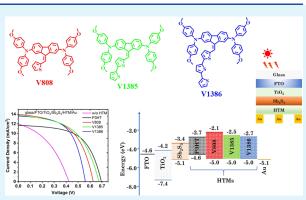
4.9% Efficient Sb₂S₃ Solar Cells from Semitransparent Absorbers with Fluorene-Based Thiophene-Terminated Hole Conductors

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terminating thiophene units are explored, for the first time, for antimony sulfide (Sb₂S₃) solar cells. These HTMs possess largely simplified synthesis processes and high yields compared to the conventional expensive hole conductors making them reasonably economical. The thiophene unit-linked HTMs have been successfully demonstrated in ultrasonic spray-deposited Sb₂S₃ solar cells resulting in efficiencies in the range of 4.7-4.9% with an average visible transmittance (AVT) of 30-33% (400-800 nm) for the cell stack without metal contact, while the cells fabricated using conventional P3HT have yielded an efficiency of 4.7% with an AVT of 26%. The study puts forward cost-effective and transparent HTMs that avoid a post-coating activation at elevated temperatures like P3HT, devoid of



parasitic absorption losses in the visible region and are demonstrated to be well aligned for the band edges of Sb_2S_3 thereby ascertaining their suitability for Sb_2S_3 solar cells and are potential candidates for semitransparent applications.

KEYWORDS: antimony sulfide, solar cells, thiophene, hole transport material, semitransparent

1. INTRODUCTION

The energy crisis is among the major problems of the present day's growing world with digitalization and the internet of things. In addition, the limited availability of fossil fuels, the dominant sources of current days' energy and the associated problems of pollutant emissions by burning fossil fuels is a bigger concern. Considering the UNO's policies of net-zero emissions by 2050 coupled with the goals set by many countries worldwide to produce clean energy necessitates the requirement of alternative energy sources that are clean, abundant, and easy to harvest. Renewable energy technologies are the need of the hour to meet these demands and among them, solar photovoltaics (PV) is a technology that has tremendous potential to be a major yet important energy harvesting source. Solar energy is abundantly available and PV is a green technology that facilitates the conversion of sunlight to electricity in a clean manner once the solar panels are in fully operational condition. Furthermore, building-integrated photovoltaics (BIPV) is an emerging technology in PV wherein solar panels could be integrated into buildings that produce energy to meet some portion of energy demand within. Energy-efficient buildings are going to be the future with the growing smart city infrastructure and contribute effectively toward net-zero emissions. Semitransparent solar cells could play an important role in BIPV and could be integrated into

windows. Intense investigations are underway to explore novel materials and device architectures that are suitable for semitransparent solar cells.

Among various emerging materials, antimony sulfide (Sb_2S_3) is the promising candidate for the photovoltaic community owing to the earth-abundant and environmentally friendly constituent elements alongside appropriate optoelectronic properties such as a desirable band gap of ≈ 1.7 eV, large absorption coefficient ($\approx 10^5$ cm⁻¹) and long-term stability.^{1–3} The Shockley–Queisser limit of the power conversion efficiency (PCE) of the single junction Sb_2S_3 solar cells is 28.64%.¹ Sb_2S_3 absorbers are potential contenders for semitransparent solar cells due to their relatively wide band gap and the possibility to yield higher PCEs even with very thin absorber layers (≈ 100 nm).⁴ Further, the wide band gap of Sb_2S_3 also makes them a wiser choice for tandem cell applications.^{5–7} The highest reported PCE of Sb_2S_3 solar cells

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Table 1. Reported Device Parameters of Sb₂S₃ Solar Cells with Different HTMs, Device Configurations, Methods of Preparation, and Thicknesses^a

| | | | | solar cell parameters | | | | |
|------------------------------------|---|----------------------------------|--|-------------------------|--|-----------|------------|------------|
| HTM | device configuration | Sb_2S_3 (deposition technique) | Sb ₂ S ₃ thickness (nm) | V _{OC} (mV) | J _{SC} (mA/cm ²) | FF (%) | PCE (%) | reference |
| РЗНТ | ITO/ZnO/Sb ₂ S ₃ /P3HT/Ag | TE | 210 | 450 | 12.6 | 42 | 2.4 | 21 |
| | ITO/TiO ₂ /Sb ₂ S ₃ /P3HT/Au | CBD | 200 | 630 | 6.1 | 35 | 1.4 | 22 |
| | ITO/TiO ₂ /Sb ₂ S ₃ /P3HT/Ag | ALD | 155 | 732 | 9.3 | 62 | 4.3 | 2 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /P3HT/Au | SC | | 616 | 8.1 | 46 | 2.3 | 16 |
| | ITO/TiO ₂ /Sb ₂ S ₃ /P3HT/Au | USP | 150 | 618 | 6.0 | 51 | 1.9 | 23 |
| | ITO/TiO ₂ /Sb ₂ S ₃ /P3HT/Au | USP | 100 | 693 | 13.8 | 58 | 5.5 | 6 |
| Spiro-OMeTAD | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Au | CBD | 84 | 690 | 13.4 | 50 | 4.6 | 17 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Au | SC | 400 | 660 | 13.1 | 59 | 5.2 | 18 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Au | SC | 140 | 632 | 12.9 | 52 | 4.3 | 24 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Ag | TE | 300 | 620 | 10.7 | 56 | 3.8 | 25 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Au | SC | 162 | 650 | 17.7 | 62 | 7.1 | 26 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /Spiro-OMeTAD/Au | SC | 146 | 720 | 17.2 | 57 | 7.1 | 27 |
| | FTO/CdS/Sb ₂ S ₃ /Spiro-OMeTAD/Au | HT | 300 | 707 | 15.2 | 56 | 6.0 | 28 |
| | FTO/CdS/Sb ₂ S ₃ /Spiro-OMeTAD/Au | TE | 200 | 720 | 15.9 | 54 | 6.2 | 29 |
| | FTO/TiO ₂ /CdS/Sb ₂ S ₃ /Spiro- OMeTAD/Au | HT | 1000 | 748 | 15.3 | 57 | 6.5 | 30 |
| | FTO/CdS/Sb ₂ S ₃ /Spiro-OMeTAD/Au | CBD | 214 | 757 | 17.4 | 60.5 | 8.0 | 8 |
| P3HT/ PEDOT:PSS | FTO/TiO ₂ /ZnS/Sb ₂ S ₃ /P3HT/ PEDOT:PSS/Au | ALD | 75 | 626 | 15.7 | 52 | 5.1 | 31 |
| | FTO/TiO ₂ /Sb ₂ S ₃ /P3HT/PEDOT:PSS/ Au | ALD | 90 | 667 | 14.9 | 58 | 5.8 | 32 |
| V1236 | FTO/TiO ₂ /Sb ₂ S ₃ /HTM/Au | USP | 90 | 612 | 13.8 | 46 | 3.9 | 33 |
| ^{<i>a</i>} HTM· hole tran | sport material: TE: thermal evaporation: | CBD: chemical bath | deposition: SC: st | nin coatin | o USP ultra | sonic sn | ray pyro | lysis: HT. |

