Chemistry, 28-31 May 2017, Galway, Ireland : book of abstracts. Galway : [s.n.]. 2017, P0-21, p. 128.
-Milišiūnaitè, Vaida; Arbačiauskienė, Eglė; Šačkus, Algirdas; Holzer, Wolfgang. A straightforward approach to novel fused pyrazole systems // Chemistry and Chemical Technology : International Conference of Lithuanian Society of Chemistry : Lithuanian Academy of Science, Vilnius, Lithuania, April 28-29 2016 : book of abstracts / Fiziniụ ir technologijos mokslụ centras, Vilniaus universitetas, Lietuvos mokslų akademija, Kauno technologijos universitetas. [S.1.] : [s.n.], 2016. ISBN 9786099551135. p. 176.
-Milišiūnaitė, Vaida; Arbačiauskienė, Eglė; Bieliauskas, Aurimas; Holzer, Wolfgang; Šačkus, Algirdas. Synthesis and investigation of derivatives of pyrazolo[4',3',3', 4 ]pyrido[1,2-a]benzimidazole and related new ring systems // 16th Tetrahedron Symposium: Challenges in Bioorganic and Organic Chemistry, 16-19 June 2015, Grand Hyatt Berlin, Germany. Berlin : Elsevier. 2015, P.2.063, p. 48.

- Milišiūnaitė, Vaida; Arbačiauskienė, Eglė; Holzer, Wolfgang; Šačkus, Algirdas. Novel fused heterocyclic ring systems containing the pyrazole unit // Bioheterocycles 2015: XVI International Conference on Heterocycles in Bioorganic Chemistry, 8-11 June 2015, Metz, France: book of abstracts. Metz : [s.n.]. 2015, P037, p. 101.
-Milišiūnaitė, Vaida; Arbačiauskienė, Eglė; Šačkusa, Algirdas; Holzer, Wolfgang. Novel fused heterocyclic ring systems containing the pyrazole unit // XL International Summer School on Organic Synthesis : A. Corbella - ISOS 2015, Gargnano (BS), Palazzo Feltrinelli, 14-18 June 2015, p. 25.
-Milišiūnaitė, Vaida; Arbačiauskienė, Eglė; Šačkus, Algirdas. Kondensuotų heterociklinių sistemų sintezė iš 1-fenil-3-hidroksi-1H-pirazolo // Studentų moksliniai tyrimai 2012/2013 : konferencijos pranešimų santraukos. D. 2. / Lietuvos mokslo taryba. Vilnius : Lietuvos mokslo taryba, 2013. ISBN 9789955613541. p. 331. Prieiga per internetą:
<http://studentai.lmt.lt/DOKUMENTAI/KONFERENCIJOS/SMT\ 2013\ \ II\ tomas\ TYRIMAI.pdf.


## ACKNOWLEDGEMENTS

Prof. Dr. Habil. A. Šačkus, Department of Organic Chemistry, Kaunas University of Technology, is greatly acknowledged for the supervision and useful advices while preparing this work and giving an opportunity to work in his research group.

The author sincerely thanks Prof. Dr. W. Holzer, Department of Drug and Natural Product Synthesis, University of Vienna, for the supervision and giving the opportunity to work in his research group.

The author as well thanks Dr. Eglė Arbačiauskiené, Department of Organic Chemistry, Kaunas University of Technology, for her guidance while synthesizing, advices, encouragement, consultations, and friendship during all the years of study.

The author of this dissertation thanks Dr. Asta Žukauskaitè from the Palacký University for organizing biological activity investigations at Palacky University, her guidance, and advices to improve biological activities.

The author is grateful to colleagues from the Department of Organic Chemistry and the Institute of Synthetic Chemistry, Kaunas University of Technology: Dr. N. Kleiziené, Dr. J. Solovjova, Dr. G. Ragaité, Dr. V. Buinauskaitè, Dr. M. Dagiliené, for their consultations, encouragement, and guidance during the first steps in the organic synthesis. Warm thanks go to friends and colleagues from the Department of Organic Chemistry, Kaunas University of Technology: PhD student Elena Ščerbetkaitè, Gerda Dubickaité, Valentas Varnelis, Urtė Šachlevičiūté, for their sincere friendship, help, and support.

The author of this dissertation kindly thanks Aurimas Bieliauskas from the Department of Organic Chemistry, Kaunas University of Technology, for the NMR analysis, HRMS, and fluorescence measurements and Saulius Burinskas from the Institute of Synthetic Chemistry, Kaunas University of Technology, for the HRMS measurements.

The author is as well thankful to Kaunas University of Technology, Research Council of Lithuania for financial support.

Most of all, the author of this dissertation wants to say sincerely thank the family: parents Genovaite and Artūras and brother Povilas for their supporting love, encouragement, and understanding.

SL344. 2018-10-26, 16,75 leidyb. apsk. 1. Tiražas 16 egz. Užsakymas 294.
Išleido Kauno technologijos universitetas, K. Donelaičio g. 73, 44249 Kaunas Spausdino leidyklos „Technologija" spaustuvé, Studentų g. 54, 51424 Kaunas


[^0]:    * Data are means of at least two independent measurements.

[^1]:    4.1.2.14.4. (2,3-Difluorophenyl)(3-hydroxy-1-phenyl-1H-pyrazol-4-yl)methanone (93). White solid, yield $126 \mathrm{mg}, 84 \%, \mathrm{mp} 161-162{ }^{\circ} \mathrm{C}$ (ethyl acetate). IR $\left(v_{\max }, \mathrm{cm}^{-1}\right): 3327(\mathrm{OH}), 3100,3079,3069,3043\left(\mathrm{CH}_{\text {arom }}\right), 1628(\mathrm{C}=\mathrm{O}), 1597,1590$, $1577,1508,1495,1479,1440,1339,1311,1273,1221,1155,1051,1002(\mathrm{C}=\mathrm{C}, \mathrm{C}=\mathrm{N}$, C-F), 861, 852, 813, 771, 757, 752, 669, $621(\mathrm{CH}=\mathrm{CH}$ mono- and trisubstituted benzenes). ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta_{\mathrm{H}} \mathrm{ppm} 7.25-7.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CPh} 5-\mathrm{H}), 7.33-$ 7.36 (m, 1H, NPh 4-H), 7.38-7.43 (m, 1H, CPh 4-H), 7.44-7.48 (m, 3H, NPh 3,5-H, CPh 6-H), 7.67-7.69 (m, 2H, NPh 2,6-H), 8.05-8.06 (m, 1H, 5-H). ${ }^{13}$ C NMR (176