"HTM: hole transport material; TE: thermal evaporation; CBD: chemical bath deposition; SC: spin coating; USP: ultrasonic spray pyrolysis; HT: hydrothermal; ALD: atomic layer deposition, V1236: (N^2,N^2,N^7,N^7) -tetrakis[2,2-bis(4-methoxyphenyl)vinyl]-9,9-dihexyl-9H-fluorene-2,7-diamine).

is 8%, achieved recently through a multi-sulfur source approach using the chemical bath deposition method (CBD).⁸ The overall efficiencies are rather limited by the deficiencies in open circuit voltage ($V_{\rm OC}$) that are attributed to the self-trapping of carriers due to lattice deformation and/or to the intrinsic defects, which sets the maximum attainable $V_{\rm OC}$ to 0.8 V.^{9,10} Antimony selenide (Sb₂Se₃), another class of antimony chalcogenide, is also a potential candidate for PV applications and is explored extensively.^{11–13}

In addition to the considerable challenges in improving the PCEs, the method of Sb_2S_3 absorber fabrication is critical for upscaling. The high-efficiency Sb₂S₃ solar cells so far are mostly prepared by spin coating and CBD.^{8–15} While the spin coating technique demonstrates excellent control of parameters and reproducibility but is limited to laboratory scale and also has high material wastage (>90%). CBD, on the other hand, has promise for large-scale demonstration but needs longer deposition time at relatively low temperatures (<10 °C) to obtain the desired thickness of the absorbers. Our group has previously demonstrated the fabrication of Sb₂S₃ solar cells by the ultrasonic spray pyrolysis method, which offers large-area uniformity and is an industrially scalable technique with low capital.⁶ Further, the Sb₂S₃ solar cells are often fabricated in planar superstrate configuration wherein TiO₂ is the most commonly used electron transport layer (ETL) deposited on a transparent conducting oxide followed by the Sb₂S₃ absorber, hole transport material (HTM), and finally, the back contact, often Au. HTMs are the vital component of planar or mesoscopic Sb₂S₃ solar cells that drive the transport of photogenerated holes to the back contact and reduce recombination losses.² In general, conventional HTMs like poly(3-hexylthiophene) (P3HT), 2,16 2,2',7,7'-tetrakis[N,Ndi(4-methoxyphenyl)amino]-9,9'-spirobifluorene (Spiro-OMeTAD),^{17,18} poly{4,4-dialkyl-4*H*-cyclopenta[2,1-*b*;3,4-*b'*]dithiophene-2,6-diyl-*alt*-2,1,3-benzothiadiazole-4,7-diyl} (PCPDTBT),^{19,20} etc. are the most commonly employed in the fabrication of Sb₂S₃ solar cells. The use of these different HTMs in Sb₂S₃ solar cells where the Sb₂S₃ absorbers are prepared by different techniques and device architectures are summarized in Table 1 alongside the power conversion efficiencies achieved. It may be noted that only organic HTMs and planar configuration devices are considered for summary.

It may be noted that except for V1236, an economic and transparent hole conductor previously reported by our group,³³ other studies have majorly used conventional P3HT and Spiro-OMeTAD. These hole conductors are very expensive mainly due to their complex, expensive, and time-consuming synthesis processes³⁴ and are also moisture sensitive, and thereby could be significant barriers to the potential upscale for commercial applications. Further, tailoring HTMs with suitable thickness is also important to obtain high-efficiency solar cells but may hinder the overall transparency of the solar cell stack, an important parameter in semitransparent solar cells. For instance, the commonly employed P3HT exhibits parasitic absorption in the visible spectral region, which induces a loss in the total transmittance of the solar cell.^{2,33} In addition, P3HT also requires additional post-coating activation at temperatures around 170 °C either in a vacuum or inert atmosphere.³⁵ Therefore, it becomes imperative to explore low-cost novel HTMs, which could be easily synthesized using available commercial starting materials while possessing suitable properties like good thermal stability, desired solubility, and hole mobility.

In view of the above, the study herein proposes to fabricate Sb₂S₃ solar cells by a scalable and industrially benign ultrasonic

spray pyrolysis (USP) method by employing novel, low-cost, and transparent hole conductors from fluorene-based molecules with terminated thiophene units. The 4,4'-dimethoxydiphenylamine-substituted fluorene fragment is used due to its wider application in the organic HTMs, in particular, Spiro-OMeTAD, while thiophene units are added mimicking P3HT HTM. The new materials follow simple synthesis processes with high yields thereby having considerably lower costs than conventional HTMs. Sb_2S_3 solar cells with new HTMs are fabricated in the standard superstrate configuration wherein ETL and absorber are deposited by USP while HTMs are spin coated followed by the thermal evaporation of Au. Comprehensive characterization of solar cells through different techniques clearly demonstrates that the cells with new HTMs yield similar or better PCEs compared to the conventional P3HT-based devices and have exhibited enhanced average visible transmittance in the 400-800 nm range. Further, the band energetics show that the new HTMs possess favorable band alignment with the Sb₂S₃ absorber thereby validating their efficacy toward semitransparent solar cells.

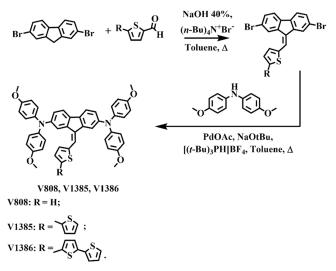
2. EXPERIMENTAL SECTION

2.1. Materials. The following materials were used: FTO substrate (7 Ω sq⁻¹), titanium(IV) tetraisopropoxide (TTIP)—99 wt % (Acros Organics), acetylacetone—99 wt % (Acros Organics), ethanol—96.6 vol % (Estonian Spirit), methanol—99.9 vol % (Sigma-Aldrich), antimony trichloride—99.99 wt % (Sigma-Aldrich), thiourea—99 wt % (Sigma-Aldrich), chlorobenzene—99.5 vol % (Sigma-Aldrich), poly(3-hexyl-thiophene-2,S-diyl) (P3HT)—100 kDa, >90% regioregular (Sigma-Aldrich), and materials related to HTM synthesis as mentioned in the S1. The materials were used as received.

2.2. Solar Cell Fabrication. Antimony sulfide (Sb₂S₃) solar cells were fabricated in superstrate configuration (glass/FTO/TiO₂/Sb₂S₃/ HTM/Au). The fabrication process started with thorough cleaning of the glass/FTO substrate with DI water, acetone, and IPA in an ultrasonic bath for 10 min each followed by boiling the substrates in DI water and purging with N2 gas. The TiO2 electron transport layer (ETL) and Sb₂S₃ absorbers were deposited by an indigenously made ultrasonic spray deposition system. TiO2 was sprayed from a solution mixture of 0.2 M TTIP with 0.2 M acetylacetone in ethanol by using a spray rate of ≈ 2.5 mL/min for 30 min on a hot plate maintained at around 340 °C. The sprayed TiO₂ samples were then annealed at 450 °C for 30 min in air. Sb₂S₃ thin films were spray-deposited from a solution mixture of 60 mM SbCl₃ and 180 mM thiourea in methanol. The films were deposited at a spray rate of 2 mL/min while maintaining a substrate temperature of ≈ 200 °C. The as-deposited Sb_2S_3 films were annealed at around 250–260 $^\circ C$ for 5 min under nitrogen. The hole transport materials (HTM) explored in this study are the fluorene-based molecules with linked thiophene units N^2, N^2, N^7, N^7 -tetrakis(4-methoxyphenyl)-9-(thiophen-2-ylmethylene)-9H-fluorene-2,7-diamine (V808), 9-([2,2':5',2"-terthiophen]-5-ylmethylene)- N^2 , N^2 , N^7 , N^7 -tetrakis(4-methoxyphenyl)-9H-fluorene-2,7diamine (V1385), and 9-([2,2'-bithiophen]-5-ylmethylene)- N^2 , N^2 , N^7 , N^7 -tetrakis (4-methoxyphenyl)-9H-fluorene-2, 7-diamine (V1386). The detailed synthesis procedure of V808, V1385, and V1386 is described in Section S1 while the synthesis process is presented in Scheme 1. The HTM layers were spin coated on glass/ FTO/TiO₂/Sb₂S₃ samples followed by the thermal evaporation of Au contacts. The active area of solar cells was 7.06 mm².

2.3. Characterization. The crystal structure and phase constitution of the as-deposited and annealed Sb_2S_3 thin films were examined using a Rigaku Ultima IV X-ray diffractometer with a Cu $K\alpha$ source ($\lambda = 1.5406$ Å). In addition, micro-Raman spectroscopy was further used to ascertain the crystalline purity while also detecting the possible presence of the minute antimony oxide phase and micro-Raman spectra were recorded at room temperature using a Horiba Labram HR 800 in backscattering mode with a He Ne laser of 532

Scheme 1. Synthetic Route to the Novel Hole Transporting Materials V808, V1385, and V1386



nm. A Zeiss HR FESEM was used to analyze the cross-sectional morphologies of the layers. The optical absorption and transmittance measurements were made using a Jasco V-670 ultraviolet-visible spectrophotometer (UV-VIS) in the range of 300-1000 nm. For the band diagram construction, photoelectron emission spectroscopy (PES) and photoconductivity methods were used. For these methods, the measurement system consisted of an ENERGETIQ Laser-Driven Light Source (LDLS EQ-99) as a light source, a Spectral Products DK240 1/4m monochromator, and a Keithley 617 electrometer. The measurements were performed in vacuum at a pressure of $\sim 10^{-5}$ mbar. Both methods are described in detail in S2. The hole transport materials were characterized comprehensively using thermogravimetric analysis (TGA), differential scanning calorimetric (DSC) analysis, and absorption spectroscopy. The work function of Au contacts was measured using a Kelvin probe. The current-voltage (I-V)characteristics of the fabricated solar cells were measured using Wavelabs LS-2 LED solar simulator with an AM1.5G (100 mW cm⁻² light source. A Newport 69911 system with a 300 W Xenon lamp was used to measure the external quantum efficiency (EQE) of the fabricated solar cells.

3. RESULTS AND DISCUSSION

The fluorene-based thiophene-linked hole transport materials (HTMs) are explored in this study for the fabrication of Sb_2S_3 solar cells. The design of these p-type photoconductors has been inspired by recently published findings that thiophene can coordinate with Sb atoms in $Sb_2(S,Se)_3$ leading to the improvement of the interfacial properties between $Sb_2(S_2Se)_3$ and HTM.^{36,37} This interaction provides desirable channels for carrier transport, which significantly enhances the open circuit voltage and fill factor. In a single thiophene unit-linked molecule (V808), 4,4-dimethoxydiphenylamine chromophores are symmetrically connected on both sides of the fluorene core, which was easily modified chemically by the thiophene endgroup. V808 is synthesized through a simple two-step process (Scheme 1) devoid of expensive/complex materials/steps with considerably good yields thereby resulting in low synthesis costs. On a similar note, to further investigate the possibility of better interaction, the molecules are synthesized by adding 2 and 3 thiophene units to the fluorene core, which are thus named V1385 and V1386, respectively. The detailed synthesis procedure of V808, V1385, and V1386 is described in Section S1 and the cost estimation is given in Tables S1-S3, respectively. It may be observed that the synthesis cost of a

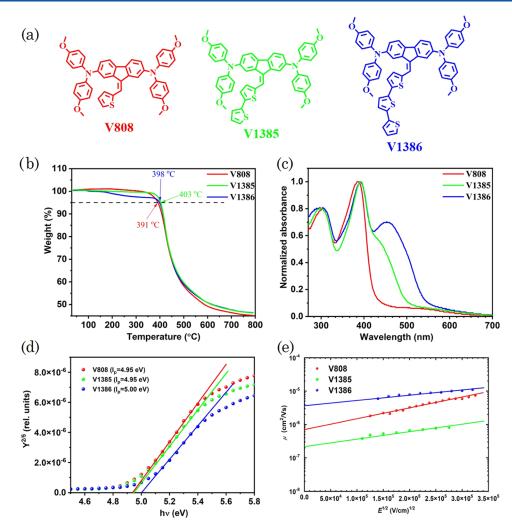


Figure 1. (a) Chemical structures of new fluorene-based hole transport materials with terminating thiophene units (V808, V1385, and V1386), (b) thermogravimetric analysis (TGA) data, (c) normalized absorbance spectra of the HTMs deposited on glass, (d) photoemission spectra in air for determining the ionization potential, and (e) electric field dependencies of the hole mobility of V808, V1385, and V1386.

single thiophene linking molecule (V808) is $\approx 13 \in$ per gram while those of the ones with 2 and 3 thiophene units are 32 and 177 € per gram, respectively, indicating the increase in cost is majorly owing to the increased price of the starting thiophenecarbaldehyde. However, the cost of single and double thiophene unit-linked molecules (V808 and V1385) is considerably cheaper in comparison to the conventional hole conductors like P3HT and Spiro-OMeTAD. Irrespective of the cost, all three materials are comprehensively characterized and are explored as HTMs in Sb₂S₃ solar cells. The typical chemical structures of V808, V1385, and V1386 with 1, 2, and 3 linked thiophene units, respectively, are shown in Figure 1a. Figure 1b shows the thermogravimetric analysis (TGA) curves of V808, V1385, and V1386, and as the plots reveal all three materials start to decompose at temperatures >390 °C, which is far above the temperature for conventional device operation. The differential scanning calorimetric (DSC) curves of new HTMs are shown in Figure S1. All compounds have shown T_{α} slightly less than 100 °C (92, 97, and 99 °C for V808, V1385, and V1386, respectively). Comparison of T_g of new materials reveals that an increase in thiophene fragments in the molecule leads to slightly increased T_g, and could be attributed to the increased molecular weight. In addition, it is important to note that the melting process is observed just for V808, while other

new synthesized compounds (V1385 and V1386) have shown no endothermic peaks, indicating that they have only an amorphous state, which is an advantage for the formation of homogenous films as it eliminates the possibility of crystallization of films during preparation/operation of devices. The normalized absorbance spectra of the films, deposited on glass, are shown in Figure 1c. As can be seen, the spectra of new compounds display intensive $\pi - \pi^*$ absorption bands at 270-430 nm, which might be attributed to the 4,4'dimethoxydiphenylamine-substituted fluorene fragment. In addition, with the increase in the number of thiophene units, the $(\pi - \pi^*)$ transition associated with this conjugated part exhibits a strong bathochromic shift. For compound V1386, this results in a band with the absorption maximum at 453 nm, for V1385 it appears as a shoulder to the main peak, while for V808 it overlaps with the main peak. The normalized absorbance spectra of new HTMs are also compared with the conventional P3HT as presented in Figure S2, which shows that P3HT absorbs significantly in the 400-650 nm region while the new HTMs possess much less absorption in the same region thereby demonstrating better visible light transparency.

The solid-state ionization potential (I_p) of HTMs is determined using photoelectron emission spectroscopy

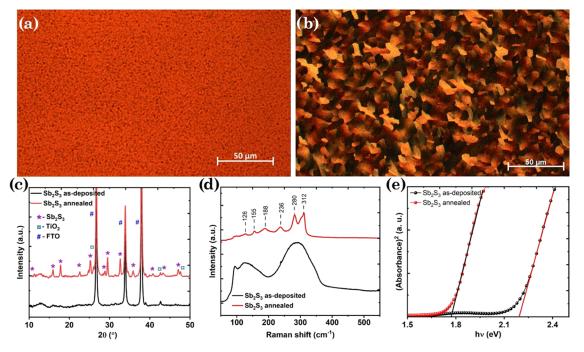


Figure 2. Morphological, structural, and optical characteristics of Sb_2S_3 thin films deposited by ultrasonic spray pyrolysis: Optical microscopy images of (a) as-deposited and (b) annealed Sb_2S_3 , (c) XRD patterns, (d) Raman spectra, and (e) (absorbance)² vs $h\nu$ plots of as-deposited and annealed Sb_2S_3 .

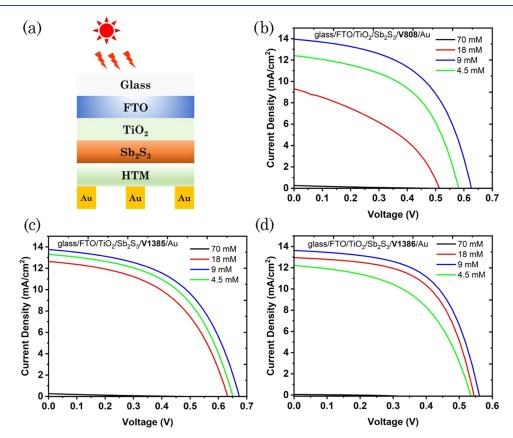


Figure 3. (a) Schematic illustration of device configuration (glass/FTO/TiO₂/Sb₂S₃/HTM/Au) of Sb₂S₃ solar cells (Note: Not to scale), J-V characteristics of solar cells with (b) V808, (c) V1385, and (d) V1386 as HTMs with the variation in precursor concentrations from 4.5 to 70 mM in each case.

(PES) of the thin films (Figure 1d), which clearly reveals that the I_p levels of **V808**, **V1385**, and **V1386** are determined to be -4.95 eV, -4.95, and -5.00 eV, respectively. Almost similar I_p

values can be explained by the unfavorable arrangement of thiophene fragments. As can be seen from absorbance spectra, the increase of the conjugated system is very slight with the

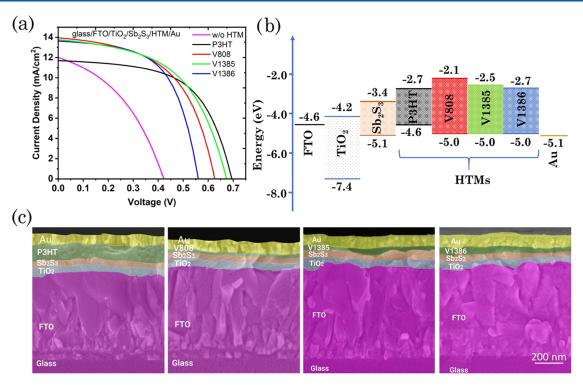


Figure 4. (a) J-V characteristics of champion cells, (b) band energy diagram and (c) cross-sectional images of Sb₂S₃ solar cells with P3HT, V808, V1385, and V1386 as HTMs, respectively.

increasing number of thiophene moieties in the molecule, therefore the highest occupied molecular orbital (HOMO) levels can be very similar. Further, hole drift-mobility of HTMs is estimated using the time of flight (ToF) method with Ekspla PL kHz ps tunable laser as a light source and a Tektronix MDO3024 oscilloscope for signal registration. As shown in Figure 1e, the inferred zero-field hole mobilities are 7.07×10^{-7} , 2.15×10^{-7} , and $3.68 \times 10^{-6} \text{ cm}^2/(\text{V}\cdot\text{s})$, for V808, V1385, and V1386, respectively.

Sb₂S₃ thin films are deposited by the USP process on glass/ FTO/TiO₂ samples. Optical microscopic images are recorded to ascertain the conformal coverage of $\mbox{Sb}_2\mbox{S}_3$ films on \mbox{TiO}_2 as shown in Figure 2a and as the image, which is recorded for a considerably larger area, unveils that the films are conformally covered devoid of any pin holes or crystallization during the deposition. The crystallization of Sb₂S₃ thin films during deposition in the USP process is a commonly observed feature, which is a detrimental process that reduces the quality of films for solar cell applications as this crystallization during deposition is an uncontrollable process and the resulting crystallinity is not the desired form too, which is absent in the present study. Further, the as-deposited films are characterized by XRD and Raman as shown in Figure 2c,d. The XRD pattern shows the sharp peaks pertaining to FTO and TiO₂ layers while the Raman spectra reveal broad characteristic peaks of Sb₂S₃ confirming that the as-deposited films are amorphous in nature. Upon annealing under a nitrogen atmosphere at 250 °C, the Sb₂S₃ films exhibit crystallinity. Optical microscopy images of annealed Sb₂S₃ films show a large grain structure as shown in Figure 2b without any voids or pin holes. The XRD patterns of annealed Sb₂S₃ show several sharp peaks in addition to the existing FTO and TiO₂ peaks, which correspond to the orthorhombic stibnite Sb₂S₃ phase (ICDD PDF 01-075-4013). On a similar note, the Raman spectra of annealed Sb_2S_3 show sharp peaks characteristic of Sb_2S_3

thereby corroborating the XRD analysis and confirming the formation of the crystalline stibnite Sb_2S_3 phase. Further, the band gap of Sb_2S_3 absorbers is inferred from (absorbance)² vs $h\nu$ plots as shown in Figure 2e. The band gap of amorphous Sb_2S_3 is determined to be around 2.2 eV while that of crystallized one is 1.7 eV, which agree well with the values reported previously.³³ Therefore, it may be inferred that the crystalline Sb_2S_3 absorbers with the desired band gap are successfully fabricated by the area-scalable USP process followed by annealing.

In the next step, the Sb₂S₃ solar cells are fabricated in the planar configuration, as shown in Figure 3a, using conventional P3HT and new V808, V1385, and V1386 as HTMs, which are spin coated onto glass/FTO/TiO₂/Sb₂S₃ samples and Au is thermally evaporated. While the concentration of P3HT has been previously optimized for the best efficient Sb₂S₃ solar cells,³³ the new HTMs require a similar process. Therefore, prior to comparison with P3HT-based solar cells, the concentration of the precursors in spin coating has been optimized for new HTMs. Figure 3b-d shows the current density-voltage (J-V) characteristics of solar cells with concentration variation in each case for V808, V1385, and V1386, respectively. The variation in the associated device parameters with concentration is shown in Figure S3. For an initial concentration of 70 mM V808, the solar cells exhibit very low PCEs, essentially due to the negligible photocurrent. The concentration of V808 is therefore diluted considerably in steps of ca. 4× resulting in 18, 9, and 4.5 mM. As shown in the I-V curves in Figure 3b, the efficiencies of solar cells for V808 concentrations of 70, 18, 9, and 4.5 mM are 0.1, 2.0, 4.7, and 3.8%, respectively. The device parameters, $V_{\rm OC}$, $J_{\rm SC}$, and FF have followed an identical trend that the respective values are maximum for 9 mM V808 and are lower on either side (see Figure S3a,b). To identify the plausible reasons, the series resistance (R_s) of the devices is determined from the I-V data

that a very large R_s (>200 $\Omega \cdot cm^2$) is observed for the 70 mM V808 device, which decreased significantly with a decrease in the concentration resulting in considerable improvement in $J_{\rm SC}$. The large $R_{\rm s}$ may be attributed to the larger thickness of V808 arising from the large concentration. It may therefore be concluded that 9 mM V808 is considered optimal resulting in the highest PCE among all. On a similar note, the power conversion efficiencies of the solar cells with 70, 18, 9, and 4.5 mM concentrations of V1385 are obtained to be 0.2, 4.0, 4.9, and 4.5%, respectively. For the case of V1386, the obtained efficiencies are 0.15, 4.1, 4.5, and 3.4% for concentrations of 70, 18, 9, and 4.5 mM. The variation trends in the associated device parameters are also similar as shown in Figure S3c-f. It may therefore be inferred that the 9 mM concentration in each case has resulted in the best device efficiencies of 4.7, 4.9, and 4.5% for solar cells with V808, V1385, and V1386 as HTMs, respectively.

While the new molecules **V808**, **V1385**, and **V1386** have demonstrated reasonable efficiencies, it is important to compare them with existing conventional proven HTMs. Therefore, to further validate the efficacy of new molecules as the HTM layer in Sb₂S₃ solar cells, they are compared with conventional P3HT-based devices. The J-V characteristics of champion Sb₂S₃ solar cells with P3HT, **V808**, **V1385**, and **V1386** are shown in Figure 4a. In addition, to understand the need for an HTM in devices, solar cells fabricated without the HTM layer are also shown. The device parameters are listed in Table 2. The solar cells without HTM have exhibited a PCE of

Table 2. Device Parameters of Champion Sb₂S₃ Solar Cells with P3HT, V808, V1385, and V1386 HTMs

| HTM | V _{OC} (V) | $J_{\rm SC}$ (mA/cm ²) | FF | PCE (%) | $(\Omega \cdot cm^2)$ | $R_{ m SH} \ (\Omega \cdot m cm^2)$ |
|--------|------------------------|------------------------------------|------|------------|-----------------------|--------------------------------------|
| no HTM | 0.43 | 12.0 | 0.41 | 2.12 | 1.33 | 167 |
| P3HT | 0.69 | 11.7 | 0.58 | 4.68 | 2.1 | 1190 |
| V808 | 0.63 | 13.9 | 0.54 | 4.73 | 1.01 | 586 |
| V1385 | 0.68 | 13.7 | 0.53 | 4.94 | 2.3 | 569 |
| V1386 | 0.56 | 13.6 | 0.59 | 4.50 | 0.89 | 844 |

2.12% with a reasonably low $V_{\rm OC}$ of 0.43 V, a $J_{\rm SC}$ of 12.0 mA/ cm², and an FF of 0.41. Such characteristics are often classically observed for bare Sb₂S₃-based solar cells devoid of HTM. The band energy positions of different device layers explored in the present study are determined using photoelectron emission spectroscopy and the band energy diagram is presented in Figure 4b. The Sb_2S_3 absorber possesses the valence (VBE) and conduction band edges (CBE) at -5.1 and -3.4 eV, respectively, and with Au and TiO₂ on either side of the absorber in solar cells without HTM, the transport of photogenerated electrons and holes is reasonably smooth owing to the favorable band edge positions. However, with Au being the back contact there is no barrier for electrons from the conduction band of Sb_2S_3 at the Sb_2S_3 -Au interface thereby leading to recombination often resulting in low $V_{\rm OC}$ and FF. The devices also exhibit a very low shunt resistance (R_{SH}) of $\approx 167 \ \Omega \cdot cm^2$. On the other hand, the solar cells with conventional P3HT have demonstrated a PCE of 4.68% with a $V_{\rm OC}$ of 0.69 V, a $J_{\rm SC}$ of 11.7 mA/cm², and an FF of 0.58. It can be observed that the device parameters are considerably increased, particularly the $V_{\rm OC}$ and FF. The presence of an HTM contributes in various ways to enhance the conversion efficiency of solar cells as they improve band energetics of the

device for improved transport of photogenerated carriers while reducing recombination losses as they block the electrons by passivating the interface across Sb₂S₃-HTM. As seen from Figure 4b, P3HT possesses the VBE and CBE at -4.6 and -2.7 eV, respectively, that its presence could provide a barrier at the Sb₂S₃-HTM interface for the electrons to travel back and therefore, reduce the recombination losses. The devices have shown a significantly enhanced $R_{\rm SH} \approx 1.2 \text{ k}\Omega \cdot \text{cm}^2$, an almost 7-fold increment compared to the bare cells. It may also be observed that the series resistance (R_s) of the P3HT-based devices has slightly increased compared to the bare cells and could be due to the additional P3HT layer, which is a reasonably resistive material. However, due to the enhanced charge transport and reduced recombination losses owing to the favorable band alignment, the $V_{\rm OC}$ is significantly increased and J_{SC} has not changed hugely despite the higher R_s .

The solar cells with new HTMs have demonstrated efficiencies of 4.73, 4.94, and 4.50% for V808, V1385, and V1386, respectively. Primarily, it is important to note that the new HTMs have exhibited efficiencies either on par or slightly higher than the conventional P3HT, which clearly demonstrates their validation. Another major observation from Figure 4a and Table 2 is that the solar cells with V808, V1385, and V1386 have yielded considerably higher J_{SC} , almost close to 14 mA/cm² over the P3HT devices (11.7 mA/cm²). However, the $V_{\rm OC}$ values of the devices with new HTMs are relatively lower than that of P3HT devices. As can be seen, the $V_{\rm OC}$ of solar cells with V808, V1385, and V1386 are 0.63, 0.68, and 0.56, respectively, while the $V_{\rm OC}$ of P3HT-based solar cells is 0.69 V. The band edge positions of new HTMs are compared with P3HT alongside other device layers as shown in Figure 4b and it can be observed that all the new HTMs possess the same VBE at -5.0 eV while that of P3HT is -4.6 eV. For the transport of photogenerated holes, the VBE of P3HT is in a favorable position with respect to the Sb₂S₃ absorber compared to the new HTMs, however, with the back contact (Au) energy level being -5.1 eV, the relatively smoother transport of holes is feasible in solar cells with new HTMs compared to P3HT. In addition, the CBE positions of new HTMs are -2.1, -2.5, and -2.7 eV for V808, V1385, and V1386, respectively, compared to P3HT (-2.7 eV). As may be observed, the interface barrier at the Sb₂S₃-HTM interface is larger for new HTMs over P3HT, which is responsible for electron blocking at the interface and leads to reduced recombination losses. While the higher J_{SC} may be ascribed to the favorable band alignment of new HTMs, the lower $V_{\rm OC}$ values compared to P3HT are not as expected especially considering the larger band gap of new HTMs than P3HT. Particularly, considerably lower $V_{\rm OC}$ (0.56) V) in solar cells with V1386 with a deficiency of almost 100 mV or more compared to other solar cells despite its reasonably higher mobility is still a question, although it may be speculated that this is owing to the increase in the number of thiophene units, the coverage might not be as uniform as compared to other layers leading to agglomerated conducting paths that may act as trap centers for photogenerated carriers. On that note, the wettability of new HTMs alongside P3HT in their precursor form is explored on the Sb₂S₃ surface. It is observed that all the HTMs including P3HT have shown excellent wettability with close to 0° contact angle for new HTMs and 2° for P3HT. The visual appearance of the samples upon dropping the precursor droplet onto Sb₂S₃ indicates that the P3HT drop remains as it is while the new HTM precursor drop spreads quickly in no time, which shows the relatively

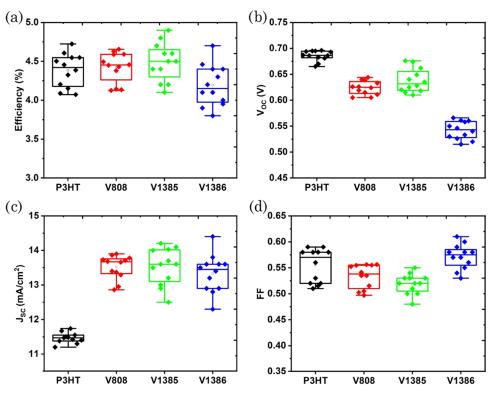


Figure 5. Reproducibility box plots of Sb₂S₃ solar cells with P3HT, **V808**, **V1385**, and **V1386** as HTMs: (a) Efficiency, (b) V_{OC} , (c) J_{SC} , and (d) fill factor (FF).

better wettability of new HTMs on the Sb₂S₃ surface. Therefore, despite possessing better interaction capability and favorable band edge positions, the solar cells with new HTMs have either marginally achieved similar V_{OC} as that of the conventional P3HT or have yielded lower values. In addition, it is known that mobility plays a crucial role in the performance of solar cells, but in this study, other factors that influence the PV parameters might be dominant. One of the important PV parameters is V_{OC} and its variation is generally ascribed to the energy levels, however, as they are quite similar, the other main aspect could be the interfacial recombination at Sb₂S₃-HTM. Now, there might be two sources of recombination—one is direct contact between the electrode and absorber layer (microshunts), due to the thinner films (which could explain lower $V_{\rm OC}$ than P3HT), second—the quality of the interface between the absorber and HTM. Thiophenes, in principle, are introduced for the better interface, for e.g., surface passivation, however, then one would expect the best performance for V1386. So, a further inspection of the Sb_2S_3 -HTM interface through advanced characterization techniques like deep-level transient spectroscopy may be needed to understand the role of interfacial defects on the performance of solar cells, and can be explored in future studies.

The cross-sectional images of solar cells with P3HT and new HTMs are shown in Figure 4c. While TiO_2 and Sb_2S_3 have an almost similar thickness in each case, P3HT is reasonably thicker (≈ 100 nm) compared to the new HTMs ($\approx 25-30$ nm). It may be noted that each individual HTM herein is optimized by its precursor concentration in spin coating to obtain the best efficient solar cells. Apparently, a thick P3HT is needed in order to get an efficiency of $\approx 4.7\%$ whereas a considerably thinner layer of each of the new HTMs is sufficient to obtain similar or slightly higher efficiencies (4.5–

4.9%). This proves the fact that the new HTMs, particularly V808 and V1385, are not only considerably cheaper but also the required materials consumption is significantly lower, which warrants the plausible fabrication of Sb₂S₃ solar cells with efficiencies close to 5% at a much cheaper cost. In addition, for the reproducibility of devices, the statistical box plots of efficiency, and the associated device parameters are shown in Figure 5. As Figure 5a unveils, the efficiency spread is almost identical in each case with P3HT and new HTMs. Further, the average efficiencies are also in a similar range in each of the optimized cases, except for the case of V1386, which shows a slightly lower efficiency owing to the low $V_{\rm OC}$. Similarly, the trend in V_{OC} , J_{SC} , and FF is almost the same as that for the champion numbers with reasonably identical spreads. The $V_{\rm OC}$ in V1386-based solar cells is considerably lower while the J_{SC} in P3HT-solar cells is the lowest and the FFs range from 0.5 to 0.6 in all the cases. As the statistics indicate, the high reproducibility of the results is obtained in the present study.

The normalized external quantum efficiency (EQE) curves of solar cells with P3HT and new HTM layers are shown in Figure 6a. In addition, the integrated photocurrent density $(J_{\text{integrated}})$ calculated using EQE spectra is also shown and it can be noted that almost identical J_{SC} values are obtained from EQE data to the values extracted from J-V curves. A pertinent aspect to be noted is the dip in EQE for solar cells with P3HT in the range of 500–650 nm, which may be attributed to the parasitic absorption losses.² When an absorber is thinner or comparable to the thickness of HTM, the light that gets transmitted through the absorber gets reflected back from the back contact and often, the reflected light could be reabsorbed in the absorber to result in photogenerated carriers. However, if the HTM is sufficiently thick, then the reflected light gets

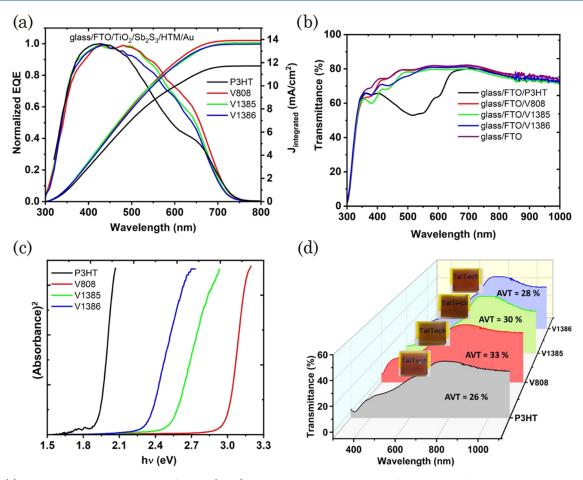


Figure 6. (a) Normalized external quantum efficiency (EQE) curves and the integrated J_{SC} of Sb₂S₃ solar cells with P3HT, V808, V1385, and V1386 as HTMs, (b) transmittance curves of glass/FTO, glass/FTO/P3HT, glass/FTO/V808, glass/FTO/V1385 and glass/FTO/V1386, (c) (absorbance)² vs $h\nu$ plots of P3HT, V808, V1385, and V1386 for band gap determination, and (d) transmittance curves of the device stack with P3HT, V808, V1385, and V1386 as HTMs without Au (inset: Respective sample photographs captured with a background showing semi-transparency).

absorbed in the HTM layer leading to parasitic absorption. In this study, P3HT is considerably thick and thus exhibits parasitic absorption while the considerably thinner new HTMS (V808, V1385, and V1386) does not show any similar dip in EQE, which could further explain the enhanced J_{SC} values in the respective solar cells. Optical transmittance studies are performed for bare P3HT and new V808, V1385, and V1386 HTM layers on glass/FTO substrates, as shown in Figure 6b, where the absorption by P3HT is corroborated with EQE results as observed from the dip in transmittance in 400-650 nm region. Further, it may also be observed that the new HTMs exhibit considerable transparency similar to that of the glass/FTO substrate. The band gaps of P3HT, V808, V1385, and V1386 are inferred to be \approx 1.9, 2.9, 2.5, and 2.3 eV, respectively, from the plots shown in Figure 6c and the absorption coefficient-based Tauc's plots in Figure S4, which ascertains the fact that new HTMs are optically more transparent in the visible region compared to P3HT. Finally, the overall transparency of device stacks without top metal contact is analyzed (see Figure 6d) and the solar weighted average visible transmittance (AVT) (400-800 nm) of devices with P3HT, V808, V1385, and V1386 HTMs are determined to be ≈ 26 , 33, 30, and 28%, respectively. In the case of new HTMs, as the thiophene units increased the band gap decreased resulting in slightly reduced transparency as

witnessed from the overall device transmittance. The sample pictures of glass/FTO/TiO₂/Sb₂S₃/HTM where HTM is P3HT, **V808**, **V1385**, and **V1386** are shown with a background and it may be observed that all the samples exhibit decent transparency with visibility being the highest for **V808**. While all device stacks exhibit reasonably good AVT values as required for semitransparent applications, the devices with new HTMs exhibit enhanced transparency, which is due to their increased transmittance over P3HT. The cost-effective fluorene-based thiophene-terminated (**V808**, **V1385**, **V1386**) HTMs are successfully demonstrated in Sb₂S₃ solar cells as potentially suitable candidates for semitransparent applications and efforts are underway to increase the PCE of solar cells and upscaling the devices.

4. CONCLUSIONS

Novel cost-effective and transparent fluorene-based thiopheneterminated hole transport materials (V808, V1385, and V1386) are demonstrated, for the first time, in semitransparent Sb₂S₃ solar cells exhibiting a power conversion efficiency of 4.5-4.9% and an average visible transmittance of 30-33%(without the metal) in the 400-800 nm range. Solar cells with the conventional P3HT as HTM have yielded 4.7% efficiency, however, they have exhibited lower transparency (26%) owing to parasitic absorption losses in P3HT. The band diagram of solar cells with new HTMs and P3HT are compared, which unveils that new HTMs have exhibited favorable band offset validating them as effective HTMs for Sb₂S₃ solar cells. The new HTMs with terminated thiophene units have demonstrated reasonably improved wettability over conventional P3HT that could provide better interaction channels for charge carrier transport thus leading to enhanced current density. However, a closer inspection of Sb₂S₃-HTM needs to be further explored to unravel the interfacial effects as the solar cells with new HTMs have exhibited reasonably lower $V_{\rm OC}$ values, particularly in the case of V808 and V1386.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsaem.2c04097.

Synthesis procedure, DSC, J-V characteristics of solar cells, and absorption spectra related data of V808, V1385, and V1386 (PDF)

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Author Contributions

S.M. conceptualization, design of experiments, analysis of the results, and manuscript writing; N.J. experiments and contribution to manuscript writing; A.K. and T.D. support in experiments and data acquisition; N.S. and S.K. supported in analysis; A.J., A.M., and V.G. synthesized and characterized new HTMs; R.G. and A.V. performed ionization potential and hole mobility measurements, M.K. and I.O.A. provided support for analysis and finalizing the manuscript draft. All authors have given approval to the final version of the manuscript.

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Notes

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HTM, hole transport material

V808, N^2 , N^7 , N^7 , N^7 -tetrakis(4-methoxyphenyl)-9-(thiophen-2-ylmethylene)-9H-fluorene-2,7-diamine

V1385, 9-([2,2':5',2"-terthiophen]-5-ylmethylene)- N^2 , N^7 , N^7 , N^7 -tetrakis(4-methoxyphenyl)-9H-fluorene-2,7-diamine

V1386, 9-([2,2'-bithiophen]-5-ylmethylene)- N^2 , N^2 , N^7 , N^7 -tetrakis(4-methoxyphenyl)-9*H*-fluorene-2,7-diamine

P3HT, poly(3-hexylthiophene)

Spiro-OMeTAD, 2,2',7,7'-tetrakis[N,N-di(4-

methoxyphenyl)amino]-9,9'-spirobifluorene

PCPDTBT, poly{4,4-dialkyl-4*H*-cyclopenta[2,1-*b*;3,4-*b*']-dithiophene-2,6-diyl-*alt*-2,1,3-benzothiadiazole-4,7-diyl}

AVT, average visible transmittance

DSC, differential scanning calorimetry

TGA, thermal gravimetric analysis

EQE, external quantum efficiency

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